Modeling Indoor Aerosol Inorganic Thermodynamics with ISORROPIA

Bryan Berman¹, Bryan Cummings¹, Anita Avery², Shannon Capps¹, Peter DeCarlo³, Michael Waring¹

¹Drexel University, Philadelphia, Pennsylvania, USA ²Aerodyne Research Inc., Billerica, Massachusetts, USA ³Johns Hopkins University, Baltimore, Maryland, USA

Motivation

- High concentrations of ambient aerosols are linked to respiratory and cardiovascular conditions
- Citizens of developed countries spend most of their time indoors
- We want to better understand how HVAC systems with mechanical ventilation influence indoor aerosol composition and chemical processing
- This work demonstrates a proof of concept for addressing these gaps with a focus on indoor inorganic aerosols (IA) from outdoor origin by expanding Cummings & Waring (2019) indoor organic model



Johnson et al., Indoor Air, 2017

ISORROPIA

- Inorganic aerosol (IA) thermodynamic equilibrium model
- Widely used to predict gas/ particle partitioning
- Inputs for forward mode are temperature (*T*), relative humidity (*RH*), and total concentration
- The stable mode can have both solid & liquid phase aerosol while the metastable mode only has liquid state aerosol
- First instance of applying ISORROPIA in an indoor model



Nenes et al., Aquatic Geochemistry, 1998; Fountoukis et al., Atmospheric Chemistry and Physics, 2007



Model Input

- Measured indoor and outdoor *T, RH*, and CO₂, as well as outdoor IA concentration data from Avery et al (2019) was used as input for this model
- Assumed the same air flow rates as Avery et al. (2019)



- Occupant NH₃ was derived from CO₂ data using methods from Li et al. (2020)
- Background NH₃ was estimated using a correlation published by Ampollini et al. (2019)



Guo et al., Journal of Geophysical Research: Atmospheres, 2016

FGA Evaluation | summer episode



HVAC Thermodynamics



- Indoor aerosol will maintain an aqueous phase in the summer, but not in the winter
- Deliquescence is encouraged by warm air and particles being cooled down
- Efflorescence is encouraged by cold air and particles being warmed up

- A 75% HNO_3 loss to the cooling coil was used and based on a parametric test
- A filter efficiency of 10% and 30% for the summer and winter respectively was used

Indoor Modeled Concentrations

summer episode



Gas Concentration (ppb)



winter episode

Comparison of Modeled & Measured Concentrations

summer episode



winter episode



Indoor-outdoor Ratio Behavior



$[I/O]_{i/SO4} = \frac{[I/O]_i}{[I/O]_{SO4}}$

Johnson et al., Indoor Air, 2017





Future Work & Summary

- Assessing model sensitivity to indoor ammonia sources (such as from third hand smoke) is ongoing
- ISORROPIA was integrated into an indoor aerosol model to predict IA partitioning for the first time
- Modeling IA partitioning in the winter could be more complicated or the FGA is only applicable to the summer set
- Serves as proof of concept towards exploring how heating, ventilating, and air-conditioning (HVAC) systems influence indoor aerosol composition and chemical processing, with a focus on indoor aerosols of outdoor origin, using ISORROPIA

Thank you!

• This work was funded by the Alfred P. Sloan Foundation



ALFRED P. SLOAN FOUNDATION



References

- Ampollini, L., et al. (2019). "Observations and Contributions of Real-time Indoor Ammonia Concentrations During HOMEChem." Environmental Science & Technology.
- Avery, A. M., Waring, M. S., & DeCarlo, P. F. (2019). Seasonal variation in aerosol composition and concentration upon transport from the outdoor to indoor environment. *Environmental Science-Processes & Impacts, 21*(3), 528-547. Article.
- Cummings, B. E., & Waring, M. S. (2019). Predicting the importance of oxidative aging on indoor organic aerosol concentrations using the two-dimensional volatility basis set (2D-VBS). *Indoor Air, 29*(4), 616-629. Article.
- DeCarlo, P. F., et al. (2018). "Thirdhand smoke uptake to aerosol particles in the indoor environment." Science Advances 4(5).
- Fountoukis, C. and A. Nenes (2007). "ISORROPIA II: a computationally efficient thermodynamic equilibrium model for K+-Ca2+-Mg2+-Nh(4)(+)-Na+-SO42--NO3--Cl--H2O aerosols." Atmospheric Chemistry and Physics 7(17): 4639-4659.
- Guo, H., Sullivan, A. P., Campuzano-Jost, P., Schroder, J. C., Lopez-Hilfiker, F. D., Dibb, J. E., et al. (2016). Fine particle pH and the partitioning of nitric acid during winter in the northeastern United States. *Journal of Geophysical Research-Atmospheres*, 121(17), 10355-10376. Review.
- Guo, H., Xu, L., Bougiatioti, A., Cerully, K. M., Capps, S. L., Hite, J. R., Jr., et al. (2015). Fine-particle water and pH in the southeastern United States. *Atmospheric Chemistry and Physics*, *15*, 5211-5228.
- Johnson, A. M., et al. (2017). "Real-time transformation of outdoor aerosol components upon transport indoors measured with aerosol mass spectrometry." Indoor Air 27(1): 230-240.
- Li, M., et al. (2020). "Human Ammonia Emission Rates under Various Indoor Environmental Conditions." Environmental Science & Technology 54: 5419-5428.
- Nenes, A., et al. (1998). "ISORROPIA: A New Thermodynamic Equilibrium Model for Multiphase Multicomponent Inorganic Aerosols." Aquatic Geochemistry 4: 123-152.