

# **ANISORROPIA:**

## **Development of the Adjoint of ISORROPIA**

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CMAQ-Adjoint Development Team

*Funding Sources*

Amir Hakami and coworkers at  
Carleton University

National Science Foundation  
Graduate Research Fellowship

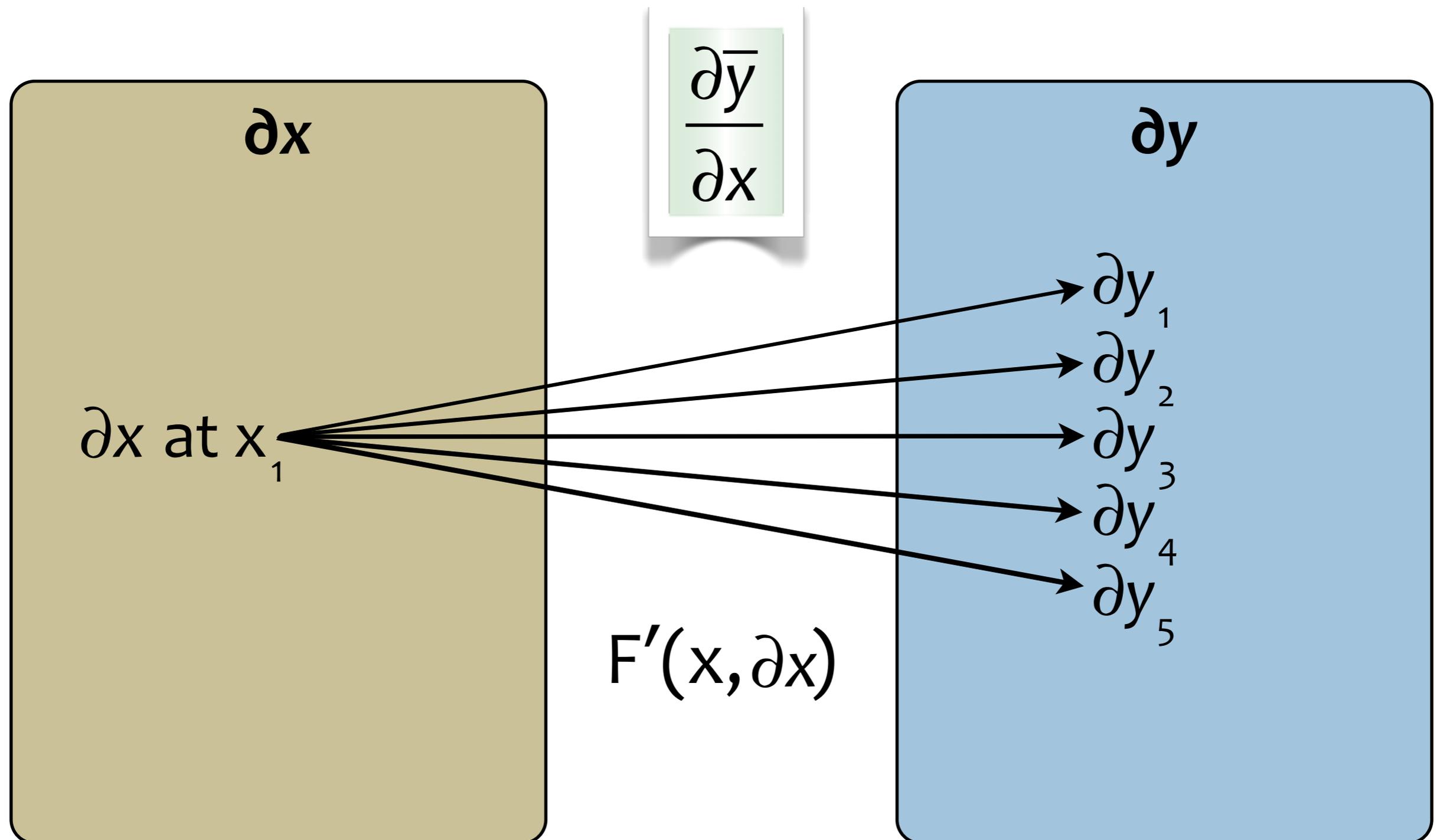
Daven Henze and coworkers at  
University of Colorado

Conoco-Phillips  
Model Development Support

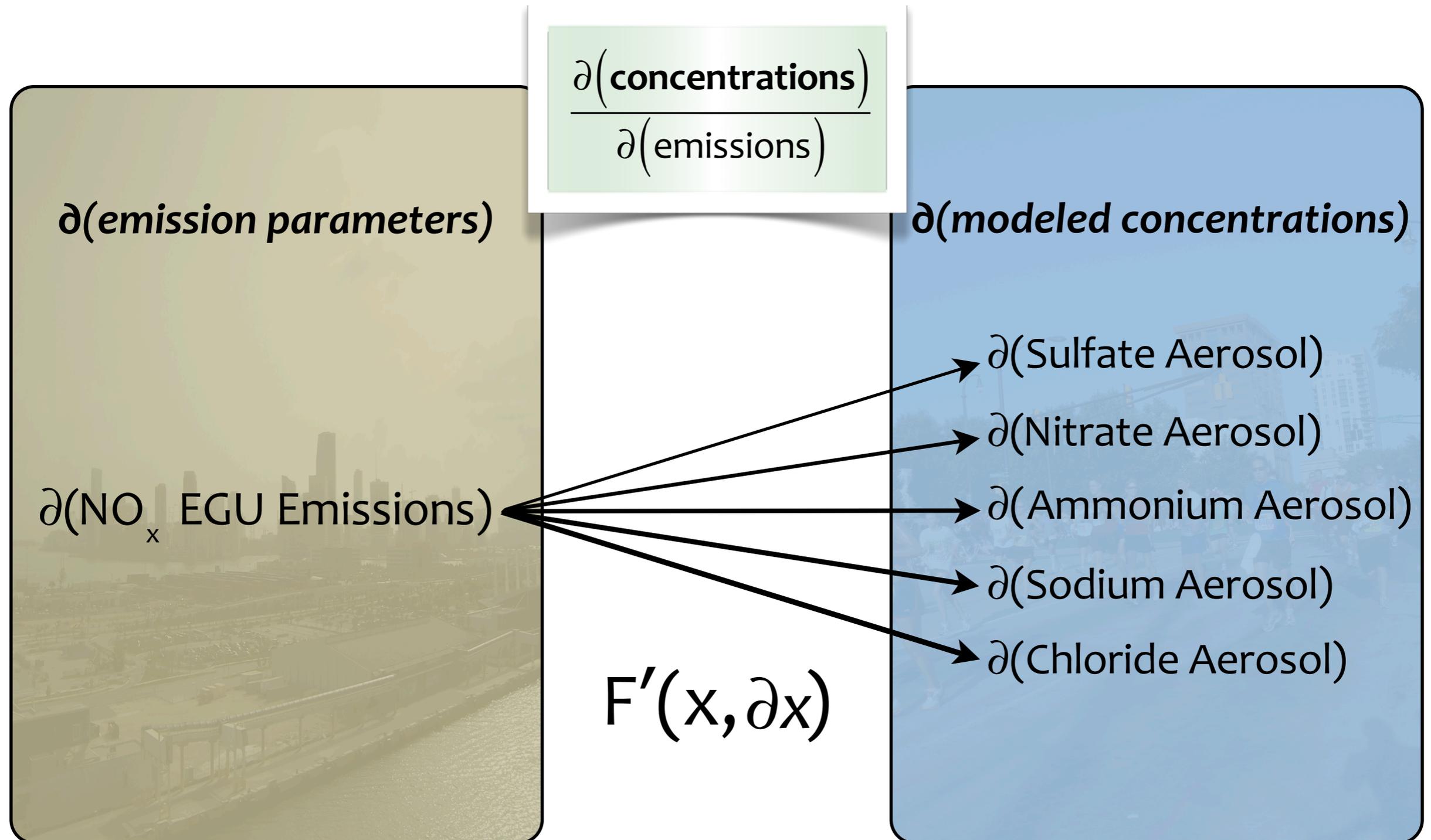


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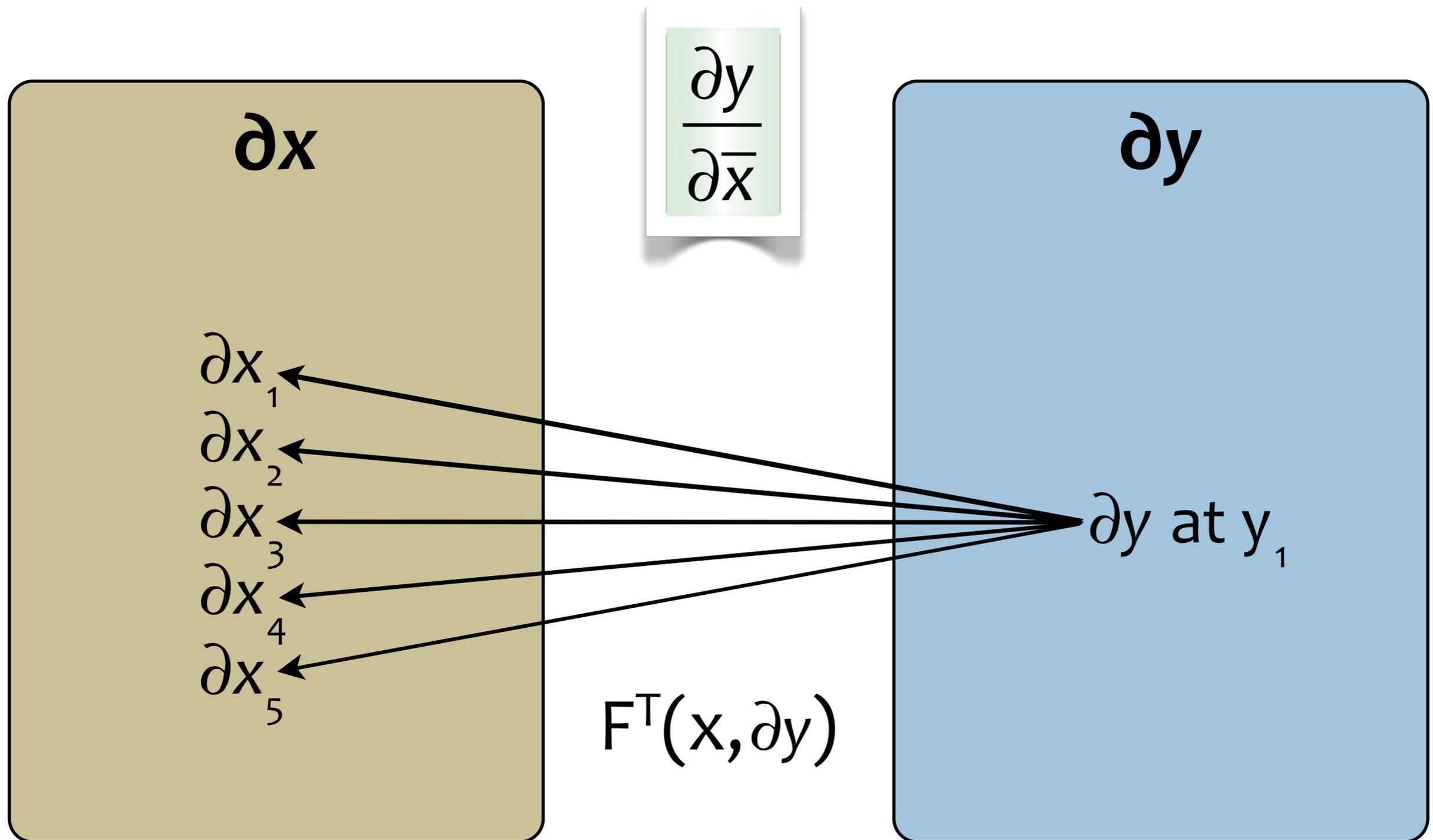
# Direct Sensitivity Analysis: Forward Method



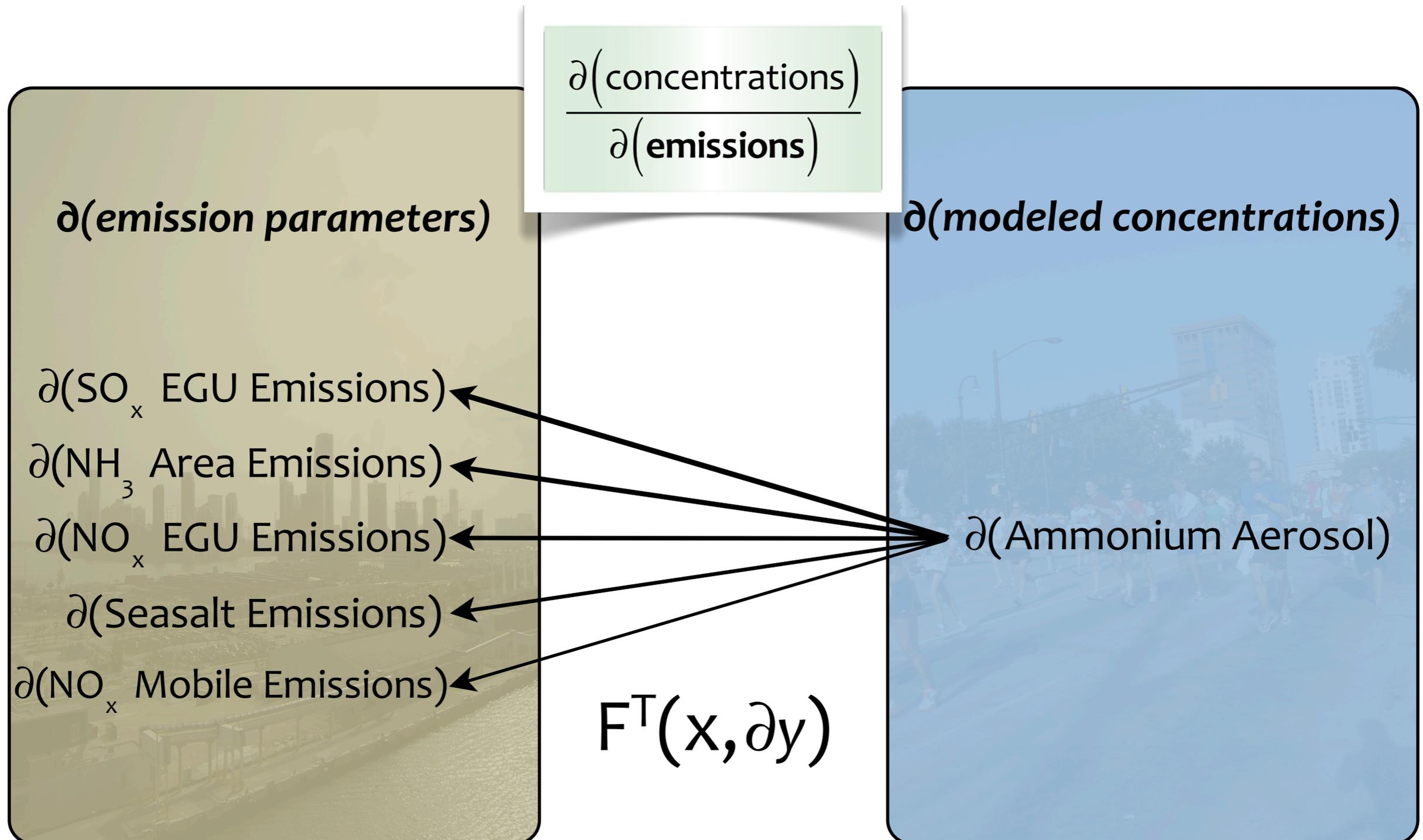
# Direct Sensitivity Analysis: Forward Method



# Direct Sensitivity Analysis: Reverse Method

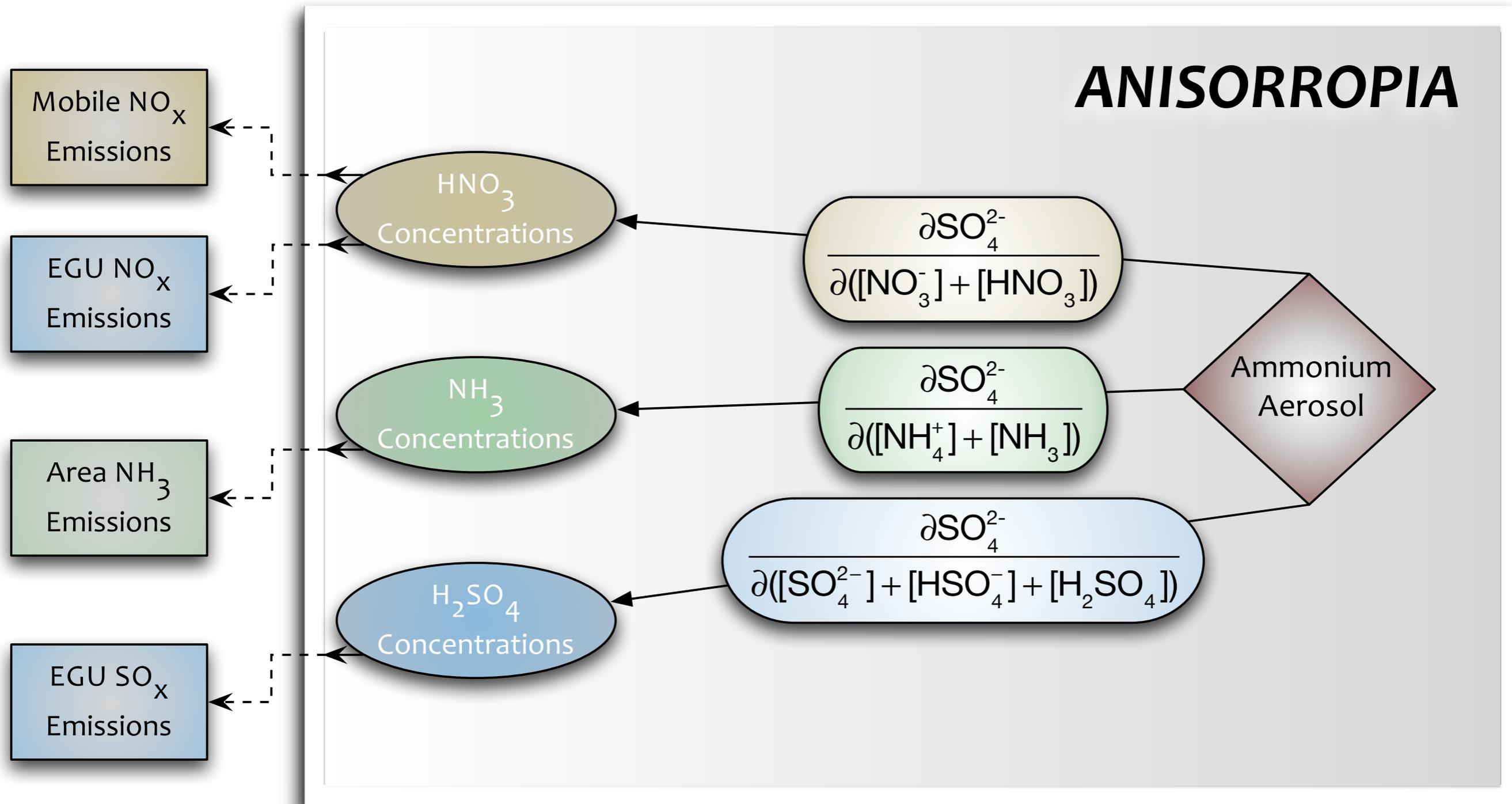


# Direct Sensitivity Analysis: Reverse Method



# CMAQ-Adjoint Sensitivities

*Inorganic Thermodynamic Equilibrium Contribution*



# Strategy for ANISORROPIA Development

acidic aerosol

neutralized aerosol

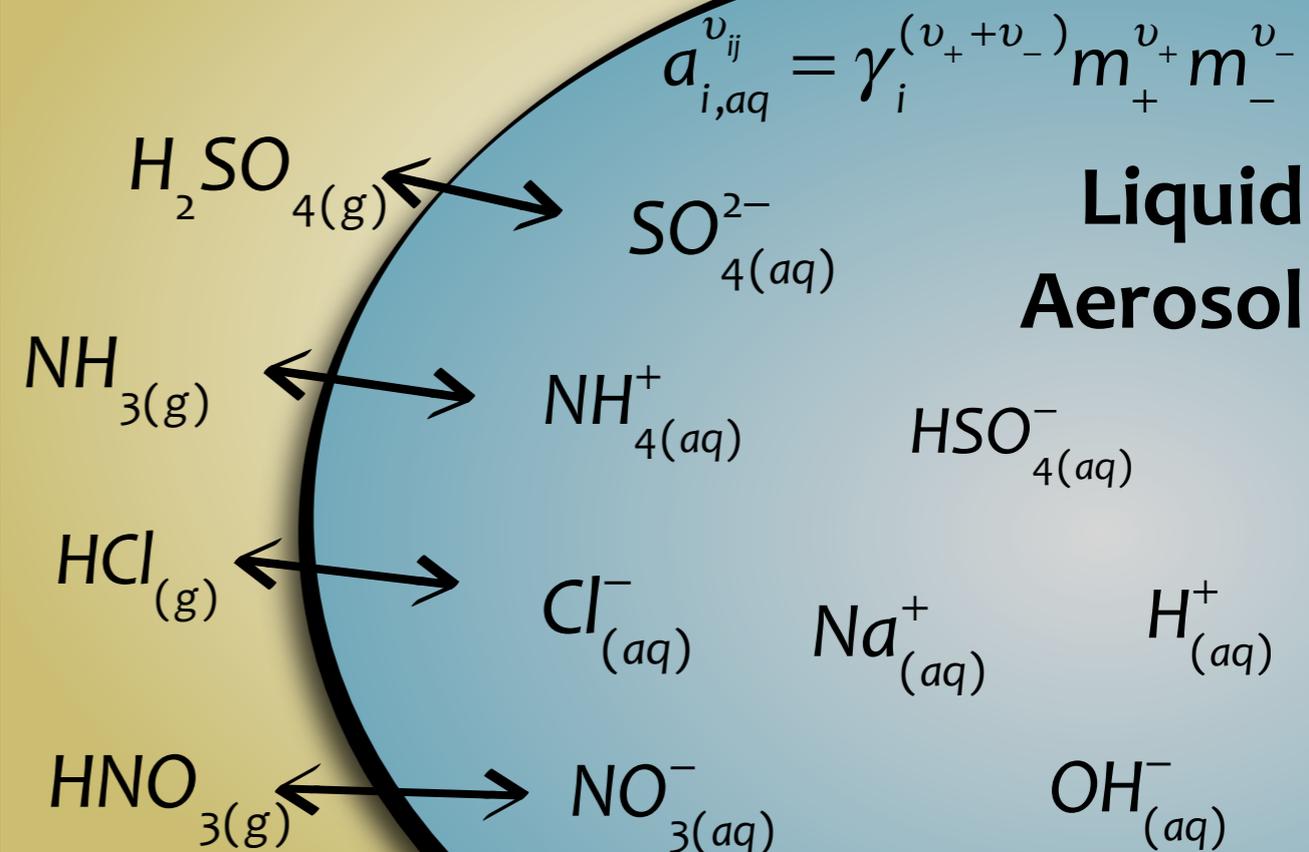
## ISORROPIA

(Nenes et al., 1998; Fountoukis et al., 2007)

Gases

$$\prod_i a_i^{v_{ij}} = K_j(T)$$

$$a_{i,\text{gas}}^{v_{ij}} = p_i$$

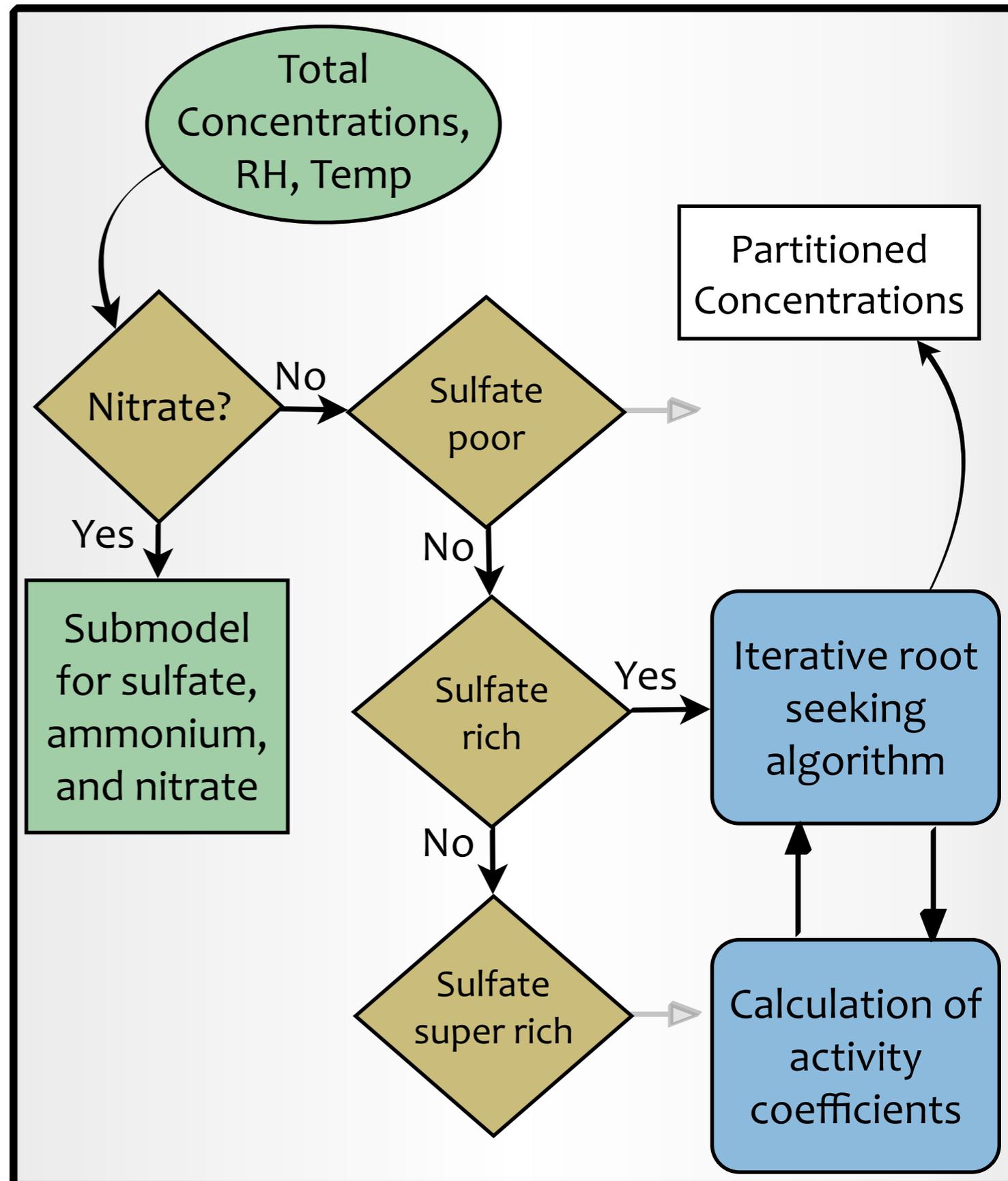


# Strategy for ANISORROPIA Development

## acidic aerosol

- ISORROPIA algorithm:  
*explicit solution,  
iteration on activity coefficients*
- Direct application of  
automatic differentiation tool,  
TAPENADE

Hascoët and Pascual, 2004



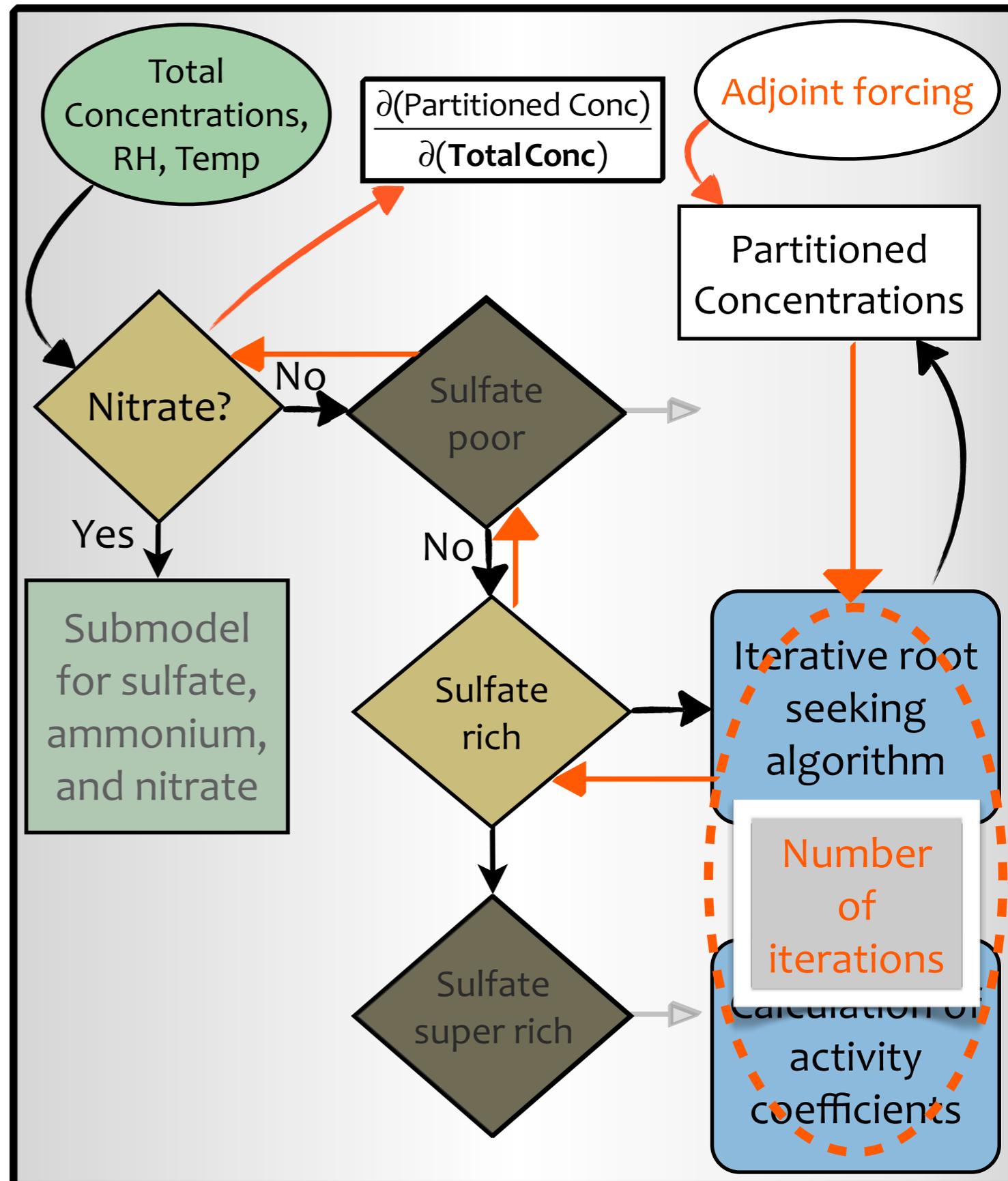
# Strategy for ANISORROPIA Development

## acidic aerosol

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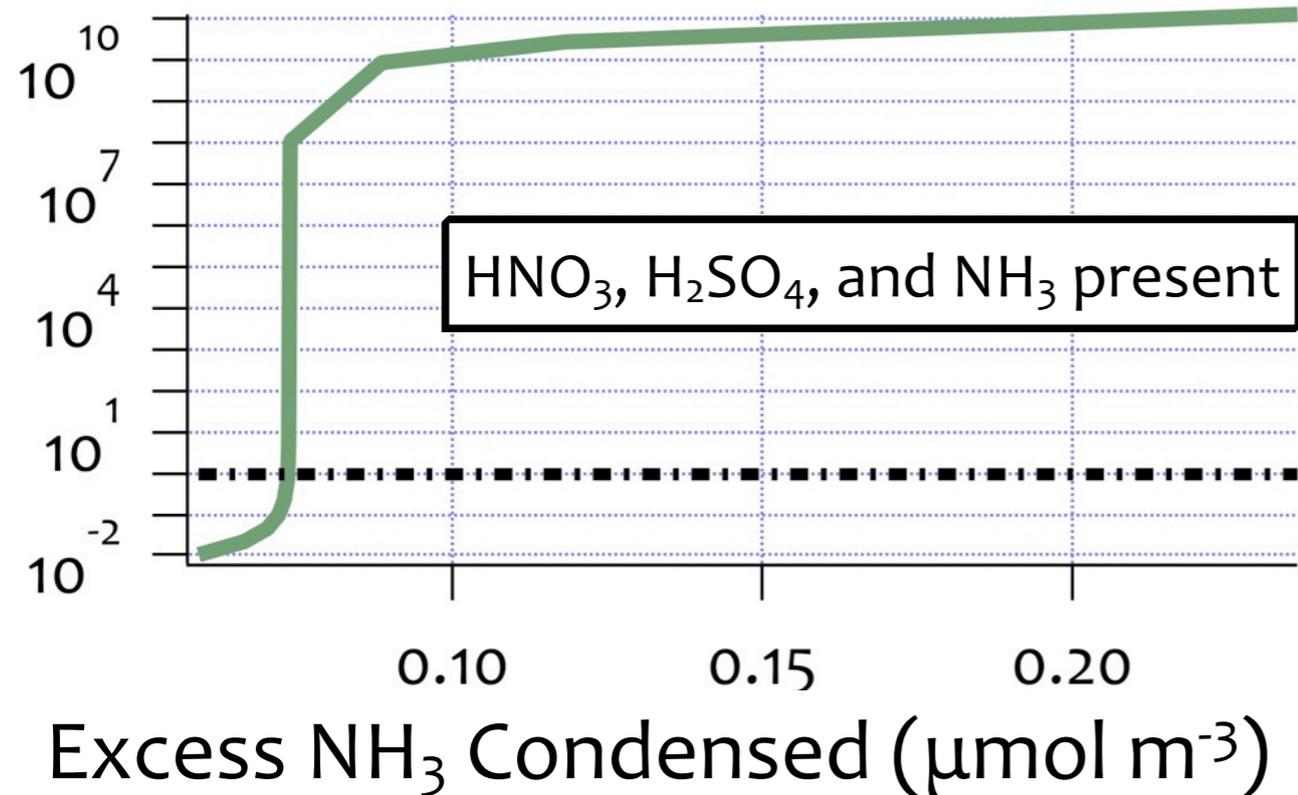
$$\frac{\partial(\text{Partitioned Conc})}{\partial(\text{Total Conc})}$$

Hascoët and Pascual, 2004



# Strategy for ANISORROPIA Development

Saturation Ratio

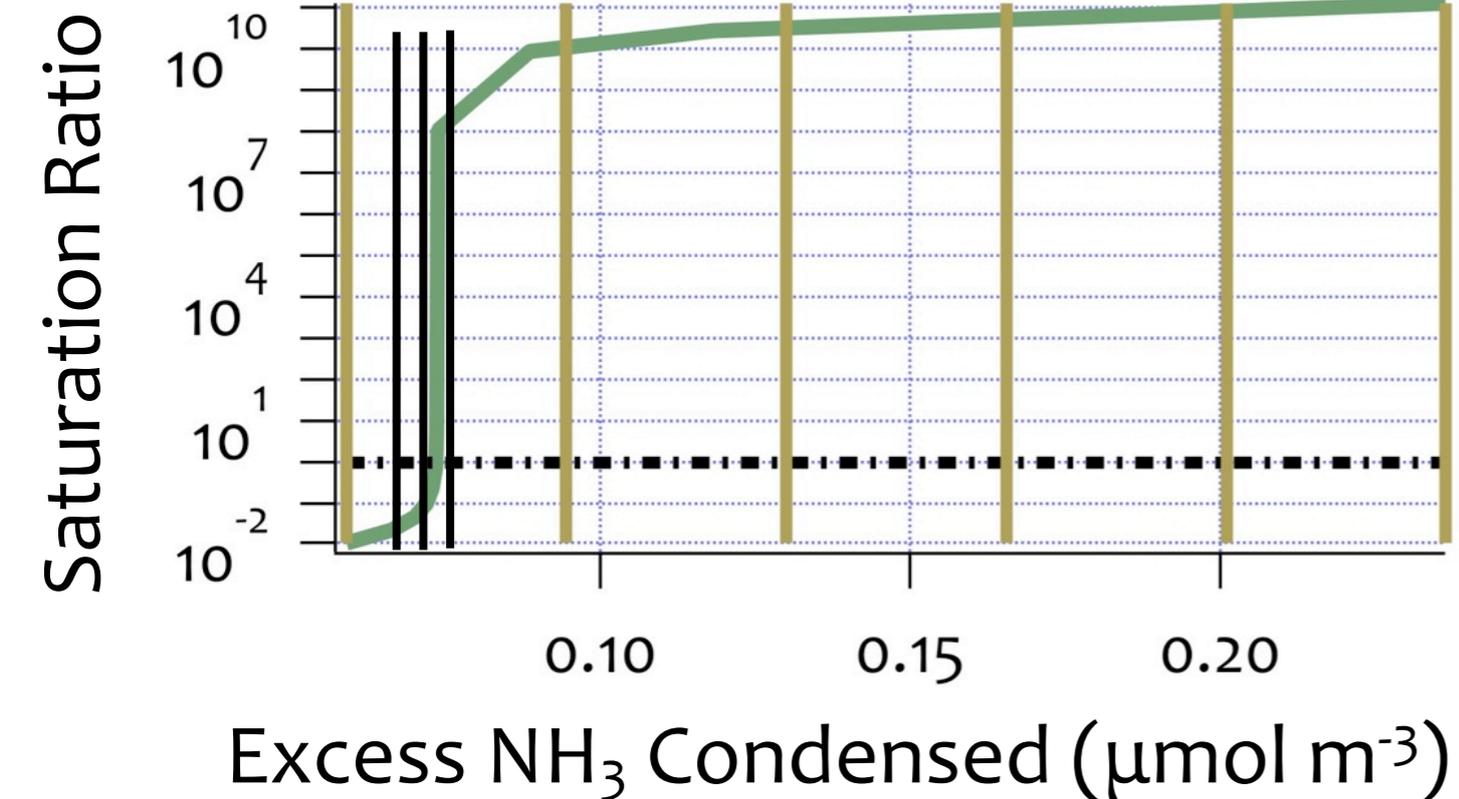


*neutralized aerosol*

- ISORROPIA algorithm:  
*bisection method*  
to minimize deviation of  
*saturation ratio from 1*

$$\text{Saturation Ratio} = \frac{[NH_{4,(aq)}^+]}{K [NH_{3,(g)}] [H_{(aq)}^+] \gamma}$$

# Strategy for ANISORROPIA Development



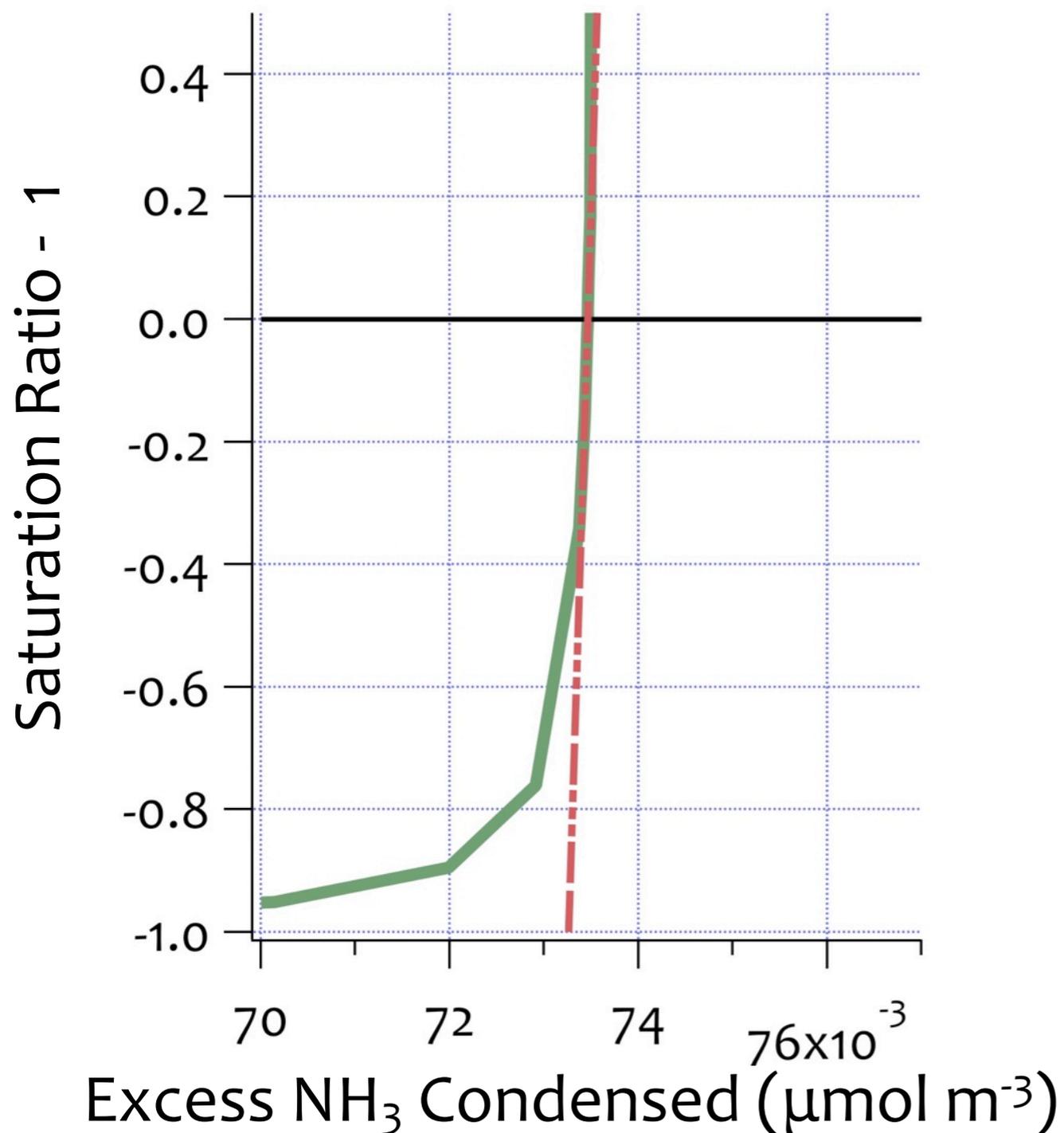
*neutralized aerosol*

- ISORROPIA algorithm:  
*bisection method  
to minimize deviation of  
saturation ratio from 1*

$$\text{Saturation Ratio} = \frac{[NH_{4,(aq)}^+]}{K [NH_{3,(g)}] [H_{(aq)}^+] \gamma}$$

- Derivative cannot be traced  
across bisection steps

# Strategy for ANISORROPIA Development

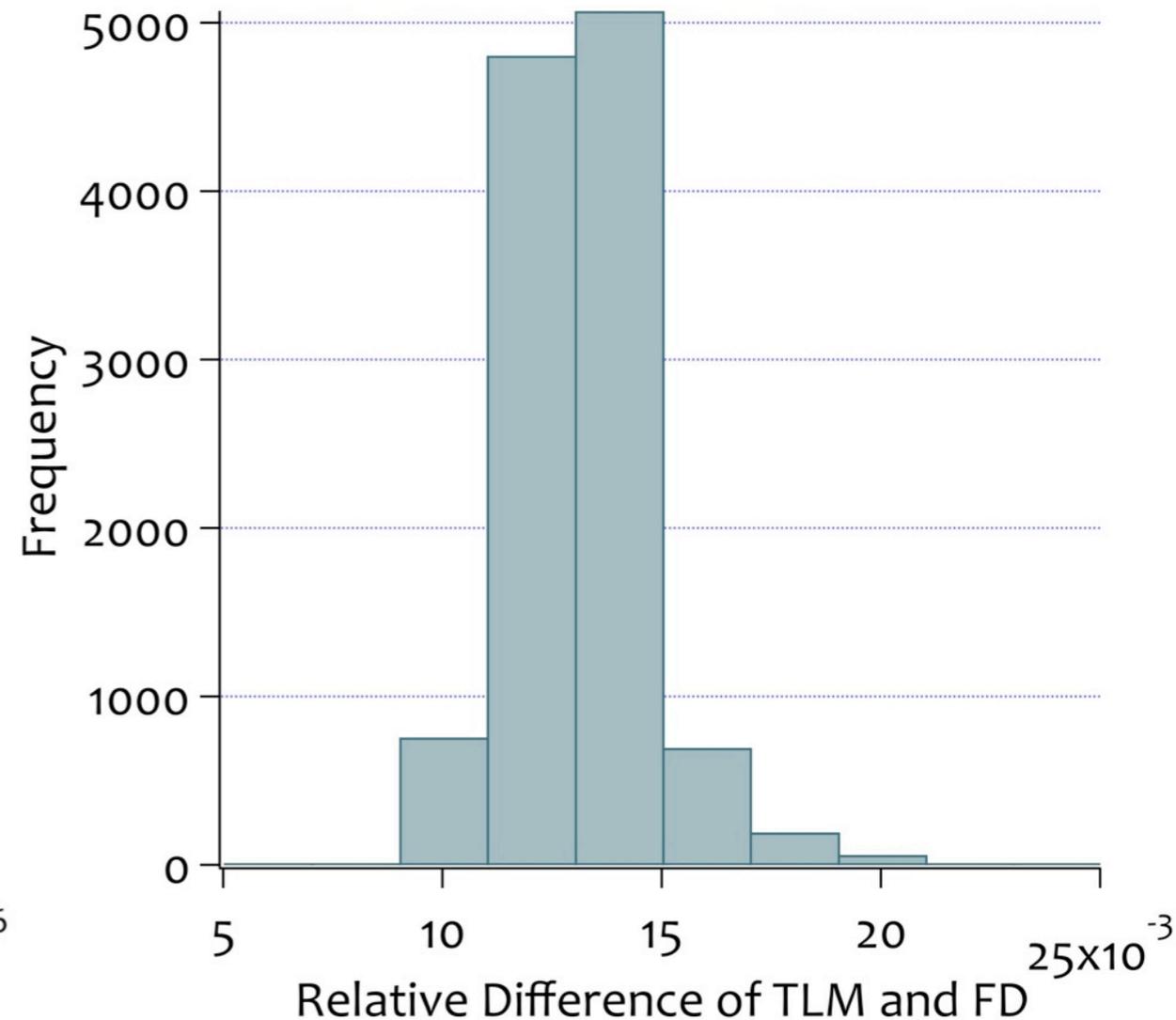
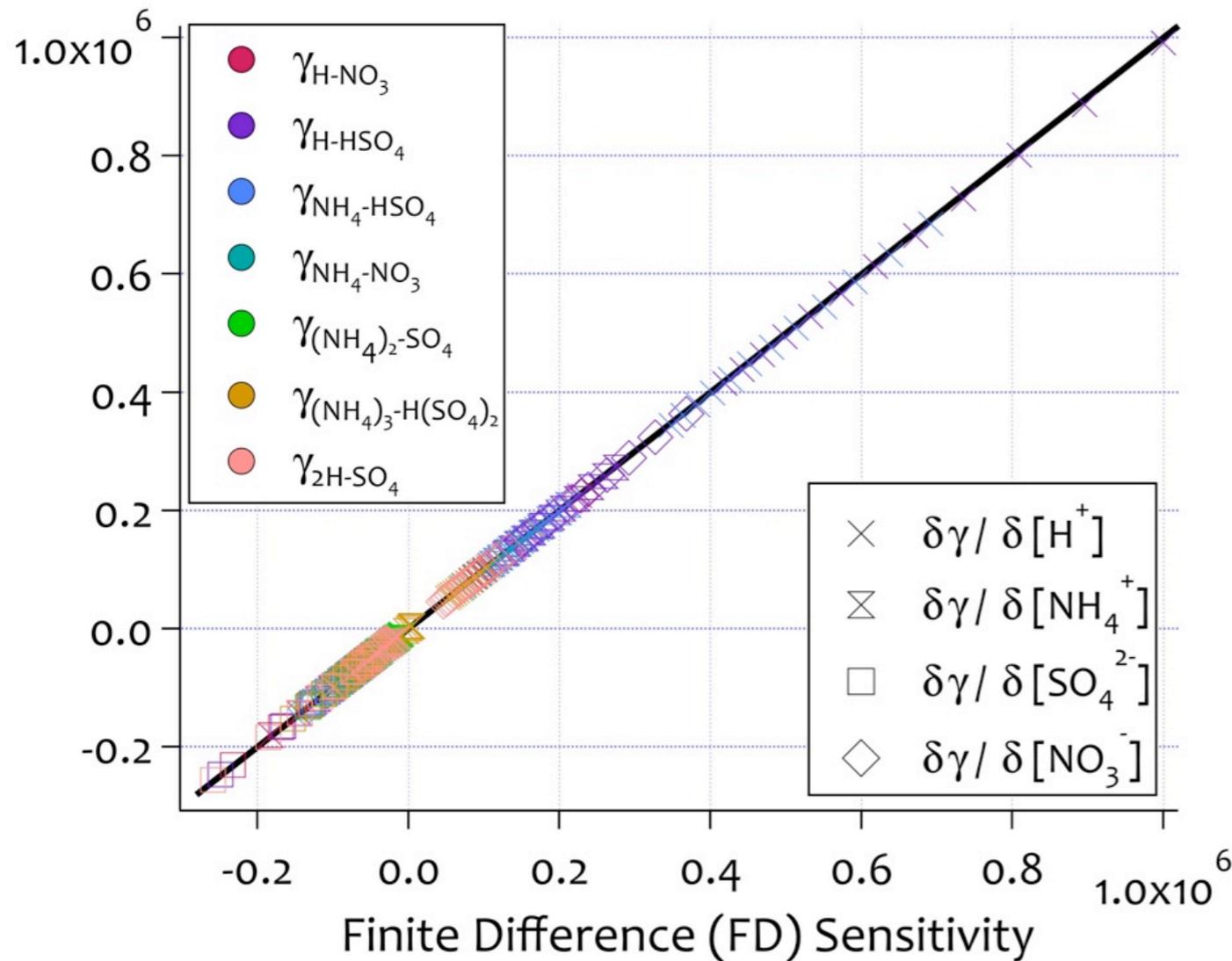


## neutralized aerosol

- Stable determination of root by bisection
- Apply single iteration of Newton-Raphson at root
  - $x_{n+1} = x_n - \frac{f(x_n)}{f'(x_n)}$
  - TAPENADE-produced forward sensitivity code
- Adjoint only of post-convergence N-R step

# Activity Coefficient Sensitivities

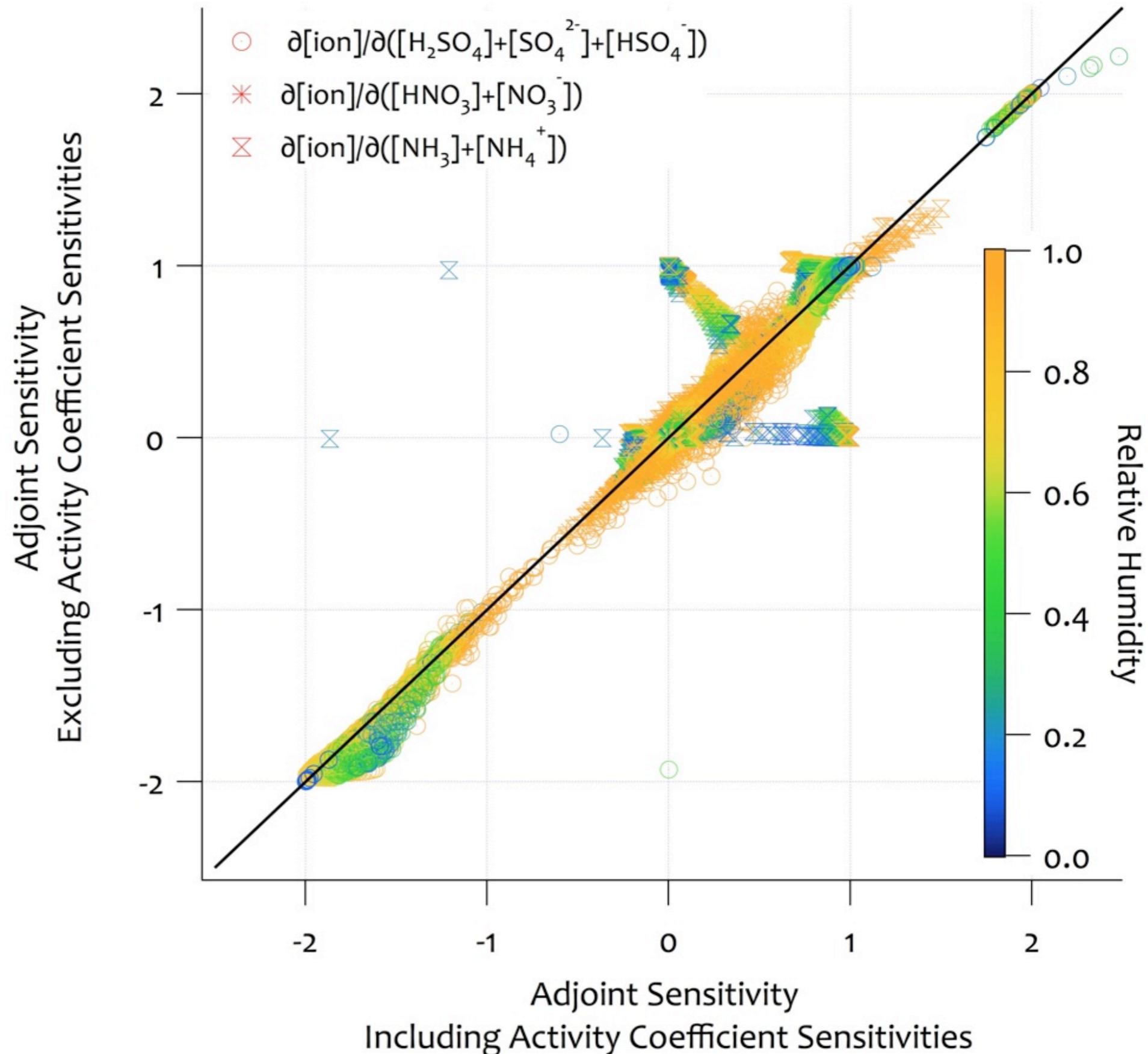
Tangent Linear Model (TLM) Sensitivity

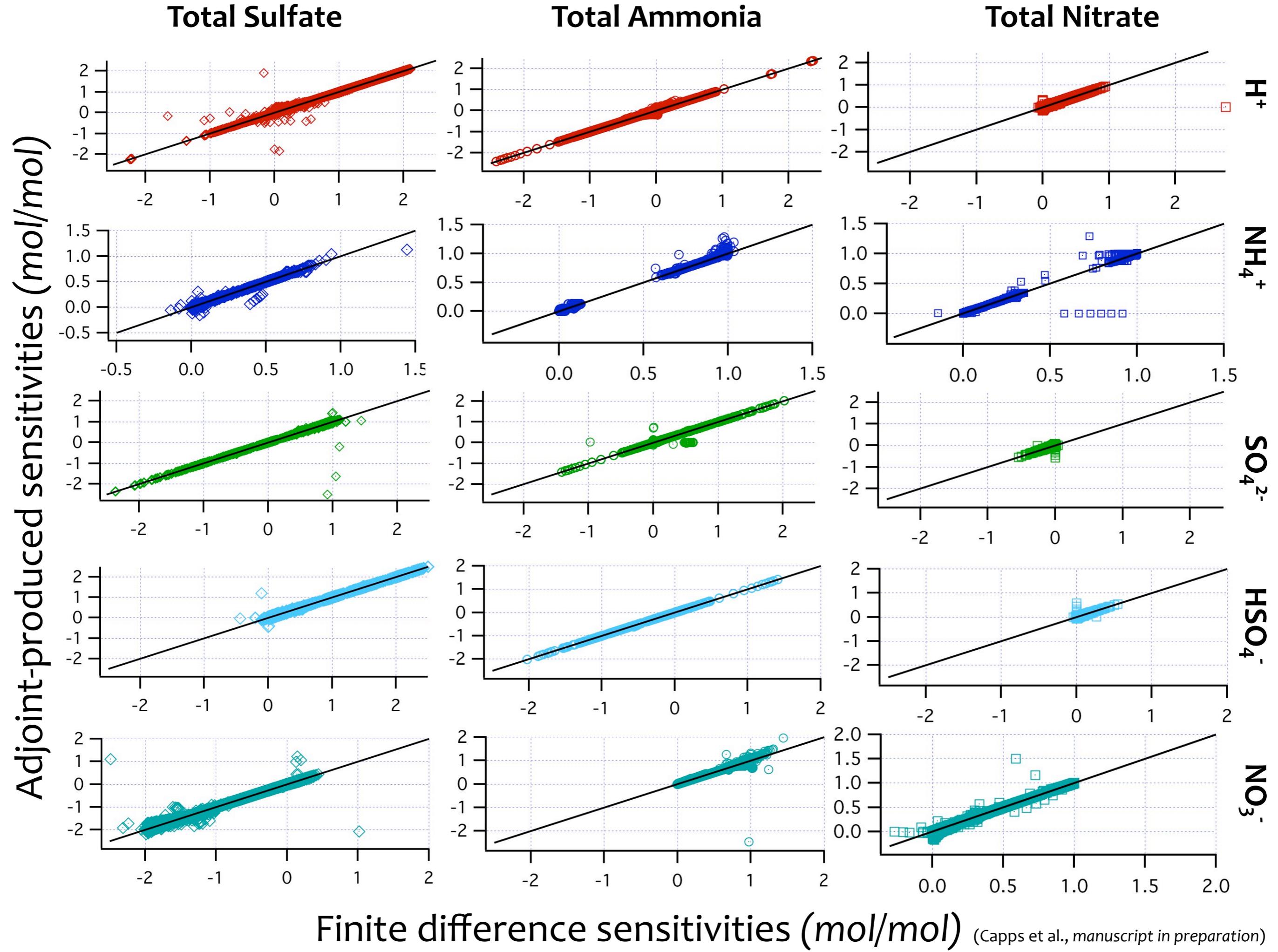


Differentiated code properly captures the sensitivity of activity coefficients involved in the calculation.

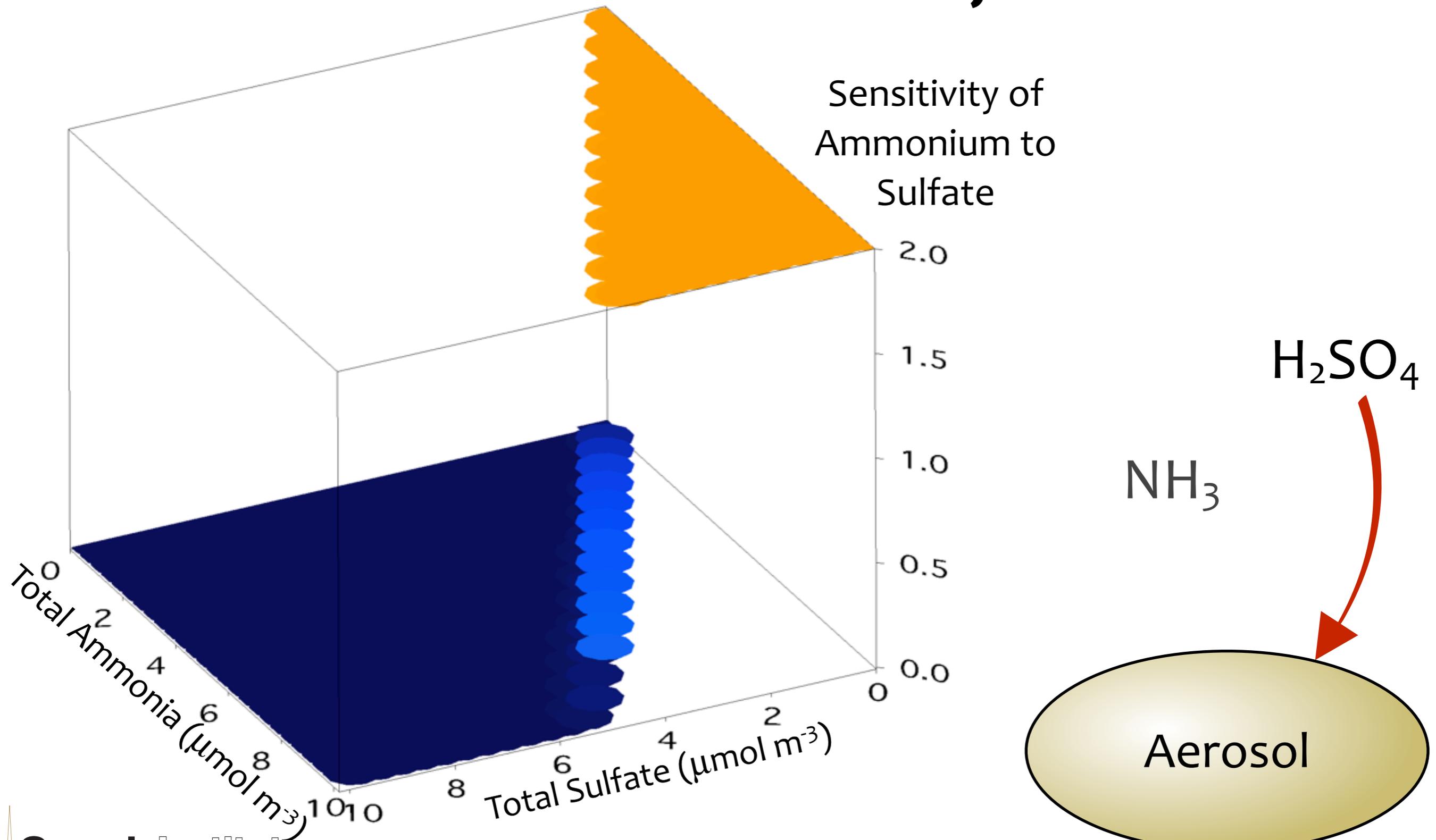
# Activity Coefficient Sensitivities

- For neutralized aerosol, accumulation of sensitivities through the activity coefficient calculations can be significant
- Lower relative humidities
- Sensitivities to total ammonia

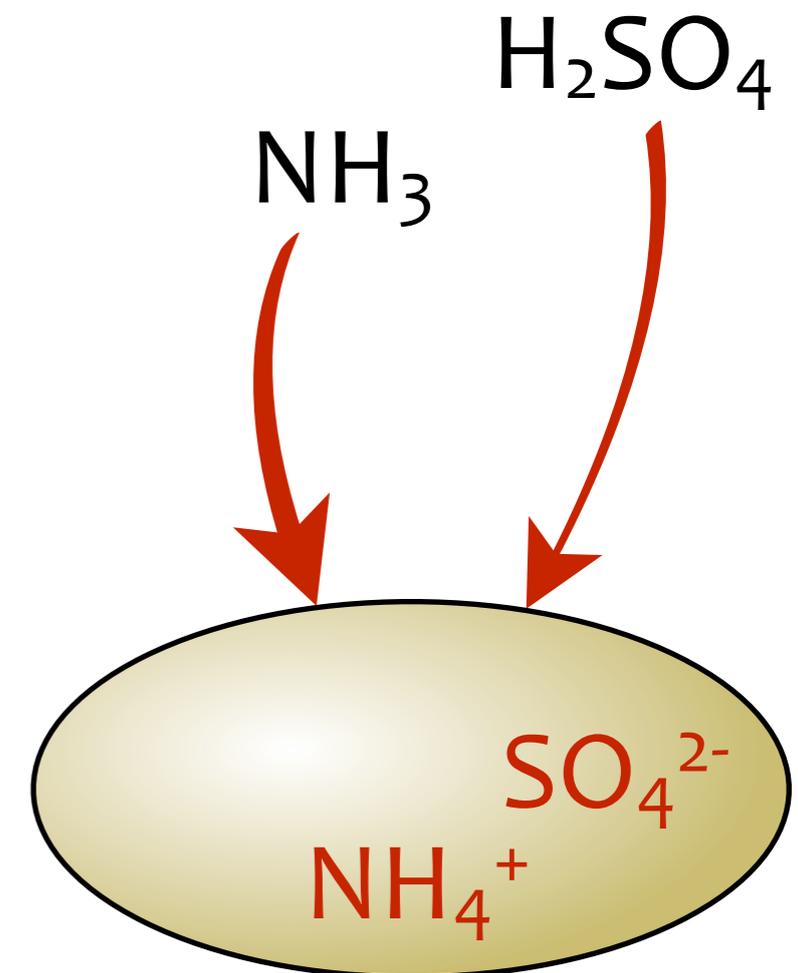
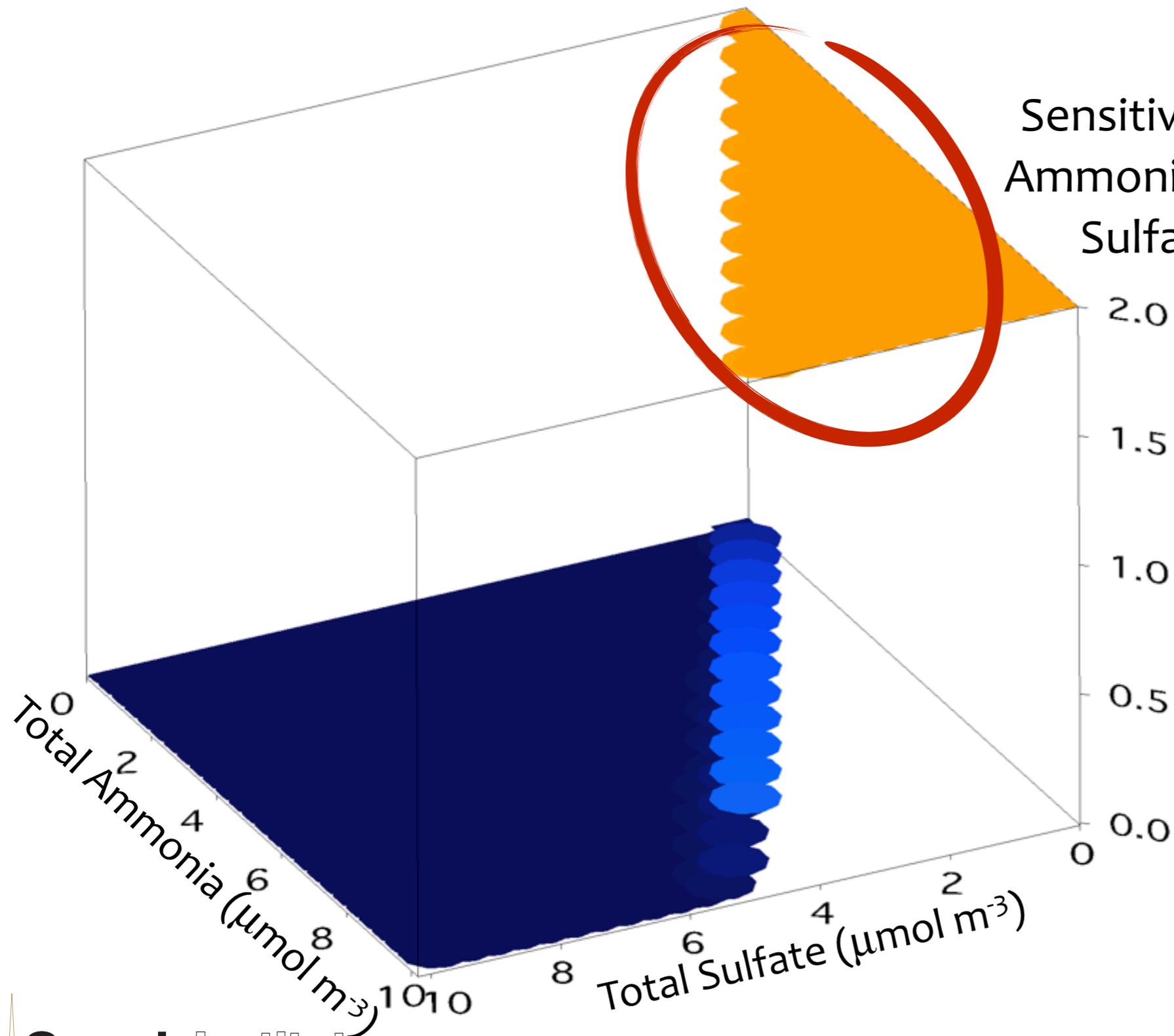




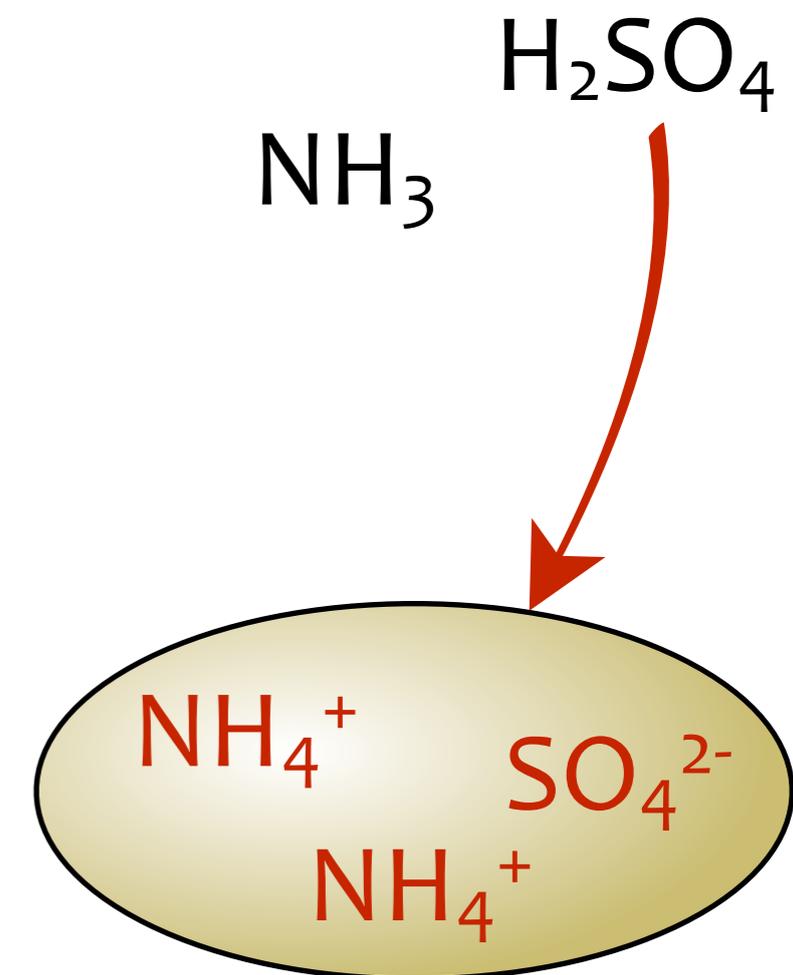
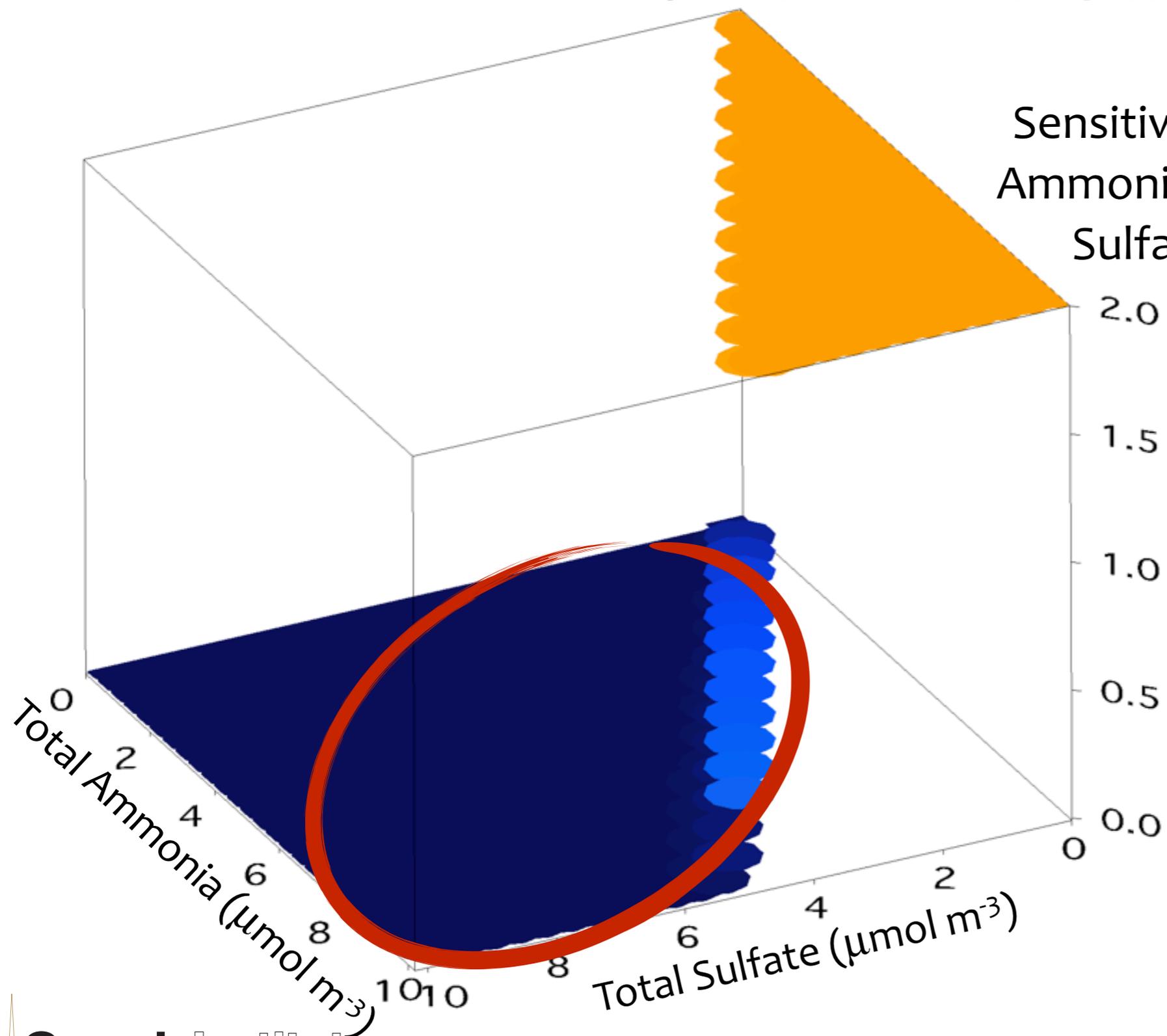
# Sensitivity Space: Ammonium and Sulfate



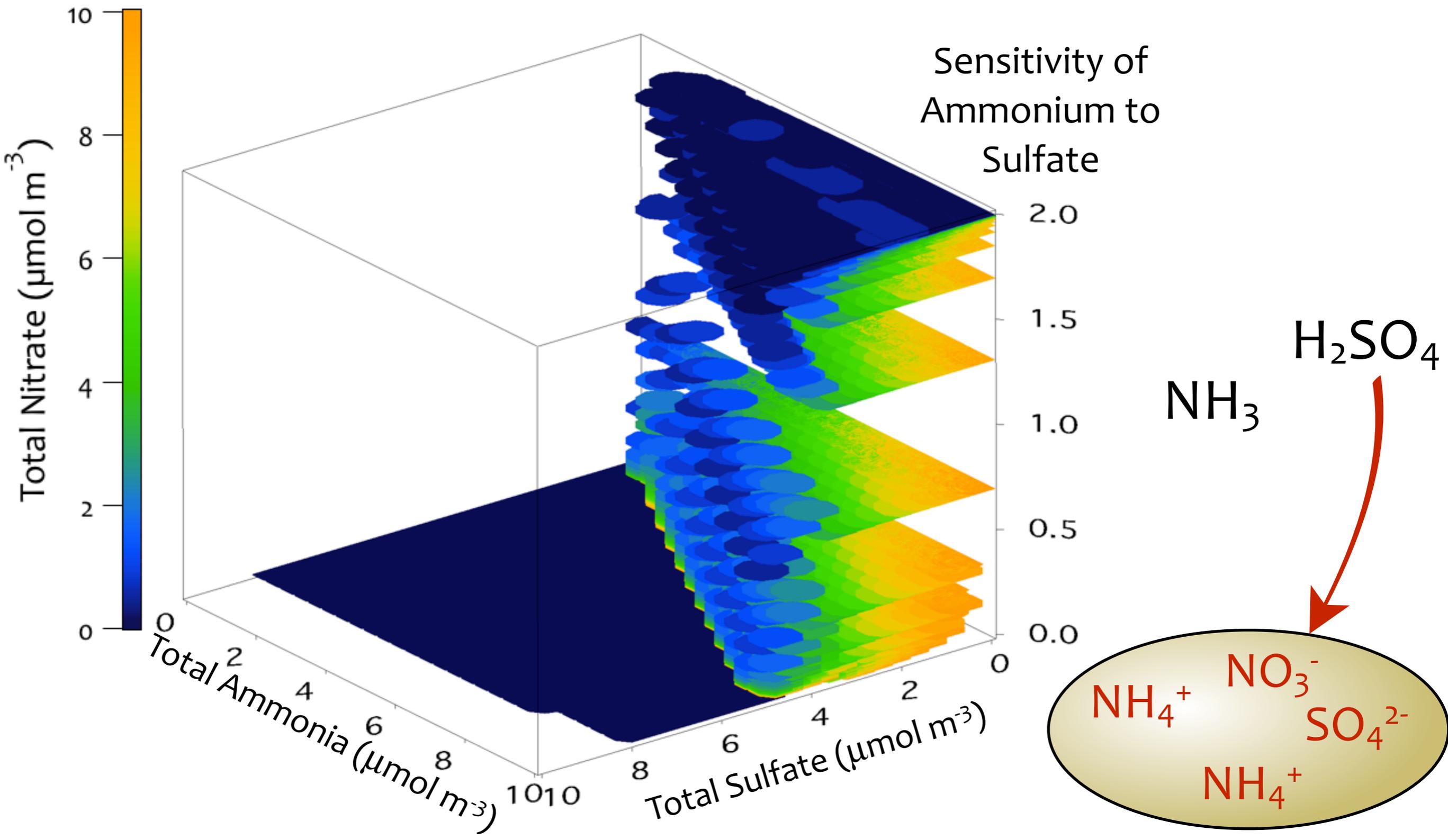
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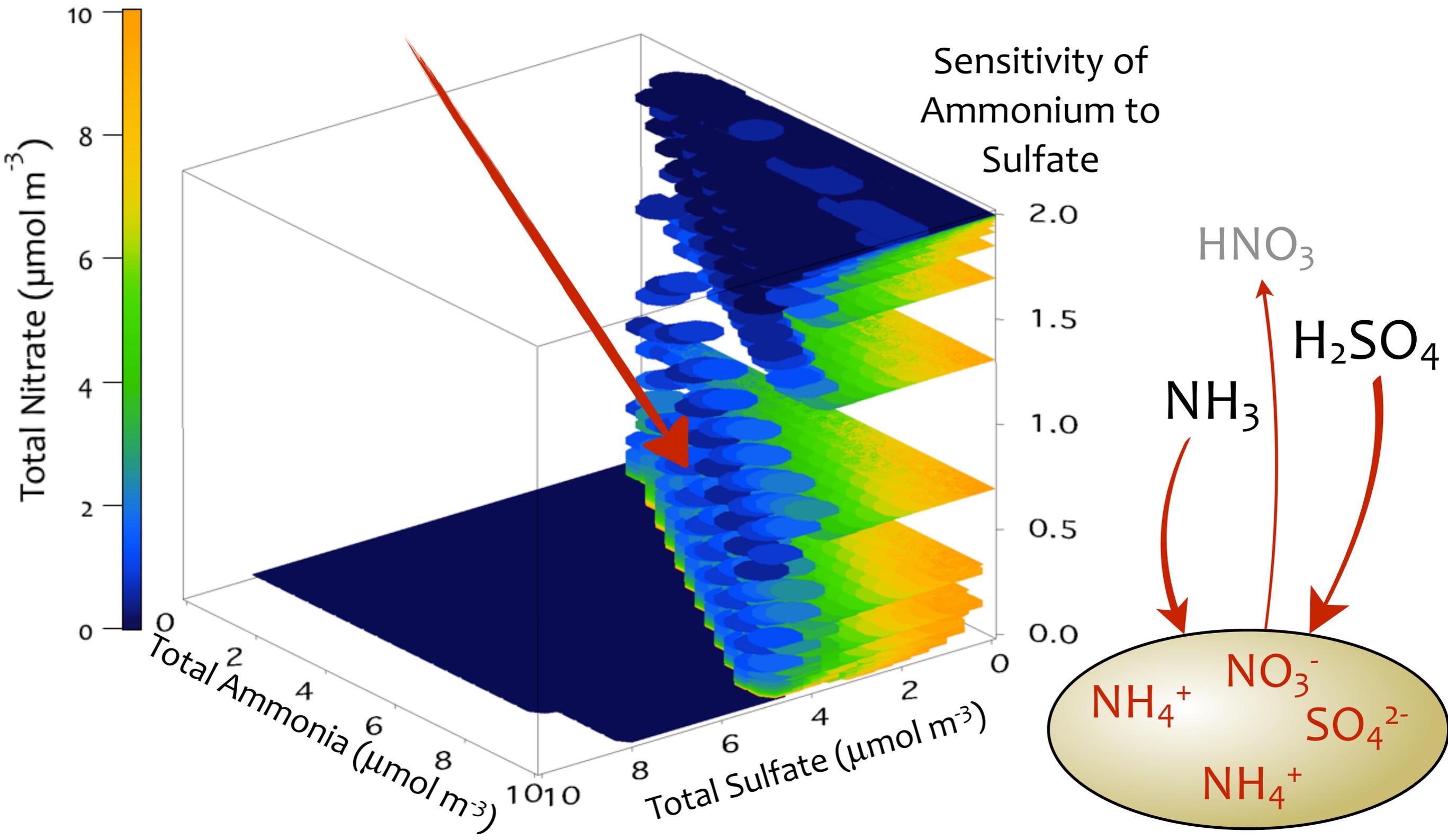
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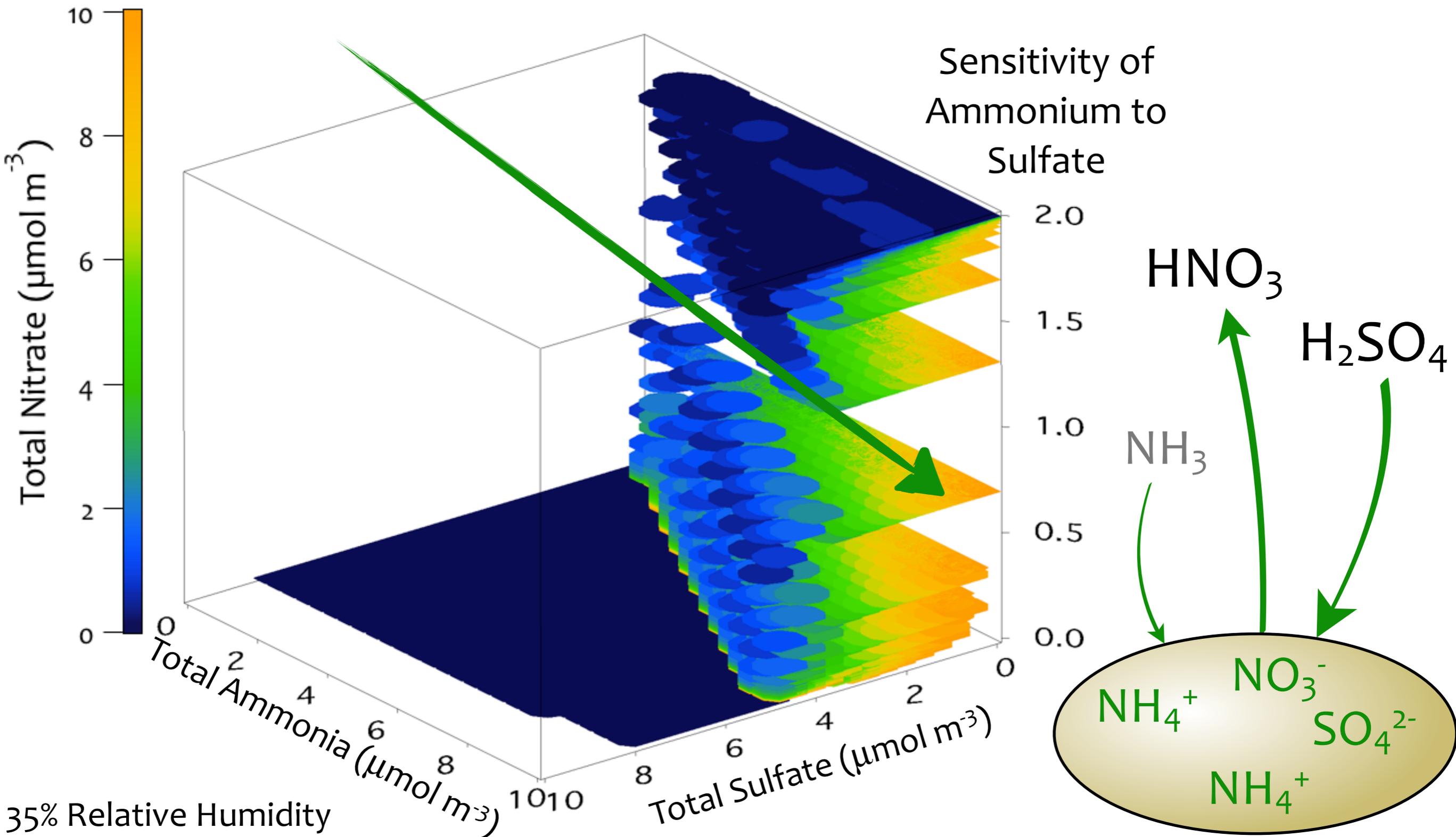
# Sensitivity Space: Ammonium, Sulfate, and Nitrate



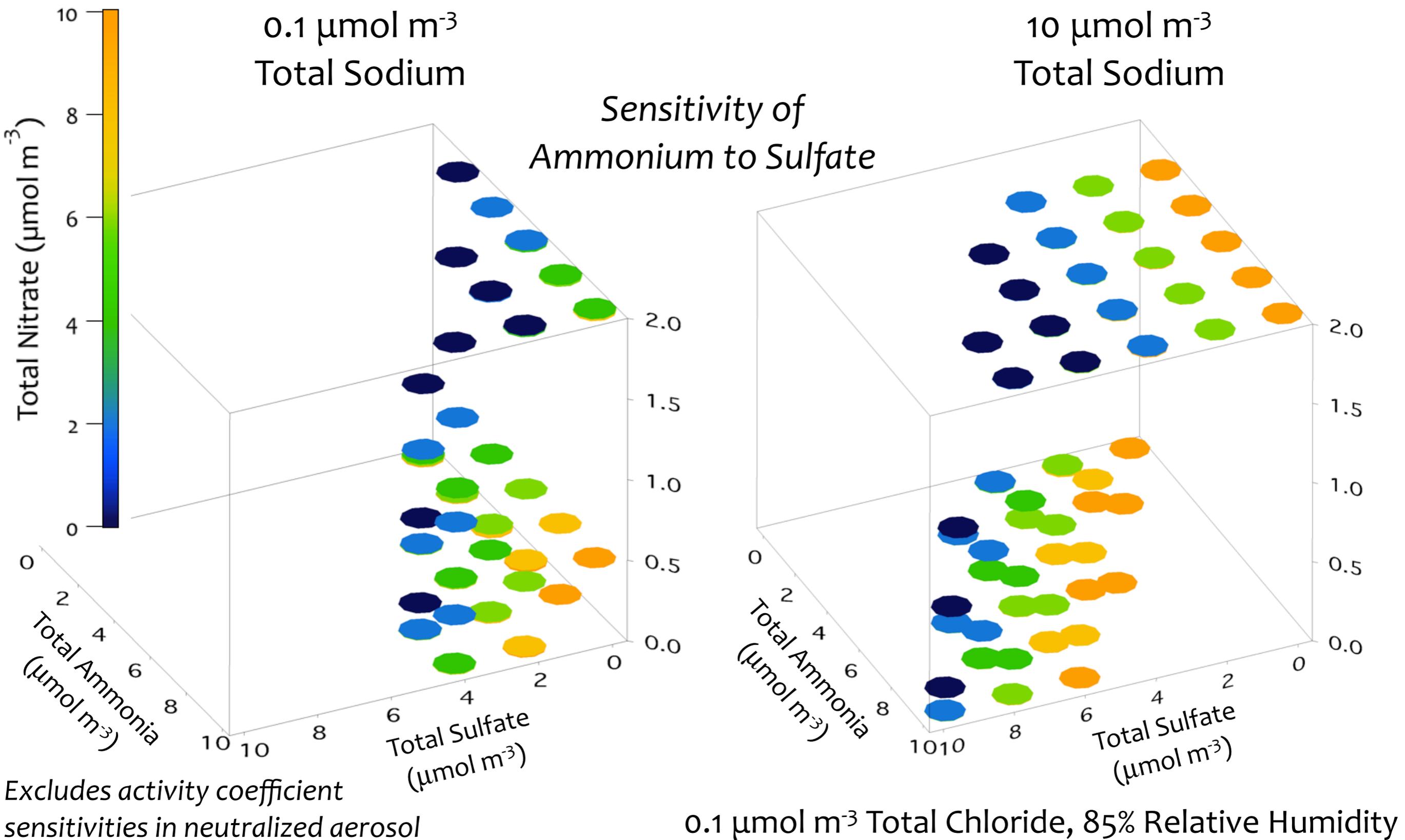
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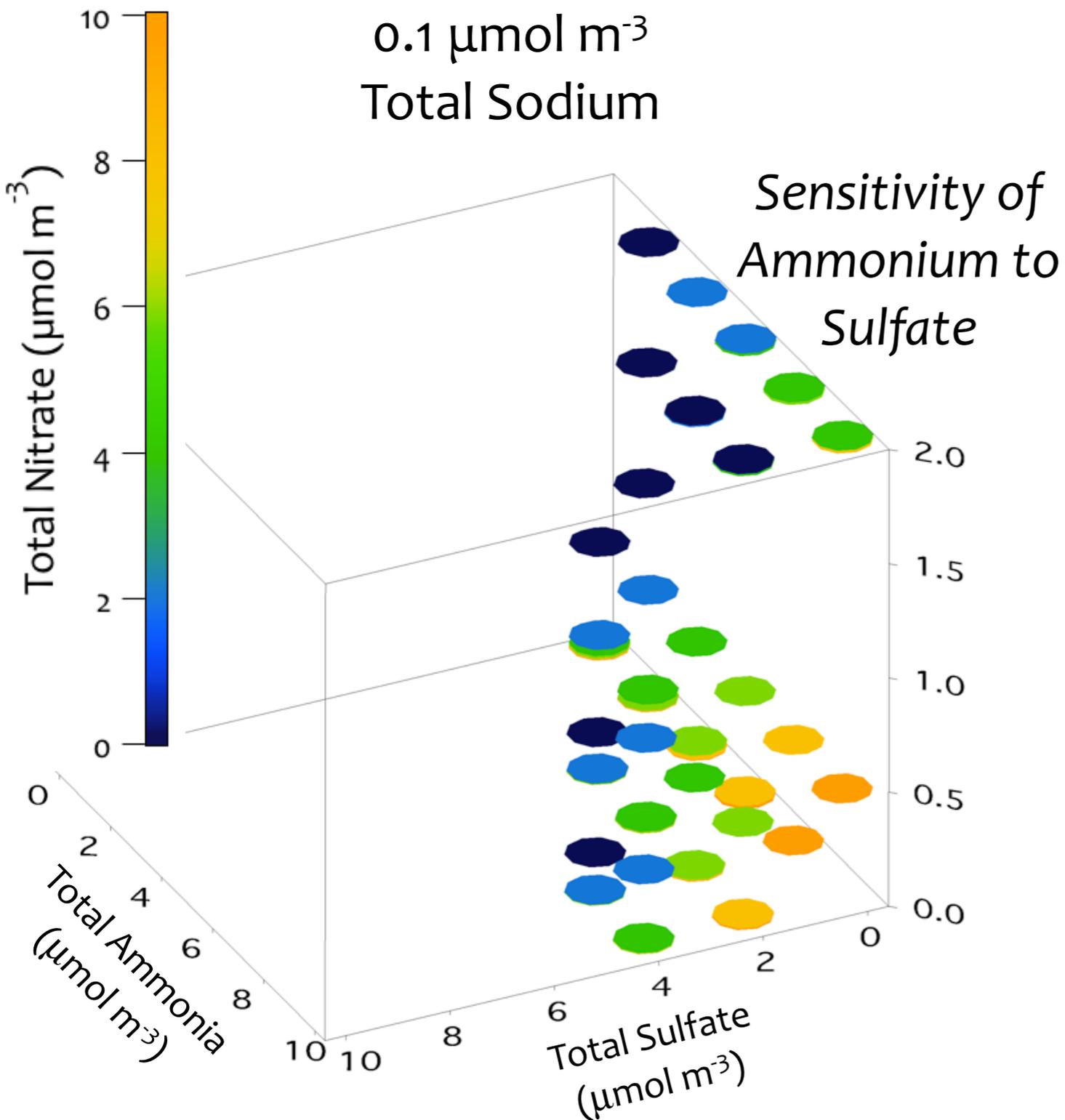


# Sensitivity Space: Ammonium, Sulfate, Nitrate, Chloride, and Sodium

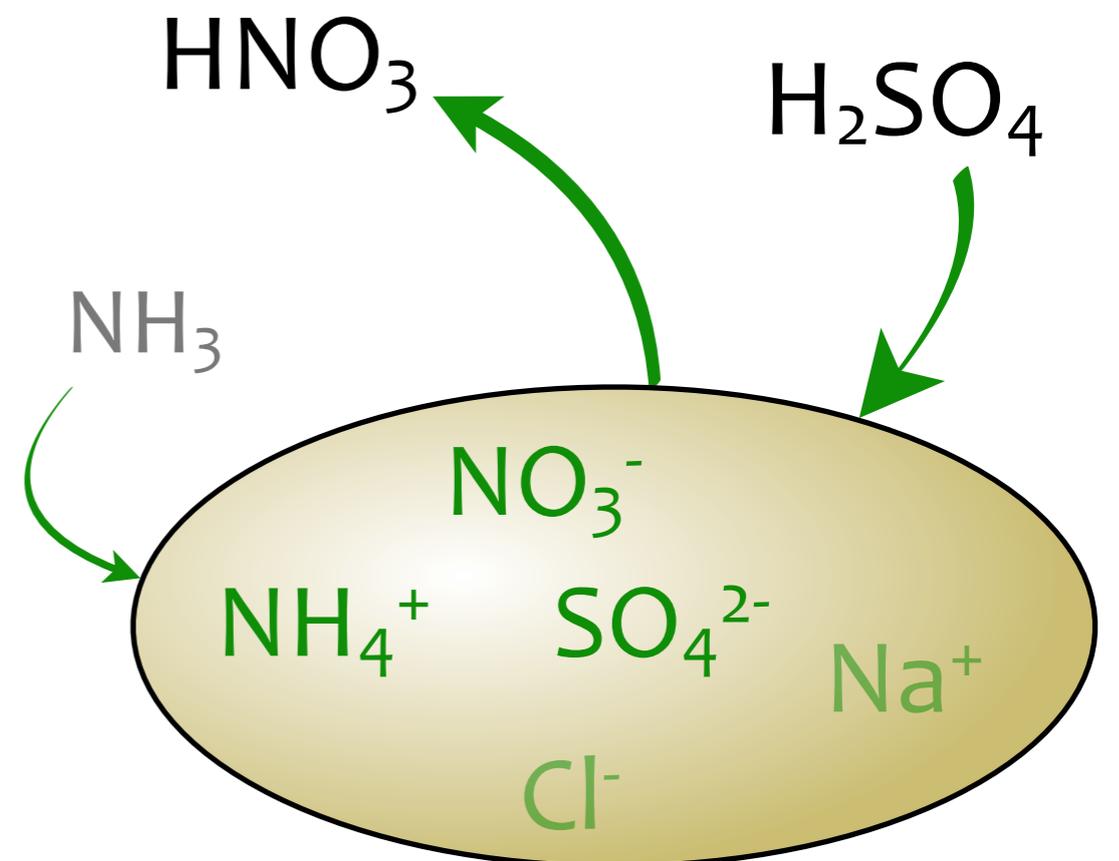


# Sensitivity Space:

## Ammonium, Sulfate, Nitrate, Chloride, and Sodium



- Limited by 2:1 ammonium to sulfate ratio
- Buffering effect of nitrate

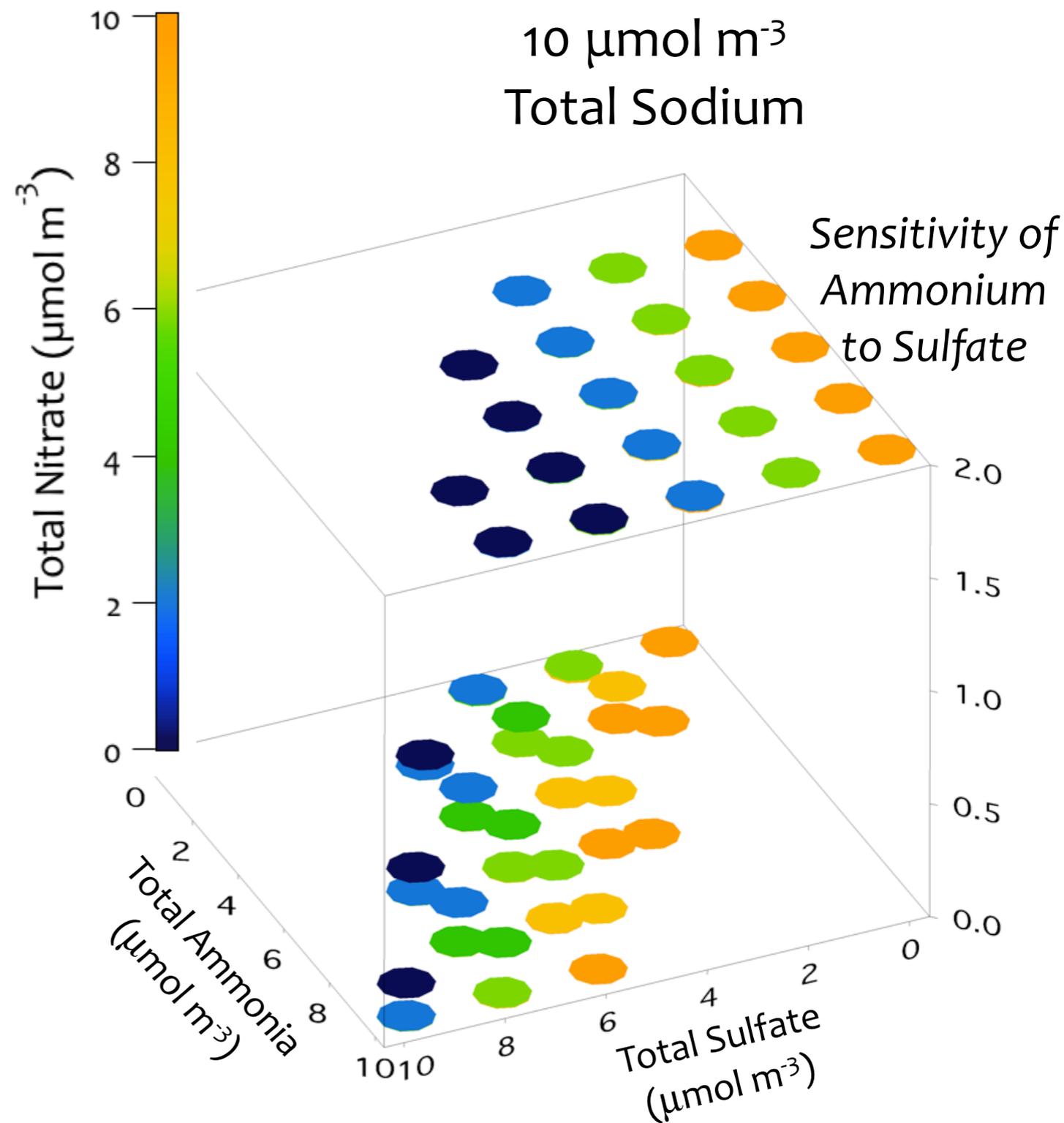
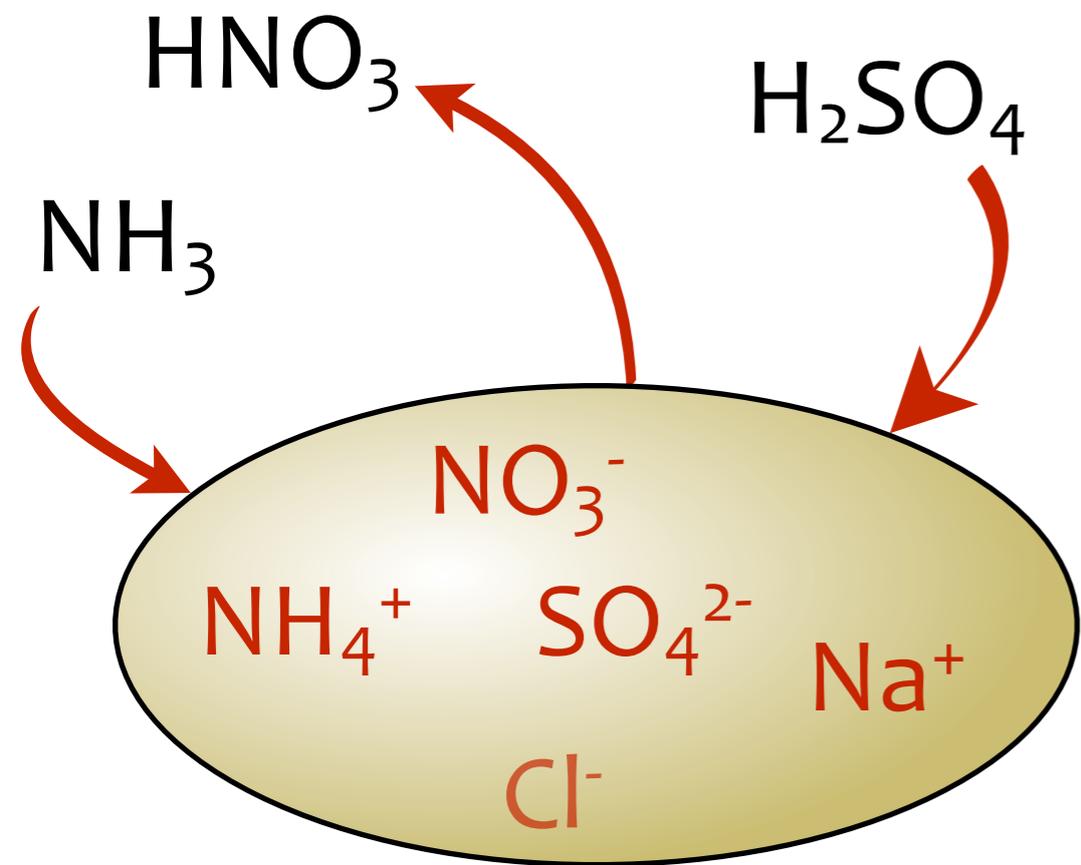


0.1  $\mu\text{mol m}^{-3}$  Total Chloride, 85% Relative Humidity

# Sensitivity Space:

## Ammonium, Sulfate, Nitrate, Chloride, and Sodium

- Sodium extends range of non-zero ammonium sensitivity
- Buffering effect of nitrate



0.1  $\mu\text{mol m}^{-3}$  Total Chloride, 85% Relative Humidity

# Future Work

- Attributing speciated aerosol concentration to sources requires implementation within the full CMAQ-Adjoint - see ShunLiu Zhao's talk at 4:40 pm

- Further refining routines with  $\text{Na}^+ - \text{Cl}^-$
- Extending to  $\text{K}^+ - \text{Mg}^{2+} - \text{Ca}^{2+}$

