

Adjoint sensitivity analysis of PM_{2.5} sources: constraining NH₃ emissions

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Collaborators



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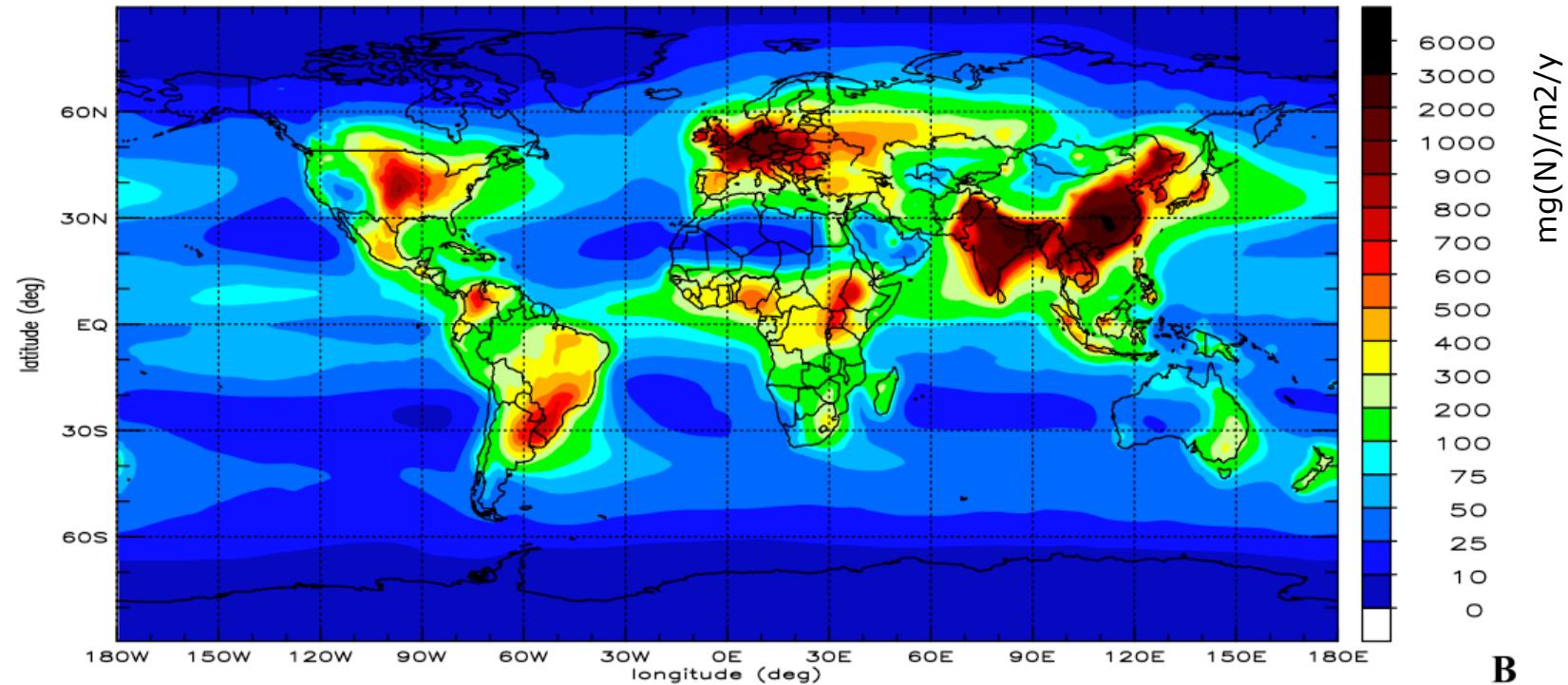
Rob Pinder, John Walker



NASA GSFC: NCCS
NASA JPL: SCC

Environmental impacts NH₃

Estimated N deposition from NHx, Dentener et al. (2006).

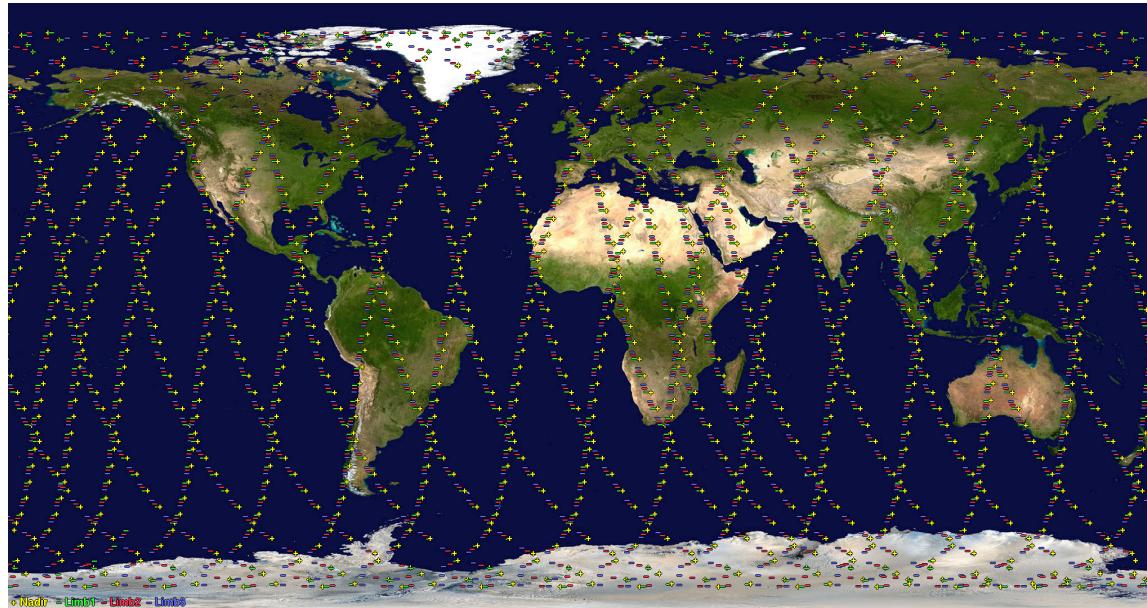


Denman et al. (2007), IPCC: NH₃ emissions to double by 2050.

Galloway et al. (2008), *Science*: Importance of atmospheric NHx transport

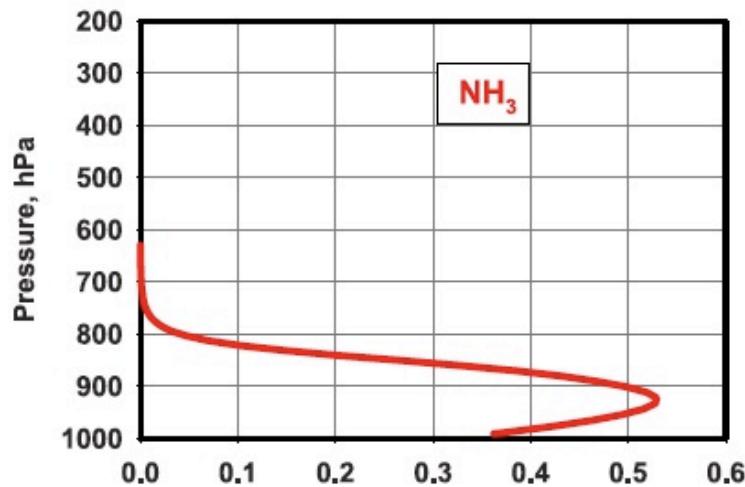
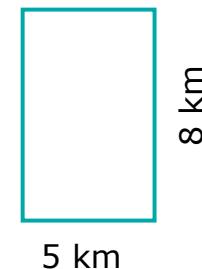
Schlesinger (2009), *PNAS*: a 46 Tg gap in N budget?

TES: remote sensing of NH₃



"covers" globe
in 16 days

Footprint size:

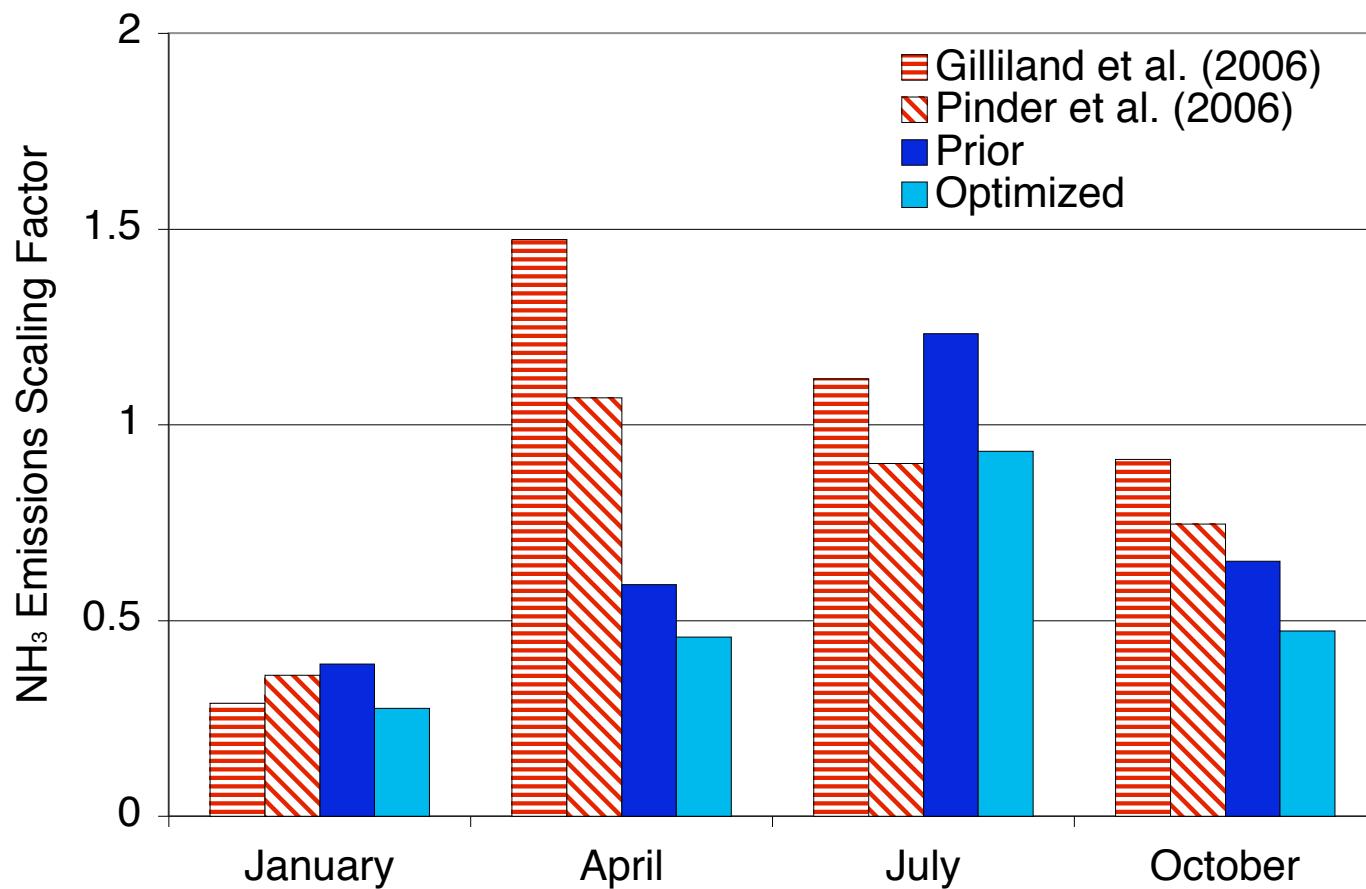


Vertical profile yields
~1 DOF near 900 hPa

Beer et al., 2008

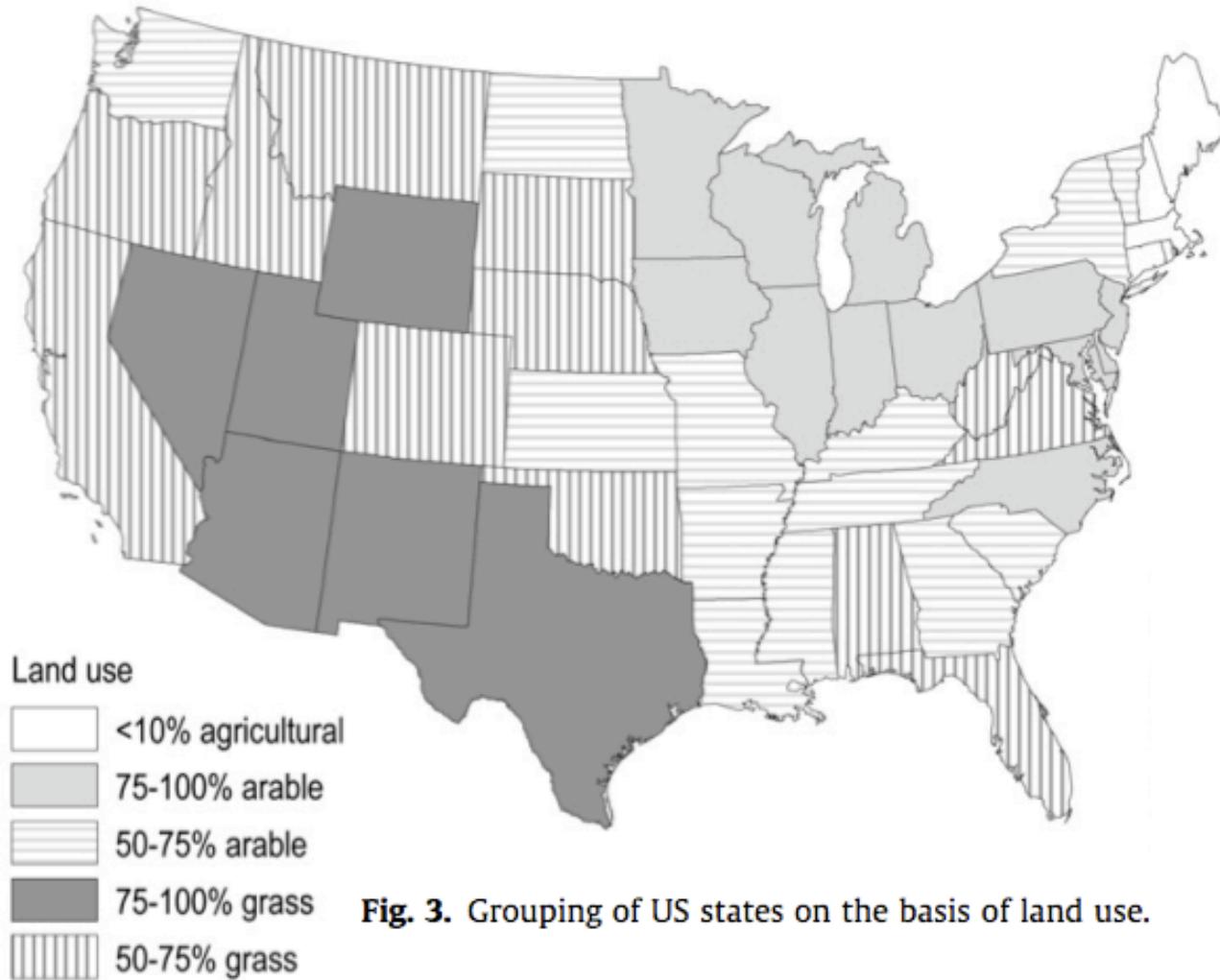
Inverse modeling: NH₃

Sum changes to NH₃ over U.S., compare to other inverse modeling (Gilliland et al., 2006) and bottom up (Pinder et al., 2006)

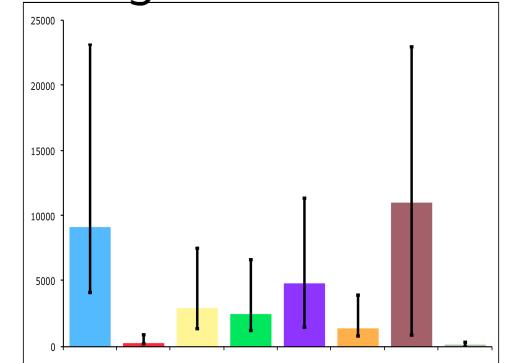


Seasonal peak in April or July? (Henze et al., 2009)

NH₃ emissions variability and uncertainty: Beusen et al. (2008)



Global animal
NH₃ emissions



Source types

- housing mixed
- housing pastoral
- grazing mixed
- grazing pastoral
- spreading cropland
- spreading grassland
- fertilizer cropland
- fertilizer grassland

Model sensitivity

Model: estimates, c_i , and parameters, p_i

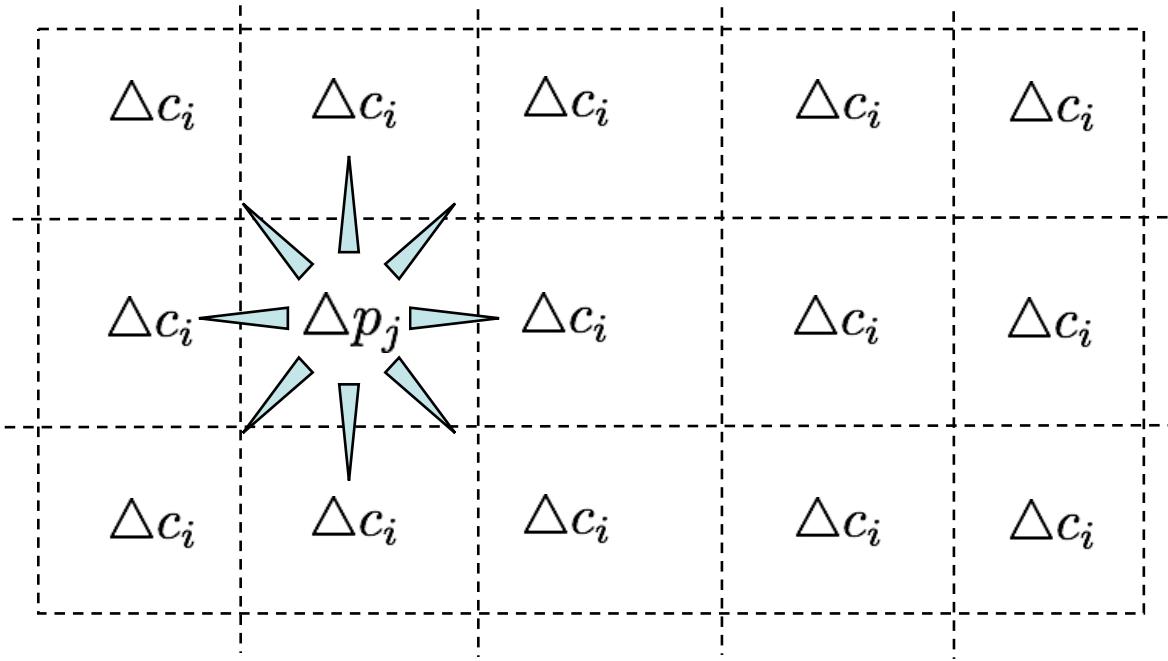
c_i, p_i				
c_i, p_i				
c_i, p_i				

Ideally, want model Jacobian,

$$\frac{\partial \mathbf{c}}{\partial \mathbf{p}} = \begin{bmatrix} \frac{\partial c_1}{\partial p_1} & \cdots & \frac{\partial c_1}{\partial p_J} \\ \vdots & \ddots & \vdots \\ \frac{\partial c_I}{\partial p_1} & \cdots & \frac{\partial c_I}{\partial p_J} \end{bmatrix}$$

but it is generally much too large to calculate.

Forward sensitivity



$$\frac{\partial \mathbf{c}}{\partial p_j} = \begin{bmatrix} \frac{\partial c_1}{\partial p_1} & \cdots & \frac{\partial c_1}{\partial p_J} \\ \vdots & \ddots & \vdots \\ \frac{\partial c_I}{\partial p_1} & \cdots & \frac{\partial c_I}{\partial p_J} \end{bmatrix} \mathbf{u_j}$$

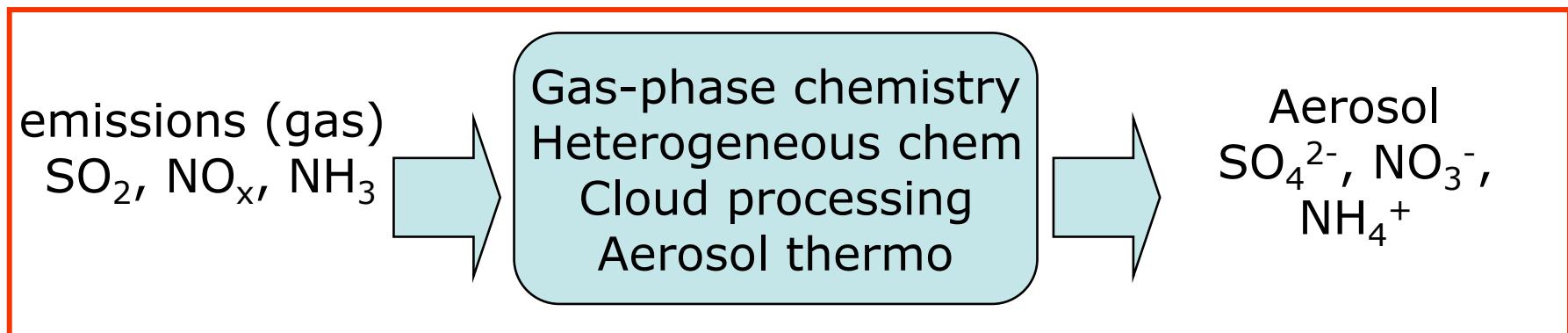
Adjoint sensitivity

The diagram shows a 3x5 grid of dashed boxes, each containing the symbol δp_j . A specific row in the middle of the grid is highlighted with red arrows pointing to the right, indicating the adjoint sensitivity vector \mathbf{u}_i .

$$\begin{bmatrix} \frac{\partial c_1}{\partial p_1} & \cdots & \frac{\partial c_1}{\partial p_J} \\ \vdots & \ddots & \vdots \\ \frac{\partial c_I}{\partial p_1} & \cdots & \frac{\partial c_I}{\partial p_J} \end{bmatrix}^T \mathbf{u}_i = \frac{\partial c_i}{\partial \mathbf{p}}$$

4D-Var with GEOS-Chem Adjoint Model

Forward model v6-02-05 (*Bey et al., 2001; Park et al., 2004*)



All included in adjoint (*Henze et al., 2007*)

Calculates sensitivity of single model response w.r.t. all model parameters in $t = 3 \times t_{forward}$

Testing the Adjoint Model: Gradient Check

Check gradient using finite difference calculation

$$\lambda \approx \frac{J(\sigma + \delta\sigma) - J(\sigma - \delta\sigma)}{2\delta\sigma}$$

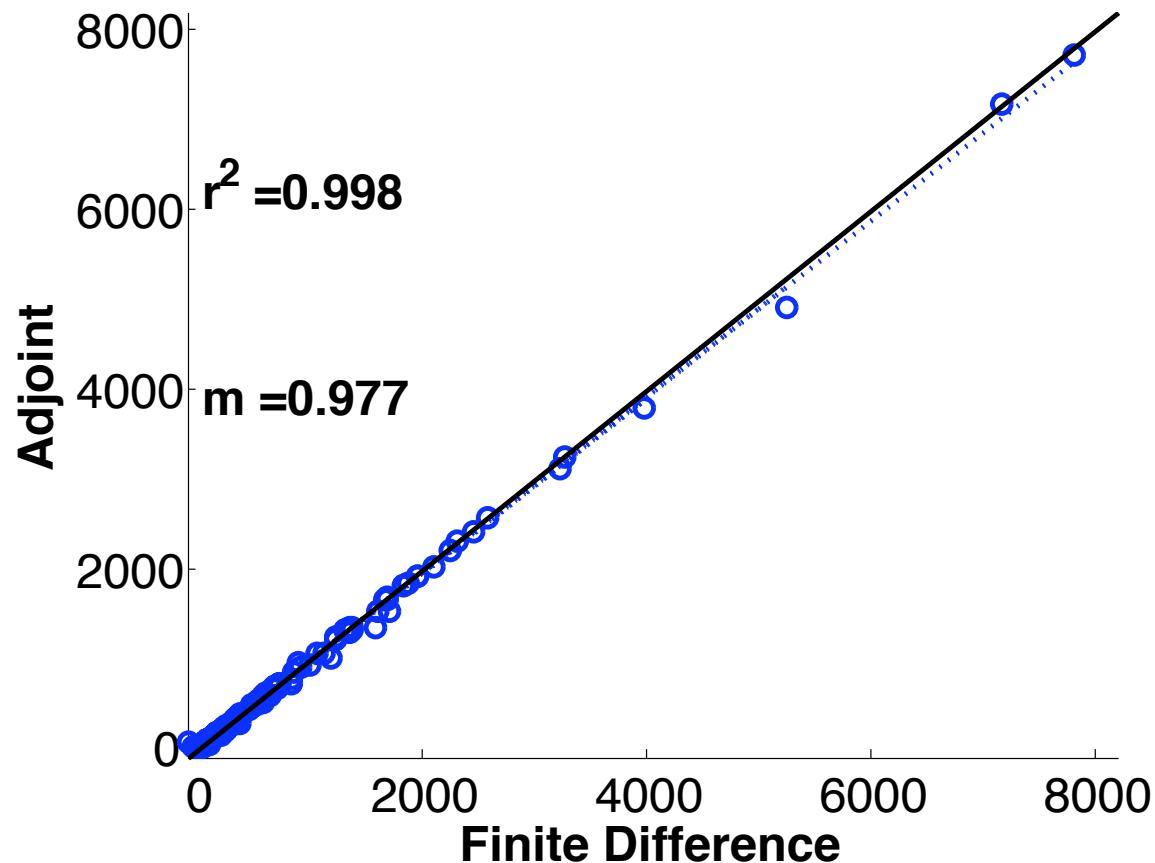
J = model response

σ = control parameter (emissions)

λ = adjoint sensitivity

Component-wise analysis affords domain wide points-of-comparison (e.g., Hakami et al., 2007)

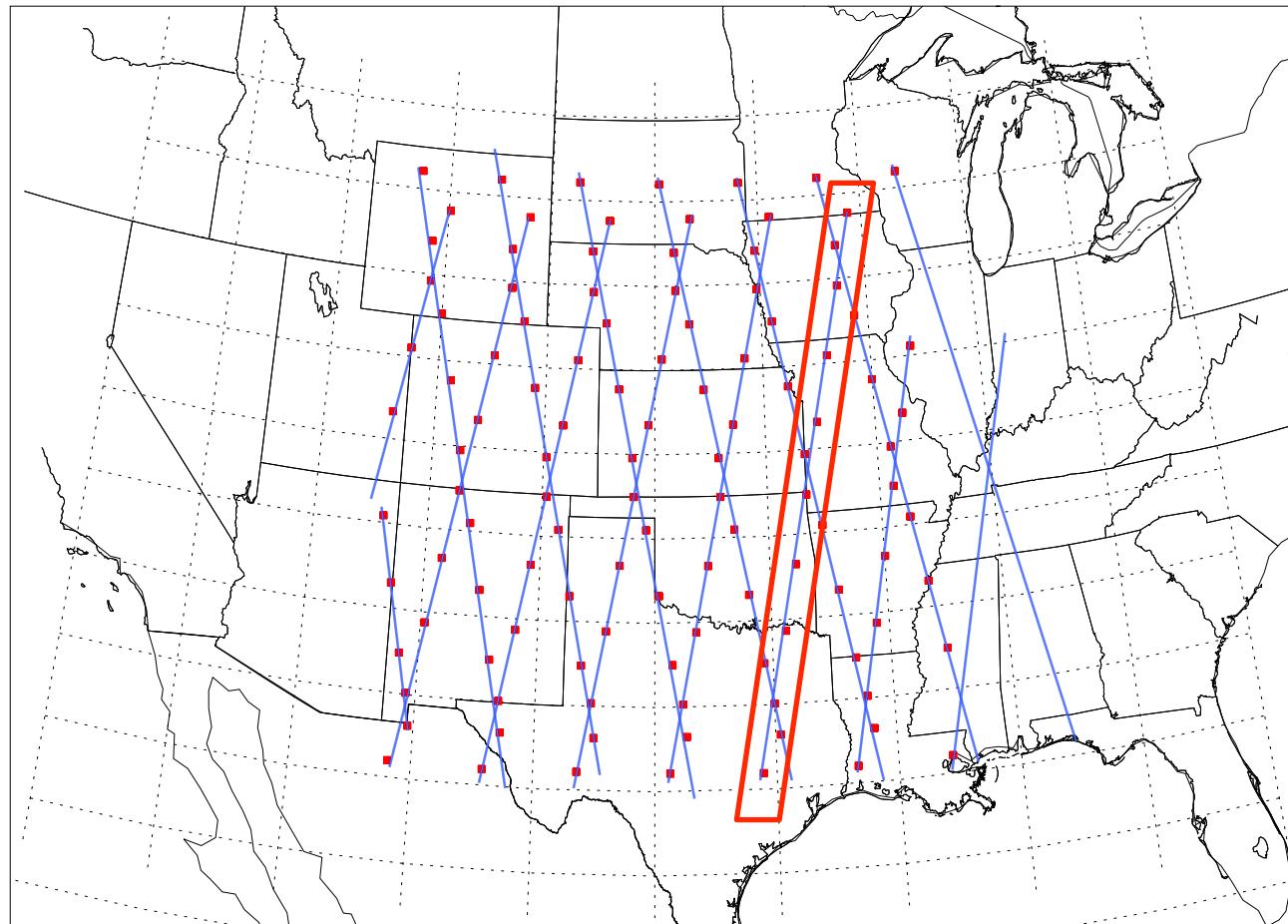
Adjoint model validation



Sensitivity of nitrate aerosol w.r.t ammonia emissions,
1 week, all processes other than horizontal transport

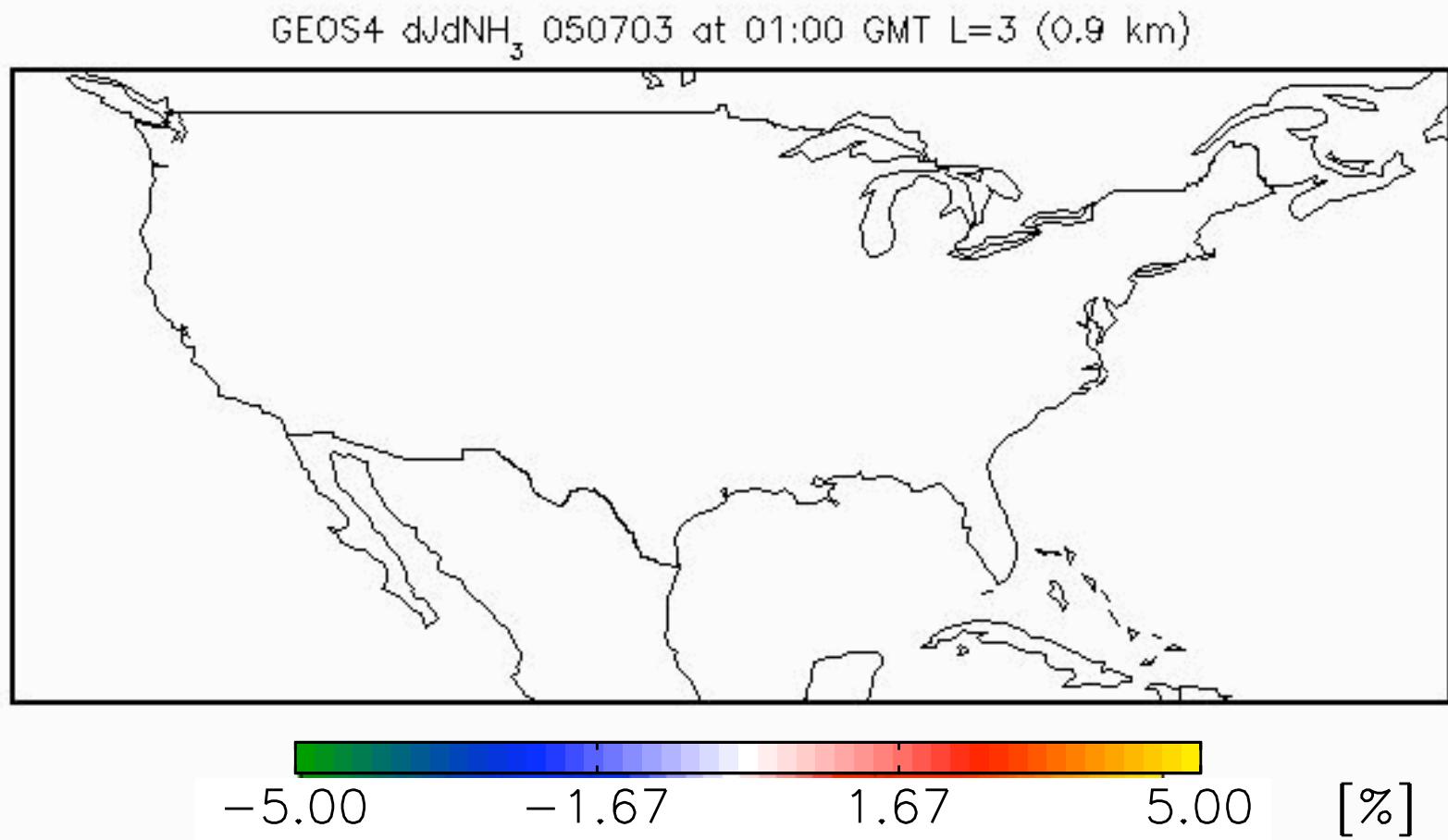
Potential for further constraints: TES coverage for 2 weeks in July 2005

TES GS Footprints, July 4-19, 2005



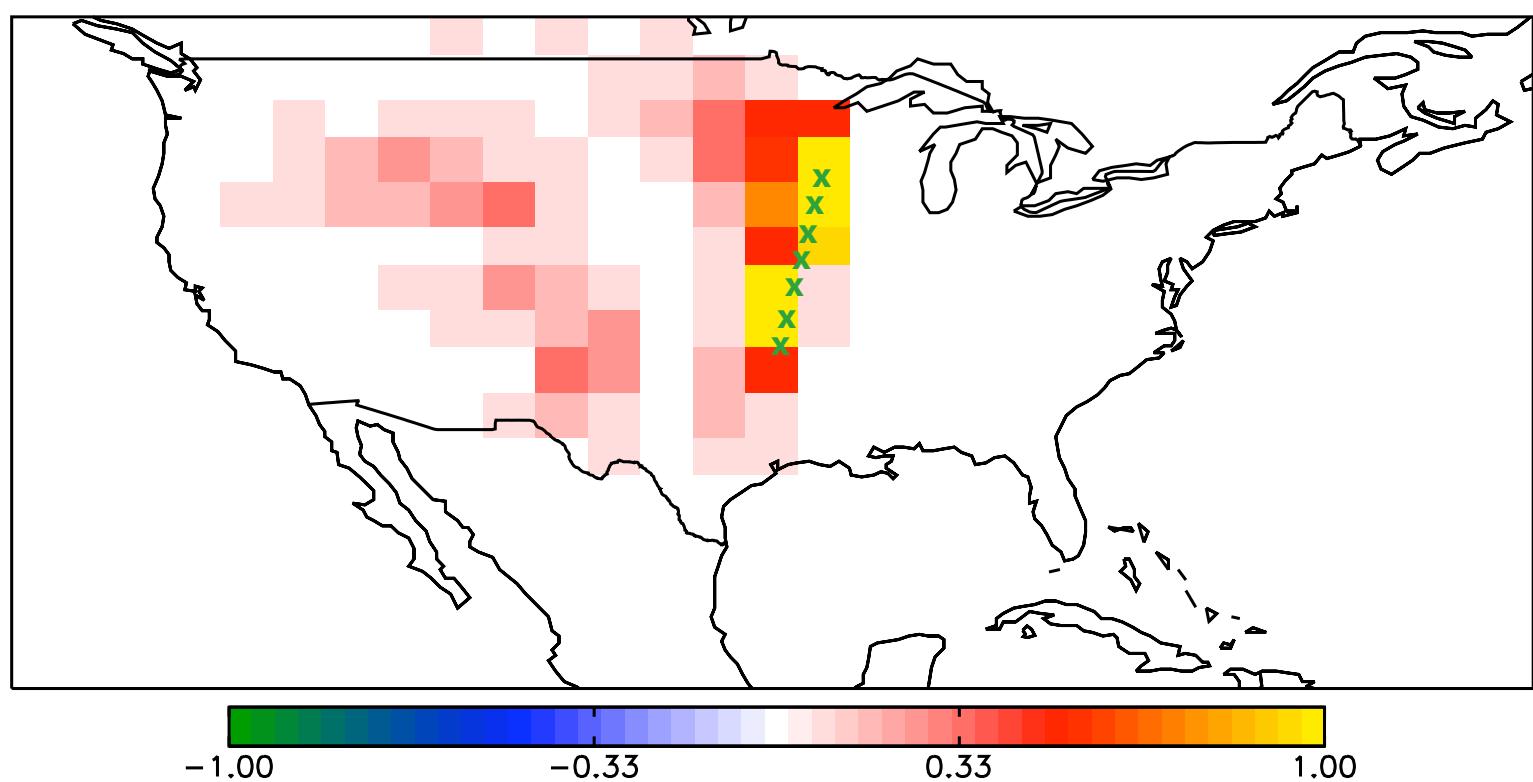
data selection: lat = [30.,45.], lon = [-110, -90.]; total number of profiles: 115

Adjoint sensitivities of modeled NH_3 retrievals



Sensitivities show the origins of the NH_3 that eventually will be “observed” by TES

Sensitivity of TES observation in the track highlighted on previous slide to NH_3 emissions from the week prior



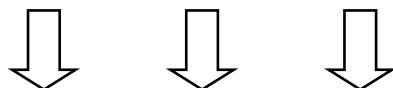
Inverse Modeling: twin experiment

Parameter Estimate

$$\mathbf{p} = 2\mathbf{p}_a$$

Forward Model

$$t_0 \longrightarrow t_f$$



Simulated NH₃ field
 $\mathbf{c}(x, t)$

Pseudo retrievals, IMPROVE observations

Inverse Modeling using Adjoint Model

Inverse Model

Parameter Estimate

$$\mathbf{p} = \mathbf{p}_a$$

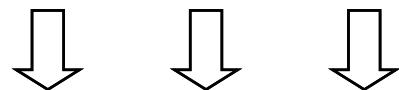


Forward Model

$$t_0 \longrightarrow t_f$$

Adjoint Model

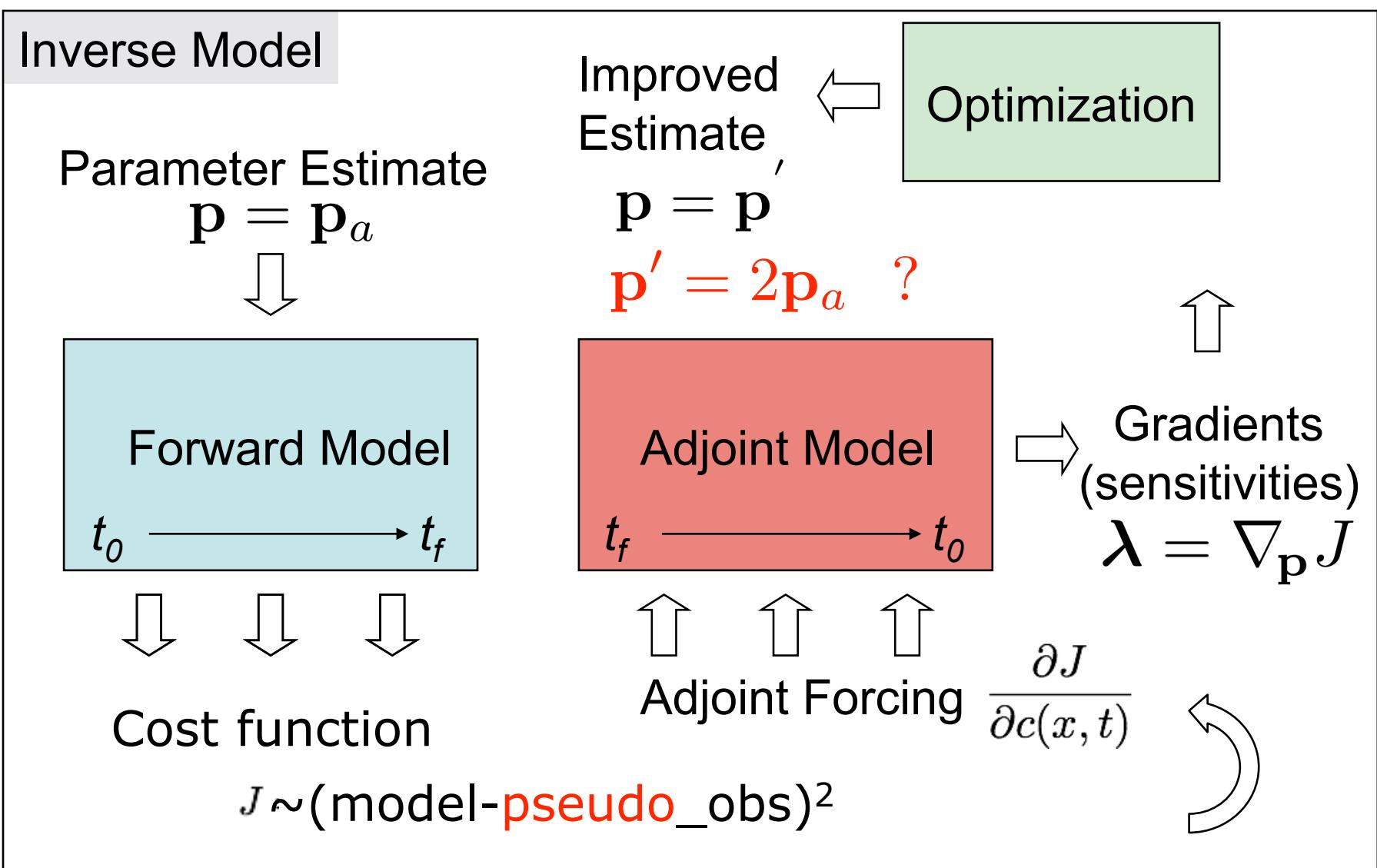
$$t_f \longrightarrow$$



Predictions

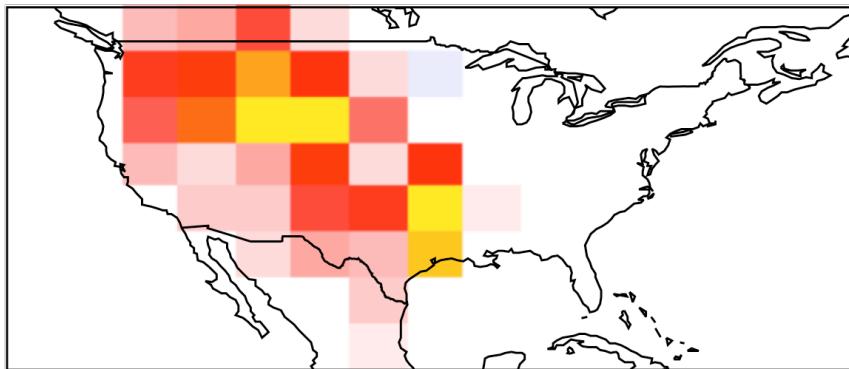
$$\mathbf{c}(\cdot^J : t)$$

Inverse Modeling using Adjoint Model

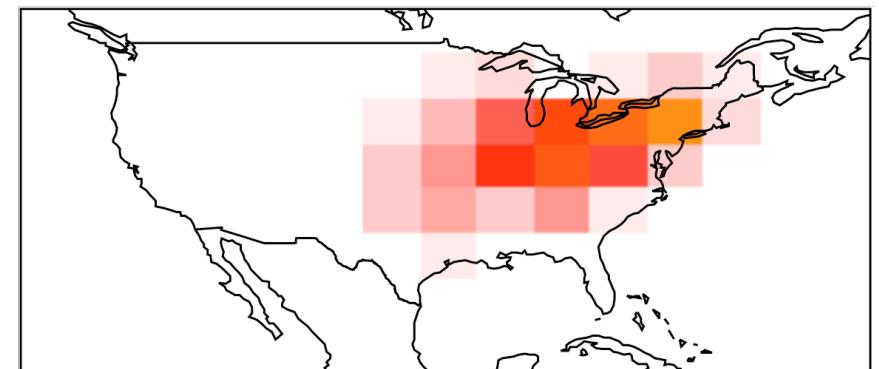


Inverse modeling tests with different obs

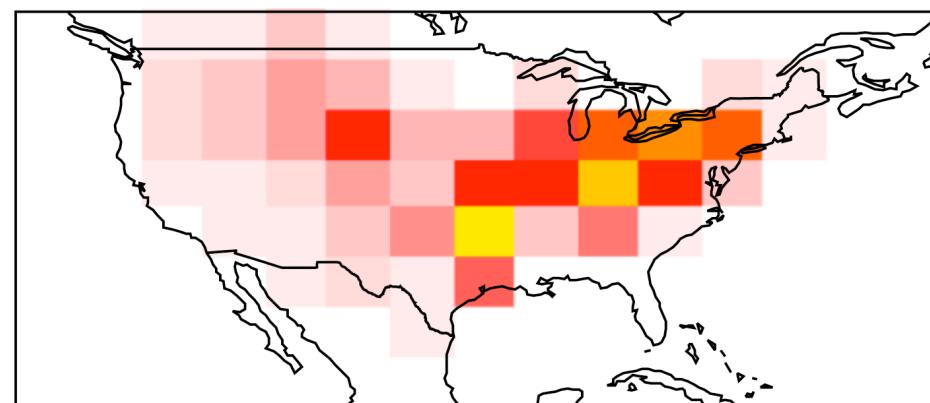
Using TES NH3



Using IMPROVE SO₄, NO₃



Using both

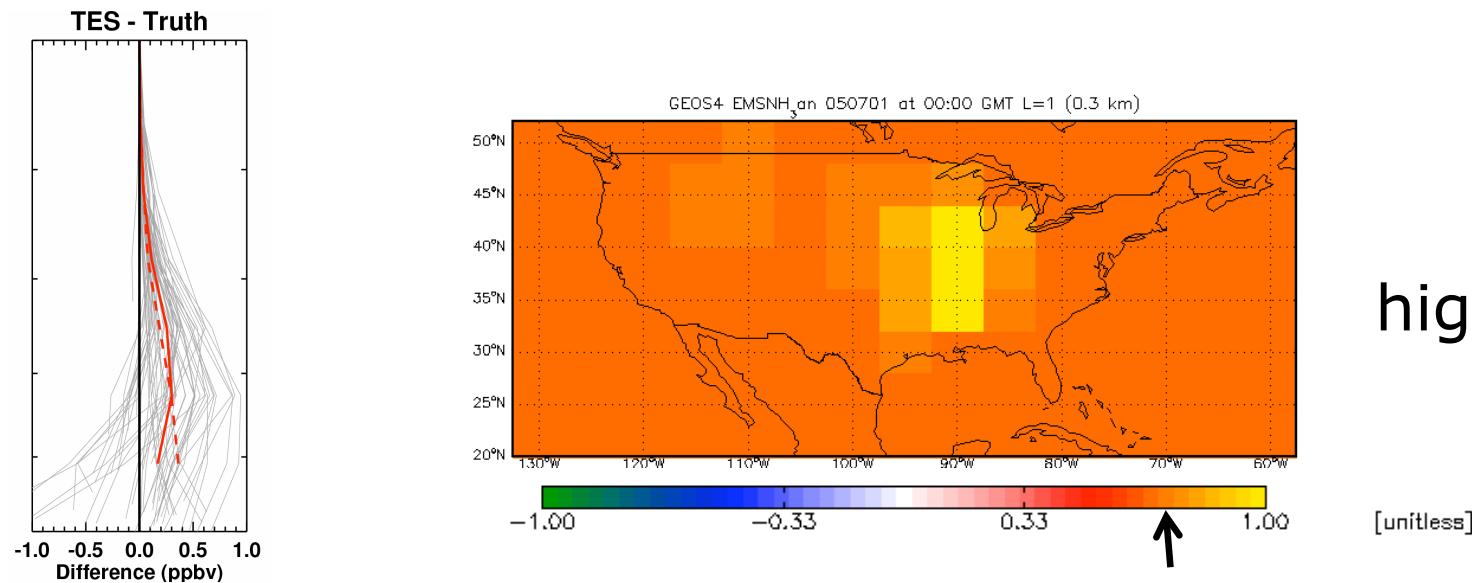


$$\sigma = \ln(p/p_a)$$

A horizontal color bar indicating the range of the variable σ . The bar is divided into two sections: "Initial Guess" (left, blue) and "Truth" (right, orange). The scale ranges from -1.00 to 1.00, with major tick marks at -1.00, -0.33, 0.33, and 1.00. The color transitions from dark blue for negative values to white for zero, and then to orange for positive values.

Ongoing efforts

- Model transport bias – use GEOS 3 / 4 / 5 met fields
- Intercomparison of TES / in situ obs / CMAQ
- Retrieval bias
 - inversion that starts with doubled emissions



high bias?

Final Remarks

- Remote sensing of NH_3 provides valuable constraints
- Multiple types of gas and aerosol observations required to constrain NH_3 emissions
- Adjoint approach spreads information from NH_3 observations across wide domain
- How well can we constrain magnitudes vs locations?
Are we aliasing for bi-directional flux?

the end

Satellite: indirect observations

Arriving at an NH₃ “observation” is an inverse problem itself.

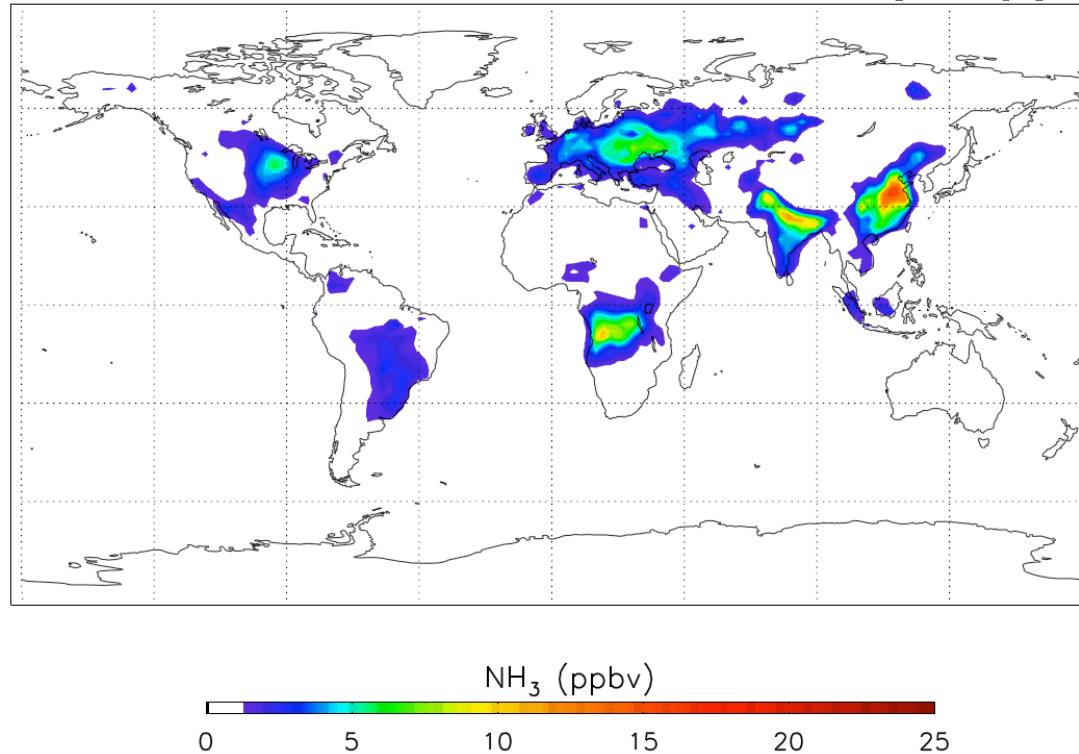
- ill posed (multiple atmospheric states look alike)
- constraints required

Satellite products are a mix of measured and modeled quantities.

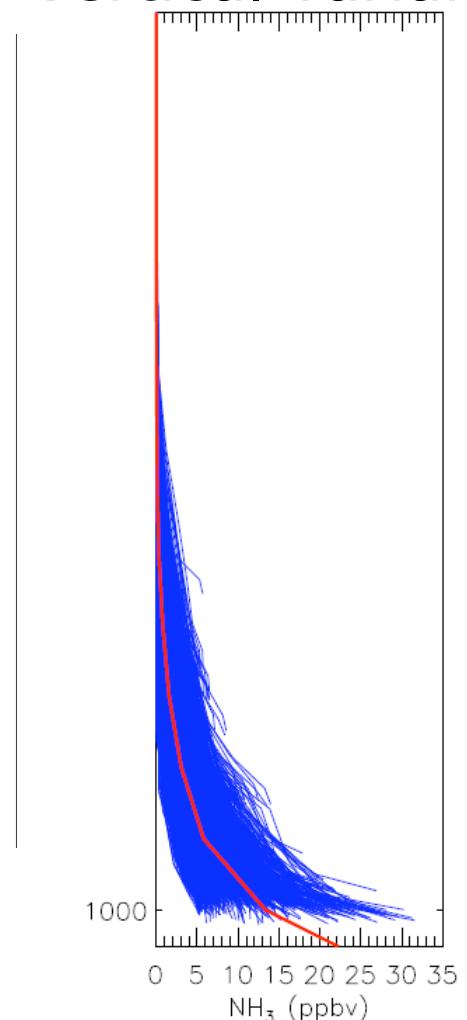
- model estimates used for initial profile
- profile scaled to match observations
- influence of model estimate can be removed

Global modeling support of NH_3 retrievals

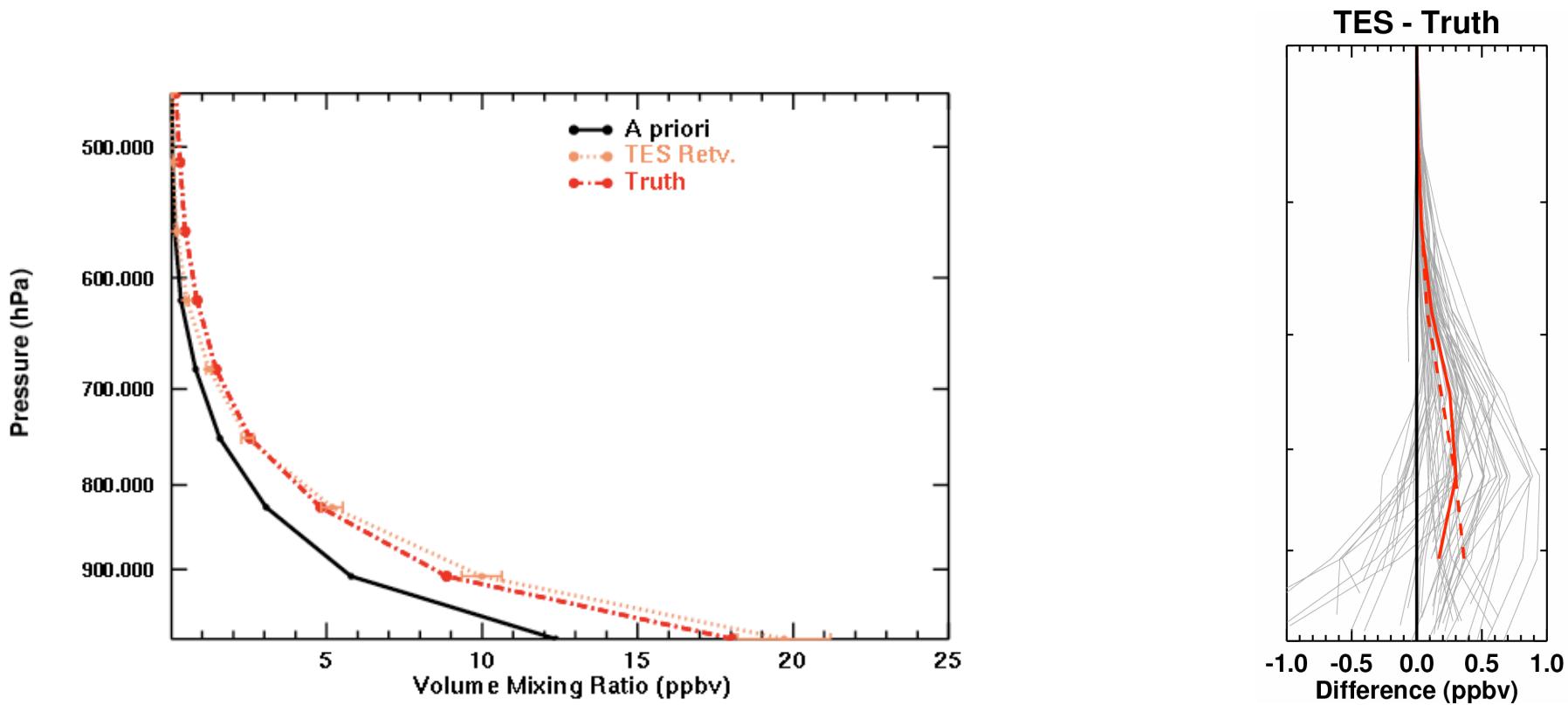
GEOS-Chem surface level (July)



Vertical variance



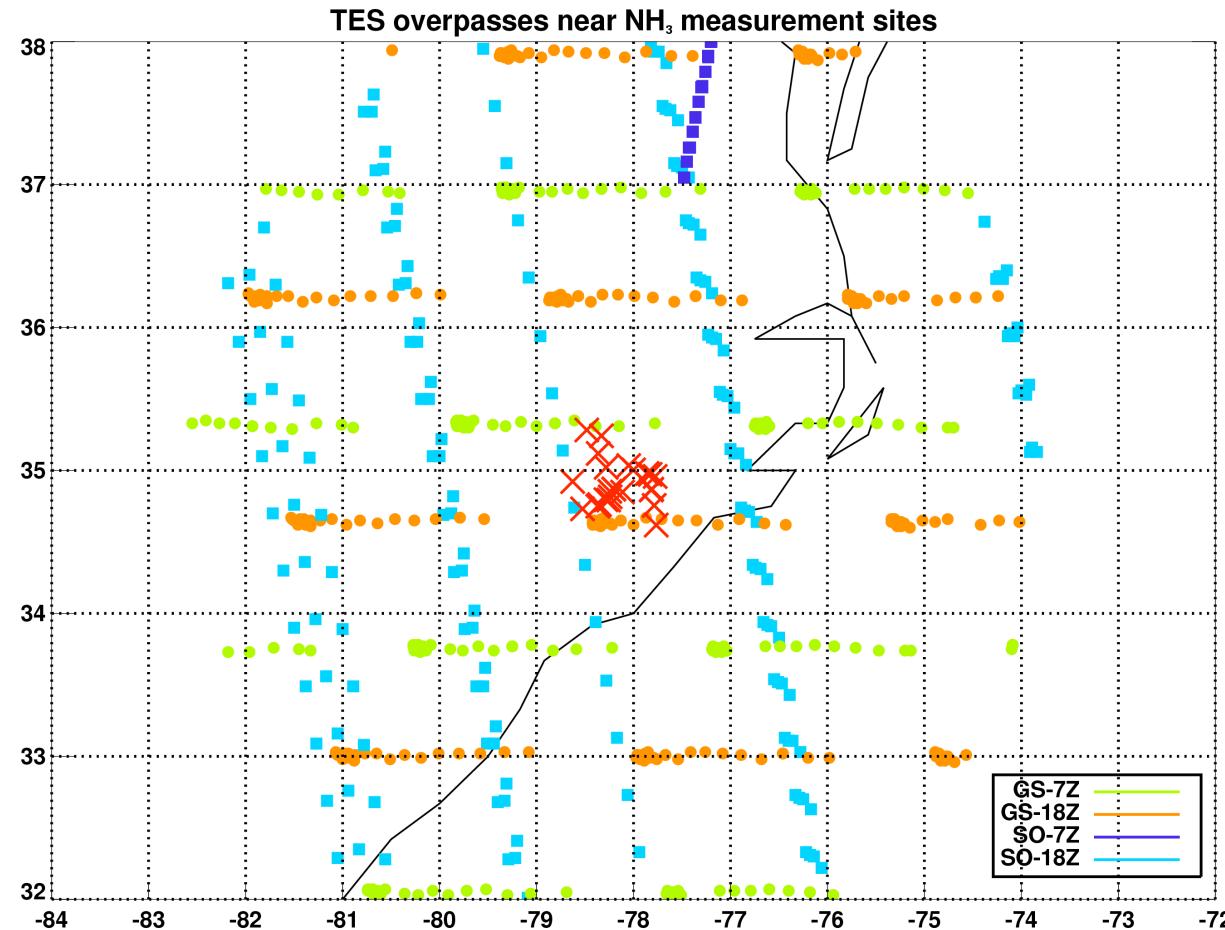
Retrieval tests: simulated NH₃



1. Simulate NH3 from a doubled emissions model run
2. Model what TES would see if “doubled” was truth
3. Starting from non-doubled run, can truth be recovered?

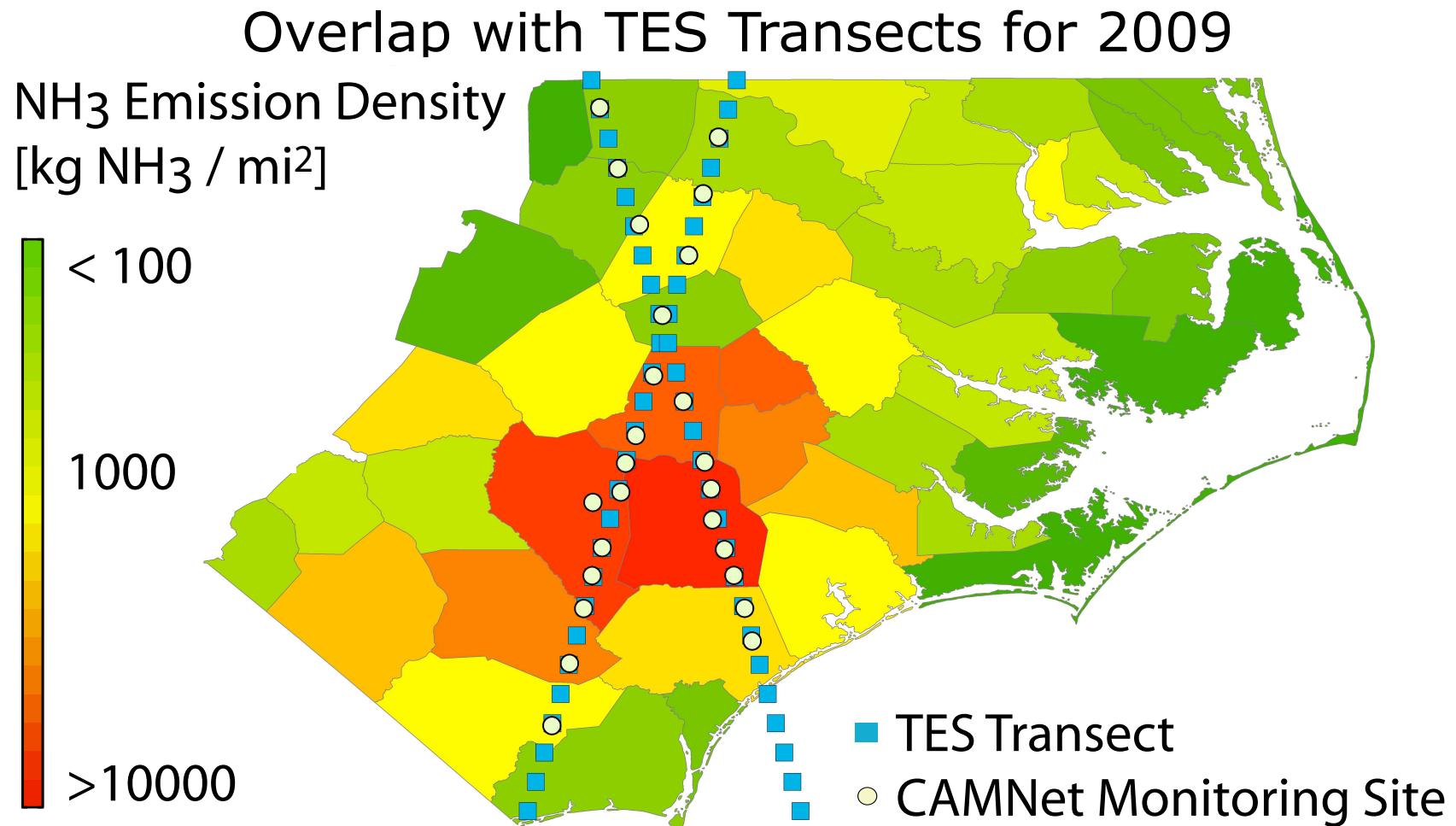
Validating TES NH_3 with surface observations

Standard overlap with North Carolina CAMNet sites



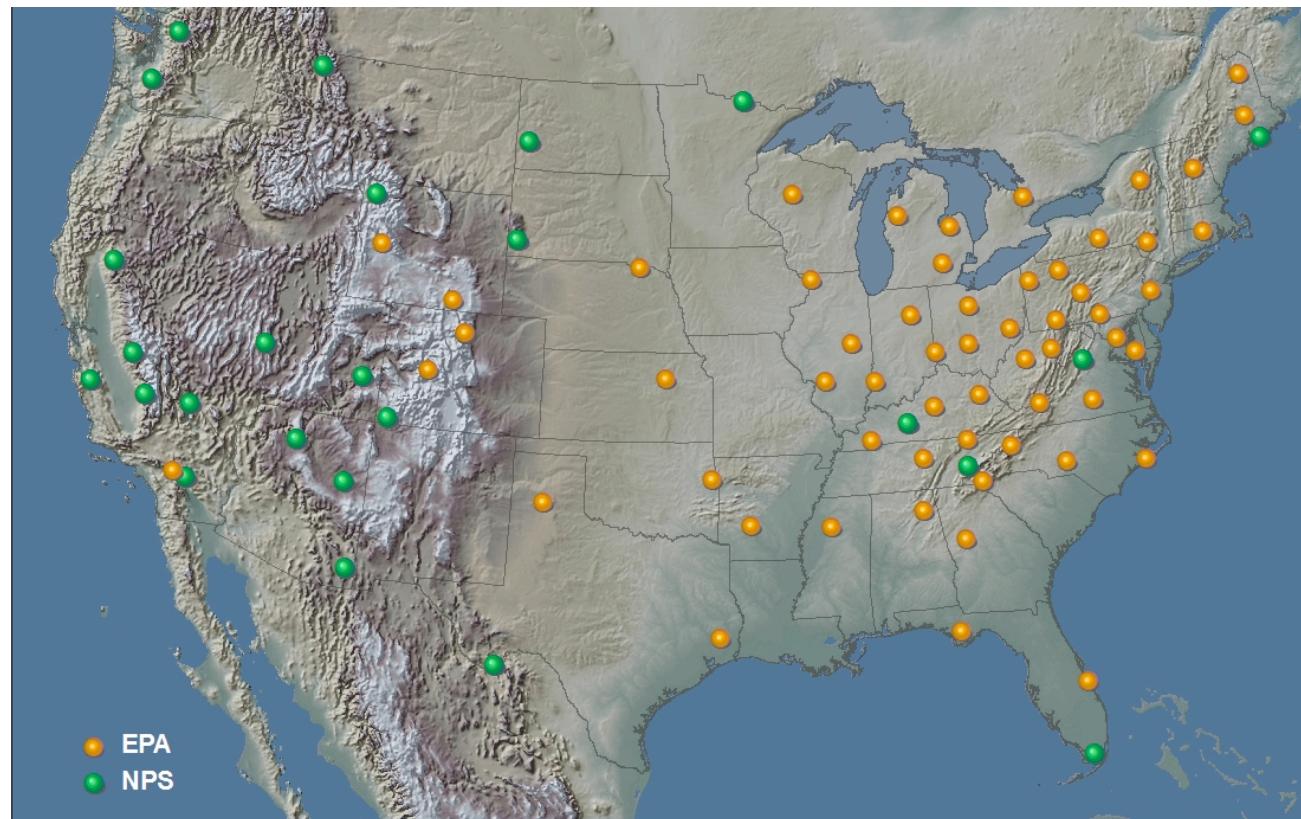
\times = CAMNet NH_3 obs (John Walker, EPA)

Validating TES NH₃ with surface observations



NH_4^+ monitoring

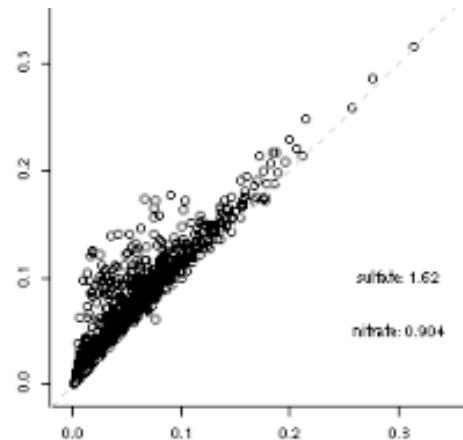
CASTNet



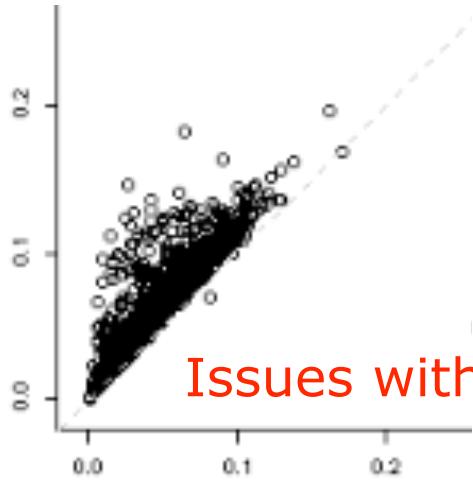
STN: another 200 sites

Checking ion balance: $n(\text{NH}_4^+) : 2n(\text{SO}_4^{2-}) + n(\text{NO}_3^-)$

CASTNet, all sites,
 2005-2006 (R. Pinder)



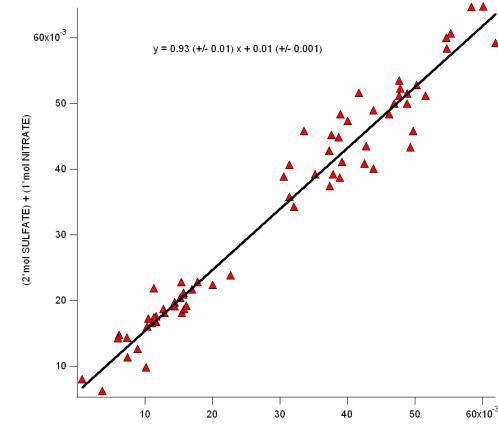
January



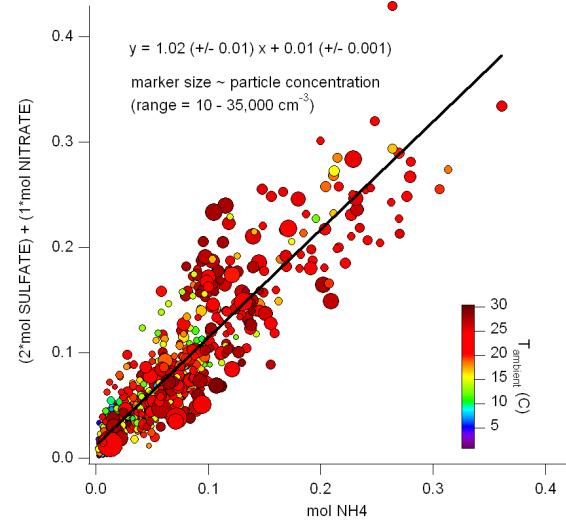
April

Issues with evaporation

Field campaigns
 (Sorooshian et al.)

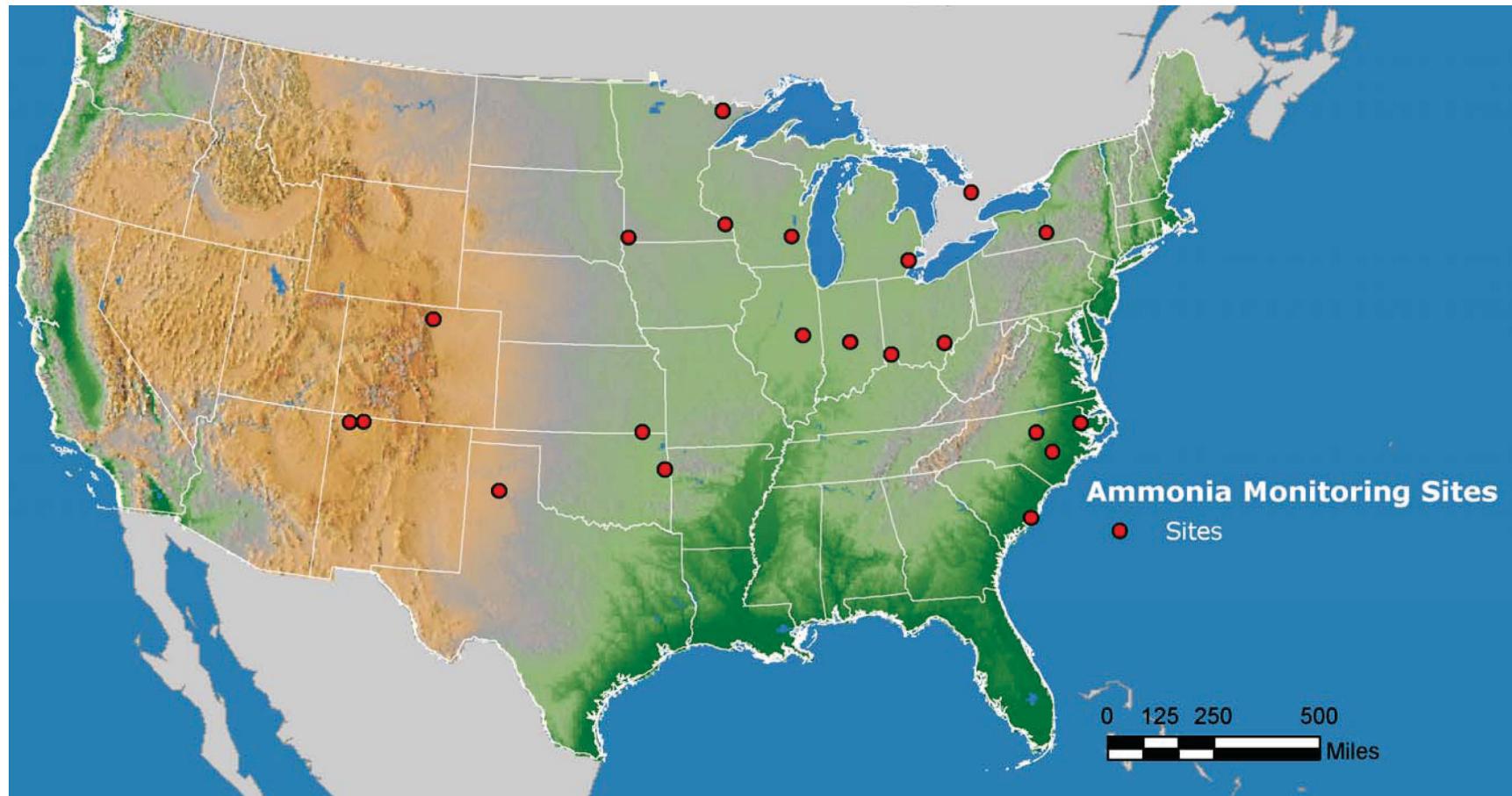


SJ
 Valley



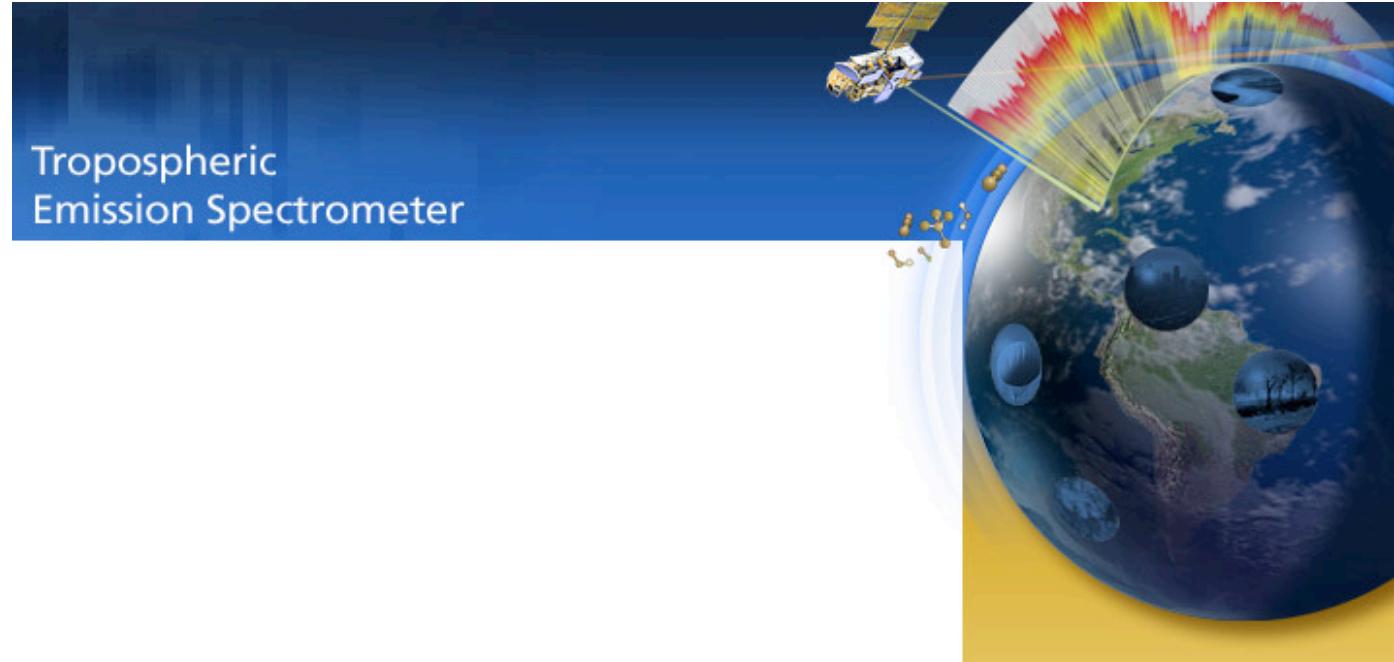
Houston

NH₃ Monitoring Sites



EPA's new AMoN sites (Gary Lear)

New NH₃ observations

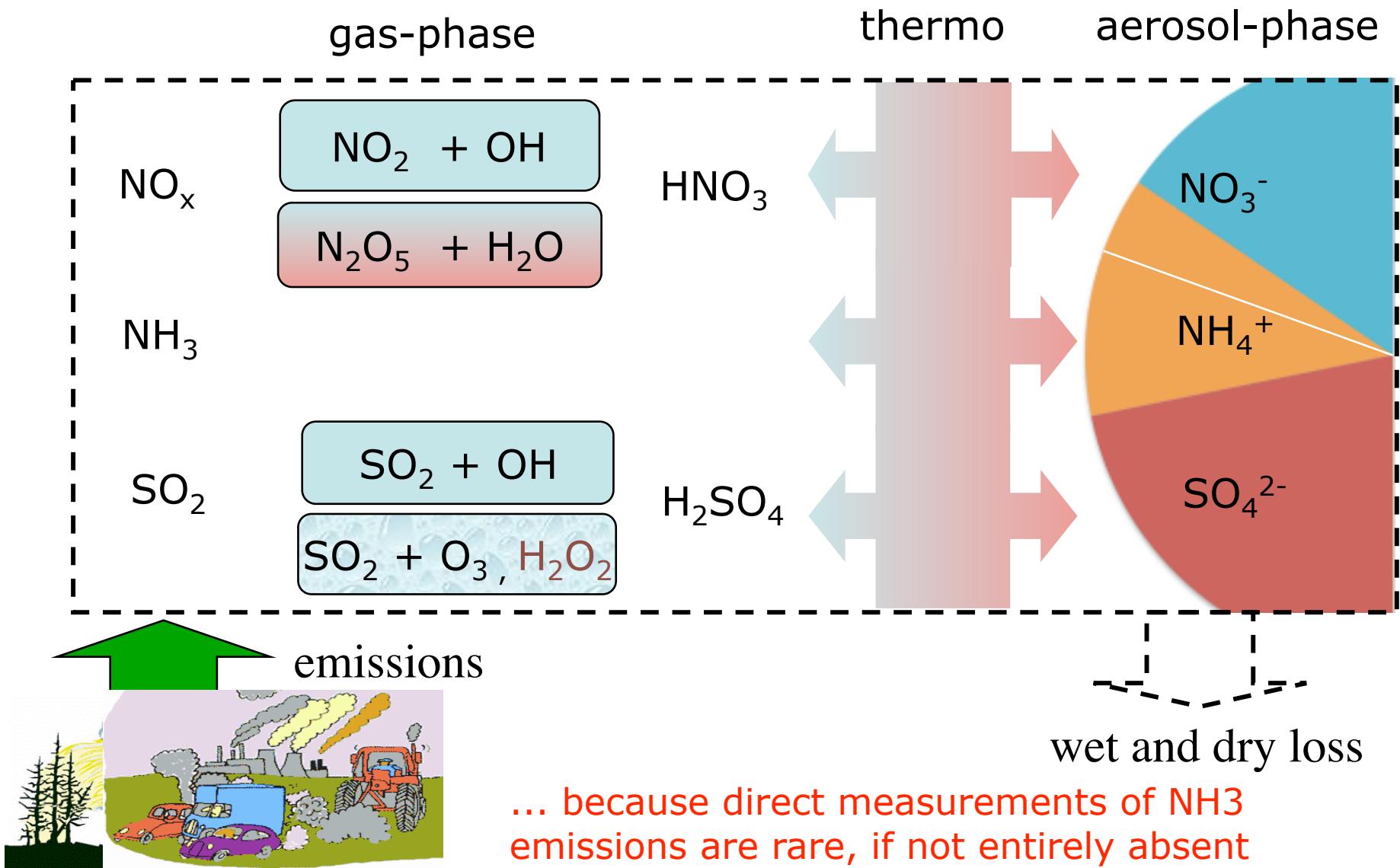


GEOPHYSICAL RESEARCH LETTERS, VOL. 35, L09801, doi:10.1029/2008GL033642, 2008

First satellite observations of lower tropospheric ammonia and methanol

Reinhard Beer,¹ Mark W. Shephard,² Susan S. Kulawik,¹ Shepard A. Clough,³ Annmarie Eldering,¹ Kevin W. Bowman,¹ Stanley P. Sander,¹ Brendan M. Fisher,¹ Vivienne H. Payne,² Mingzhao Luo,¹ Gregory B. Osterman,¹ and John R. Worden¹

Why top-down constraints?



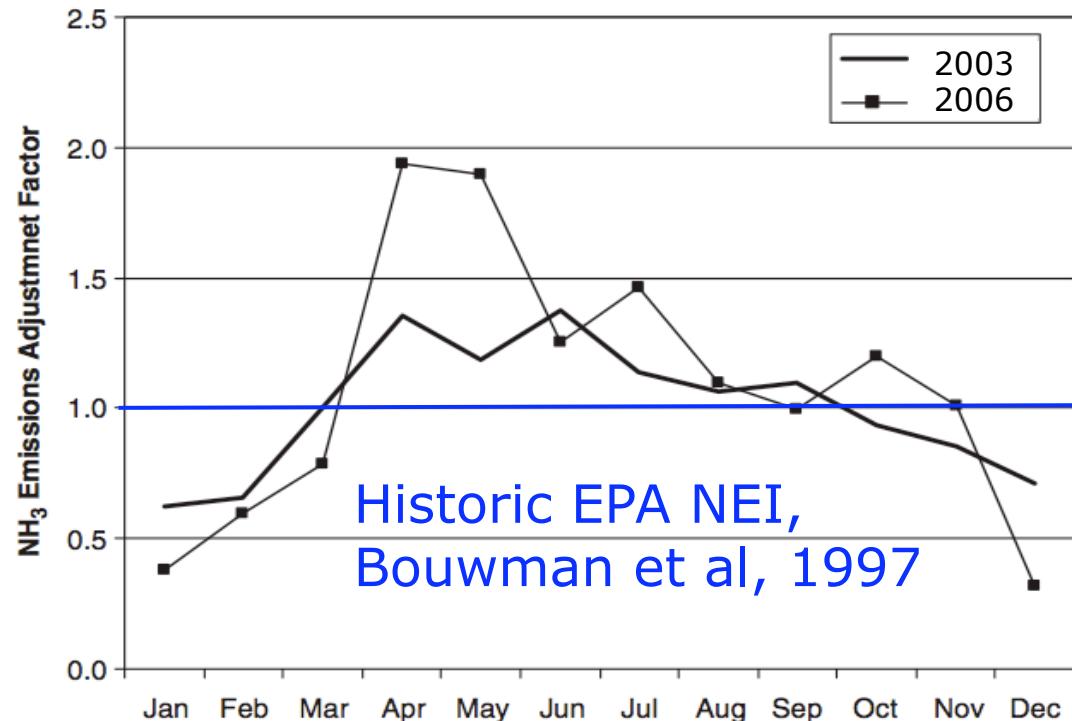
NH_3 inverse modeling: Gilliland et al.

Observations: wet $\text{NH}_x = \text{aerosol } \text{NH}_4^+ + \text{gas } \text{NH}_3$

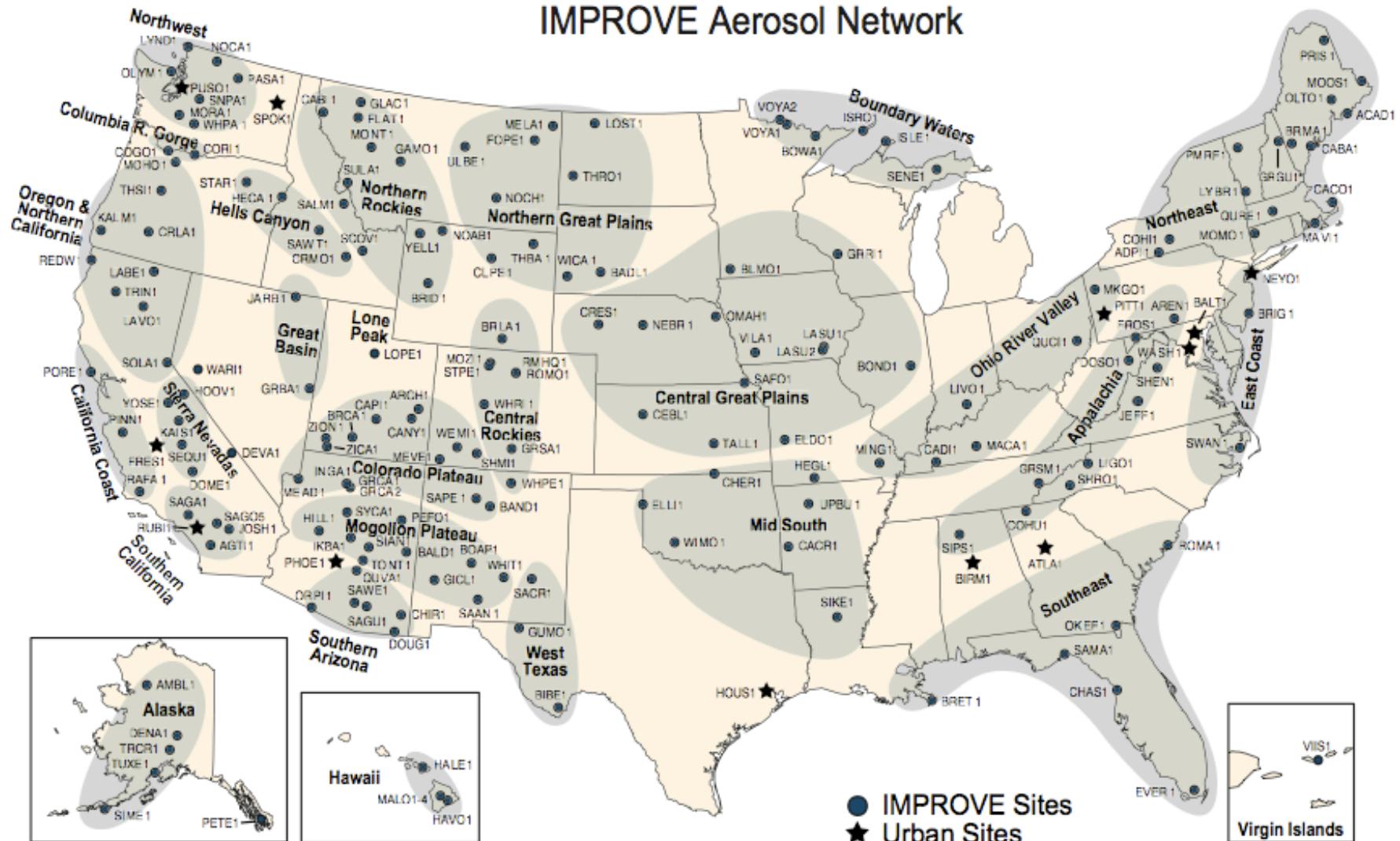
Method: Kalman filter (BF) to adjust monthly nation-wide scale factors

Results:

Gilliland et al., 2003;
Gilliland et al., 2006

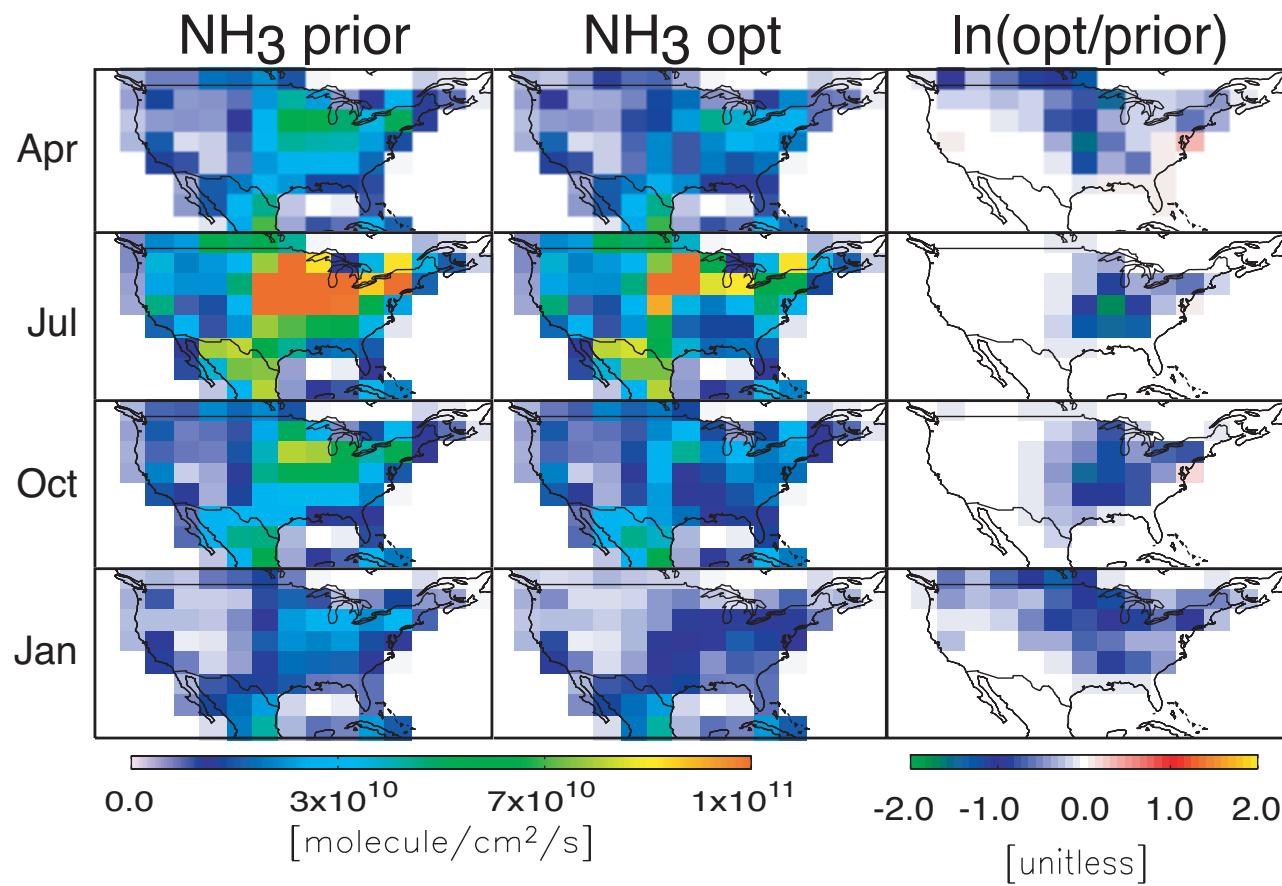


IMPROVE Aerosol Network



Measures sulfate and nitrate.

Inverse modeling: anthro NH₃ emissions



- scaling is spatially variable
- scaling generally reductions
- some increases
- Each month treated separately
- reduction in RMSE ≈ 40%

Inverse modeling: assessing the solution

Dependence on inverse modeling assumptions:

- error covariance matrices
- regularization

Estimated uncertainty of solution

- approximate inverse hessian
- std error and correlations

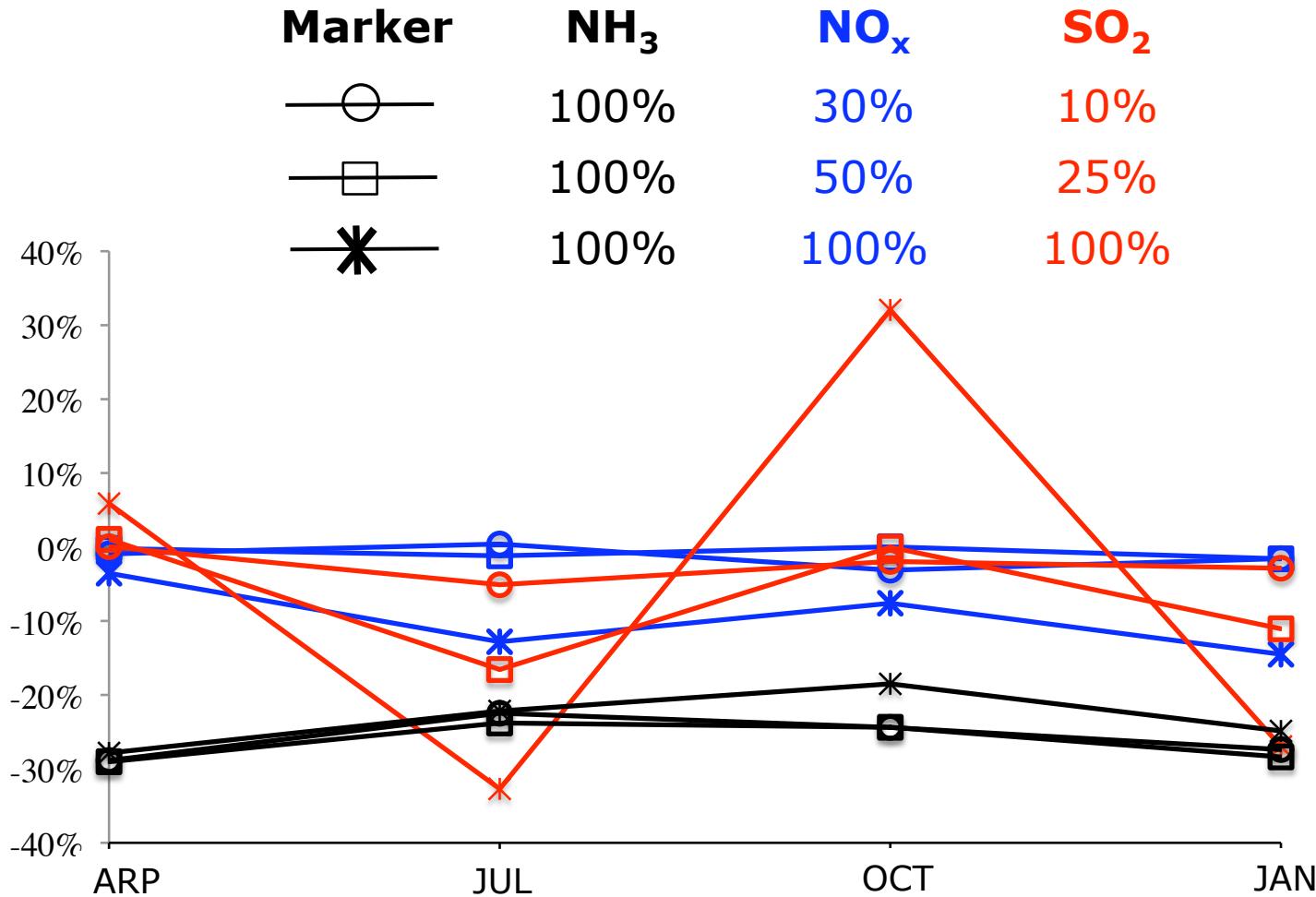
Compare to other studies

- inverse modeling
- bottom up inventories

Compare to NH₃ observations

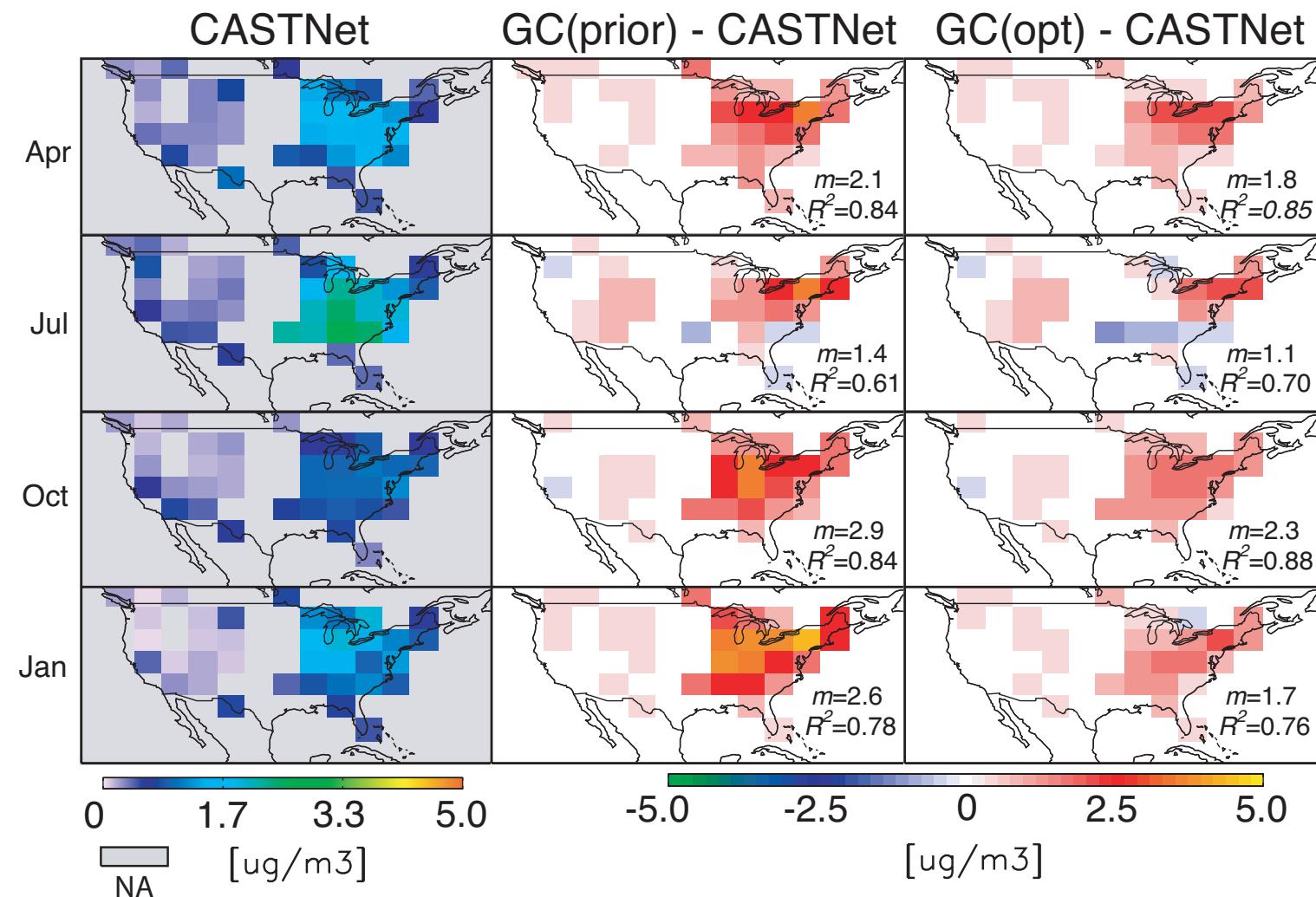
Inverse modeling: variable constraints

Repeat inversions with different emissions error estimates:



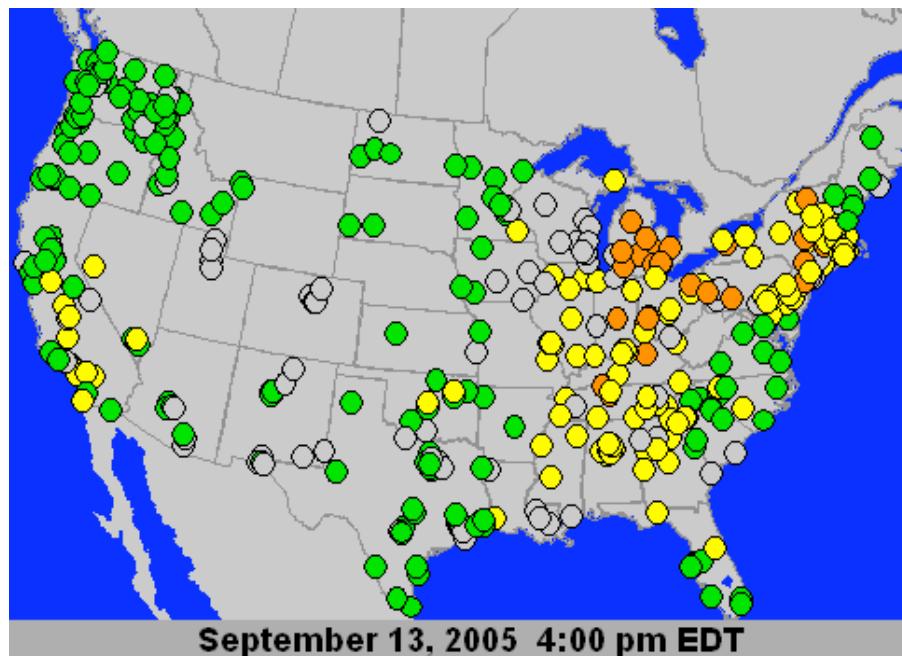
NH_3 , and to lesser extent, NO_x , emissions estimates fairly invariant

Inverse: comparison to CASTNet NH_4^+

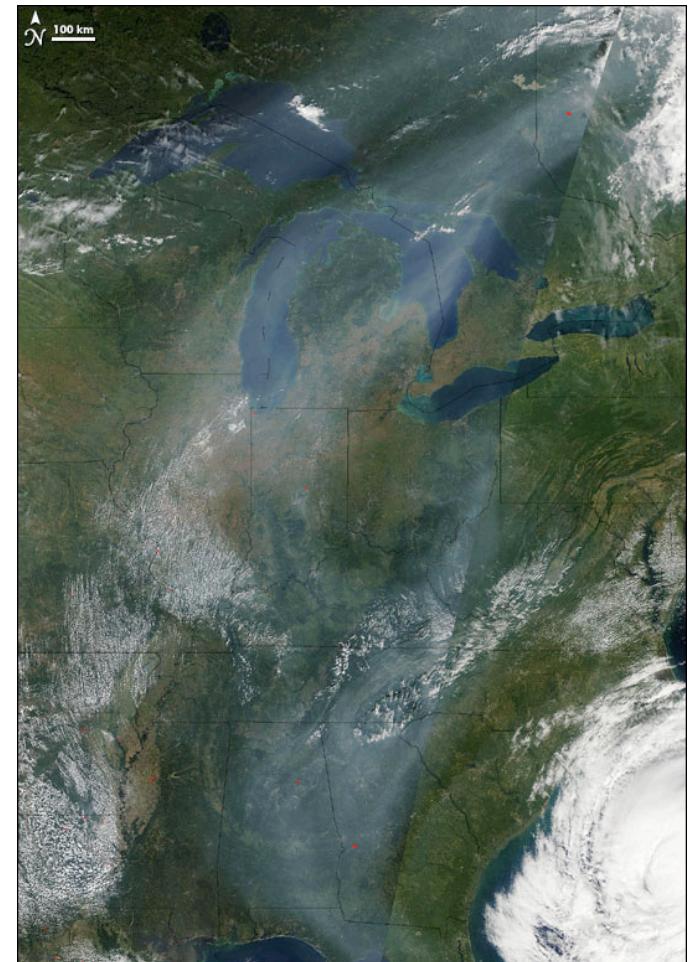


Air Pollution: Continental PM

EPA AIRNow



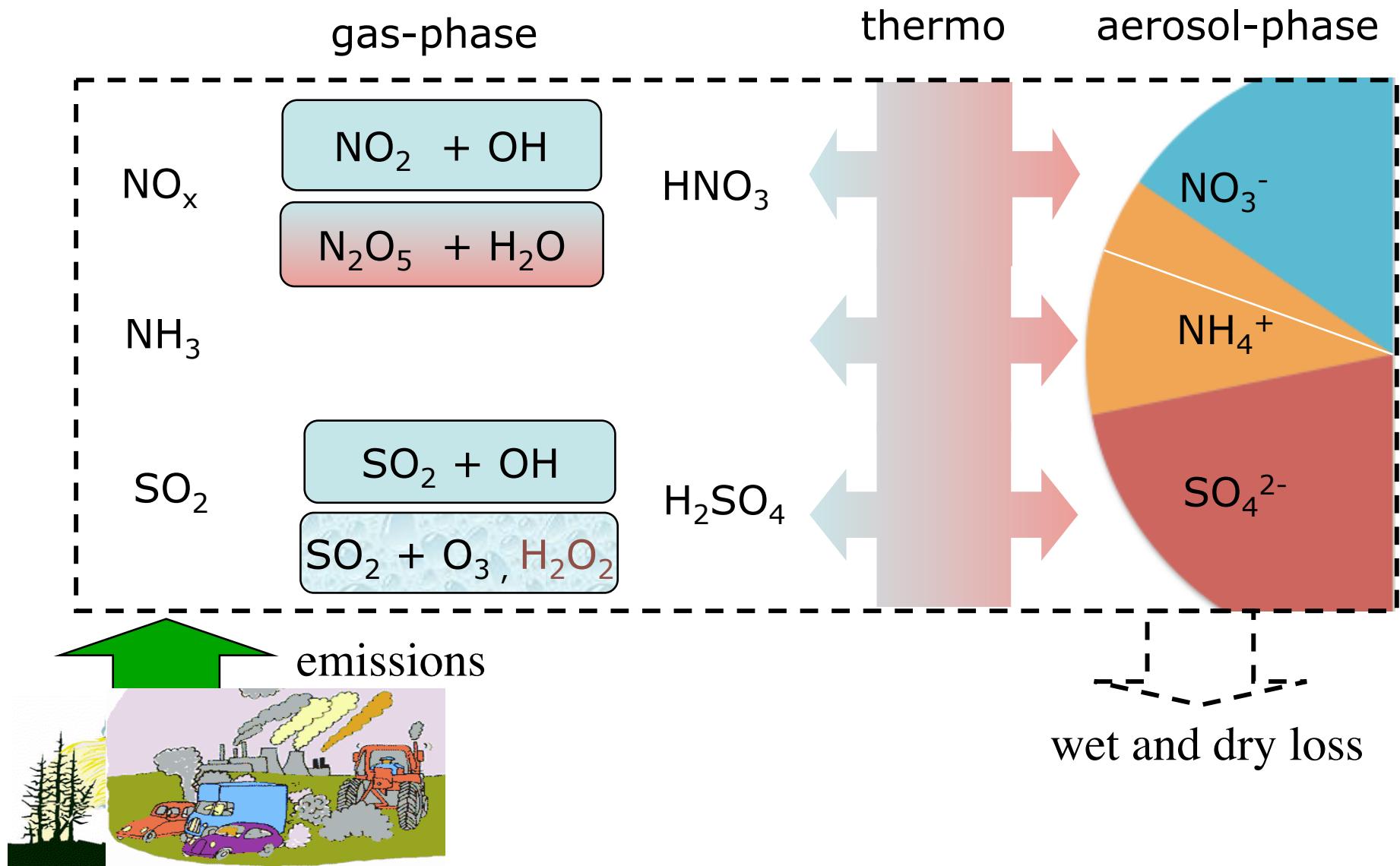
MODIS (TERRA)



(smog blog, NASA's visible earth)

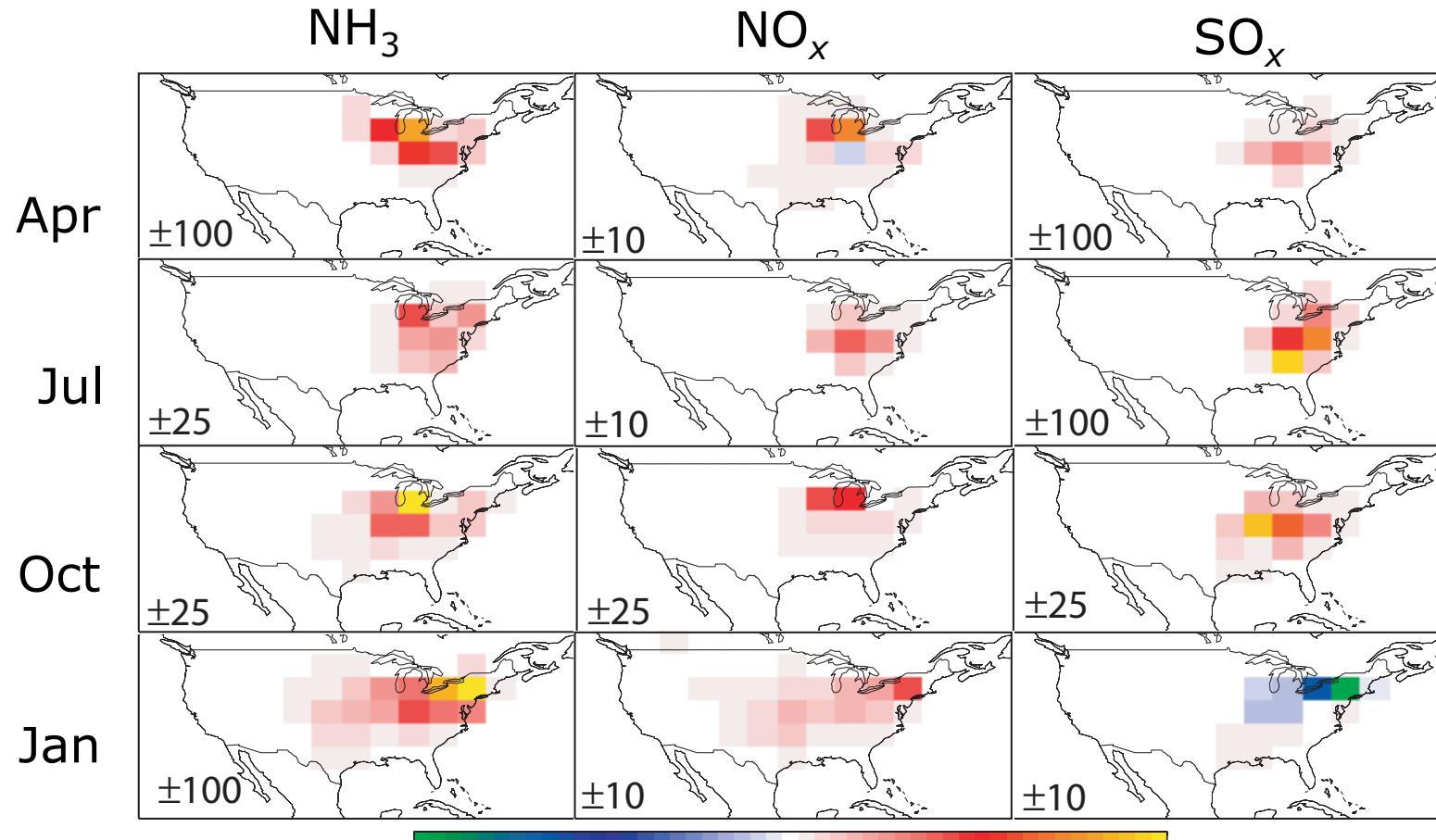
September 11, 2005

Secondary inorganic aerosol formation



AQ Attainment: seasonal variability

Seasonal variation in control effectiveness:



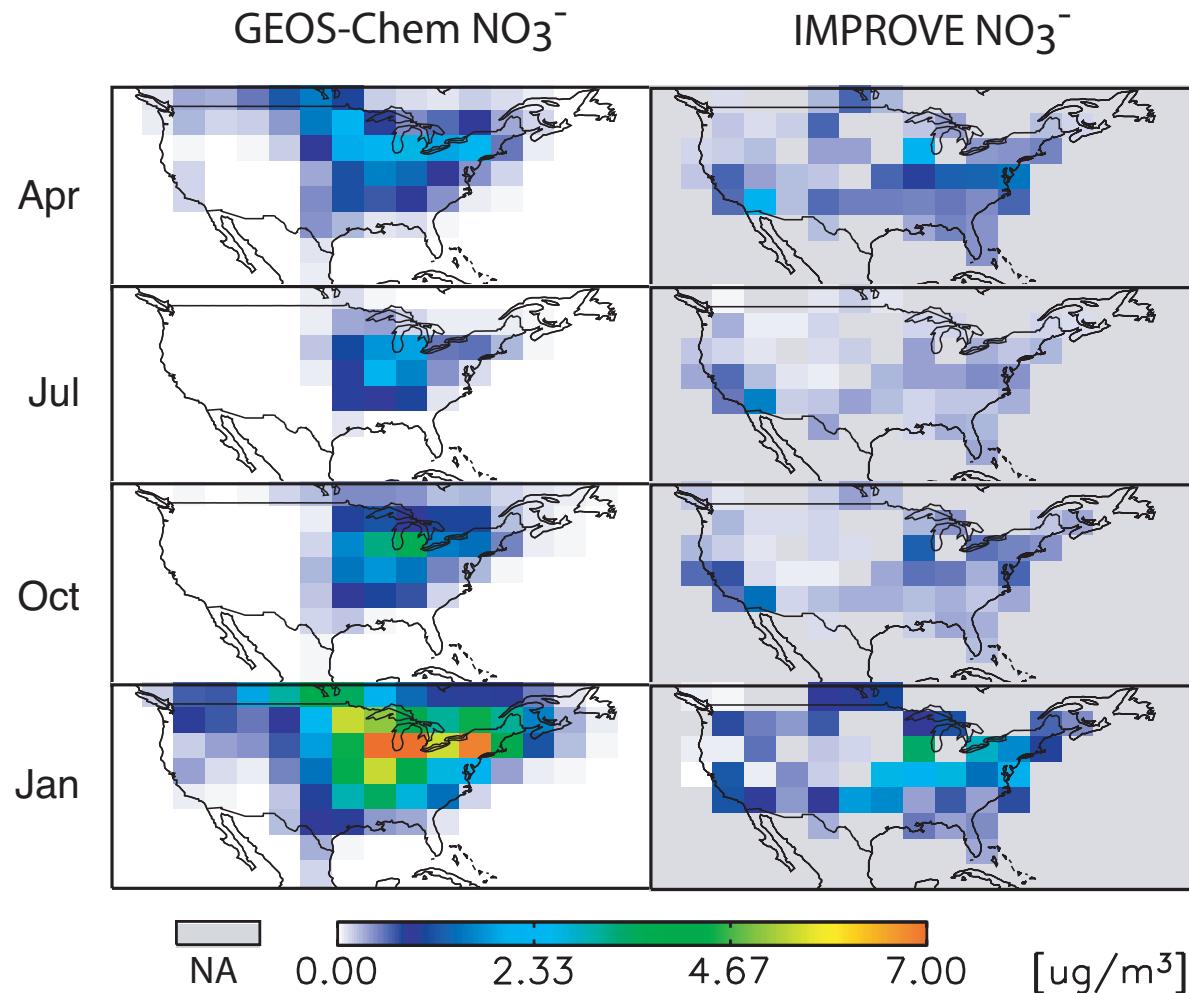
Why study NH₃ emissions?

Health impacts of PM2.5, air quality control

Environmental impacts

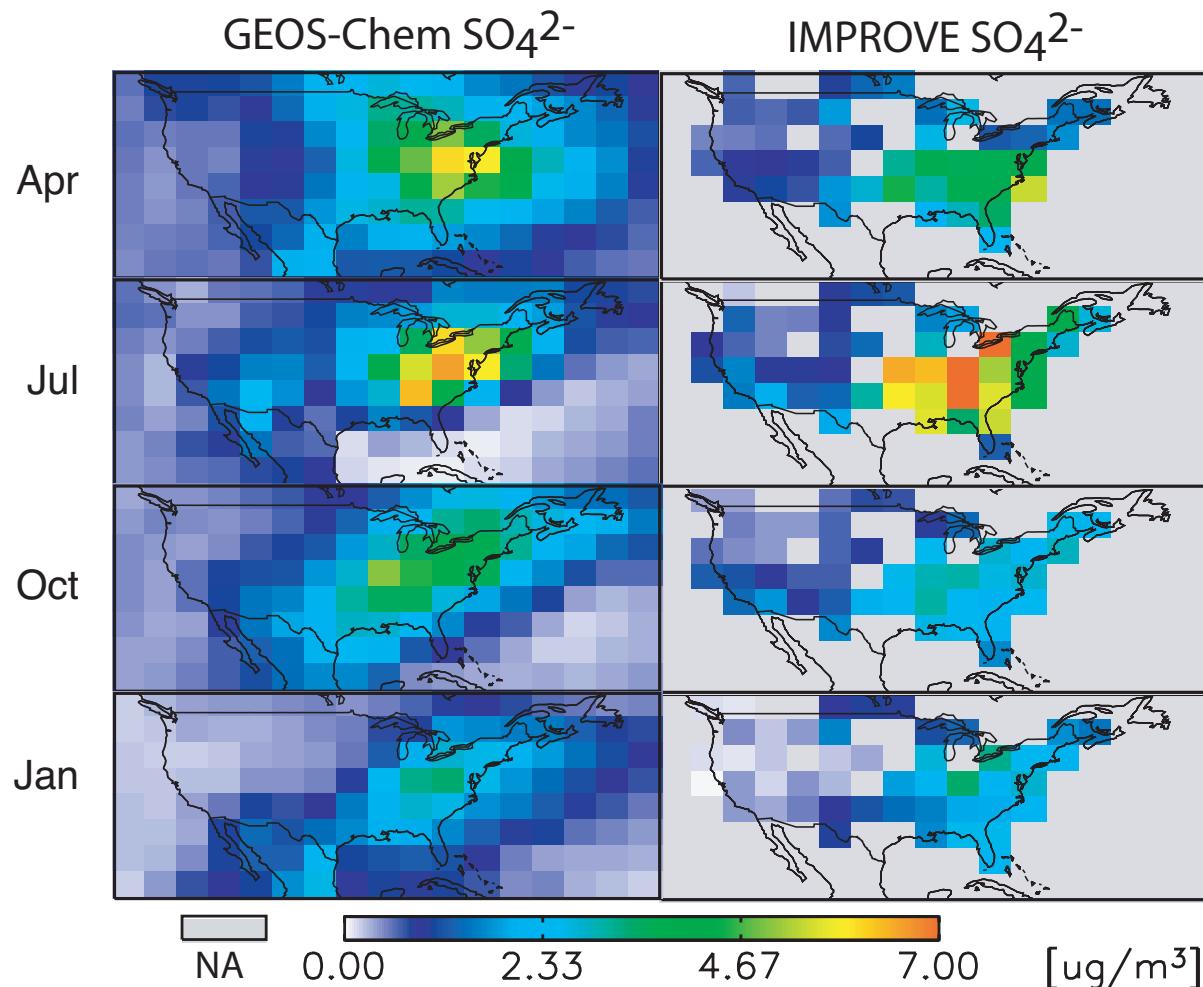
A large source of uncertainty

GEOS-Chem vs IMPROVE: nitrate

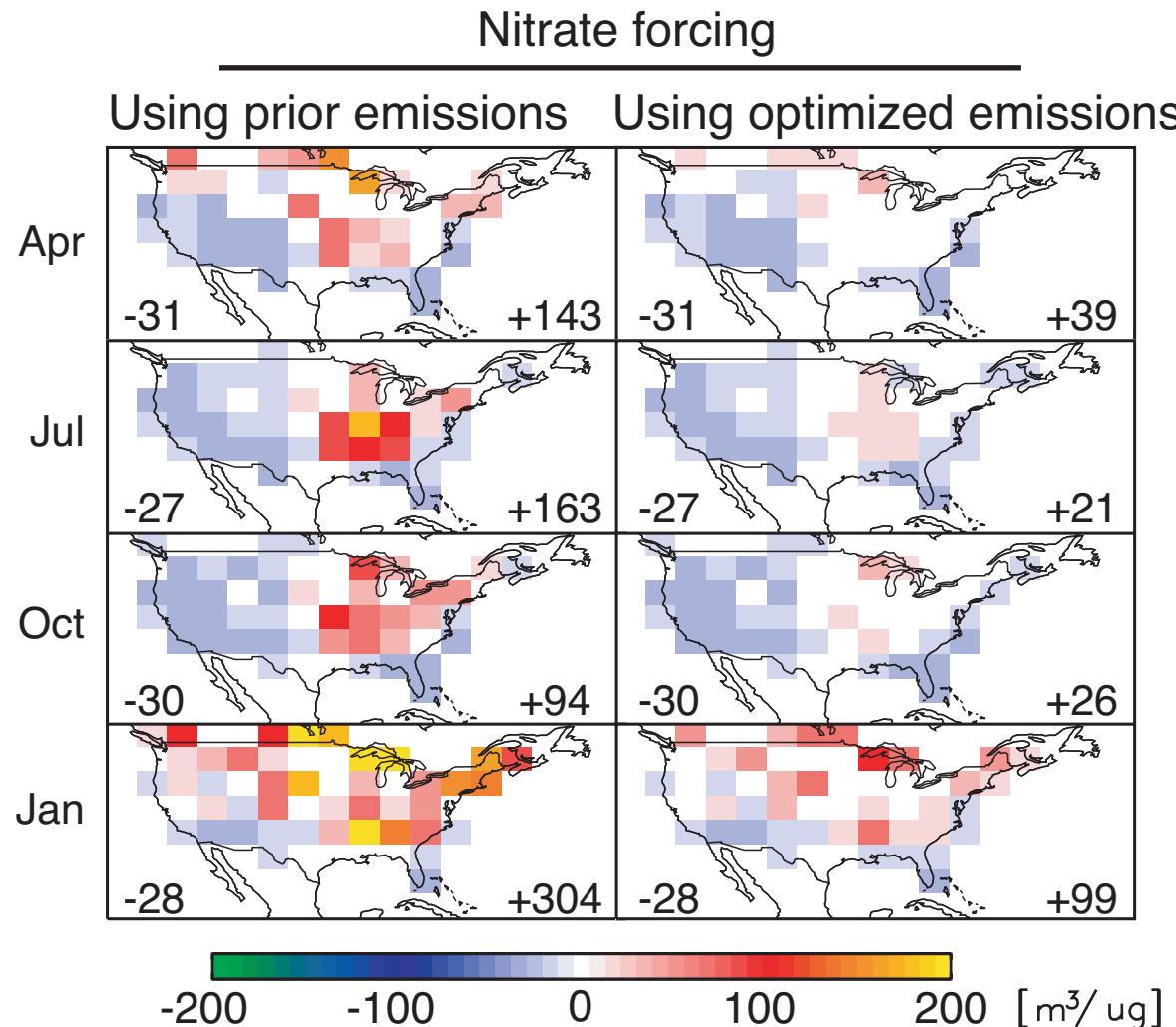


nitrate problematic: Park et al., (2004,2006), Liao et al. (2007), Pye et al. (2008)

GEOS-Chem vs IMPROVE: sulfate



Inverse modeling: reduction in error



Importance of studying NH₃ emissions

PM2.5

- Itself leads to NH₄⁺, 10-20% of PM_{2.5} mass concentration
- Governs formation of NO₃⁻, which can be 20-30% in winter

PM2.5 NAAQS Regulations

- Not a presumptively regulated species, but can be very efficient (Pinder et al., 2007; Henze et al., 2008)
- Can be regulated in place of SO₂ or NO₂

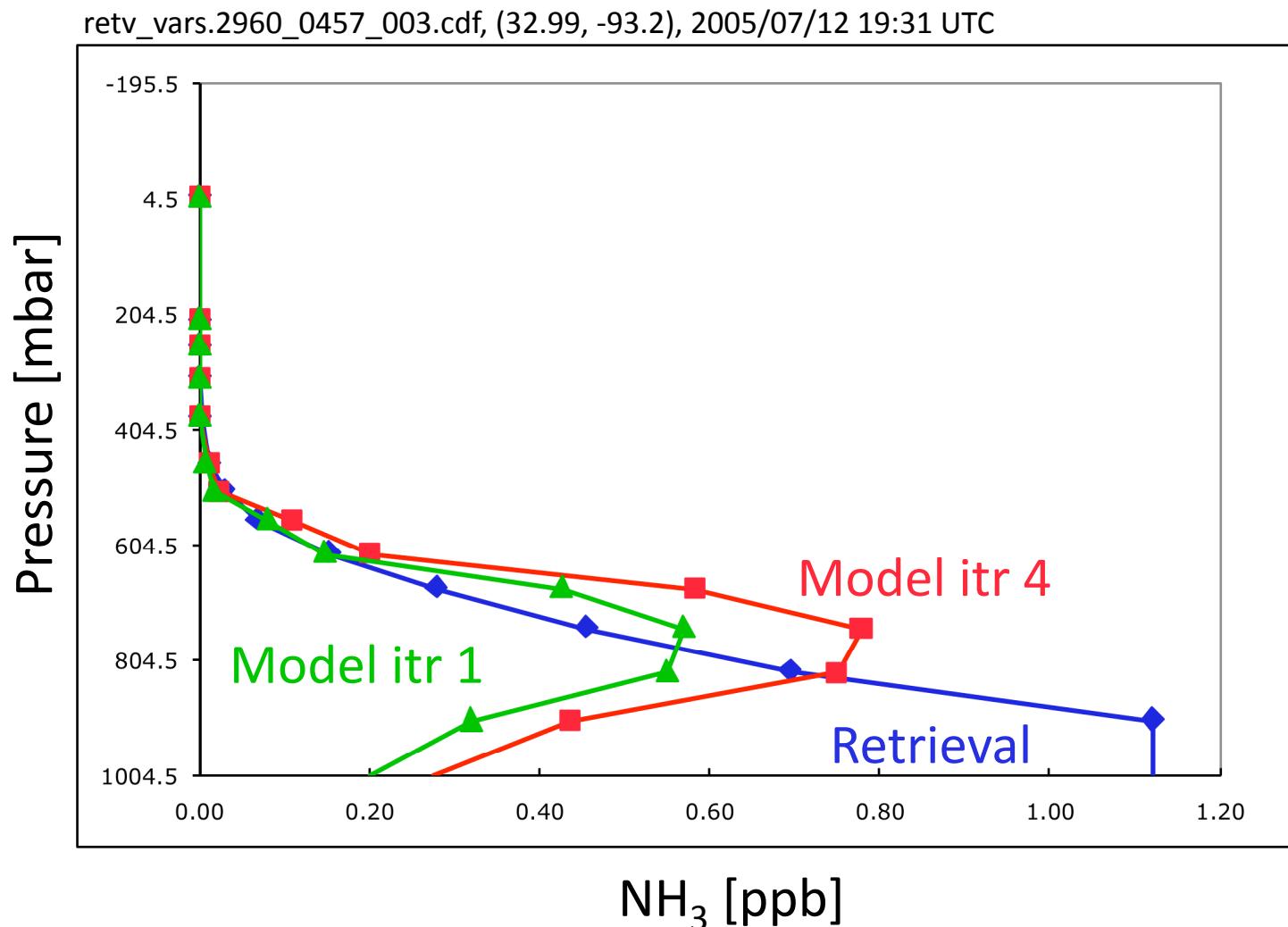
Ecosystem impacts

- 11% of worlds natural vegetation impacted by N dep (Dentener et al., 2006)
- N dep will increase 10-40% near NH₃ sources in U.S. by 2020 (Pinder et al., 2008)

Very large source of uncertainty

- estimating U.S. inorganic PM_{2.5} levels (Yu et al., 2005; Simon et al., 2008)
- global N dep. (Sutton et al., 2007)

Inverse modeling: impact on profile estimates compared to pseudo retrieval. After only 4 iterations (would usually do about 20)



Inverse modeling: cost function

Define a cost function (want to minimize):

$$\mathcal{J} = \frac{1}{2} \sum_{\mathbf{c} \in \Omega} (\mathbf{H}\mathbf{c} - \mathbf{c}_{obs})^T \mathbf{S}_{obs}^{-1} (\mathbf{H}\mathbf{c} - \mathbf{c}_{obs}) + \frac{1}{2} \gamma_r (\boldsymbol{\sigma} - \boldsymbol{\sigma}_a)^T \mathbf{S}_a^{-1} (\boldsymbol{\sigma} - \boldsymbol{\sigma}_a)$$



"error"



"penalty term"

where

\mathbf{c} = Model predictions

Ω = Domain of observations

\mathbf{S}^{-1} = Error covariance matrices

γ_r = Regularization parameter

$\boldsymbol{\sigma}$ = free parameters, $\boldsymbol{\sigma} = \ln \left(\frac{\text{emission}}{\text{emission}_a} \right)$

$\boldsymbol{\sigma}_a$ = initial guess of parameters ($= 0$)

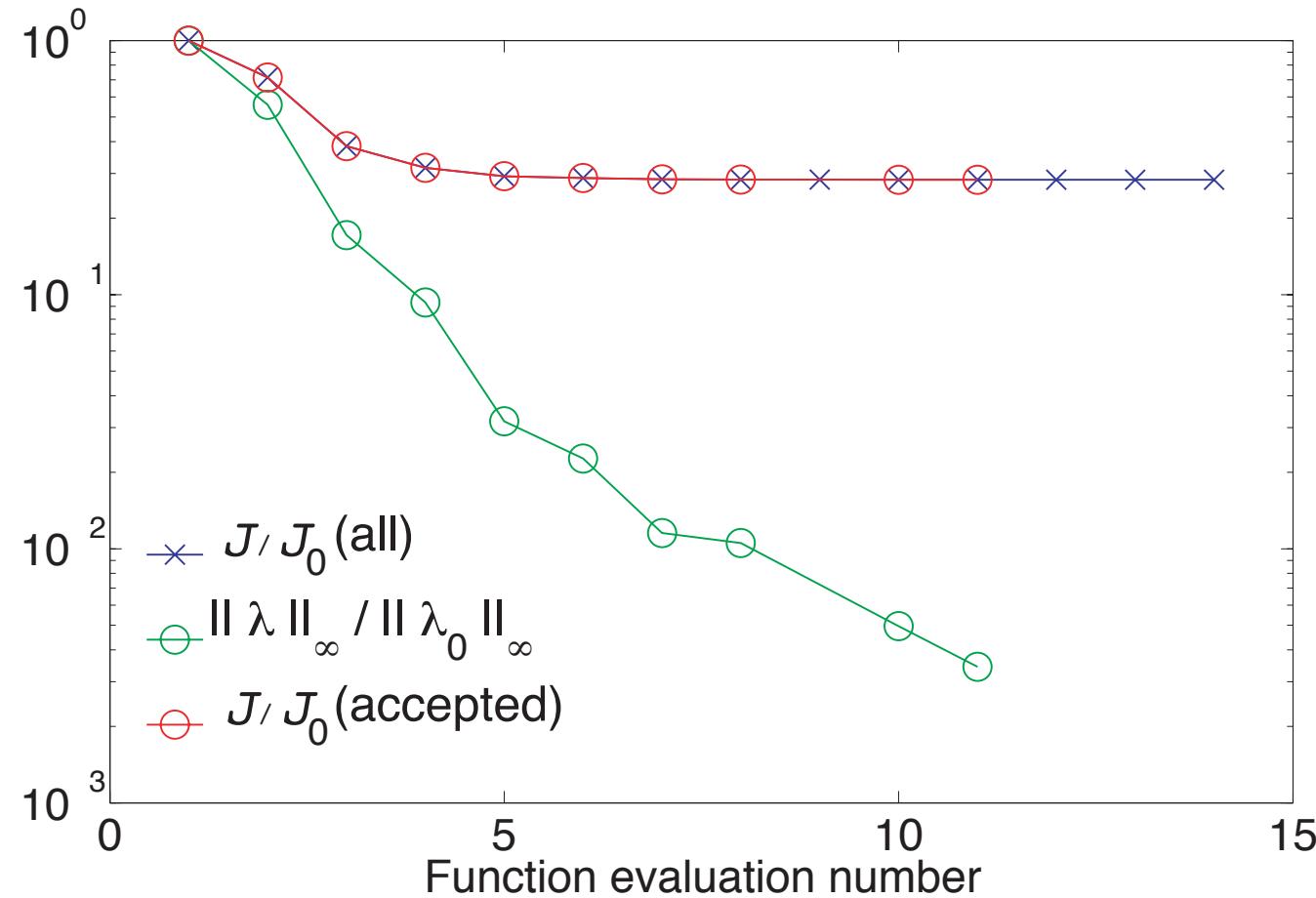
$\mathbf{S}_{obs} = 30\%$

$\mathbf{S}_a = 100\% \text{ for } \text{NH}_3$

$= 30\% \text{ for } \text{NO}_x$

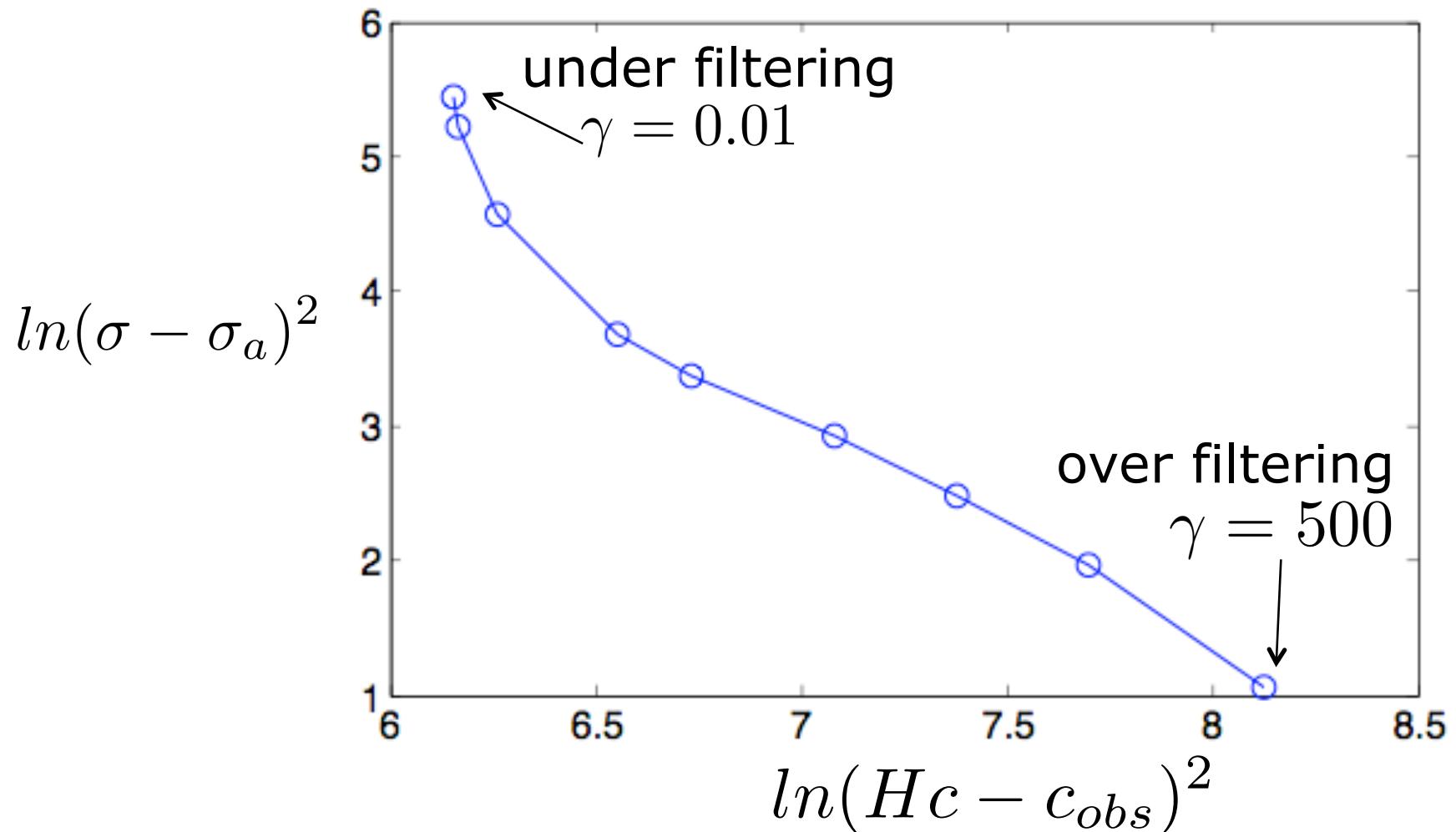
$= 10\% \text{ for } \text{SO}_x$

Inverse modeling: minimization



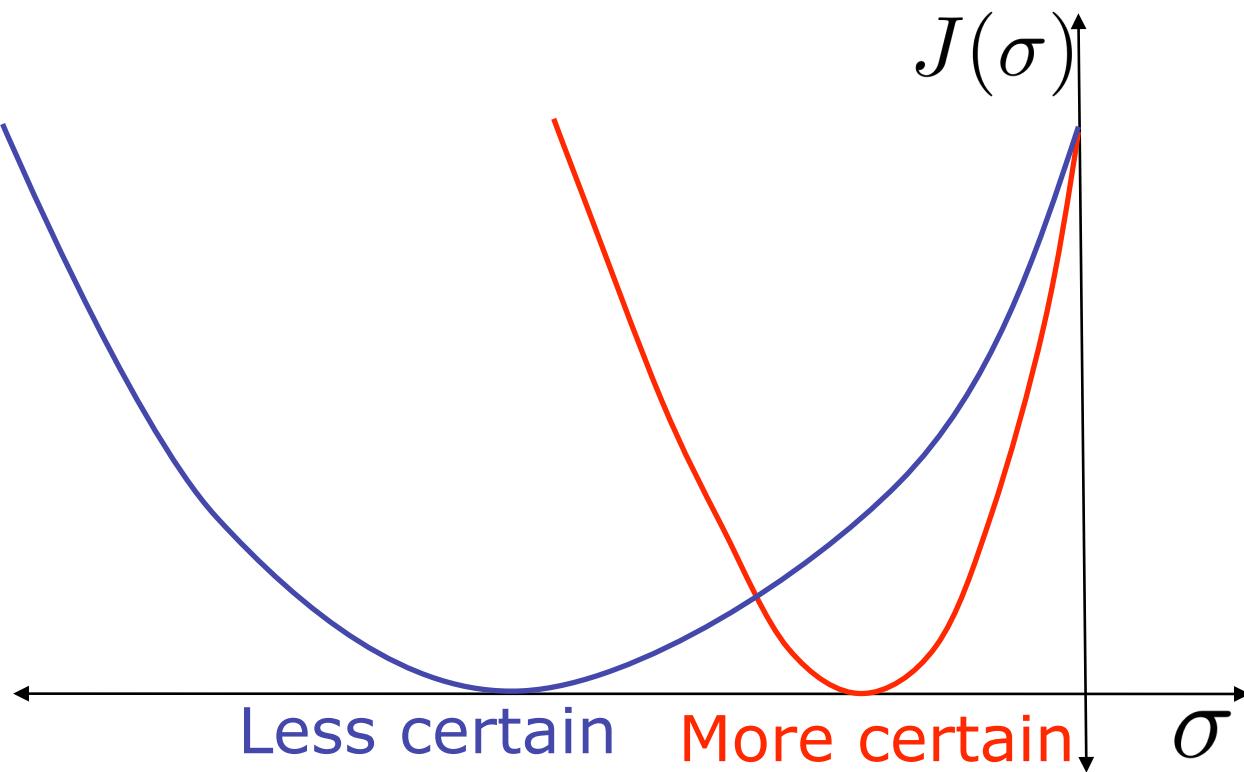
Regularization Parameter

$$J = \frac{1}{2} \sum_{\mathbf{c} \in \Omega} (H\mathbf{c} - \mathbf{c}_{obs})^T \mathbf{S}_{obs}^{-1} (H\mathbf{c} - \mathbf{c}_{obs}) + \frac{1}{2} \gamma_r (\boldsymbol{\sigma} - \boldsymbol{\sigma}_a)^T \mathbf{S}_{\boldsymbol{\sigma}_a}^{-1} (\boldsymbol{\sigma} - \boldsymbol{\sigma}_a)$$



Posterior Error Covariance Estimates

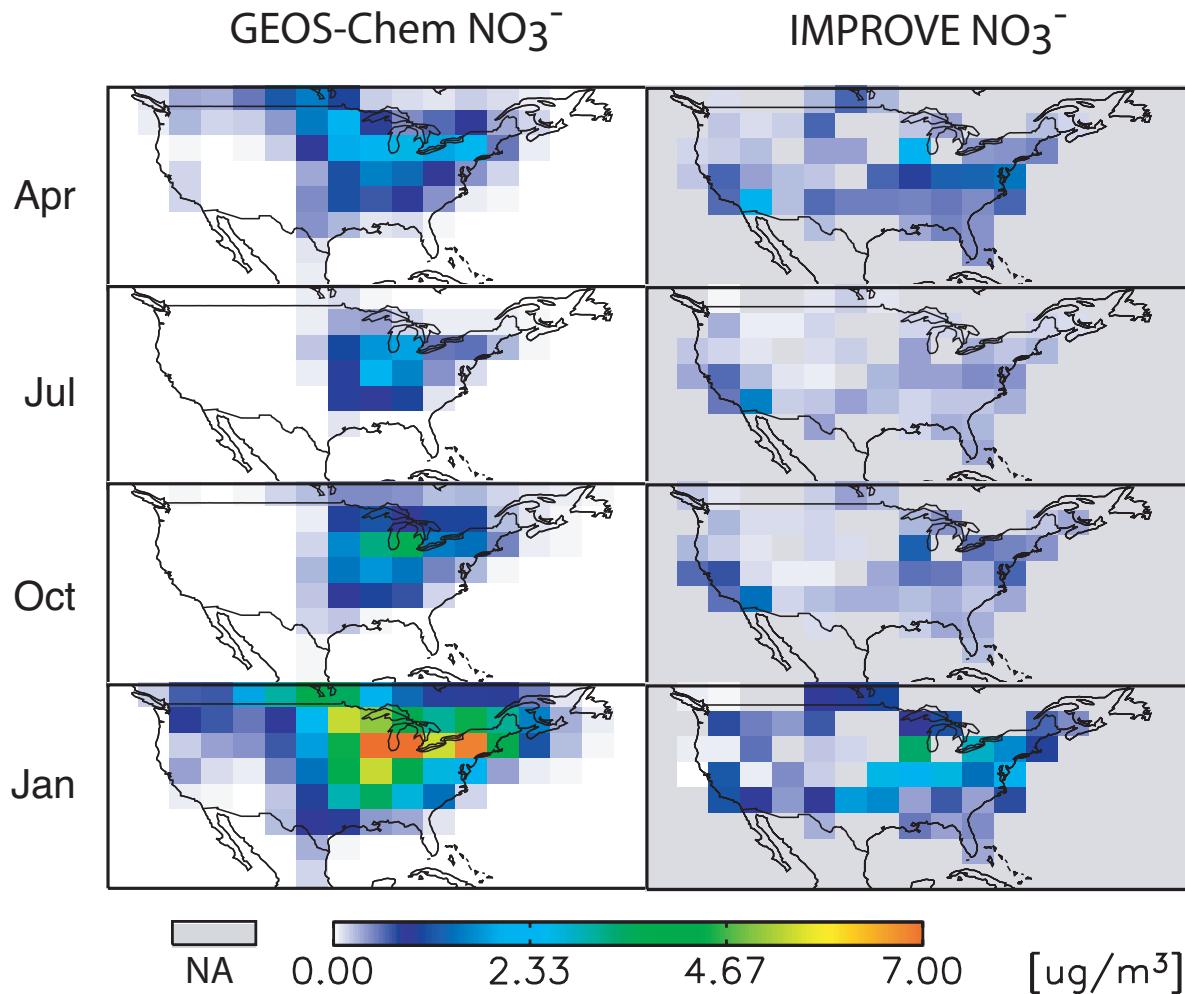
Certainty \sim curvature at minimum



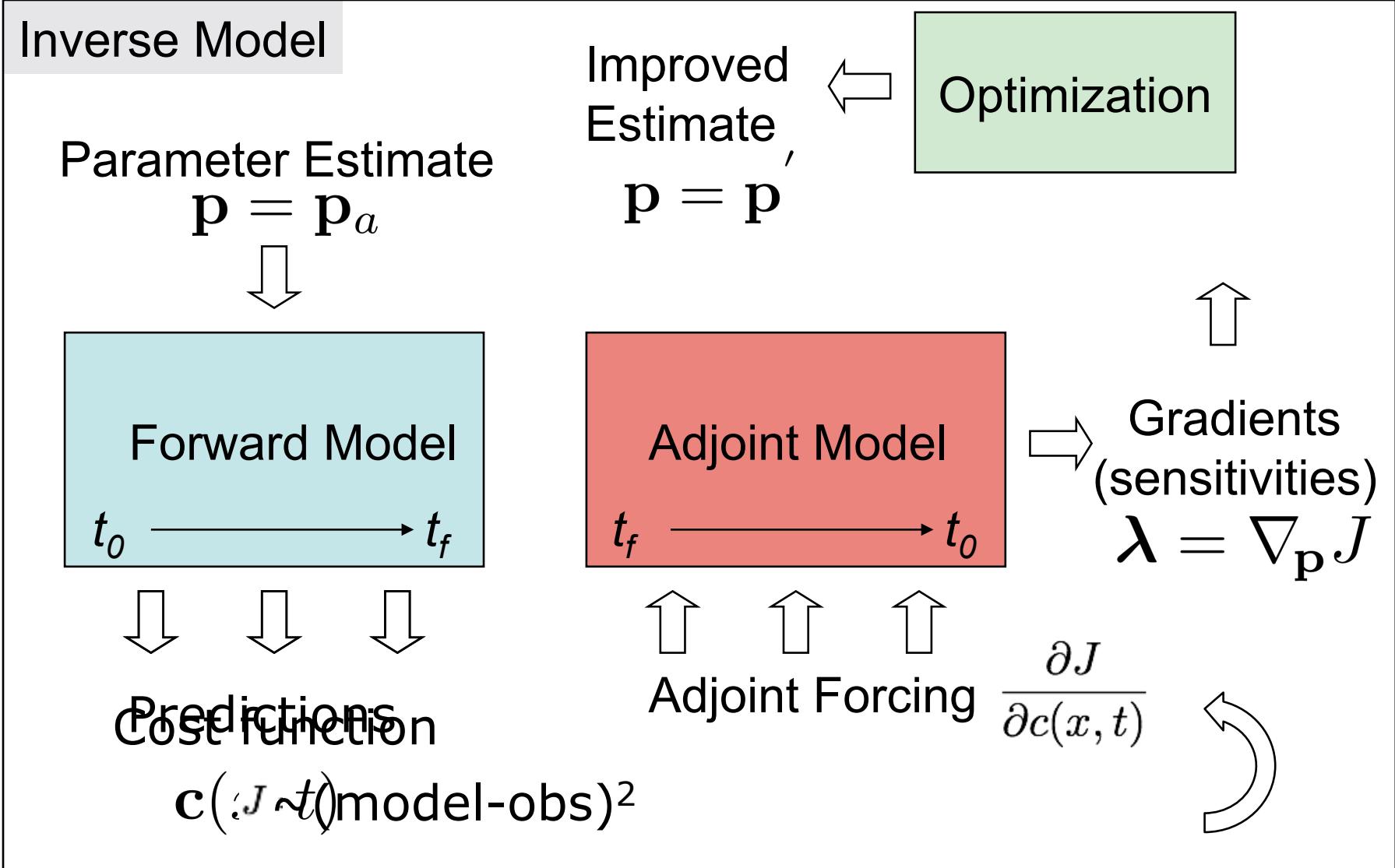
Can estimate inverse Hessian(J) by tracking the minimization

NH₃ emissions uncertainty

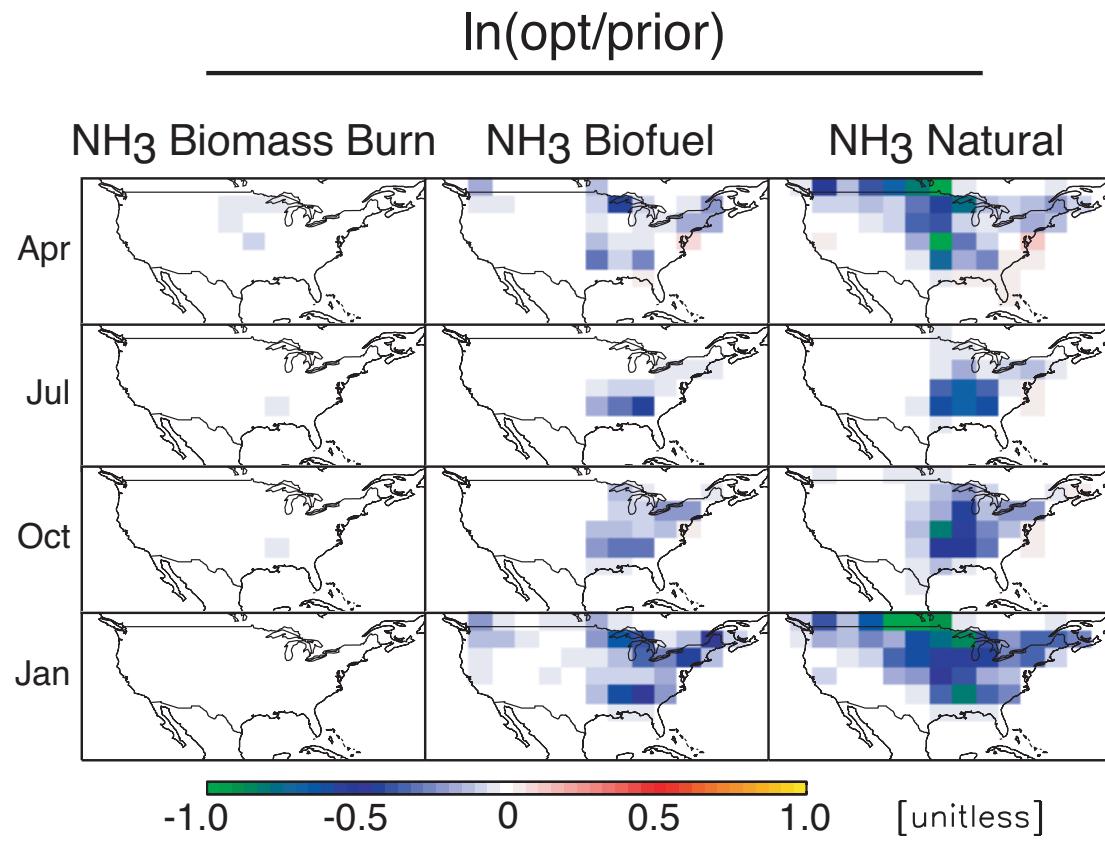
GEOS-Chem vs IMPROVE: nitrate



Inverse Modeling using Adjoint Model



Inverse modeling: other NH₃ emissions



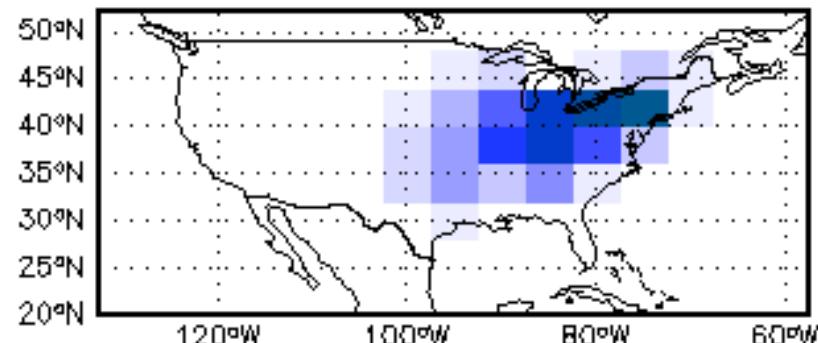
- scaling results from product of adjoints with prior emissions estimates
- reductions affect anthropogenic sources more than natural sources
- results across sectors are correlated

Can effectively distinguish between source sectors

Sensitivities of various cost functions

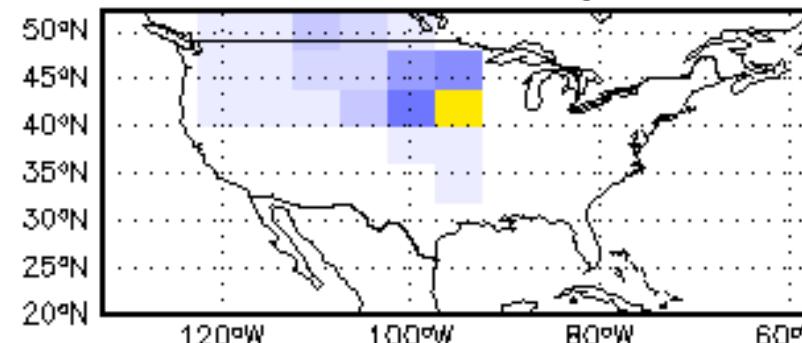
Sensitivities with respect to anthro NH₃ emissions:

IMPROVE: $J = J(\text{SO}_4^{2-}, \text{NO}_3^-)$



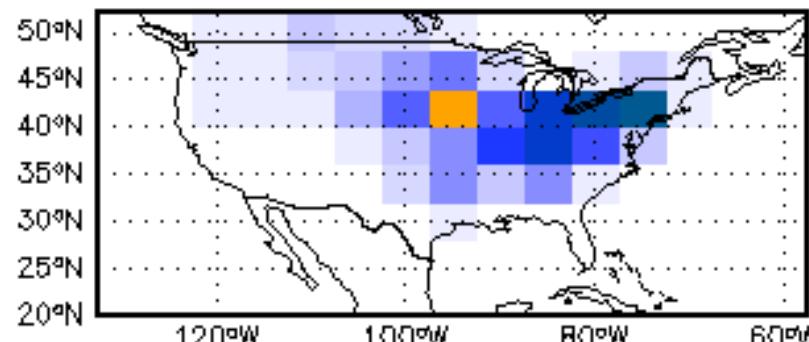
-50 -16 16 50 [dJ/ds]

TES: $J = J(\text{NH}_3)$



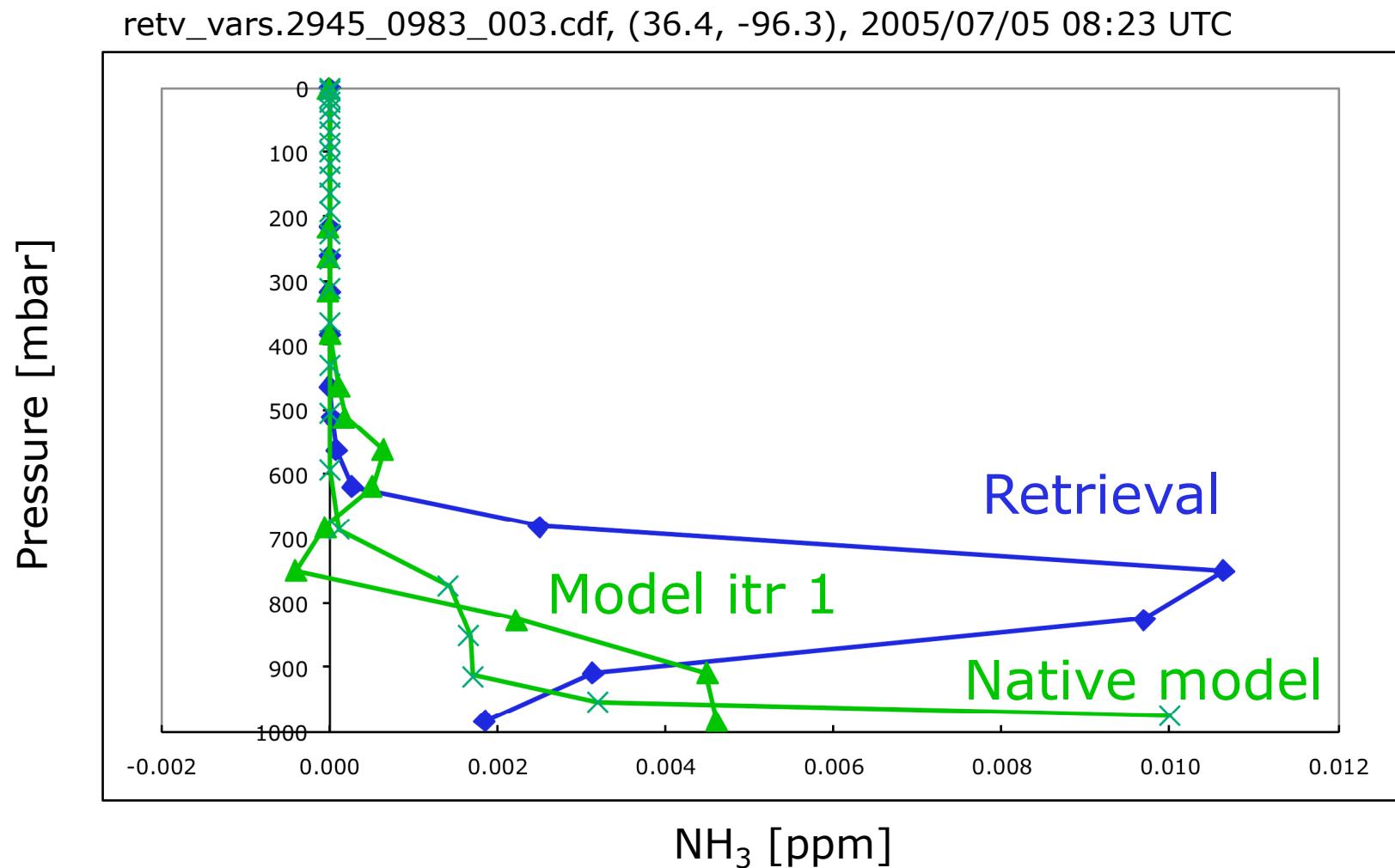
-50 -16 16 50 [dJ/ds]

BOTH: $J = J(\text{SO}_4^{2-}, \text{NO}_3^-, \text{NH}_3)$



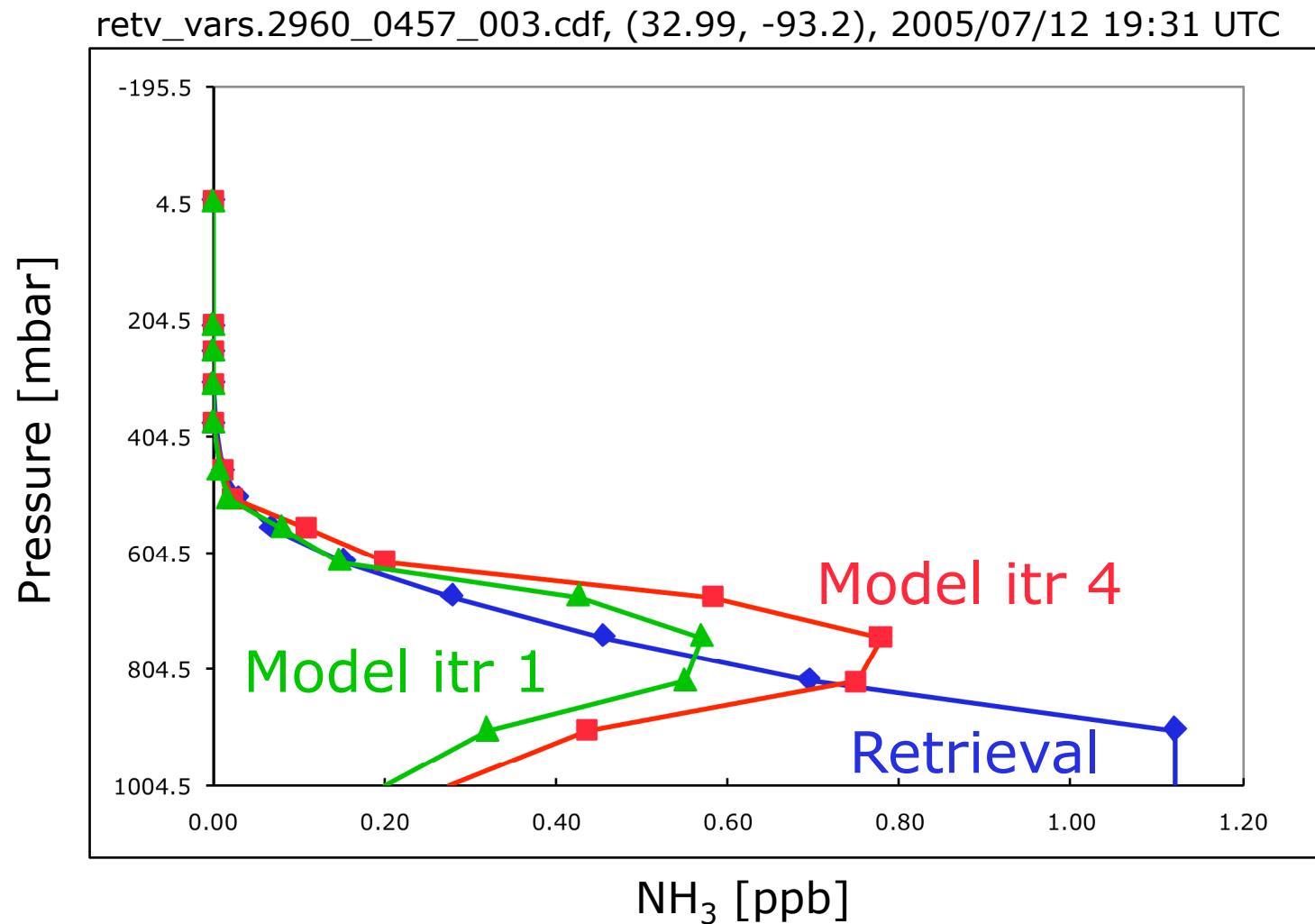
-50 -16 16 50 [dJ/ds]

Why is the sensitivity positive in one spot?

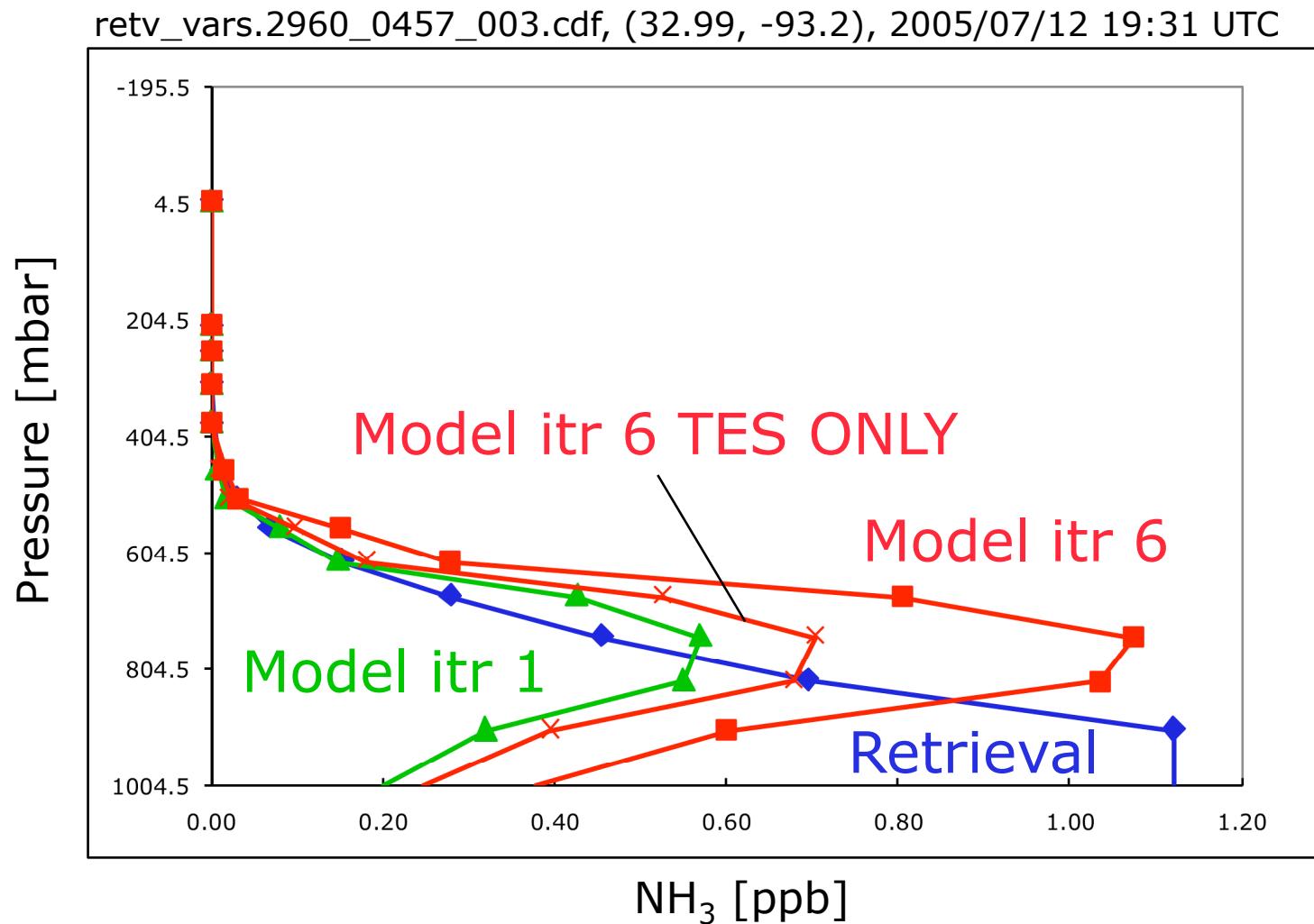


Resolution issue? Retrieval limitation?

Other locations perform better



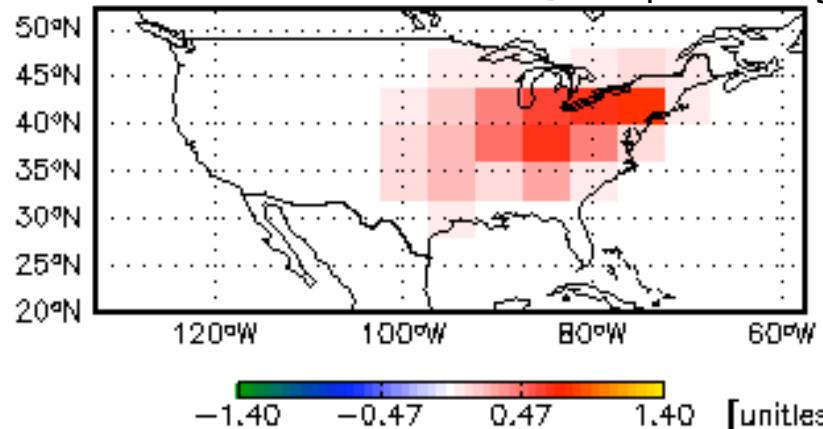
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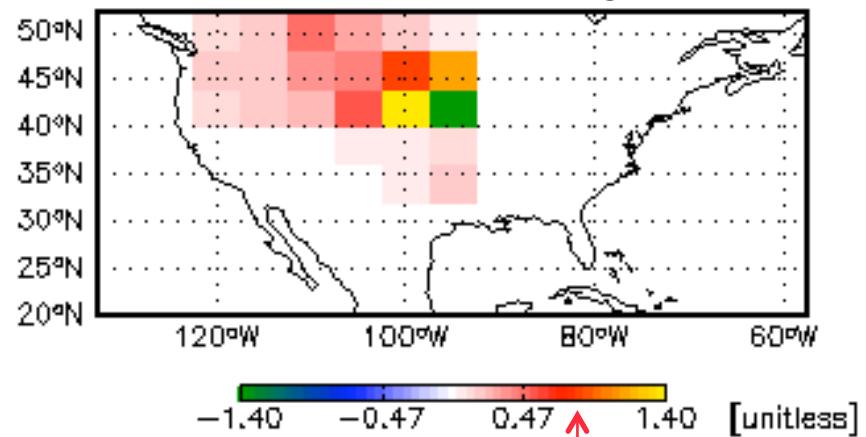
“optimized” scaling factors

Anthro NH₃ emission scaling after 6 iterations

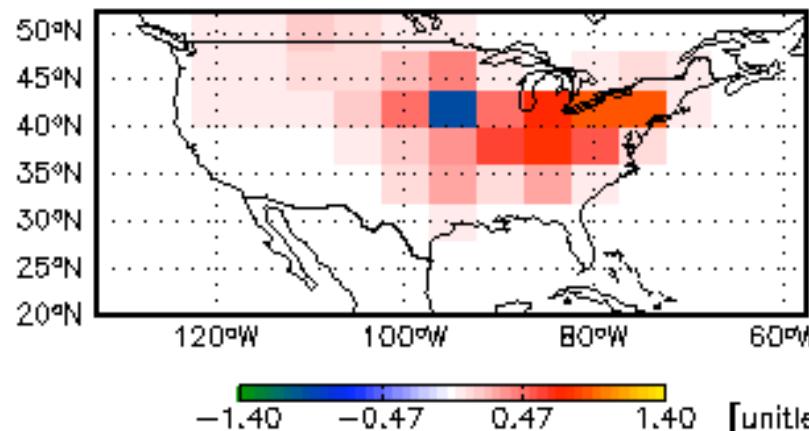
IMPROVE: $J = J(\text{SO}_4^{2-}, \text{NO}_3^-)$



TES: $J = J(\text{NH}_3)$



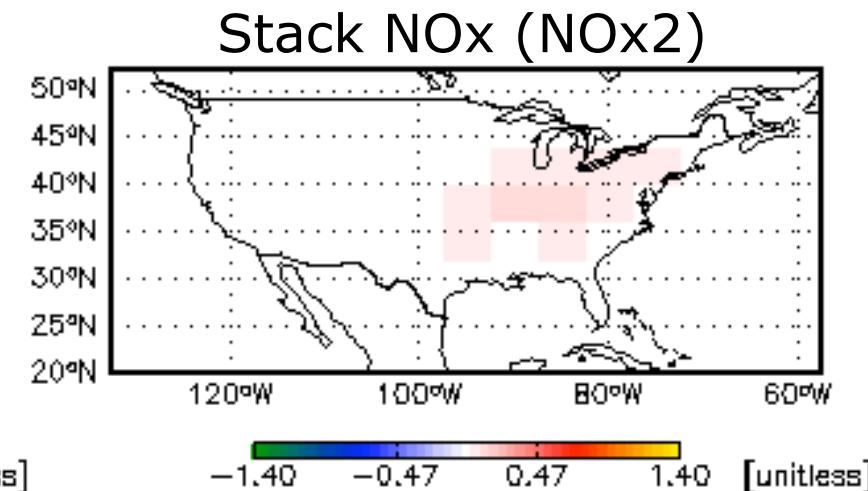
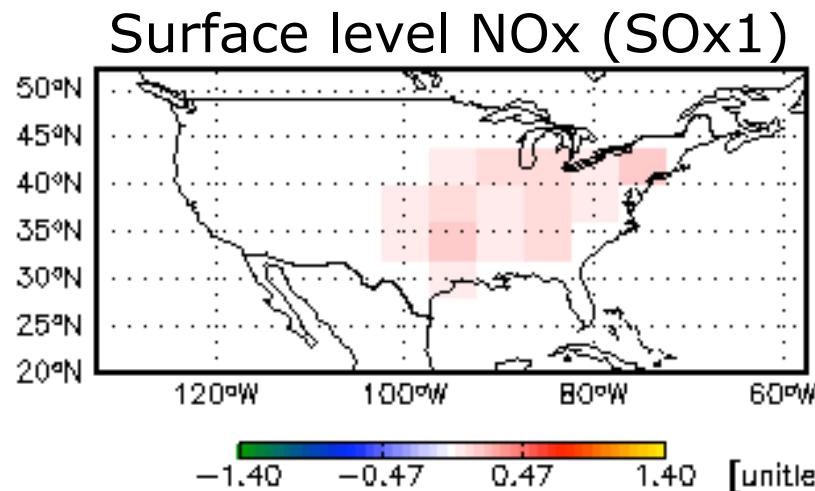
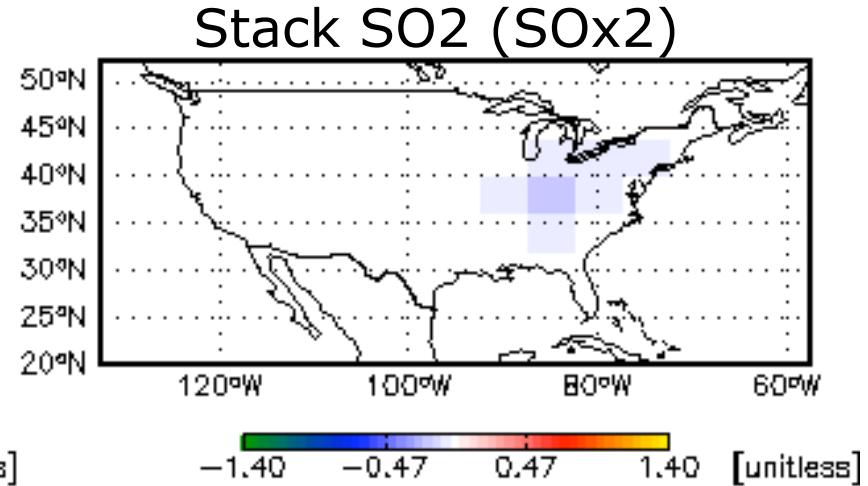
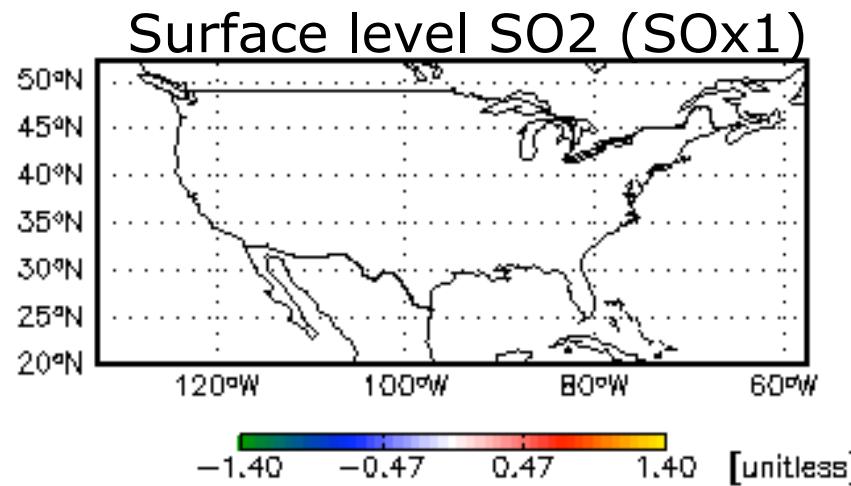
BOTH: $J = J(\text{SO}_4^{2-}, \text{NO}_3^-, \text{NH}_3)$



target solution

Scaling for other inventories

Anthro emission scaling after 6 iterations, $J=J(\mathbf{BOTH})$



NH_3 inverse modeling: Mendoza-Dominguez et al. (2001)

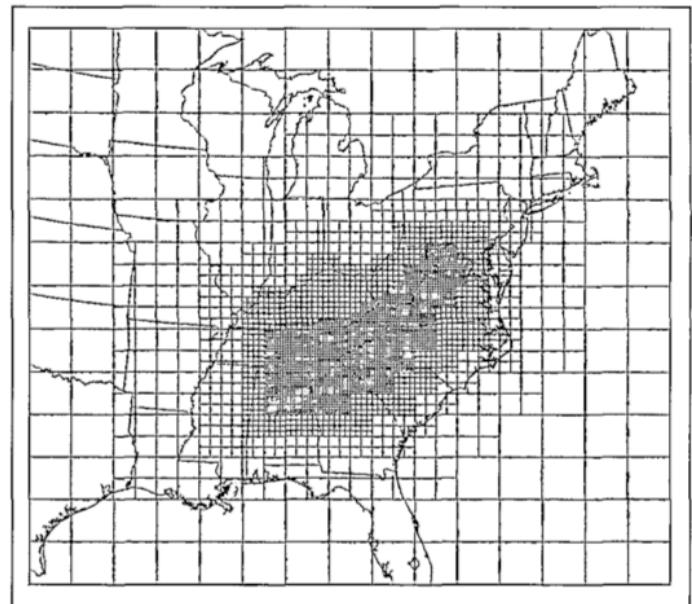
Observations: $\text{PM}_{2.5}$ (speciated and total), gas precursors (NO_x , VOCs, SO_2)

Method: Kalman filter (DDM) to adjust single domain-wide scale factors

Results:

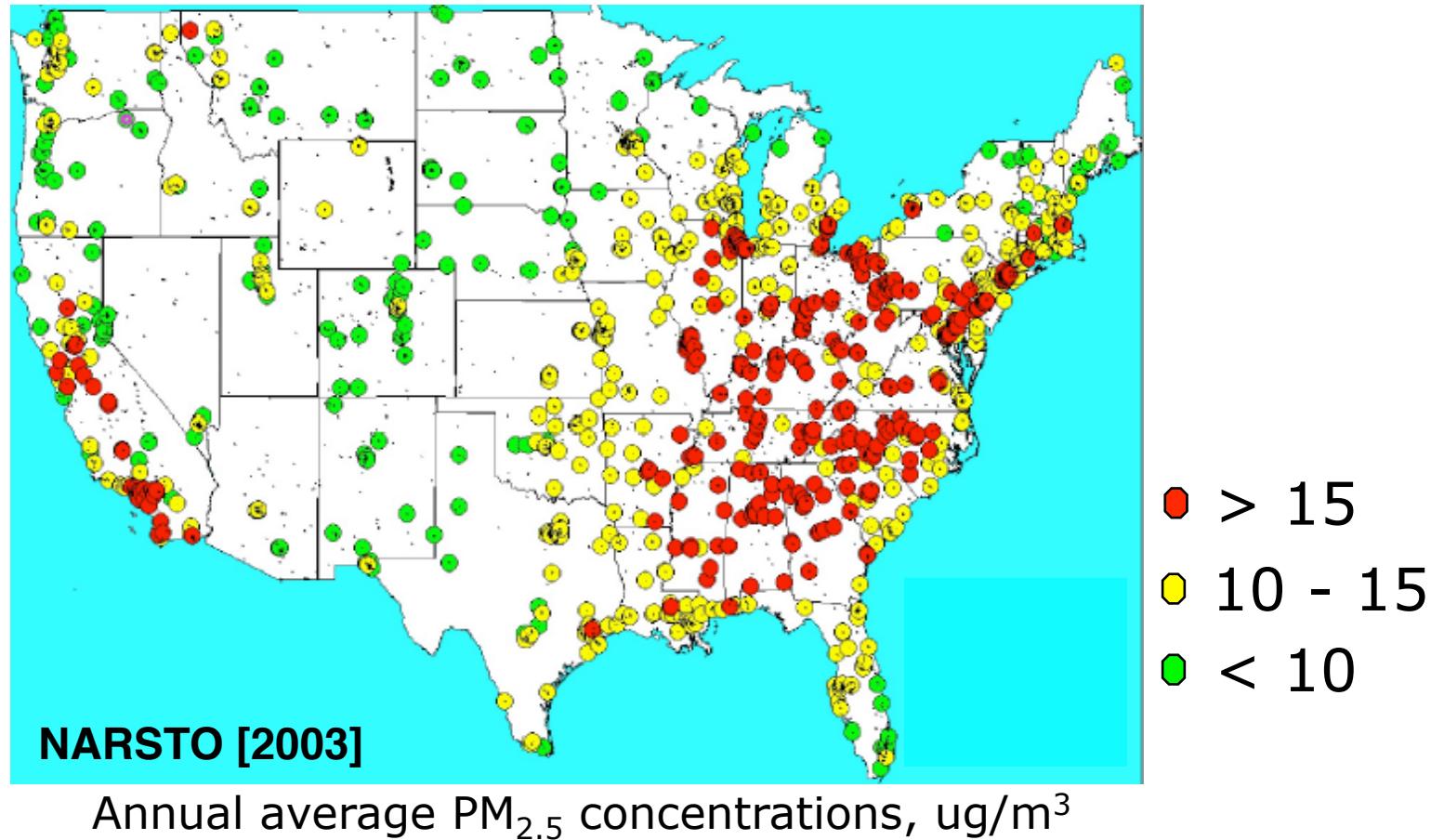
NH_3 emissions scaling factors:

- May 22-29, 1995: 0.59
- July 09-19, 1995: 0.59



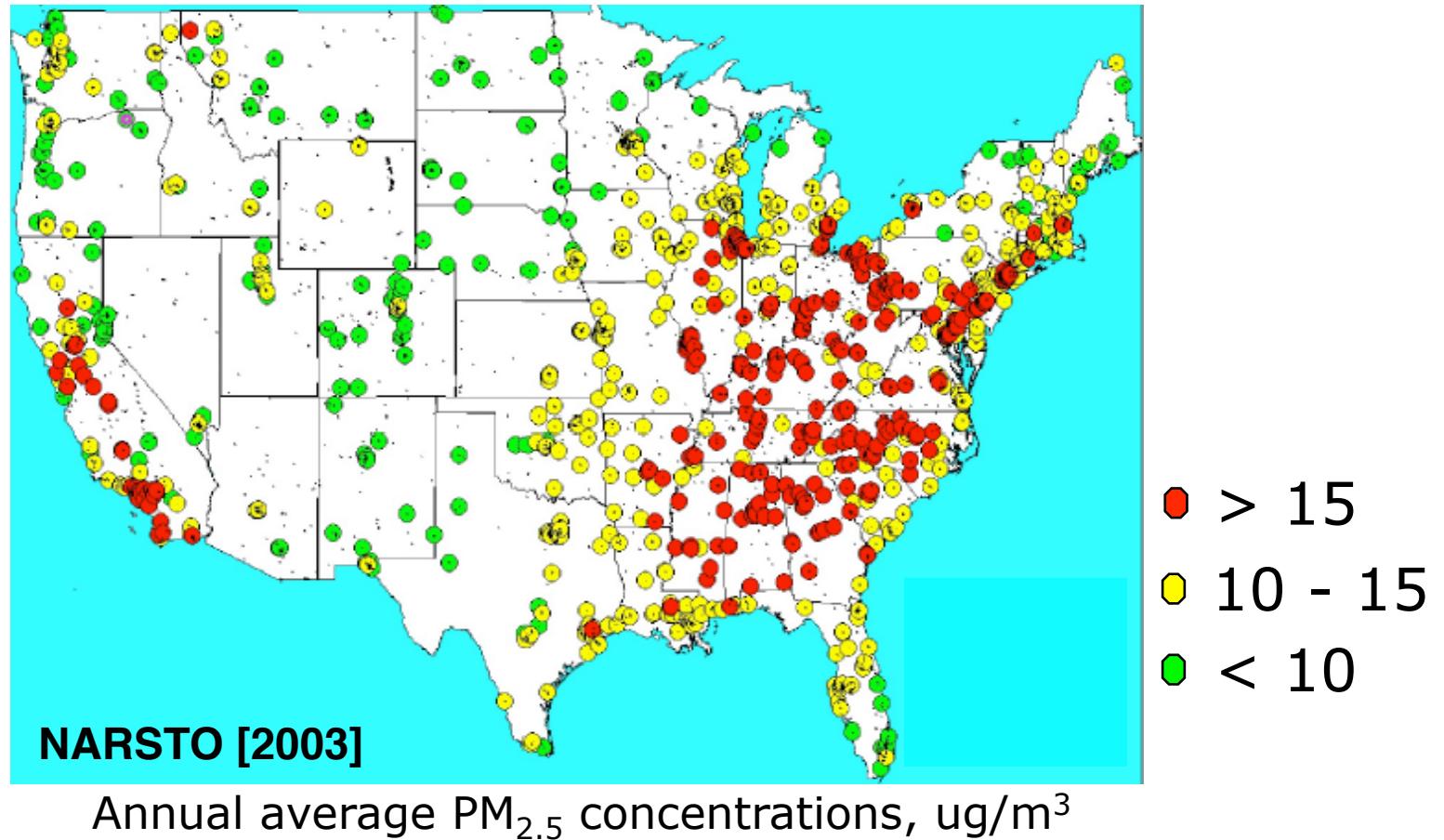
Modeling domain

Air Pollution: PM_{2.5}



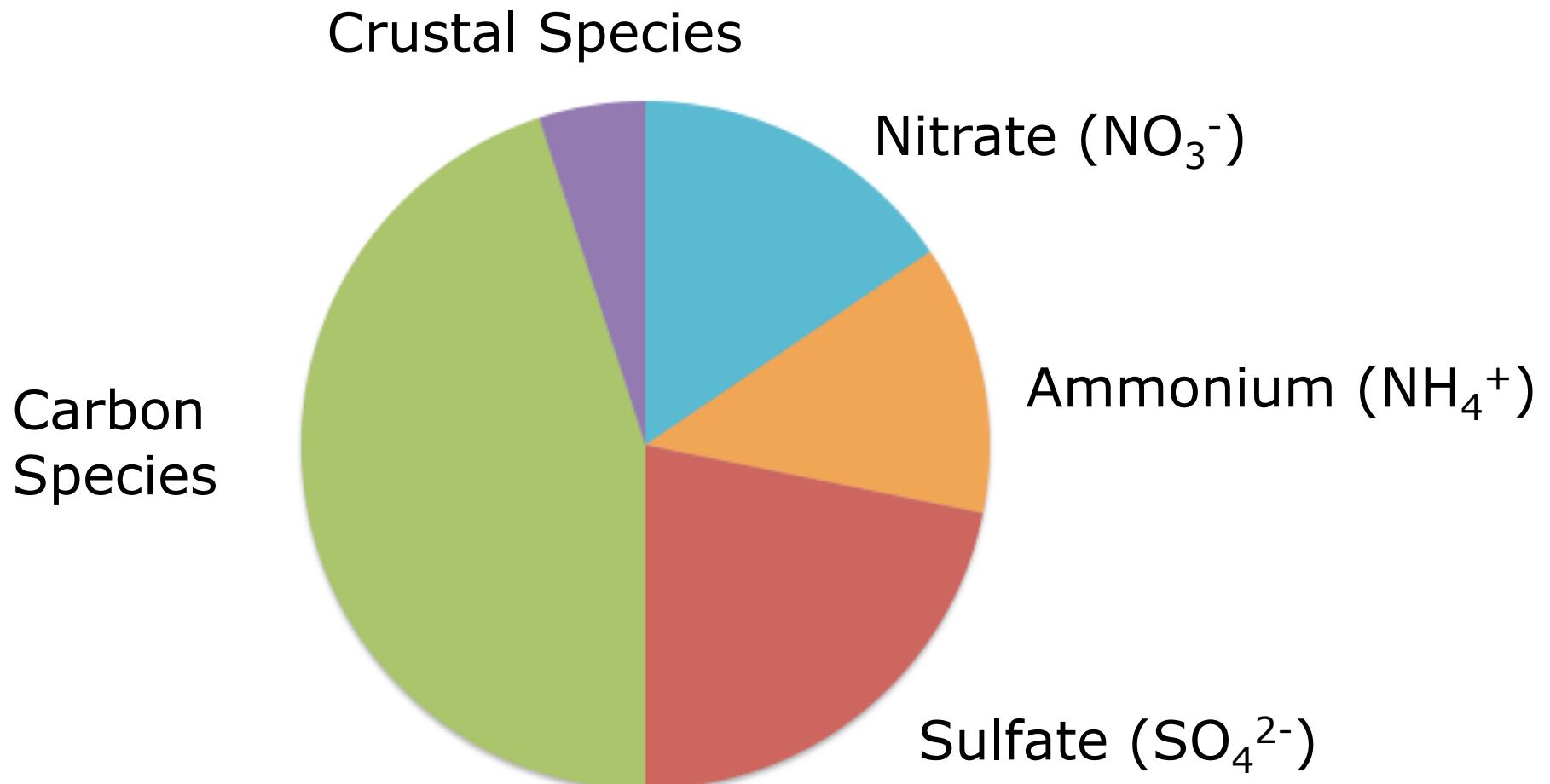
90 million people live in counties which are in exceedance of NAAQS for PM_{2.5} of 15 $\mu\text{g}/\text{m}^3$ (annual average). (EPA, 2003)

Air Pollution: PM

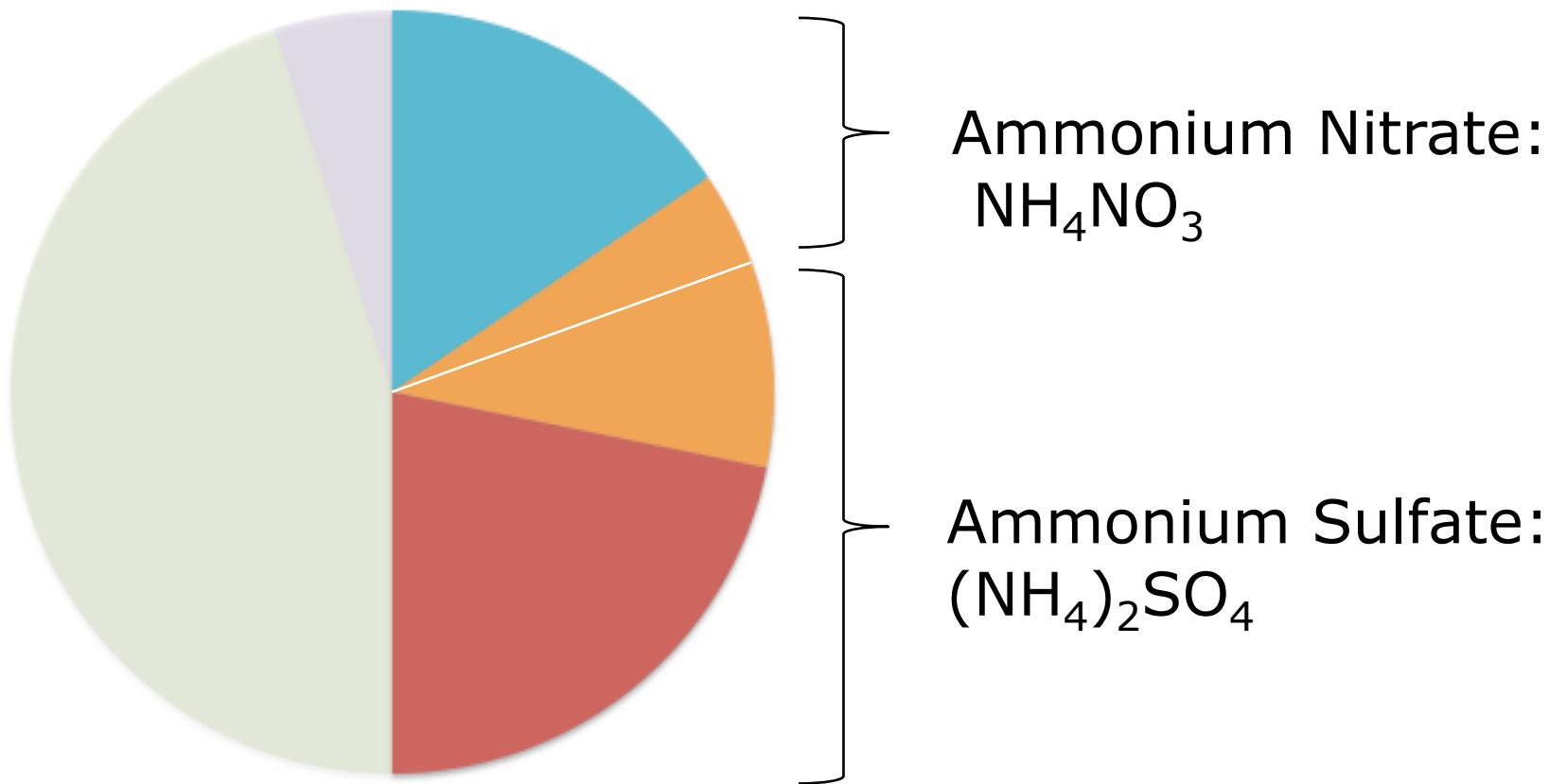


Cardiovascular disease, inhibited lung development, premature mortality
Pope, (2000); Pope *et al.*, (2002); Gauderman *et al.*, (2004)

Air Pollution: PM_{2.5} Composition



Air Pollution: PM_{2.5} Composition



Inverse modeling: additional considerations

How does solution depend upon initial error estimates?

$$\mathcal{J} = \frac{1}{2} \sum_{\mathbf{c} \in \Omega} (\mathbf{H}\mathbf{c} - \mathbf{c}_{obs})^T \mathbf{S}_{obs}^{-1} (\mathbf{H}\mathbf{c} - \mathbf{c}_{obs}) + \frac{1}{2} \gamma_r (\boldsymbol{\sigma} - \boldsymbol{\sigma}_a)^T \mathbf{S}_a^{-1} (\boldsymbol{\sigma} - \boldsymbol{\sigma}_a)$$

vary “parameters” of the inversions

Questions

What do current network observations directly tell us about NH₃ emissions?

- Not enough

How can indirect observations and modeling be used to constrain NH₃ emissions estimates?

How do emission estimates affect emissions mitigation strategies?

What new measurements required?

Questions

What do current network observations directly tell us about NH₃ emissions?

How can indirect observations and modeling be used to constrain NH₃ emissions estimates?

How do emission estimates affect emissions mitigation strategies?

What new measurements required?

Inverse modeling and data assimilation

Using observations to constrain emissions is an inverse problem.

In data assimilation, models and measurements are combined to create an optimal estimate of the state of the system

Questions

What do current network observations directly tell us about NH₃ emissions?

How can indirect observations and modeling be used to constrain NH₃ emissions estimates?

- can spatial variability be improved?

How do emission estimates affect emissions mitigation strategies?

What new measurements required?

Inverse Modeling: 4D-Var

4D Variation Data Assimilation (Kalnay, 2003):

- Optimize parameters at resolution of forward model
- Forward model equations are strong constraints

Inverse Modeling: 4D-Var

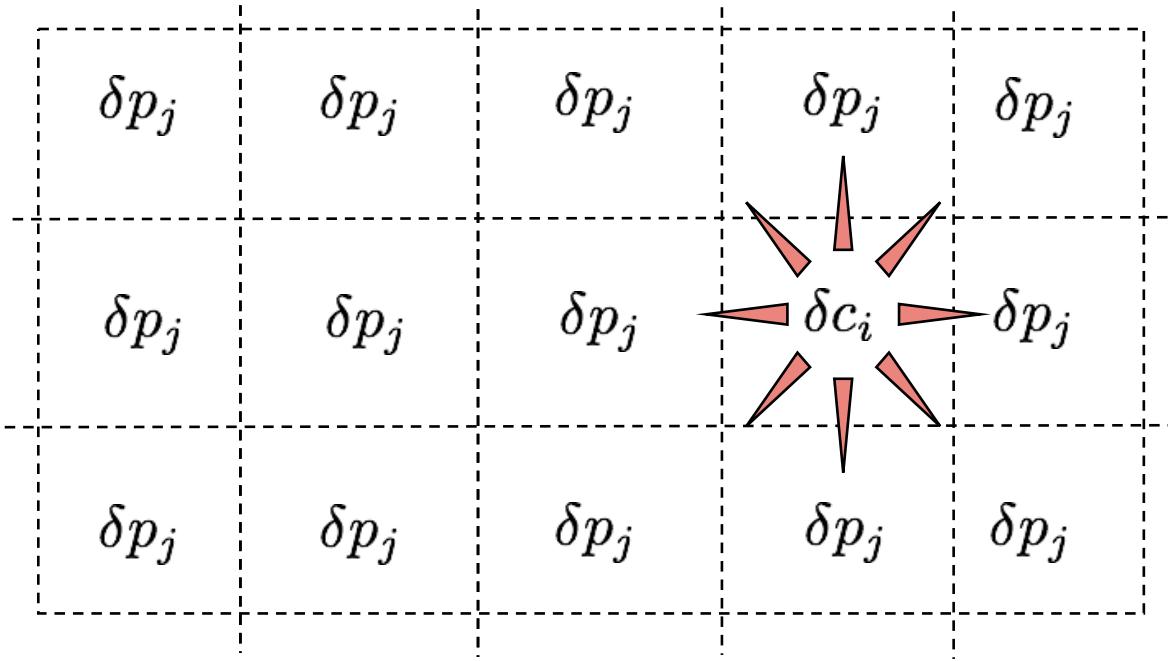
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- emissions estimates using remote sensing
 - MOPITT CO (Kopacz et al., 2008)
 - SCIAMACHY SO_2 , NO_2 (C. Lee, C. Shim, Q. Li, R. V. Martin)
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- NH_3 emissions estimates using surface obs. of sulfate and nitrate (IMPROVE)

Adjoint sensitivity



Method for calculating sensitivity of a single, scalar model response with respect to numerous (i.e., 10^6) model parameters that is very computationally efficient:

$$\text{time(adjoint model)} = 3 * \text{time(forward model)}$$

Questions

What do current network observations directly tell us about NH₃ emissions?

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Collaborators

Thanks to:



Mark Shepard, Karen Cady-Pereria,



Ming Luo, Kevin Bowman, TES team



Rob Pinder, John Walker

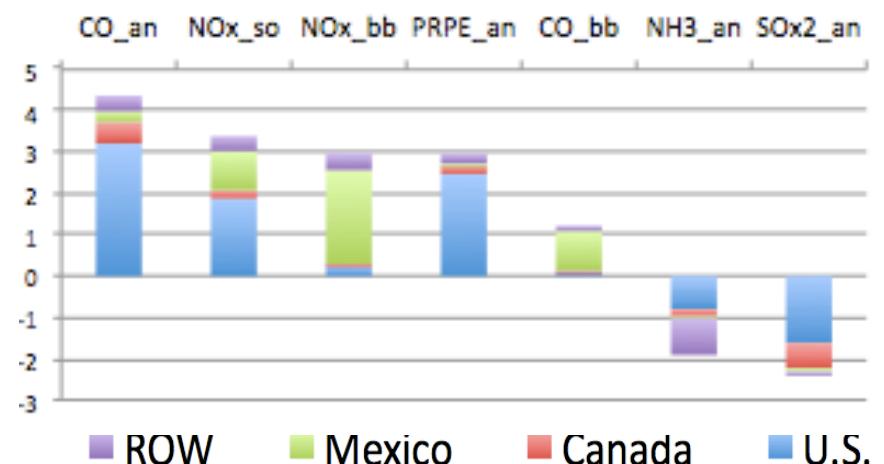
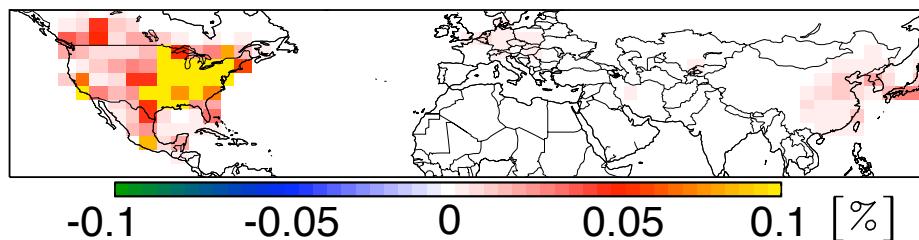


NASA GSFC: NCCS
NASA JPL: SCC

Additional projects

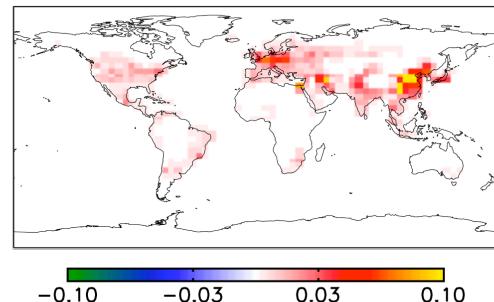
Assessing long-range influences on local air quality.

Sensitivity of US O₃ to alkane emissions

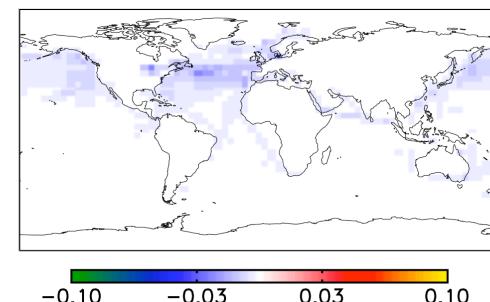


ClimateWorks: optimizing energy strategies to minimize global warming potential of aerosols.

BC fossil fuel

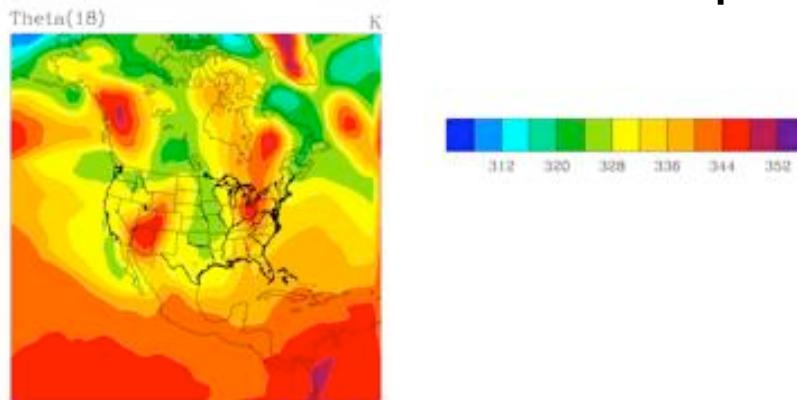


SO₂ shipping

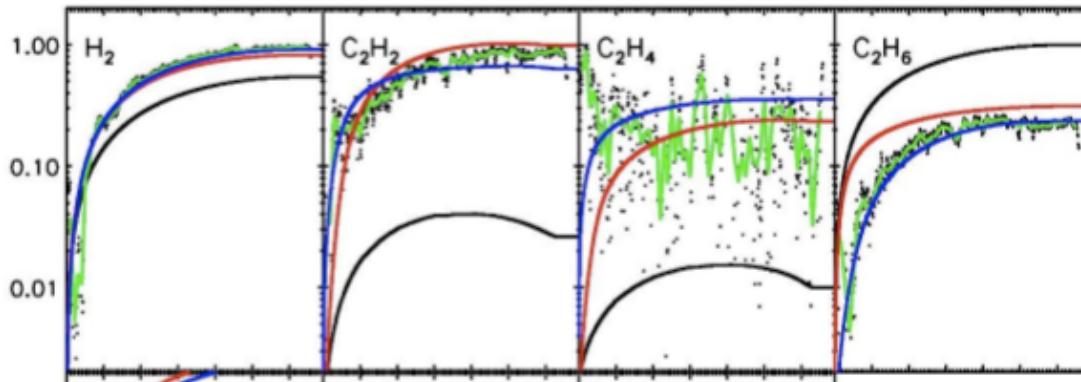


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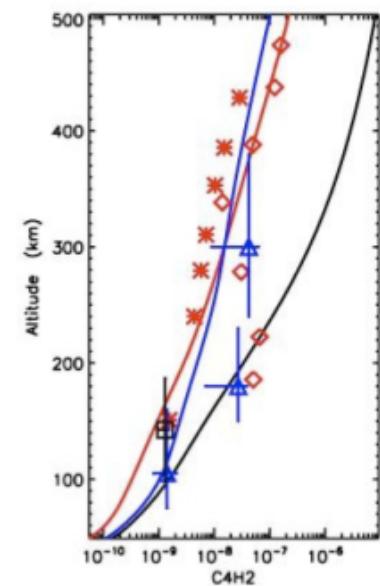
WRF-Chem / Var: how can chemical data assimilation improve weather and air quality forecasts?



Chemistry of outer planets (sub. Liang et al.)
JPL chamber



Titan

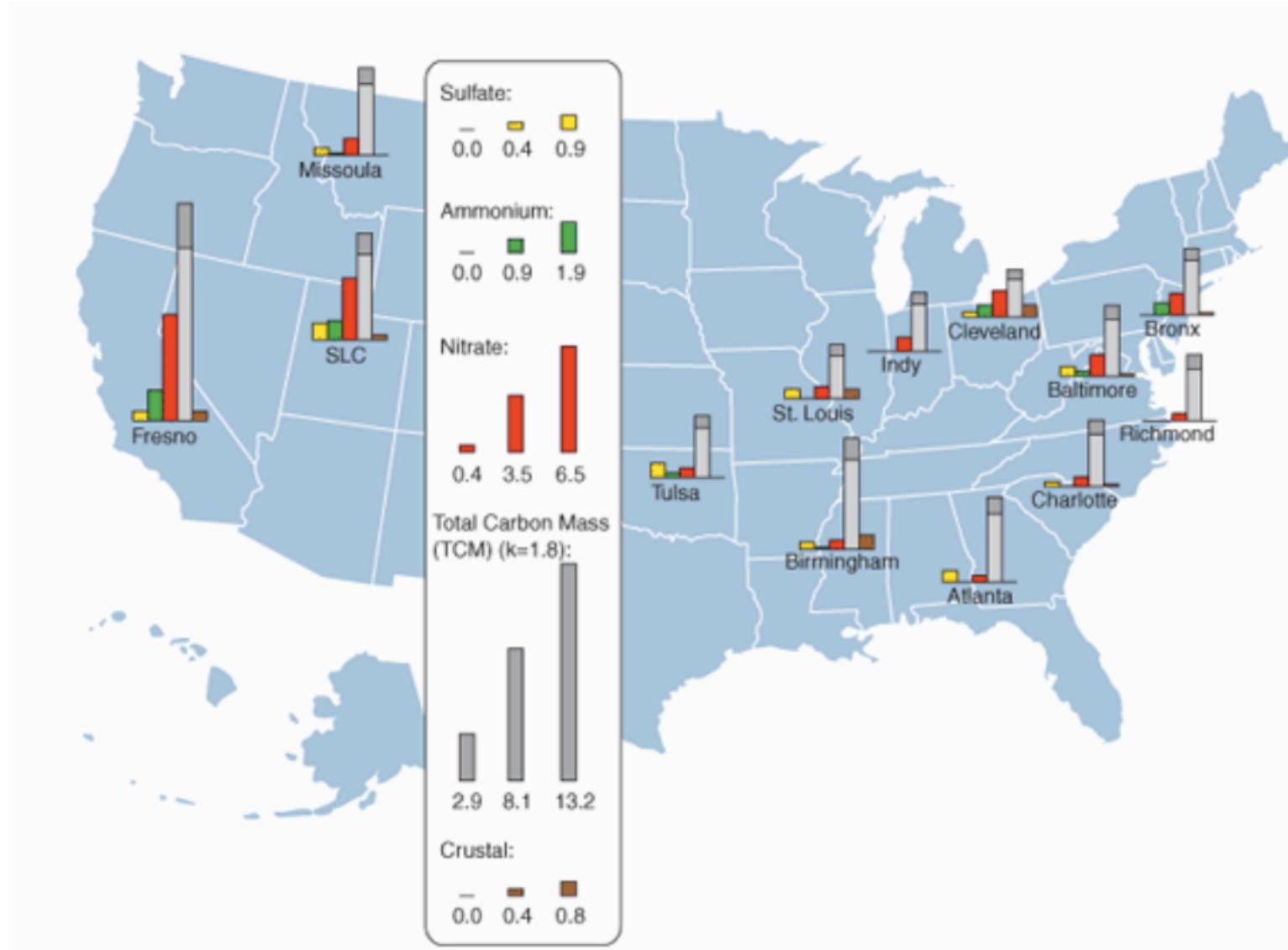


Directions

Emissions inverse modeling tests

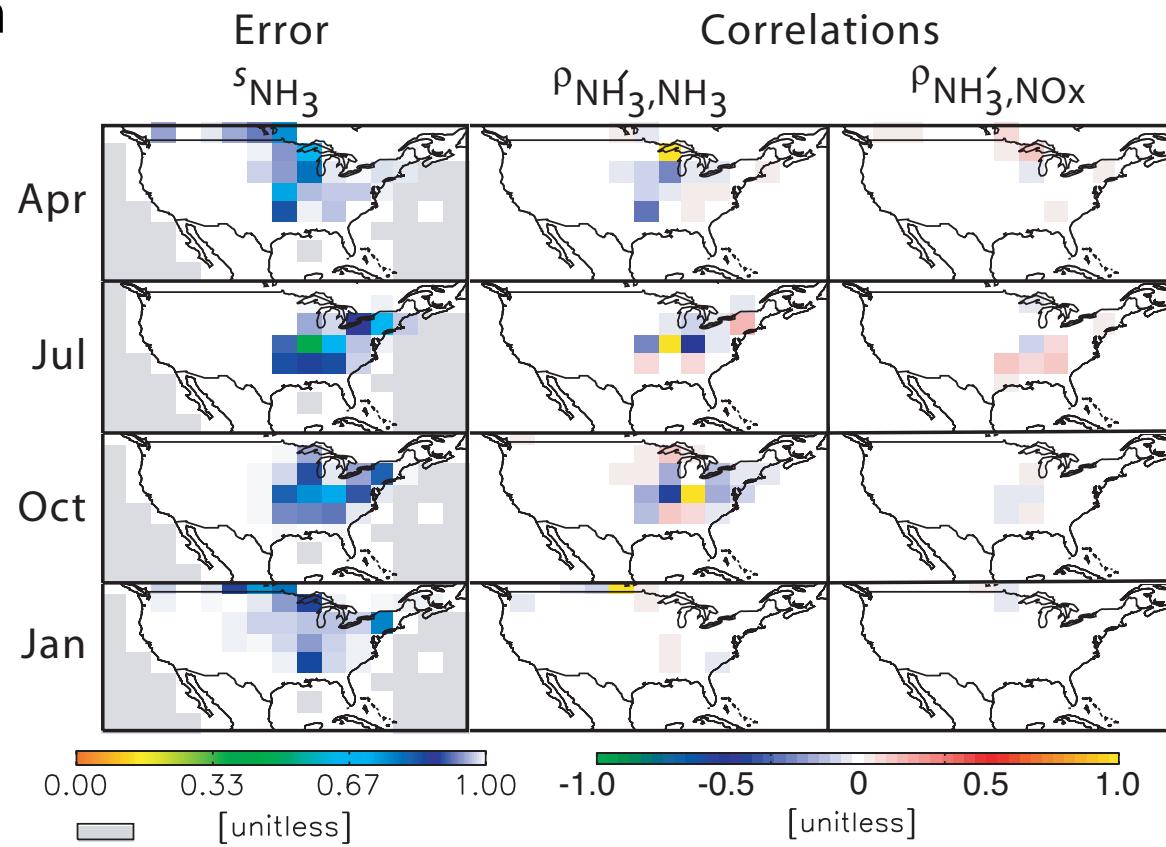
- simulate atmospheric NH₃ field
- simulate TES NH₃ observations
- can inverse model recover “true” emissions?

Air Pollution: Regional



Inverse modeling: uncertainty estimates

Inverse Hessian (IH) estimated by tracking progression towards minimum

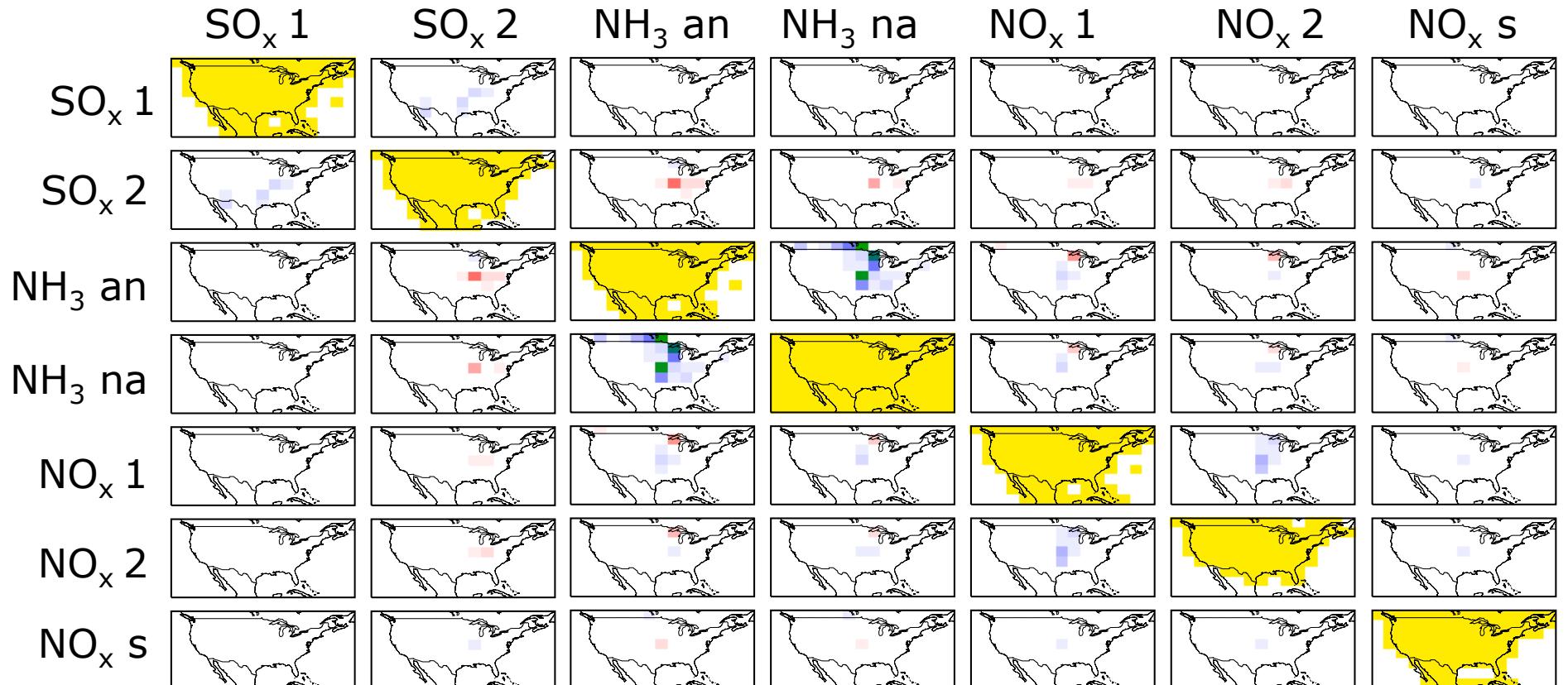


$$s_{\sigma_m} = (IH_{m,m})^{\frac{1}{2}}$$

$$\rho_{\sigma_{m_1}, \sigma_{m_2}} = \frac{IH_{m_1, m_2}}{(IH_{m_1, m_1} IH_{m_2, m_2})^{\frac{1}{2}}}.$$

Inverse modeling: uncertainty estimates

Correlation of emissions between species, same location



$$\rho_{\sigma_{m_1}, \sigma_{m_2}} = \frac{IH_{m_1, m_2}}{(IH_{m_1, m_1} IH_{m_2, m_2})^{\frac{1}{2}}}.$$

Adjoint modeling applications

Depending on “model response,” can be used for:

Sensitivity analysis: quantifying influence of uncertain model parameters (emissions, reaction rates, ...)
Response = Average concentrations of X in location Y...

Inverse modeling: using large data sets, optimizing parameters on resolution commensurate with forward model.

Response $\sim \text{sum}(\text{model} - \text{obs})^2$

Attainment studies: assessing the effectiveness of emissions changes on an air quality
Response = total amount of nonattainment

Adjoint modeling: History

From principles of functional analysis (Hilbert)

Used extensively for optimal control problems (Lions, 1971)

- nuclear reactor design (classified?!)
- oceanography (Tziperman and Thacker, 1989)
- meteorology (Derber, 1985)
- aeronautics (Giles and Pierce, 2000)

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- proposed for tracer analysis (Marchuk, 1974)
- stratospheric chemistry (Larry et al, 1995)
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Aerosols

- box model microphysics (Henze et al., 2004; Sandu et al., 2005)
- black carbon (tracer) (Hakami et al., 2005)
- coupled thermodynamics and chemistry (Henze et al., 2007)
- AOD (offline chemistry) (Dubovik et al., 2008)

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AQ Attainment

Consider a representative metric of PM_{2.5} air quality,

$$J = \frac{1}{2} \sum \text{MAX}[(\text{inorganic PM}_{2.5})_{24h} - 10\mu\text{g}/\text{m}^3, 0]^2$$

Calculate the sensitivity of this metric w.r.t. PM_{2.5} precursor emissions, E .

Map adjoint sensitivities:

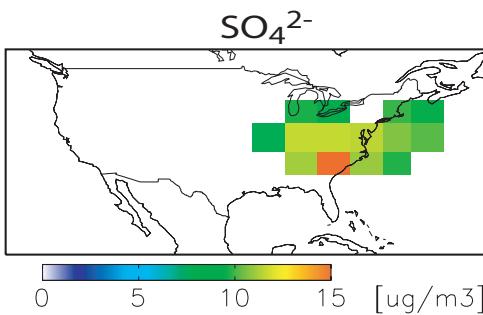
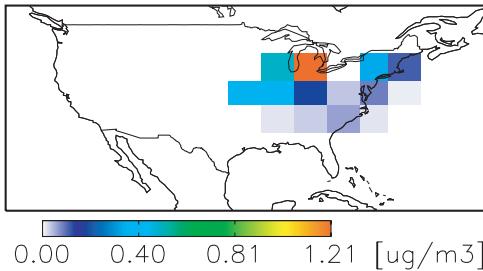
“Control effectiveness” = fully normalized sensitivities

$$\frac{\partial J}{\partial E} \times \frac{E}{J} \times 100\% \approx \frac{\Delta J[\%]}{\Delta E/E}$$

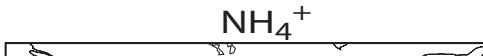
AQ Attainment: July

Nonattainment

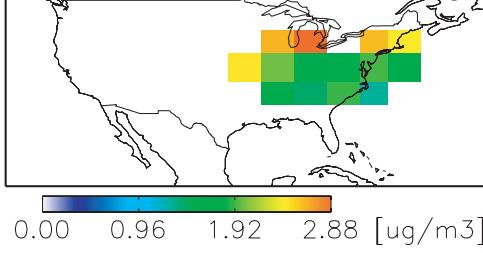
NO_3^-



SO_4^{2-}



NH_4^+



0.00 0.96 1.92 2.88 [$\mu\text{g}/\text{m}^3$]

JULY

Spatial and chemical variability in effectiveness

AQ Attainment: sector specific influences

$$\chi_m = \left(\frac{|\sum_i \lambda_{p_i, m}|}{|\sum_{i,m} \lambda_{p_i, m}|} - \frac{\sum_i p_{i,m}}{\sum_{i,m} p_{i,m}} \right) \times 100\%$$

Emission sector	January	April	July	October
SO _x surface	-11	-14	-11	-12
SO _x stack	16	17	13	13
SO _x shipping	-4	-2	-2	-1
NH ₃ anthropogenic	-10	-11	-16	-23
NH ₃ natural	9	11	12	14
NH ₃ biomass burning	-18	-9	-3	0
NH ₃ biofuel	18	9	8	10
NO _x surface	-2	-4	0	-6
NO _x stack	13	11	26	14
NO _x lightning	-6	-5	-19	-6
NO _x soil	-4	-2	-8	-2

m = sector
 i = location
 p = emission
 λ = sensitivity

AQ Attainment: long-range influences

$$J = \sum_{\text{in US}} ([\text{SO}_4^{2-}] + [\text{NH}_4^+] + [\text{NO}_3^-])_{24 \text{ h, surface level}}$$

$$\frac{\partial J}{\partial p} \frac{p}{J} \times 100\% \text{ for } p = \text{SO}_x \text{ stack}$$



SO_x surface



More than 25% of the influence of SO_x emissions on U.S. inorganic $\text{PM}_{2.5}$ comes from emissions outside the U.S.

AQ Attainment: long-range influences

Table shows the total influence of each sector (Total) and the percent of that total from each spatial domain on J. Sectors with largest influence are highlighted.

Emission sector	Total sensitivity	Percent from each region			
		U.S.	Canada	Mexico	ROW
SO _x surface	11.1	57.8	8.2	23.9	10.1
SO _x stack	30.1	75.1	16.7	3.4	4.7
SO _x shipping	2.0	67.9	6.9	6.4	19. 9
SO _x biomass burning	0.2	16.2	1.1	77.3	5.4
SO _x bio fuel	0.03	2.9	25.4	36.1	35.6
NH ₃ anthropogenic	19.6	90.0	6.0	2.3	1.7
NH ₃ natural	9.2	89.4	8.4	0.1	1.3
NH ₃ biomass burning	0.6	60.1	2.3	33.3	3.1
NH ₃ biofuel	3.48	95.4	3.9	0.4	0.2
NO _x surface	6.7	84.4	5.3	8.3	2.0
NO _x stack	2.7	97.7	1.1	0.4	0.8
NO _x lightning	0.1	68.3	1.1	24.3	6.2
NO _x soil	0.7	65.3	4.1	28.8	1.7

Final comments on adjoint sensitivities

Disadvantages

- sensitivities \neq source attribution

Advantages

- Computational efficiency
- No perturbation to forward model
 - sensitivities around current model state
 - relevant for policy (+/- 10-30% Δ emission)
- Models can be first conditioned to observations using 4D-Var
- Estimates of emissions influence side-by-side with estimates influence of other parameters

Towards actual decision making activities

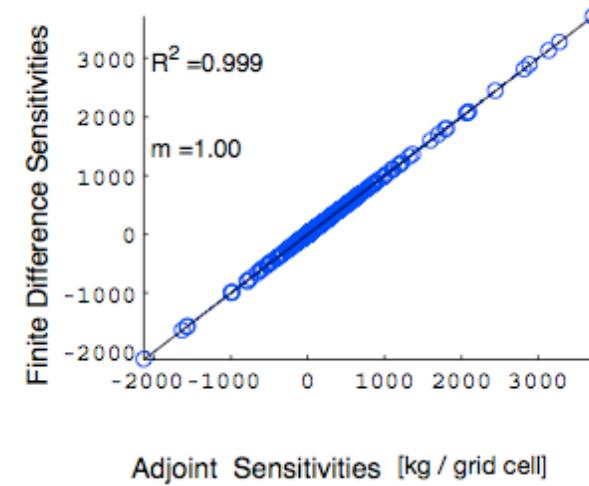
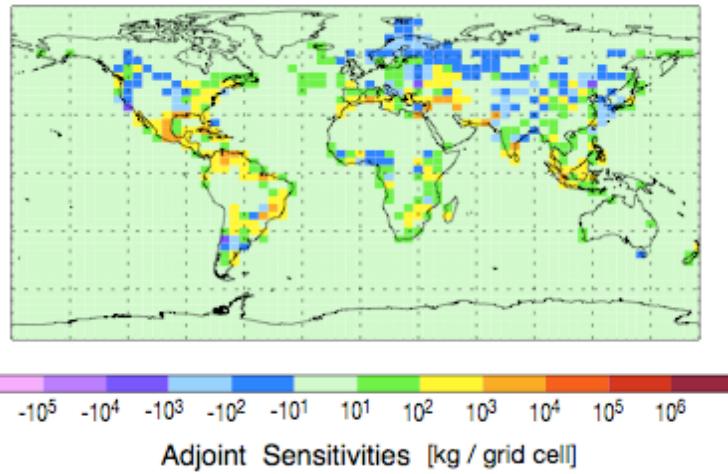
Consider additional observational constraints

- combine remote sensing with surface observations
- gas- and aerosol-phase

Testing the Adjoint: single processes, 1 week

$$\frac{\partial J(g_{NIT})}{\partial \sigma_{ESO_x}}$$

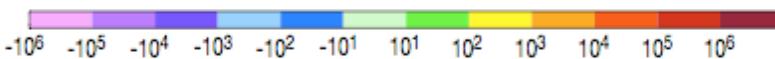
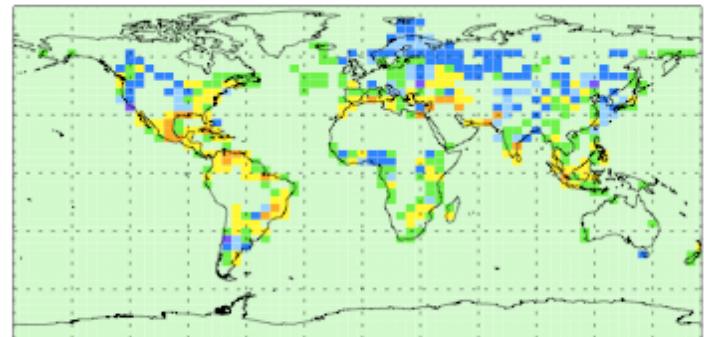
(thermo only)



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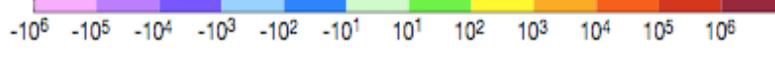
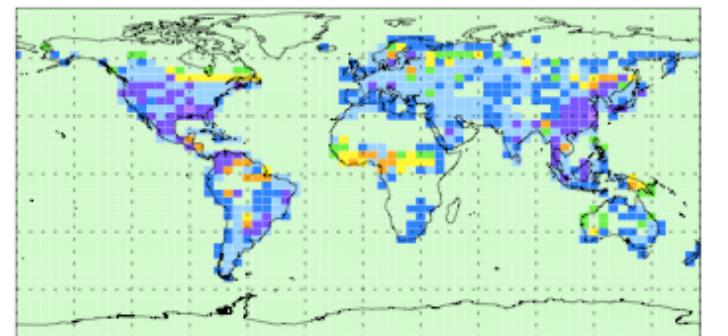
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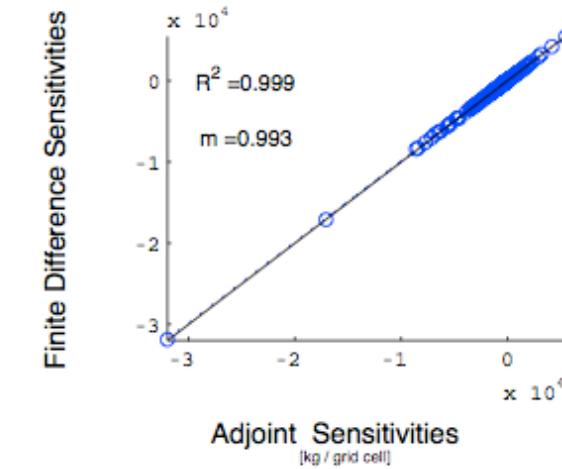
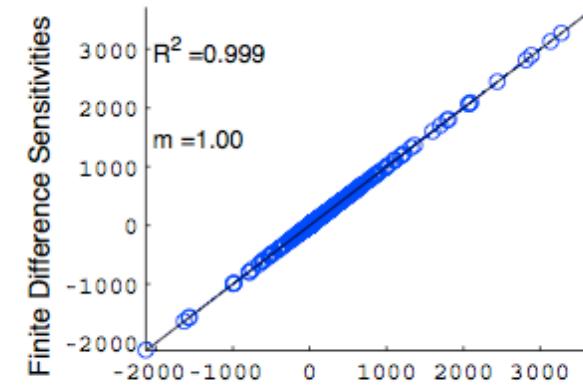
Adjoint Sensitivities [kg / grid cell]

$$\frac{\partial J(g_{MACR})}{\partial \sigma_{ENO_x}}$$

(chem only)



Adjoint Sensitivities [kg / grid cell]



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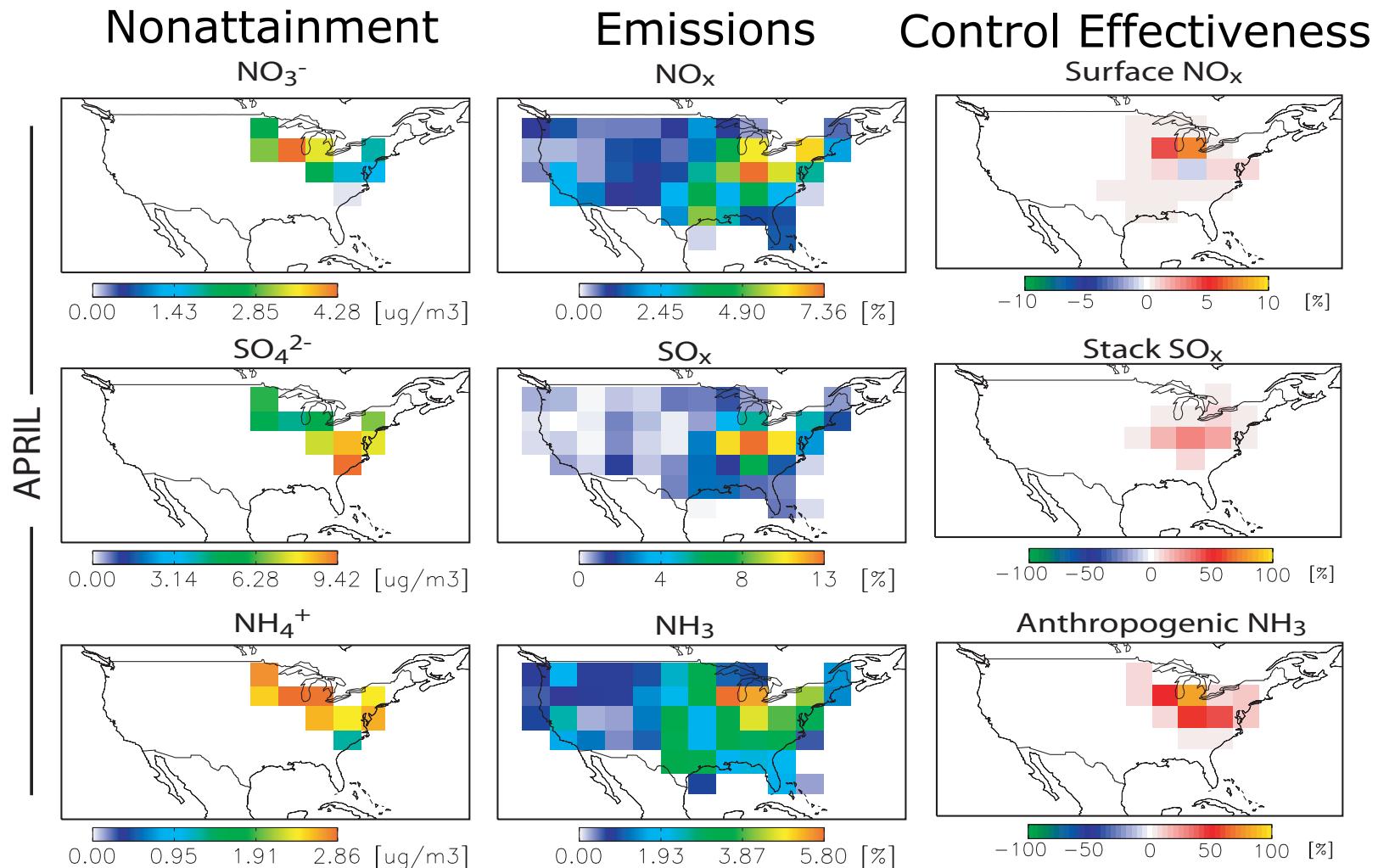
“Susceptibility” = semi normalized sensitivities

$$\frac{\partial J}{\partial E} \times \frac{100\%}{J} \approx \frac{\Delta J[\%]}{E[\text{molec/cm}^2/\text{s}]}$$

“Control effectiveness” = fully normalized sensitivities

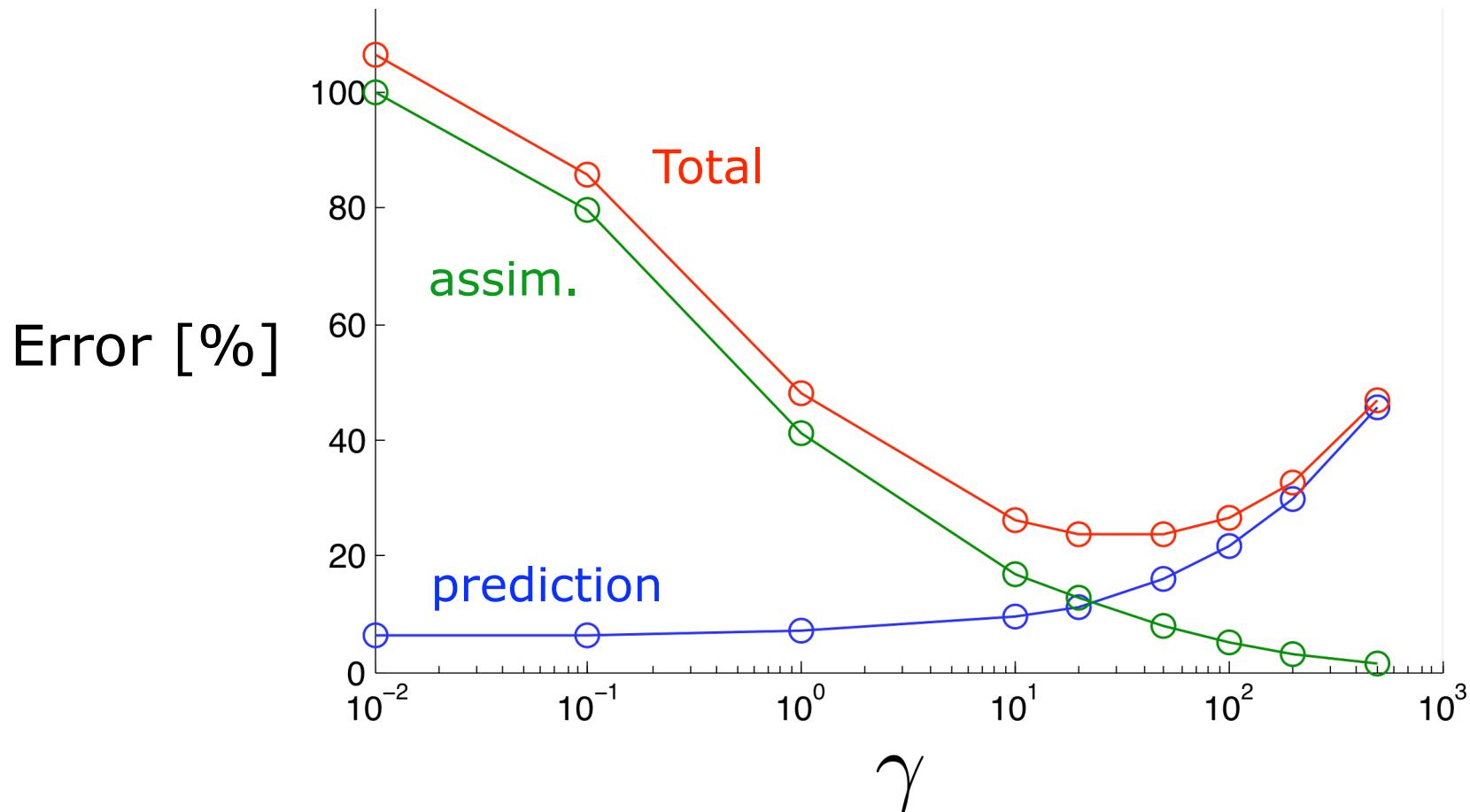
$$\frac{\partial J}{\partial E} \times \frac{E}{J} \times 100\% \approx \frac{\Delta J[\%]}{\Delta E/E}$$

AQ Attainment: April



Regularization Parameter

$$J = \frac{1}{2} \sum_{\mathbf{c} \in \Omega} (\mathbf{H}\mathbf{c} - \mathbf{c}_{obs})^T \mathbf{S}_{obs}^{-1} (\mathbf{H}\mathbf{c} - \mathbf{c}_{obs}) + \frac{1}{2} \gamma_r (\boldsymbol{\sigma} - \boldsymbol{\sigma}_a)^T \mathbf{S}_{\boldsymbol{\sigma}_a}^{-1} (\boldsymbol{\sigma} - \boldsymbol{\sigma}_a)$$



Discrete Adjoints (general)

Consider a discrete governing equation that propagates the vector of concentrations from time step n to step $n+1$:

$$\mathbf{c}^{n+1} = F^n(\mathbf{c}^n, \mathbf{p}), \quad \mathbf{c}^n = [c_1^n, \dots, c_k^n, \dots, c_K^n]^T \quad \text{Concentrations of species } k$$

$$p = [p_1, \dots, p_m, \dots, p_M] \quad \text{Parameters (emissions)}$$

$J(\mathbf{c}, \mathbf{p}) = \sum_{n=1}^N g^n(\mathbf{c}^n)$ is a model response; we are interested in sensitivity w.r.t. parameters, $\nabla_{\mathbf{p}} J$.

1. Define the local Jacobians:

$$\frac{\partial \mathbf{c}^{n+1}}{\partial \mathbf{c}^n} = \frac{\partial F^n(\mathbf{c}^n)}{\partial \mathbf{c}^n} \equiv \mathbf{F}_c^n, \quad \frac{\partial \mathbf{c}^{n+1}}{\partial \mathbf{p}} = \frac{\partial F^n(\mathbf{c}^n)}{\partial \mathbf{p}} \equiv \mathbf{F}_p^n$$

4. Solve iteratively (& backwards)

$$\text{initialize: } \lambda_c^N = \frac{\partial g^N}{\partial \mathbf{c}^N}$$

2. Define the adjoint variable:

$$\lambda_c^n = \nabla_{\mathbf{c}^n} J = \sum_{n'=n}^N \frac{\partial g^{n'}}{\partial \mathbf{c}^n}$$

iterate: DO n = N, 1, -1

$$\lambda_c^n = (\mathbf{F}_c^{n-1})^T \lambda_c^{n-1} + \frac{\partial g^{n-1}}{\partial \mathbf{c}^{n-1}}$$

END DO

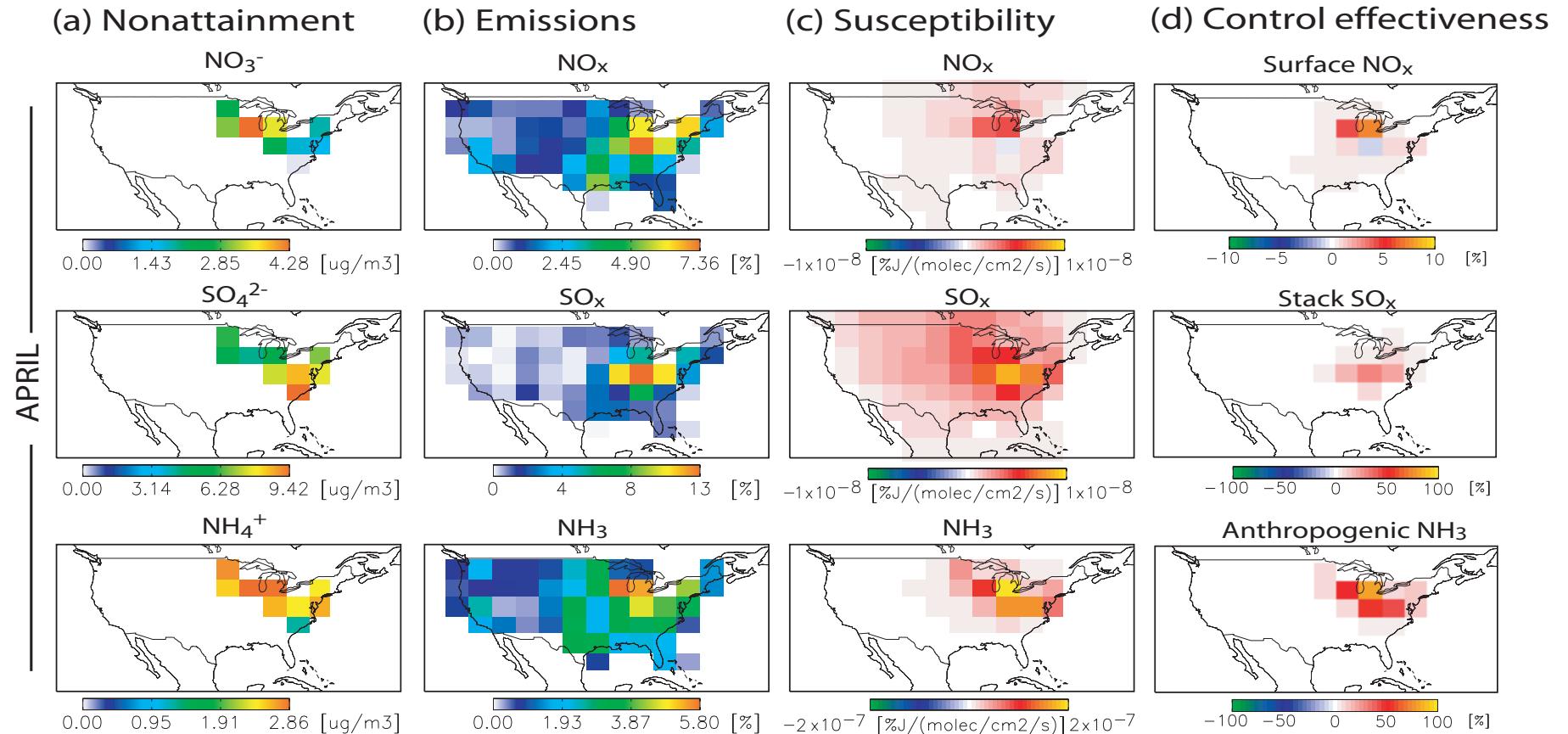
3. Expand the RHS of (2) using the chain rule

$$\lambda_c^n = \sum_{n'=n+1}^N \left(\prod_{n''=n}^{n'-1} (\mathbf{F}_c^{n''})^T \right) \frac{\partial g^{n'}}{\partial \mathbf{c}^{n'}} + \frac{\partial g^n}{\partial \mathbf{c}^n}$$

result: $\lambda_c^0 = \nabla_{\mathbf{c}^0} J$

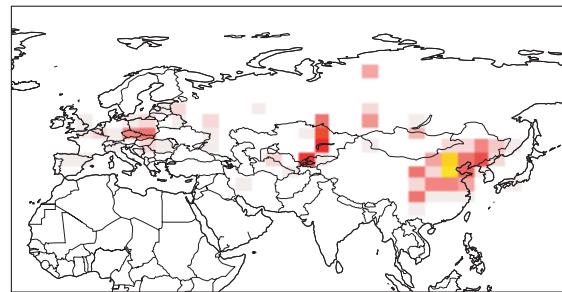
$$\lambda_p = \nabla_{\mathbf{p}} J = \sum_{n=1}^N (\mathbf{F}_p^{n-1})^T \lambda_c^n$$

AQ Attainment

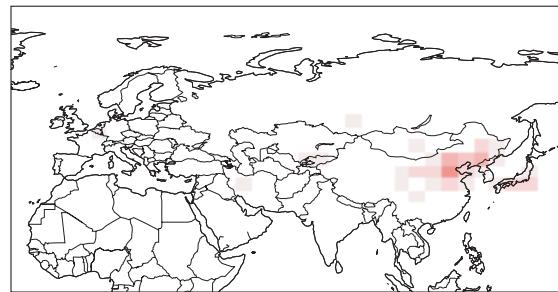


GEOS-Chem vs IMPROVE

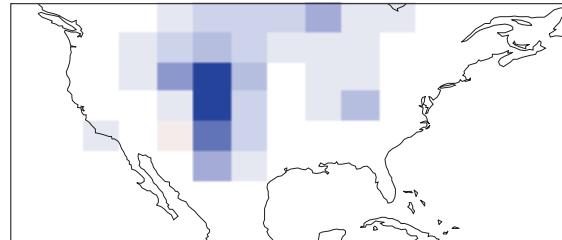
(a) Stack emissions of SO_X



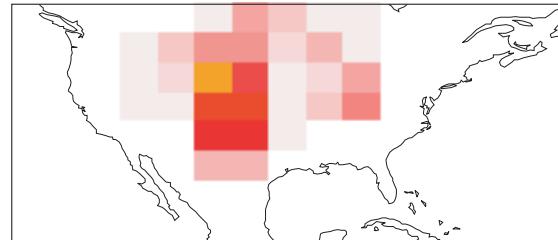
(b) Surface emissions of NO_X



(c) Initial conditions (933 hPa): SO_4^{2-}



(d) Initial conditions (933 hPa): NH_4^+



Aerosols and Radiative Forcing

IPCC 4, WGI, Ch 2, p180:

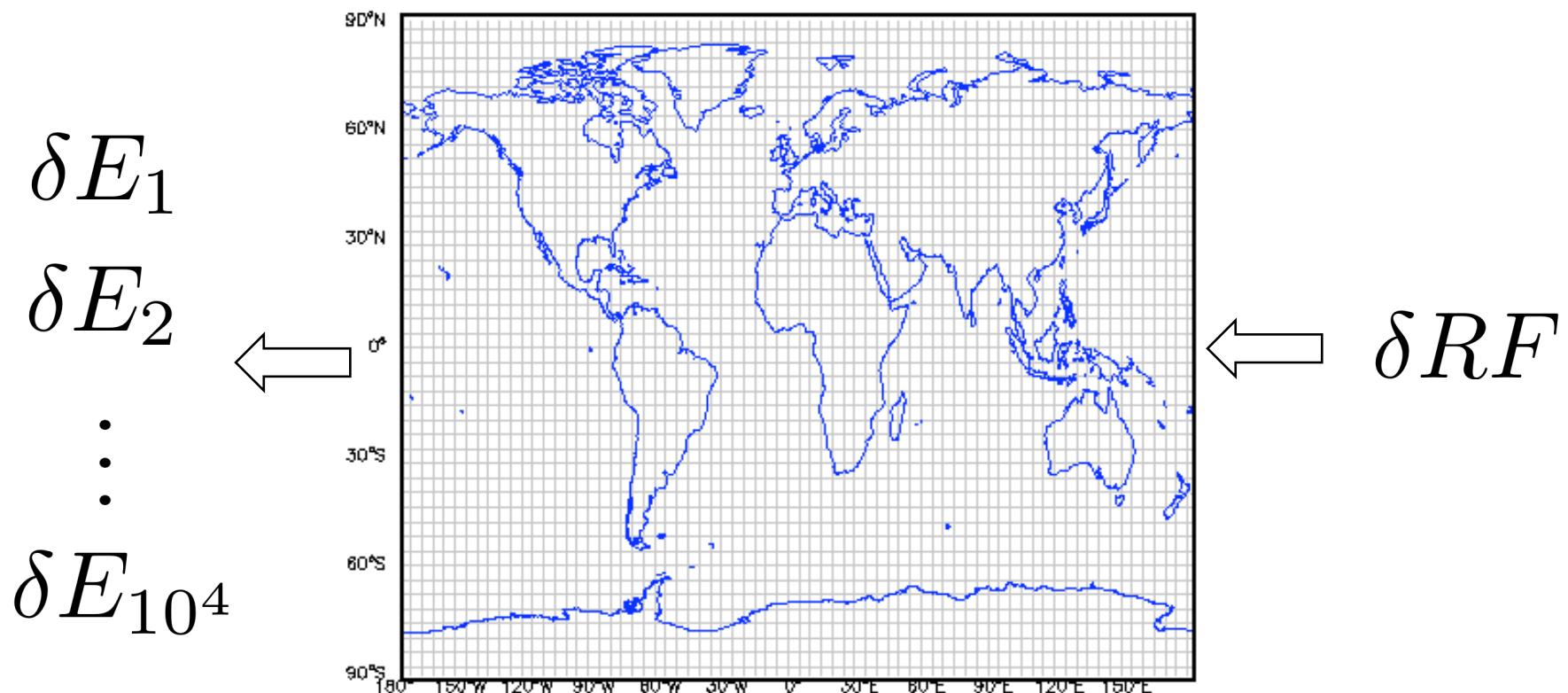
"It would be useful to identify the RF contribution attributable to different source categories (Section 2.9.3 investigates this).

However, few models have separated out the RF from specific emission source categories."

Forster et al., 2007: Changes in Atmospheric Constituents and in Radiative Forcing.
In: Climate Change 2007: The Physical Science Basis. Contribution of Working Group I to the Fourth Assessment Report of the Intergovernmental Panel on Climate Change

Aerosols and Radiative Forcing

Sensitivities from every sector and every region:



Calculated efficiently ($3 \times t_{\text{fwd}}$) with an adjoint model

GEOS-Chem direct RF

Implement radiative transfer, LIDORT, with derivative capabilities (Spurr, 2002).

Approximations made:

Macro

- Clear sky
- Only direct effects
- Only $\text{SO}_4\text{-NO}_3\text{-NH}_4\text{-H}_2\text{O}$ and BC aerosol

Micro

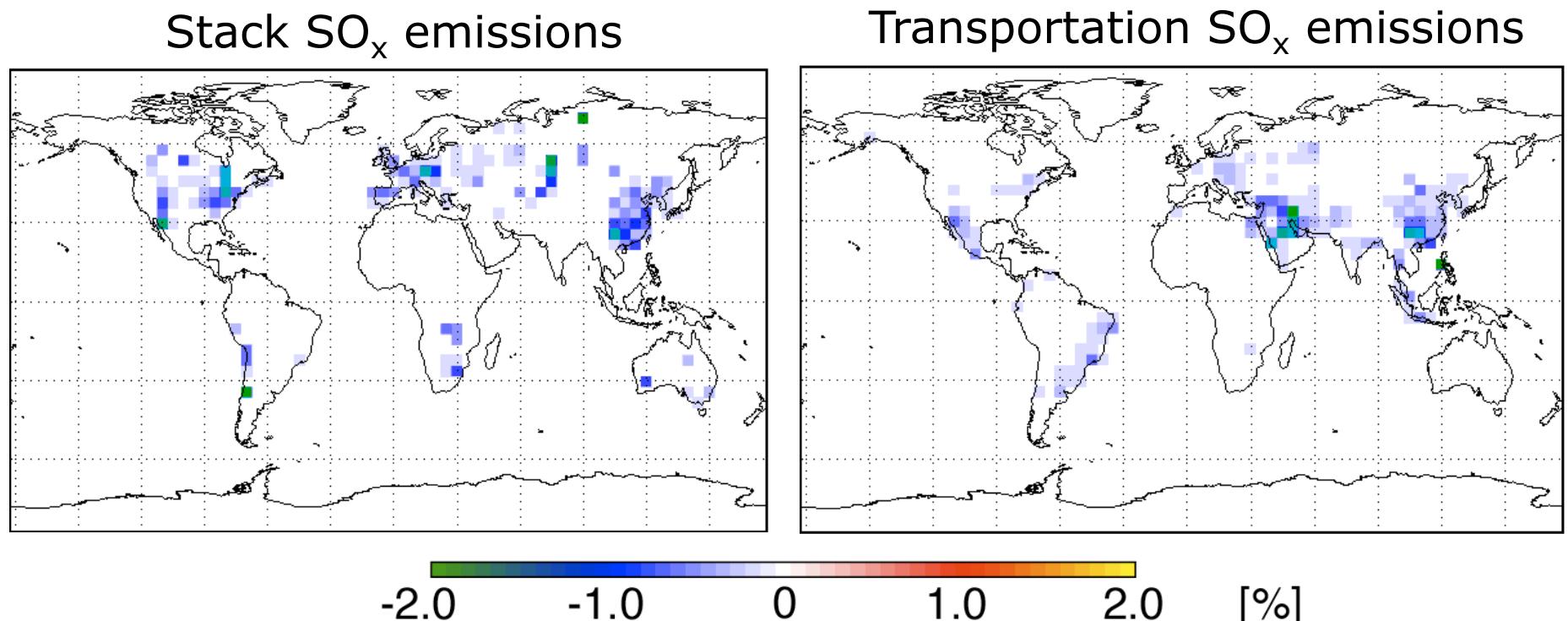
- Refractive index of $\text{SO}_4\text{-NO}_3\text{-NH}_4\text{-H}_2\text{O}$ is that of SO_4 .
- Assumed dry size
- External mixture

Timescale

- 1 week in July

Clear sky aerosol direct radiative forcing

Sensitivities with respect to different emissions sectors:

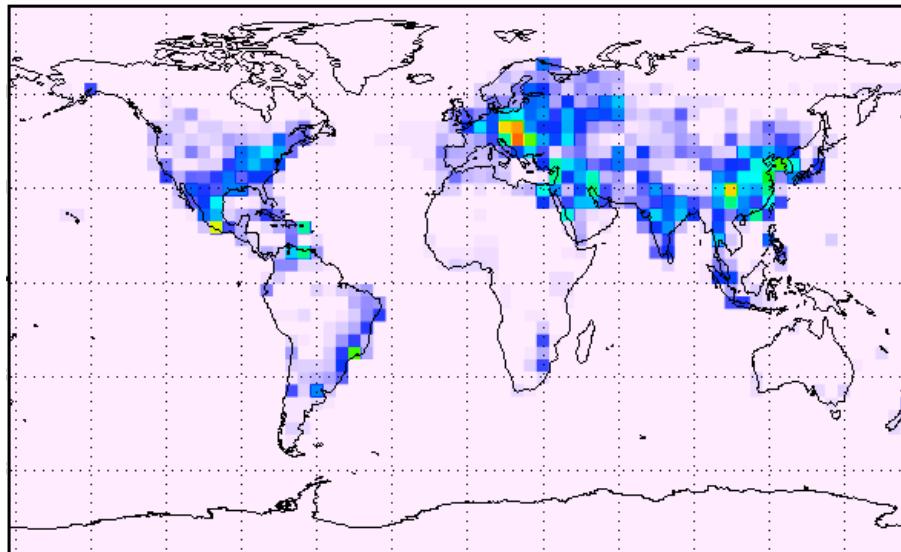


$$\text{sensitivities} = \frac{\partial RF}{\partial E} \frac{E}{RF} \times 100\% \approx \frac{\Delta RF [\%]}{100\% \Delta E}$$

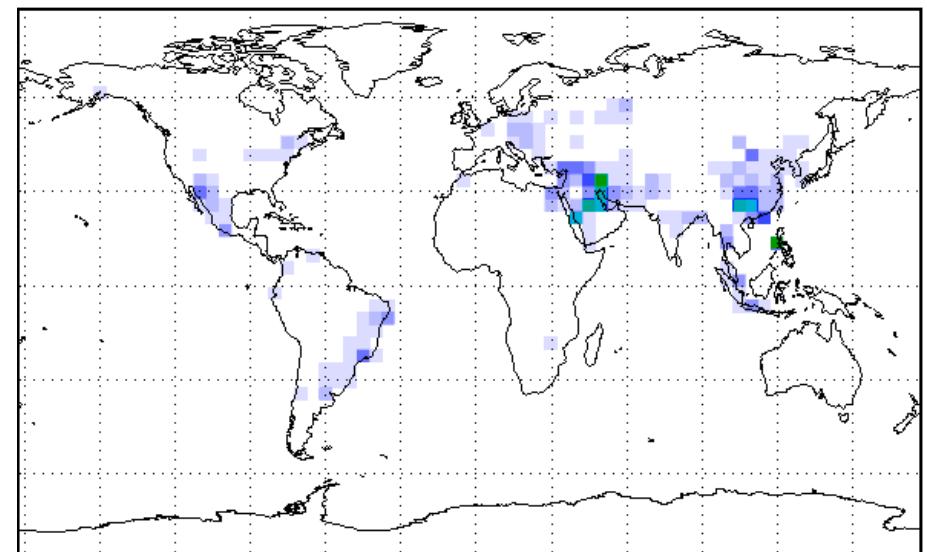
Clear sky aerosol direct radiative forcing

Consider transport sector SO_x

emissions



sensitivities

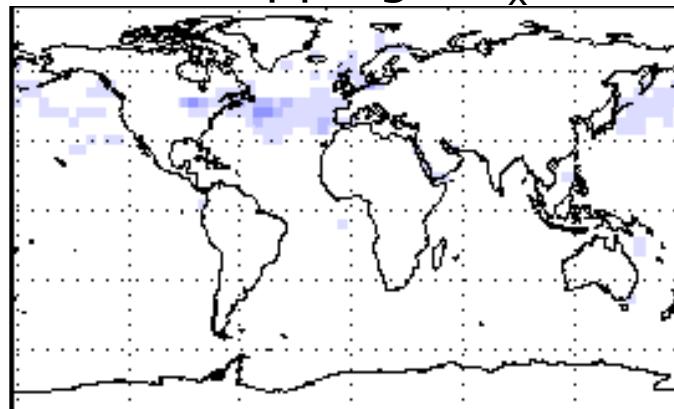


Location matters

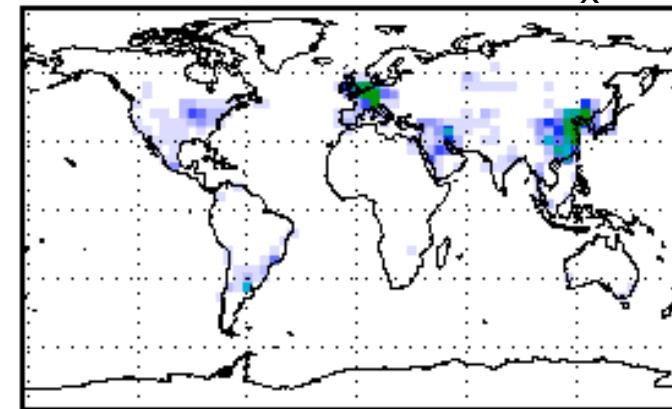
Clear sky aerosol direct radiative forcing

Sensitivities with respect to different species & sectors:

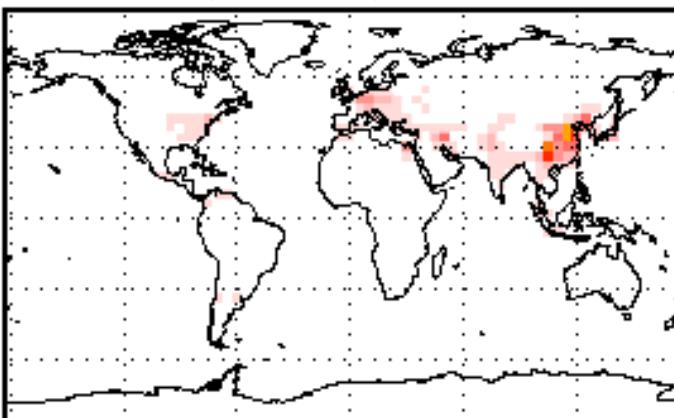
Shipping SO_x



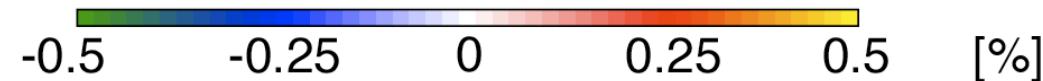
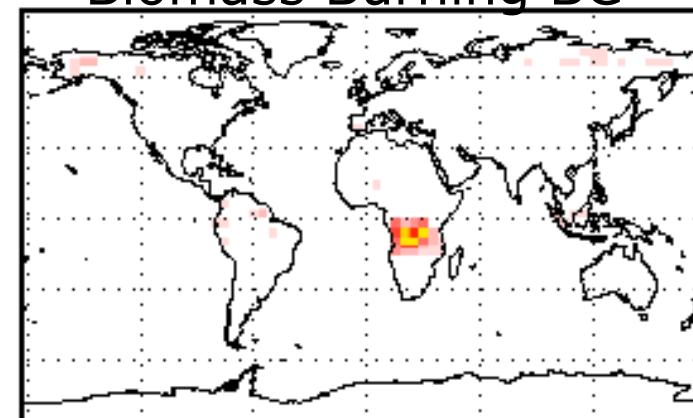
Transportation NO_x



Fossil Fuel BC



Biomass Burning BC



Future work

Additional factors to consider

- Organic carbon
- Refractive index / mixing state
- Clouds
- Additional days / seasons
- Indirect effects
- Use GISS climatology, future emissions scenarios

Final comments on adjoint sensitivities

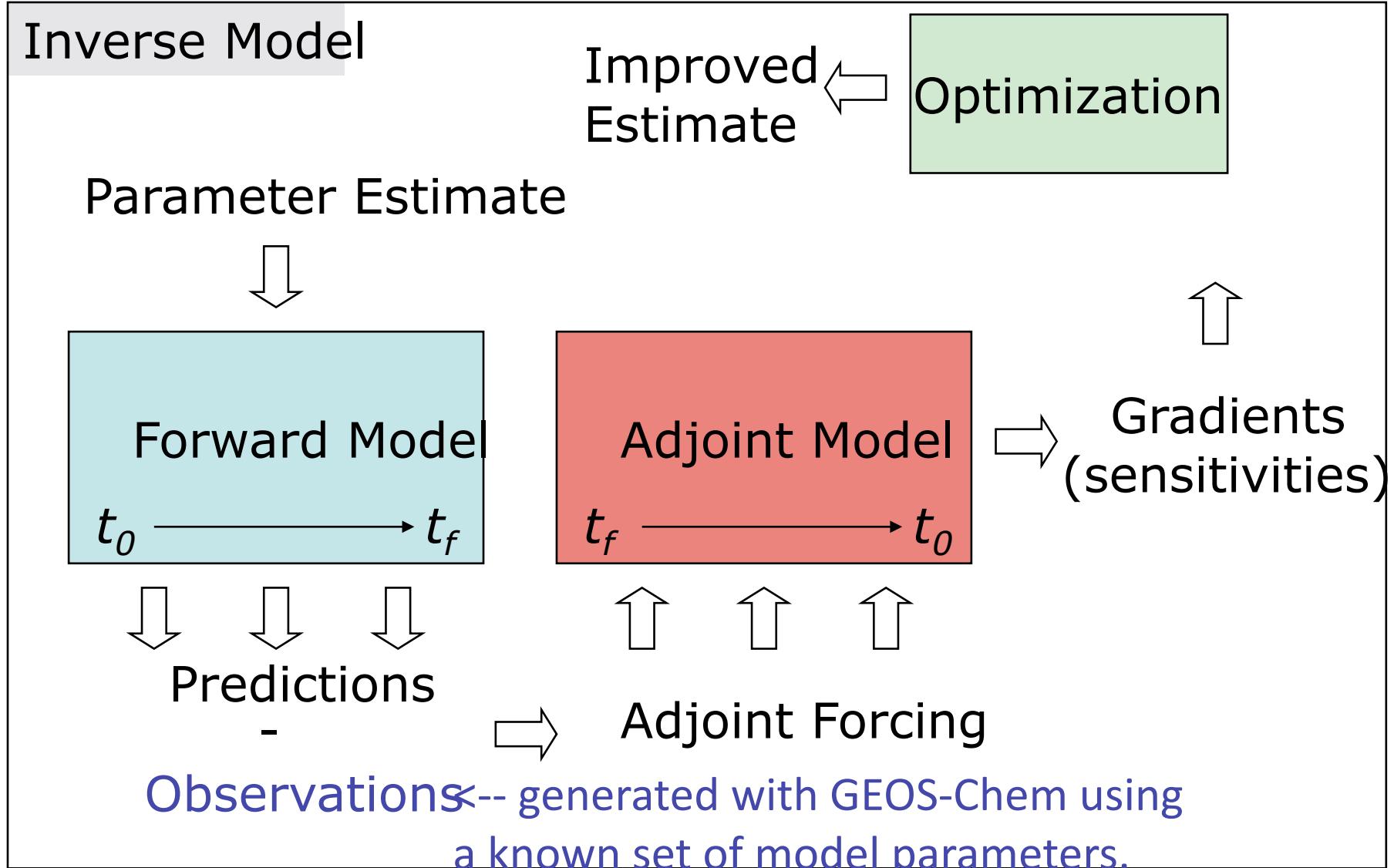
Disadvantages

- sensitivities \neq source attribution

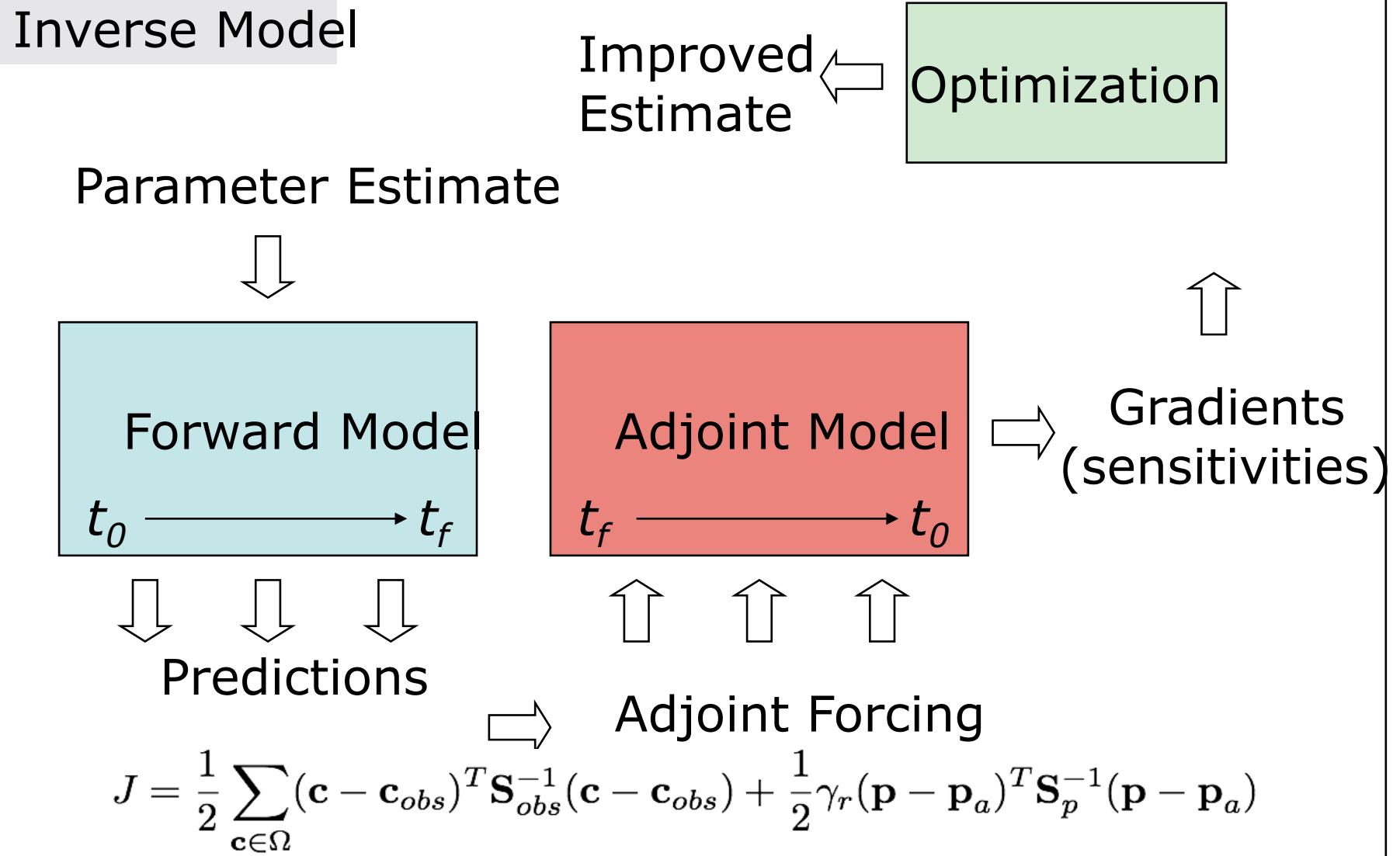
Advantages

- Computational efficiency
- No perturbation to forward model
 - sensitivities around current model state
 - relevant for policy (+/- 10-30% Δ emission)
- Models can be first conditioned to observations using 4D-Var
- Estimates of emissions influence side-by-side with estimates influence of other parameters (ex: D_{dry})

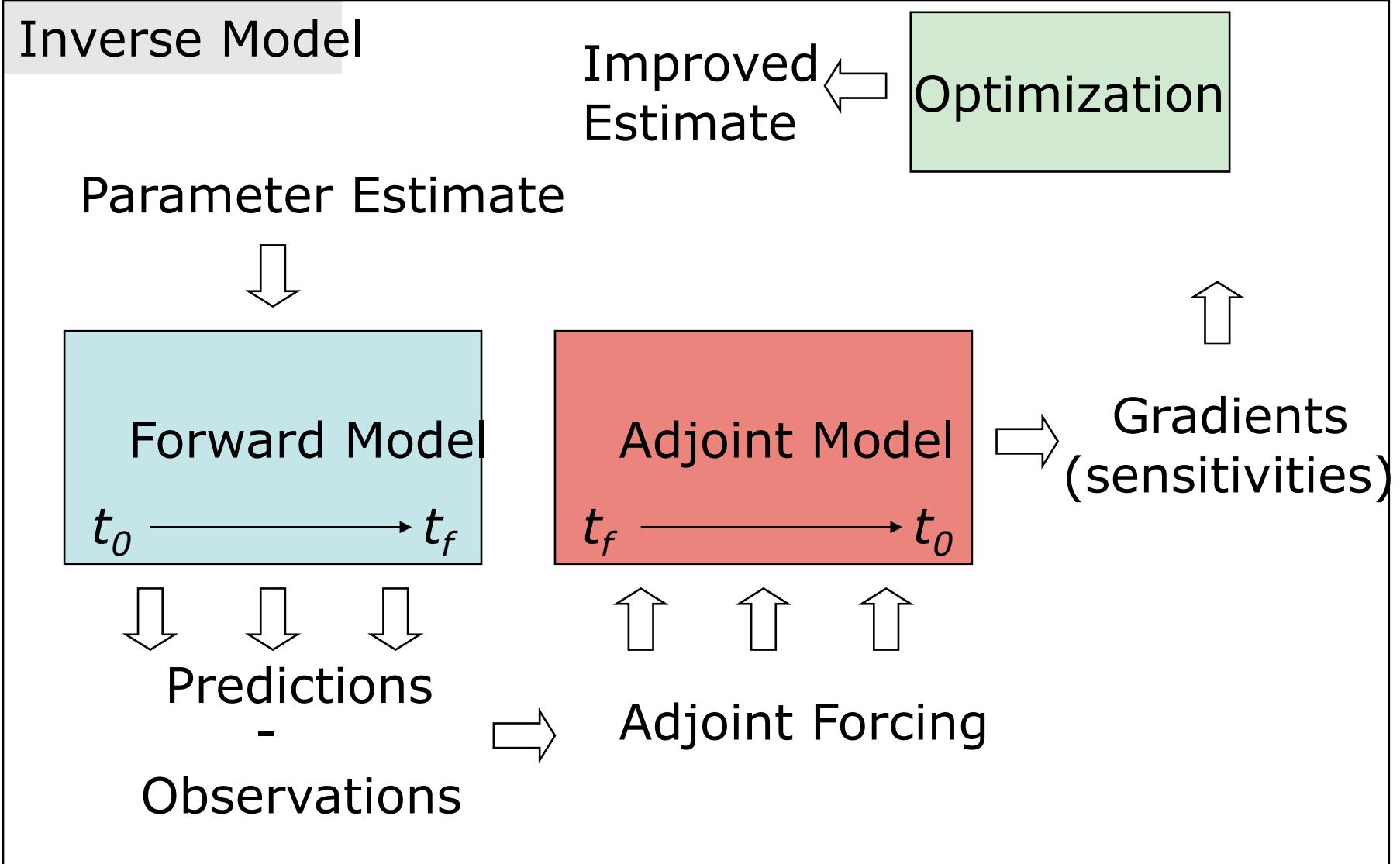
Inverse Modeling Tests: Psuedo Observations



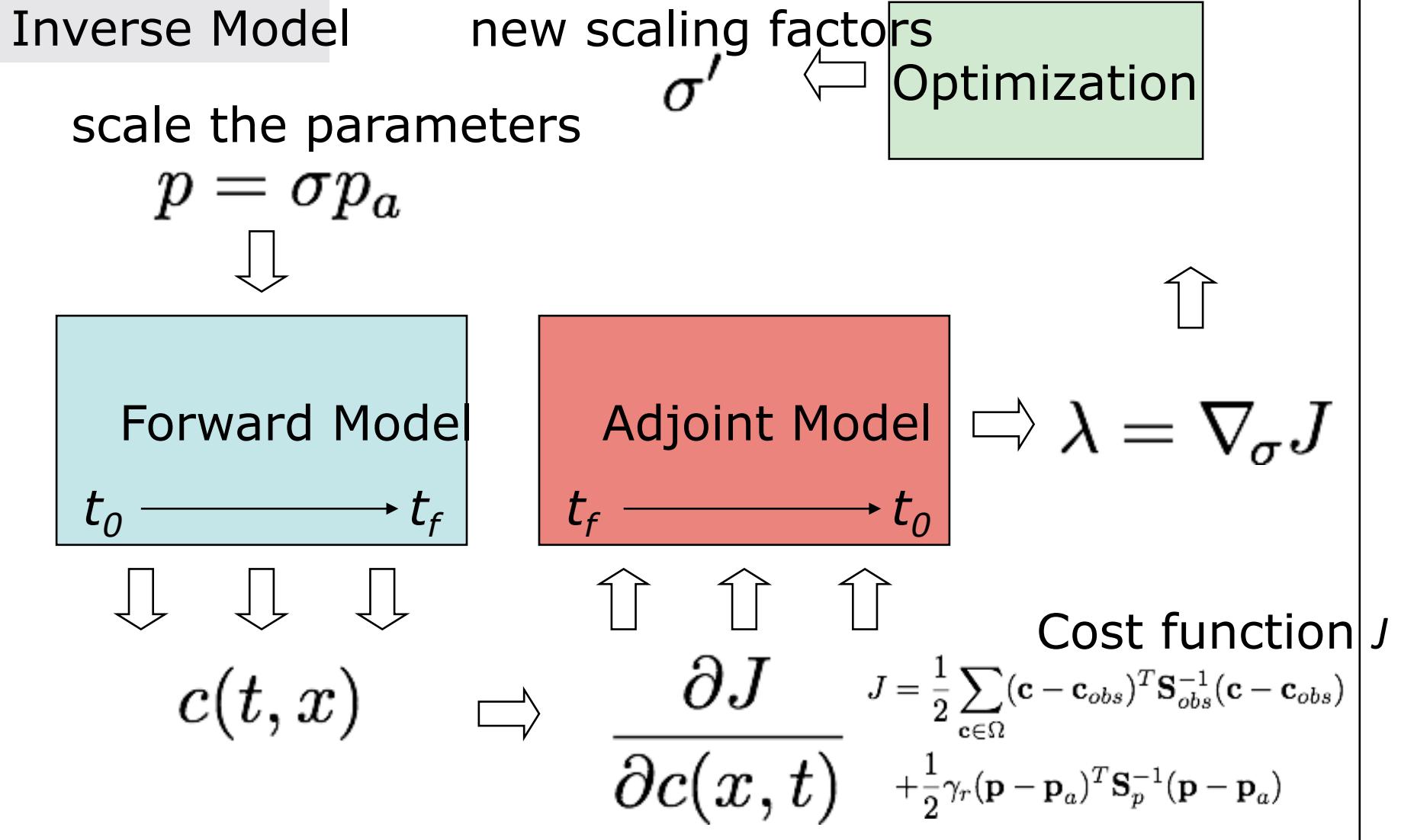
Inverse Modeling Tests: Psuedo Observations



Inverse Modeling using Adjoint Model



Inverse Modeling using Adjoint Model



Discrete Adjoint of Chemical Reaction Kinetics

Consider reaction rate equations:

$$\frac{dc}{dt} = f(c, t)$$

$$\frac{dc}{dt} = p_1 c_1 - p_2 c_2 c_3$$

Reaction rate parameters

Discrete Adjoint of Chemical Reaction Kinetics

Numerical model (Rosenbrock solver):

$$c^{n+1} = c^n + \sum_{i=1}^s m_i k_i, \quad \text{Err}^{n+1} = \sum_{i=1}^s e_i k_i$$

$$T_i = t^n + \alpha_i h, \quad C_i = c^n + \sum_{j=1}^{i-1} a_{ij} k_j$$

$$A = \left[\frac{1}{h\gamma} - \hat{J}^T(t^n, c^n) \right]$$

$$A \cdot k_i = f(T_i, C_i) + \sum_{j=1}^{i-1} \frac{b_{ij}}{h} k_j + h\gamma_i f_t(t^n, c^n)$$



*Lots of numerical tricks
for handling stiff ODEs*

Discrete Adjoint of Chemical Reaction Kinetics

Adjoint of numerical model with respect to ...

... concentrations (Sandu et al., 2002):

$$\begin{aligned}\lambda_c^n &= \lambda_c^{n+1} + \sum_{i=1}^s (H(t^n, c^n) \times k_i)^T \cdot u_i \\ &\quad + \sum_{i=1}^s \hat{J}^T(T_i, C_i) \cdot u_i.\end{aligned}$$

$$\hat{J} = f_c = \frac{\partial f}{\partial c} \quad (\text{Jacobian}) \quad H = \frac{\partial^2 f}{\partial c_i \partial c_j} \quad (\text{Hessian})$$

... reaction rate parameters (Henze et al., 2007):

$$\begin{aligned}\lambda_p^n &= \lambda_p^{n+1} + \sum_{i=1}^s (\hat{J}_p(t^n, c^n) \times k_i)^T \cdot u_i \\ &\quad + \sum_{i=1}^s f_p^T(T_i, C_i) \cdot u_i.\end{aligned}$$

KPP automatically generates simulation and direct/adjoint sensitivity code for chemistry

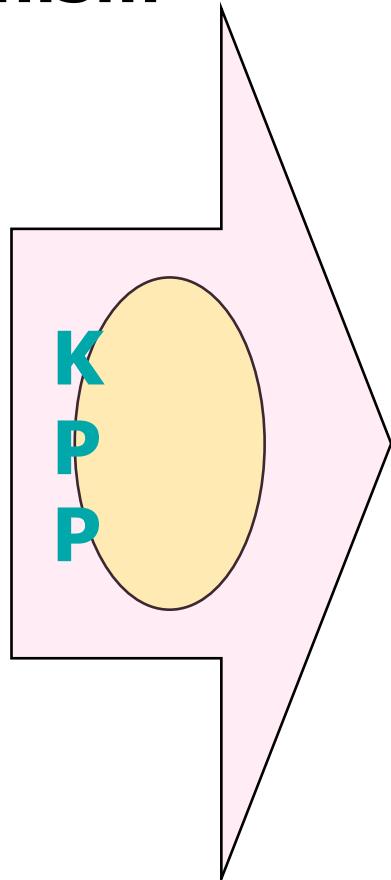
Chemical mechanism

```
#INCLUDE atoms

#DEFVAR
O = O; O1D = O;
O3 = O + O + O;
NO = N + O;
NO2 = N + O + O;

#DEFFIX
O2 = O + O; M = ignore;

#EQUATIONS { Small Stratospheric }
O2 + hv = 2O : 2.6E-10*S;
O + O2 = O3 : 8.0E-17;
O3 + hv = O + O2 : 6.1E-04*S;
O + O3 = 2O2 : 1.5E-15;
O3 + hv = O1D + O2 : 1.0E-03*S;
O1D + M = O + M : 7.1E-11;
O1D + O3 = 2O2 : 1.2E-10;
NO + O3 = NO2 + O2 : 6.0E-15;
NO2 + O = NO + O2 : 1.0E-11;
NO2 + hv = NO + O : 1.2E-02*S;
```



Simulation code

```
SUBROUTINE FunVar ( V, F, RCT, DV )
INCLUDE 'small.h'
REAL*8 V(NVAR), F(NFIX)
REAL*8 RCT(NREACT), DV(NVAR)

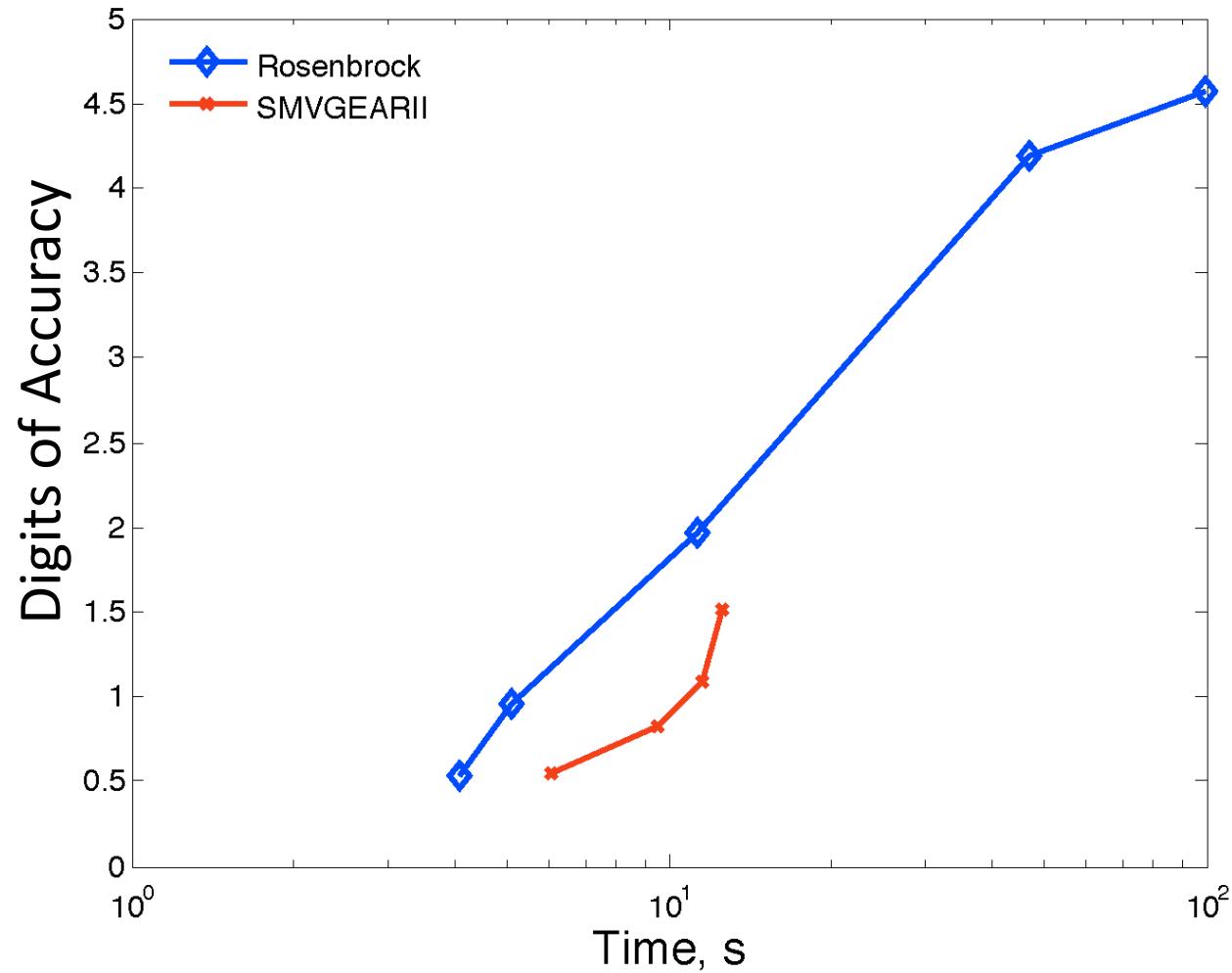
C A - rate for each equation
REAL*8 A(NREACT)

C Computation of equation rates
A(1) = RCT(1)*F(2)
A(2) = RCT(2)*V(2)*F(2)
A(3) = RCT(3)*V(3)
A(4) = RCT(4)*V(2)*V(3)
A(5) = RCT(5)*V(3)
A(6) = RCT(6)*V(1)*F(1)
A(7) = RCT(7)*V(1)*V(3)
A(8) = RCT(8)*V(3)*V(4)
A(9) = RCT(9)*V(2)*V(5)
A(10) = RCT(10)*V(5)

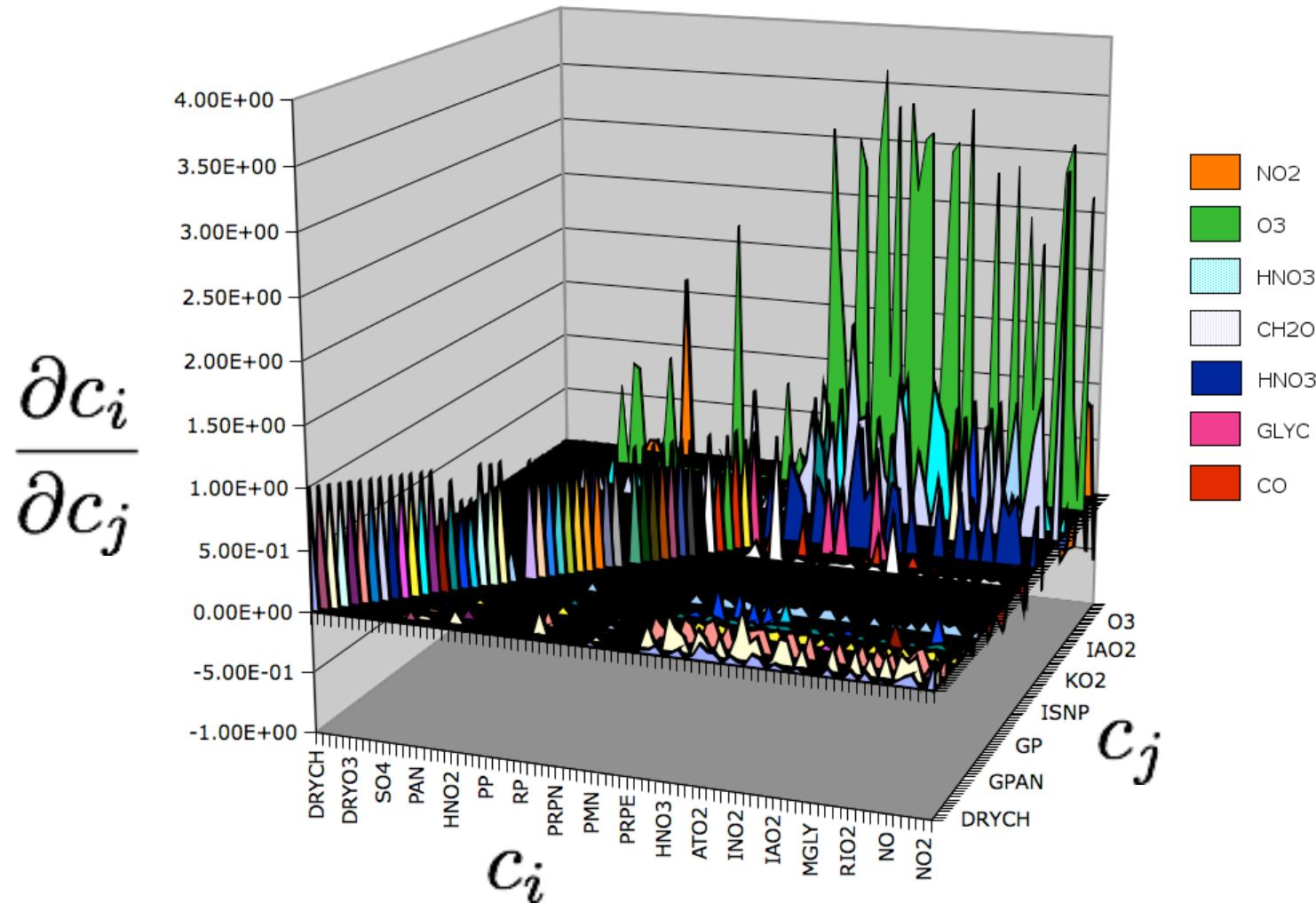
C Aggregate function
DV(1) = A(5)-A(6)-A(7)
DV(2) = 2*A(1)-A(2)+A(3)-A(4)+A(6)-&A(9)+A(10)
DV(3) = A(2)-A(3)-A(4)-A(5)-A(7)-A(8)
DV(4) = -A(8)+A(9)+A(10)
DV(5) = A(8)-A(9)-A(10)
END
```

[Damian et.al., 1996; Sandu et.al., 2002]

KPP generated code is fast: compare to widely used GEAR solver

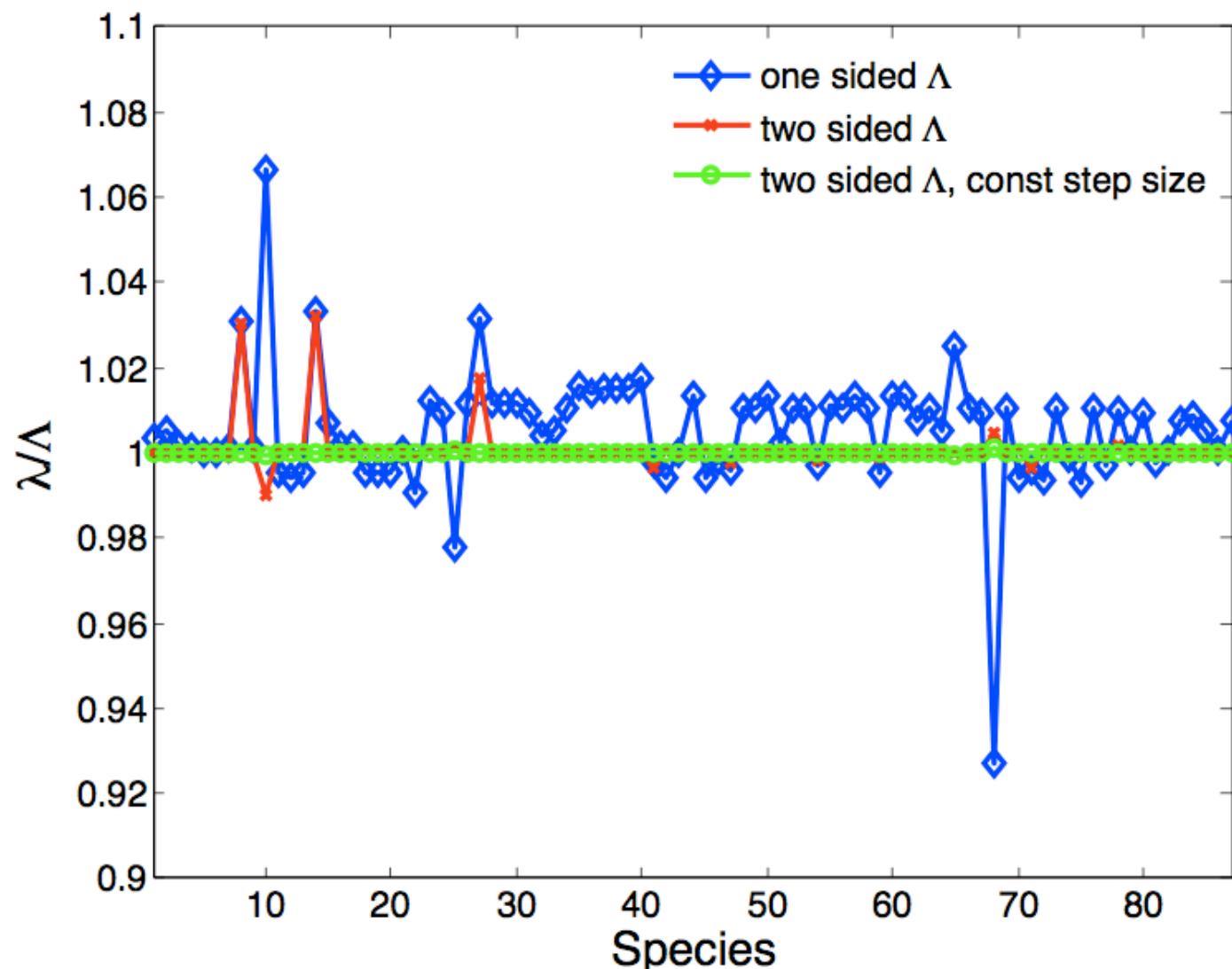


Box model test with KPP

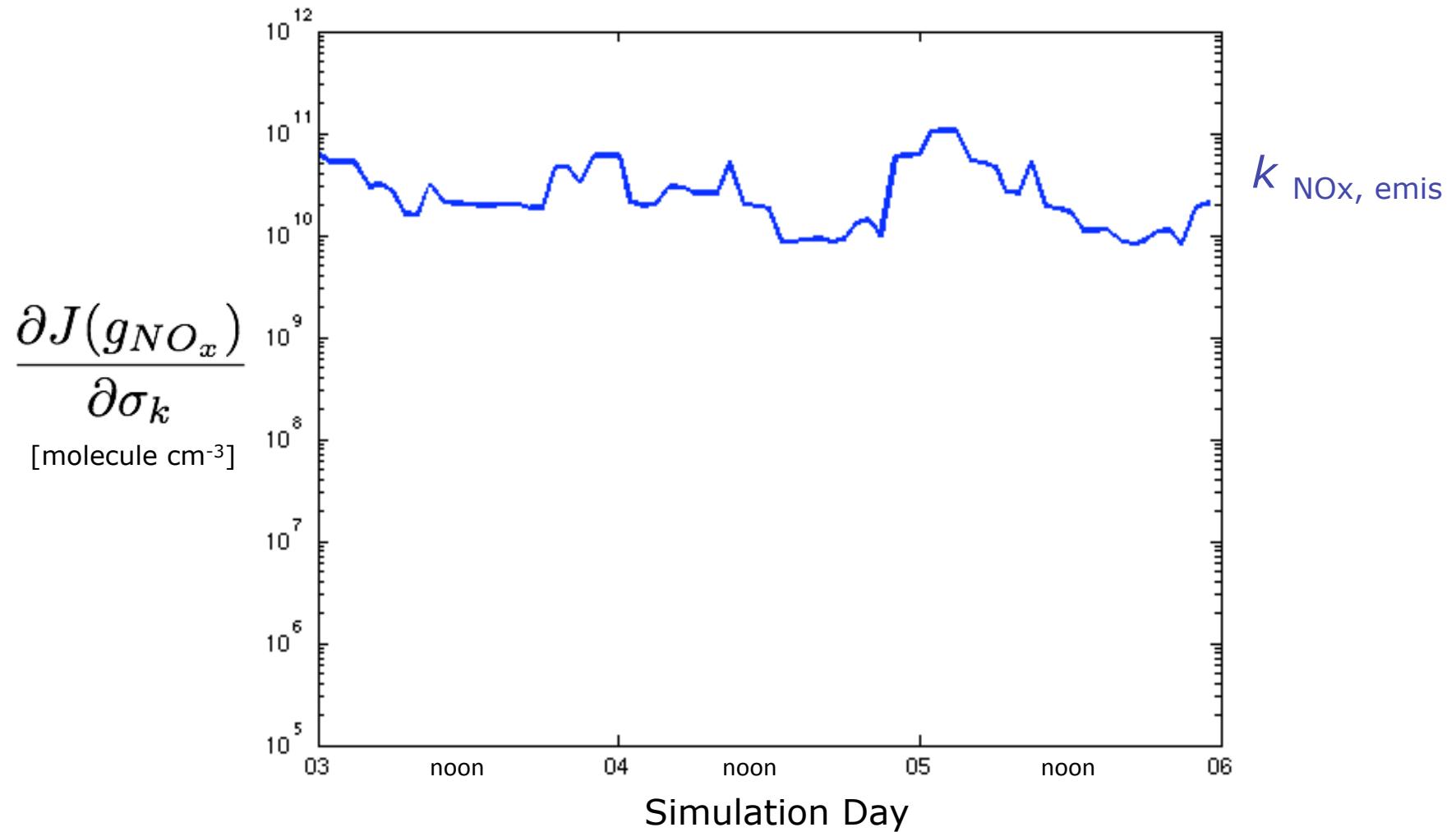


87 species, 307 reactions: Note the sparse structure --> Fast!

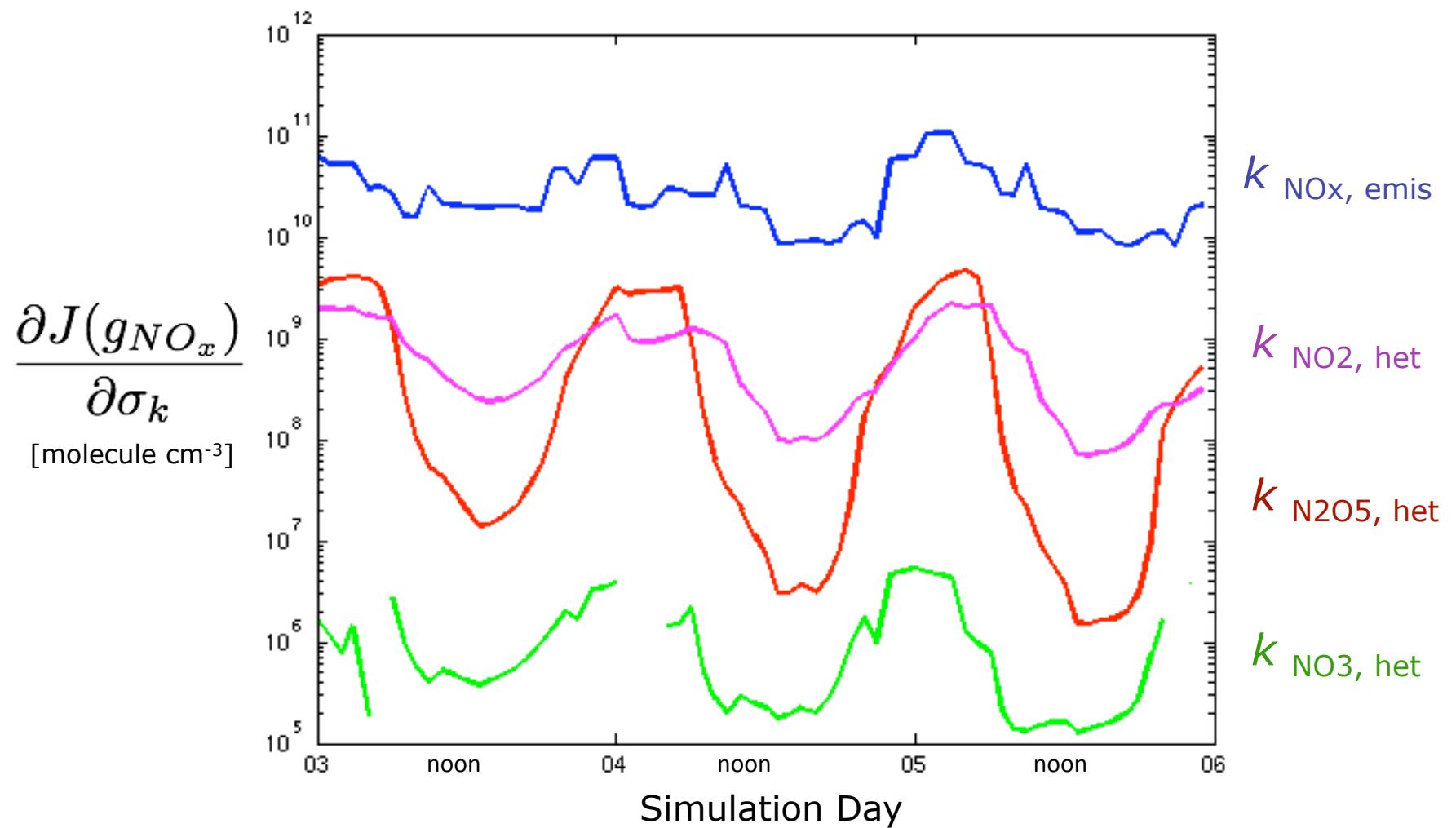
Box Model Test: **Discrete** sensitivity with respect to p_1



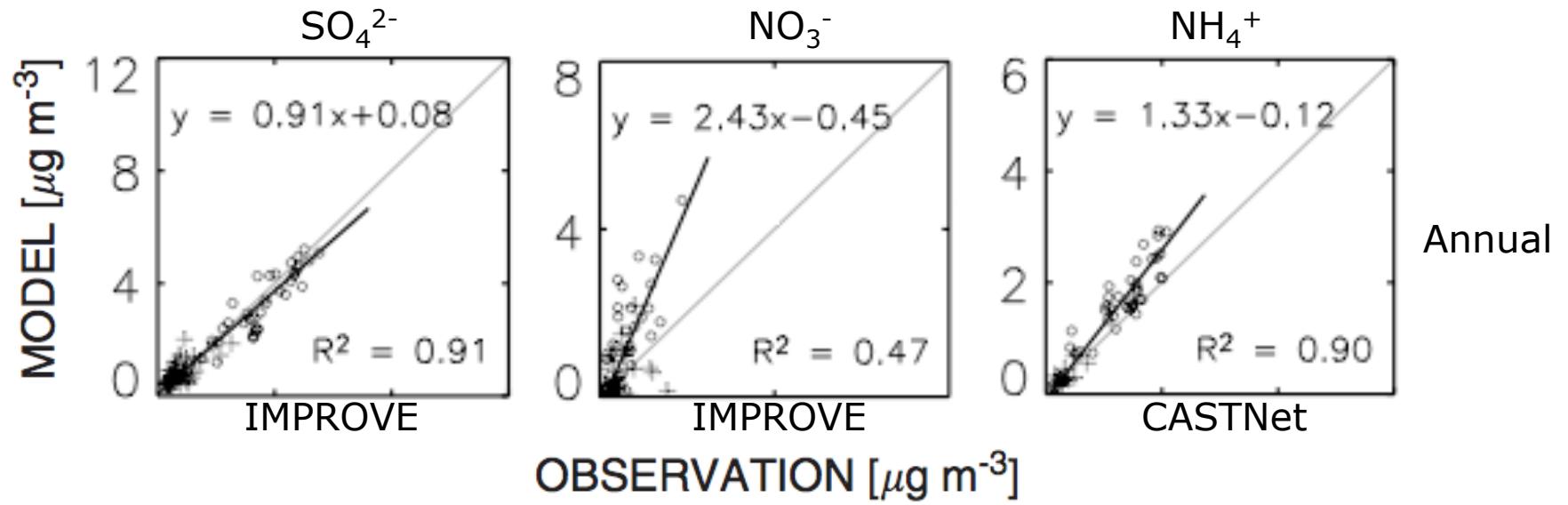
Sensitivity of NO_x on rate constants in chemical mechanism



Sensitivity of NO_x on rate constants in chemical mechanism

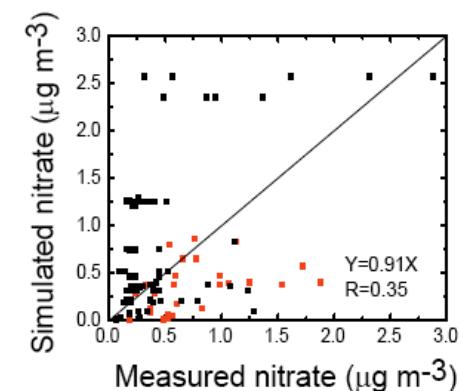
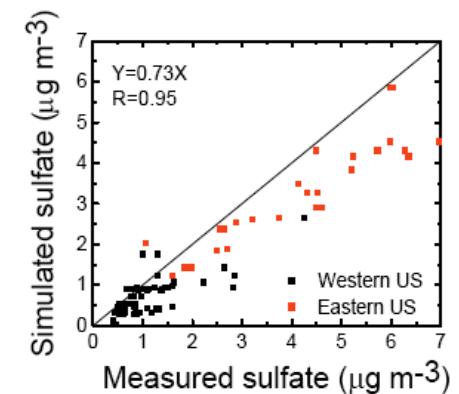
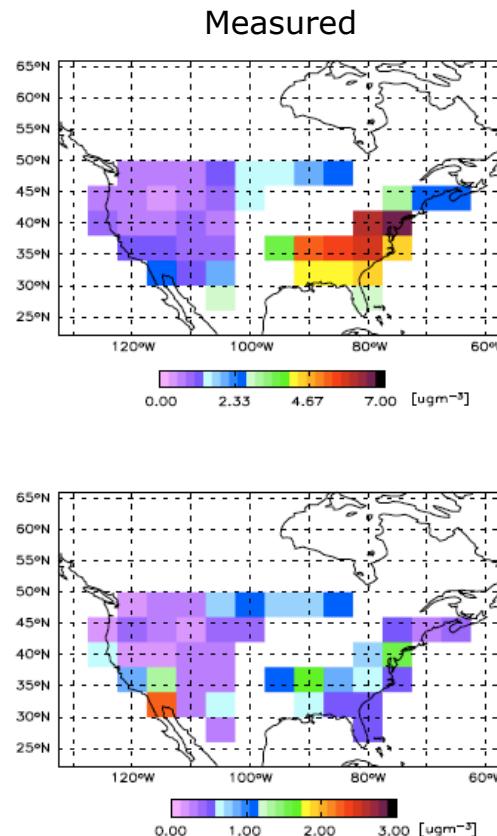
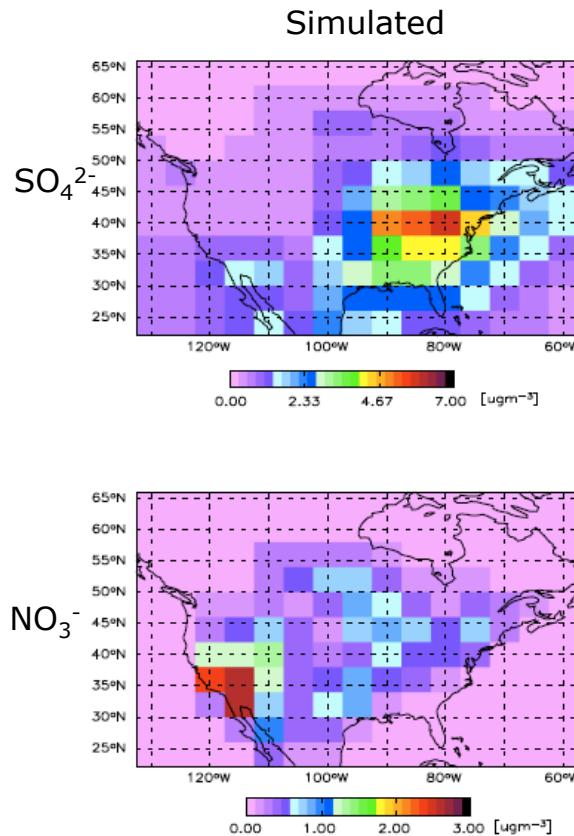


GEOS-Chem vs Observations



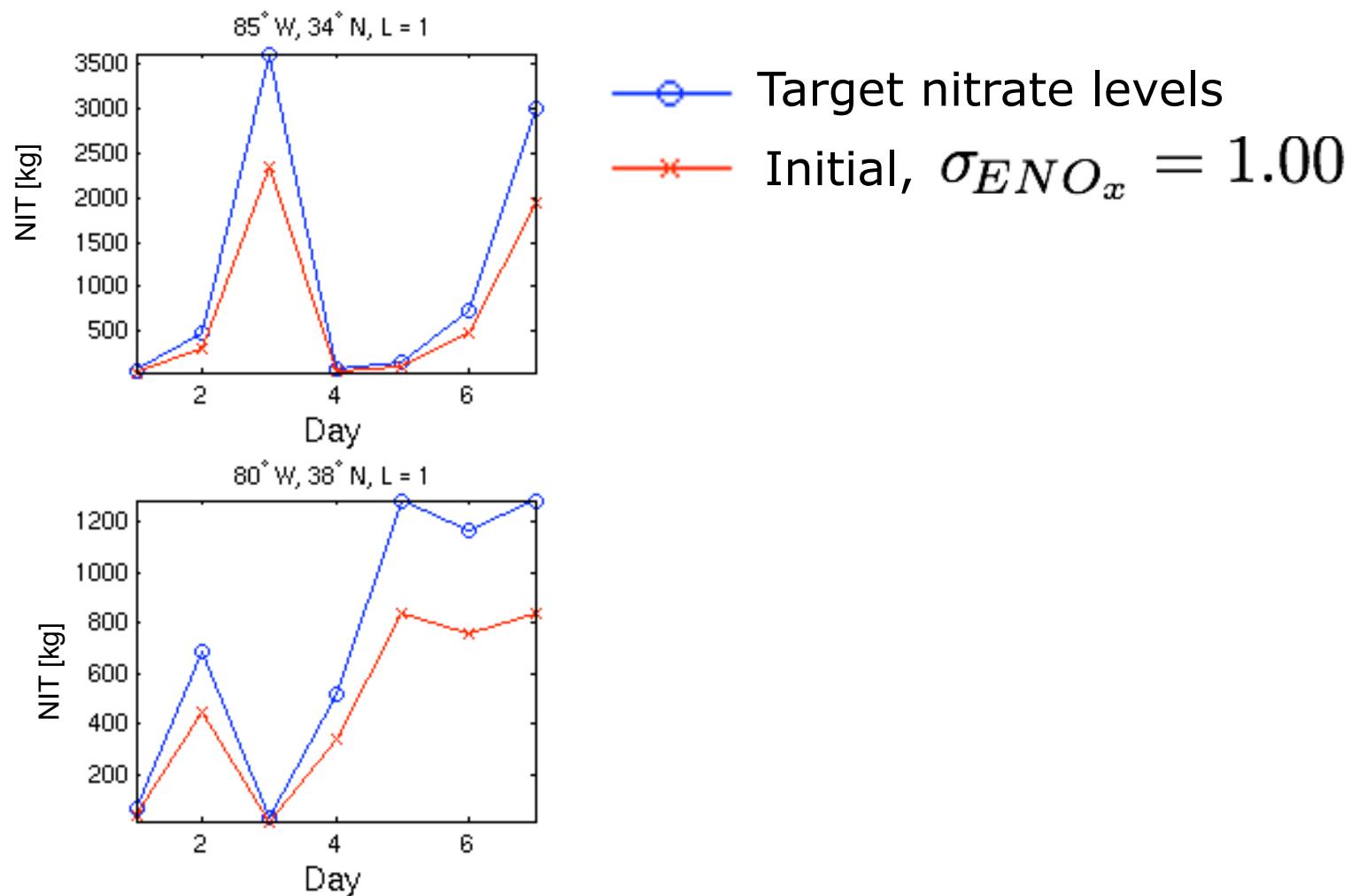
Model description: Park et al., (2004)

GEOS-Chem vs Observations



Liao et al., (2007)

Nitrate dependence on NO_x



Nitrate dependence on NO_x

