

SOURCE CONTRIBUTIONS OF VOLATILE ORGANIC COMPOUNDS TO OZONE FORMATION IN SOUTHEAST TEXAS

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1. INTRODUCTION

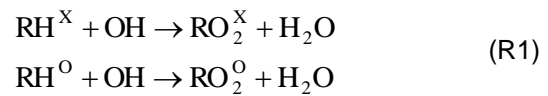
The Houston-Galveston-Brazoria (HGB) and Beaumont-Port Arthur (BPA) areas in the southeast Texas are respectively in severe and moderate non-attainment status of the National Ambient Air Quality Standards for ozone (O₃) (EPA 2004). Measures must be taken to control and further improve the O₃ air quality to better protect public health. To design an efficient emission control strategy, it is important to quantitatively determine which sources are responsible for O₃ formation in the polluted areas.

The major role of the VOCs in the O₃ formation process is in the formation of peroxy radicals that convert NO back to NO₂ and regenerate OH to allow sustained net O₃ formation (Finlayson-Pitts and Pitts 1997; Atkinson 2000). The objectives of this study are to 1) develop a source-oriented model to determine the contributions of VOCs from major emission sources to regional O₃ formation and 2) apply the model in Southeast Texas to determine source contributions to O₃ formation during a severe O₃ air pollution episode. This is the first time that the regional source contributions of VOCs to O₃ formation have been quantified using a three-dimensional source-oriented model.

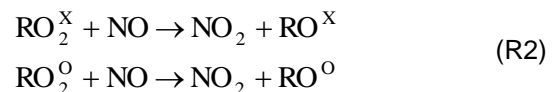
2. METHOD

The net ozone formation can be attributed to responsible VOC sources based on the contributions of the VOCs and their intermediate oxidation products to the NO to NO₂ conversion process. In this study, the sources of the directly emitted VOCs, the reactive intermediates and the RO₂ and HO₂ radicals formed from their oxidation are properly tracked in the simulation of the photochemistry as well as transport and removal processes using a revised Community Multiscale Air Quality (CMAQ) model (version 4.6) with a source-oriented SAPRC-99 photochemical mechanism. The source-oriented technique

introduces additional chemical species to represent contributions from different sources. For example, if superscript X represents VOC emissions from an explicit source X (for example, gasoline powered highway vehicles) and superscript O represents emissions from all other sources, then the reactions of a general reactive hydrocarbon (RH) from sources X and O with OH can be written as:



In this way, the contribution of source X to the peroxy radical budget in the model can be directly determined. The amount of NO₂ that is regenerated through reactions of RO₂^X and NO at each chemistry time step can be determined by keeping track of the extent of reactions (R2) using the process analysis (PA) tool included in the standard CMAQ distribution.



Similarly, by following the reactions of RO^X and RO^O with O₂, contributions to HO₂ for the two sources can be determined and the amount of NO₂ regenerated from HO₂ and NO reaction can be quantified.

The above discussion illustrates a method to track VOC from an explicit source (X) with emissions from other sources lumped into a single group (O). Theoretically, it is possible to expand the reactions so that emissions from more than one explicit source can be directly tracked simultaneously in a single model simulation. However the number of reactions in the expanded photochemical mechanism will increase almost exponentially due to the large amount of radical/radical reactions. In order to reduce the unnecessary complexity in expanding the mechanism and to make the approach more flexible to resolve a large number of sources, we decided to track only one source explicitly at a time. The final revised condensed SAPRC-99 mechanism includes 127 species and 489 reactions.

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3. MODEL APPLICATION

The CMAQ model with source-oriented SAPRC-99 extension was applied to study O₃ formation during the Texas Air Quality Study 2000 (TexAQS 2000) episode, from August 16 to September 6, 2000. A three-level nested domain was used in this study: the eastern United States was modeled with 36 km horizontal grid resolution; the east Texas and surrounding states were modeled with 12 km grid resolution and the HGB and BPA areas in southeast Texas were modeled with 4 km grid resolution (referred as the 4 km model domain hereafter).

Meteorology inputs needed to drive the CMAQ model were generated by the Meteorology-Chemistry Interface Processor (MCIP) using the output from the PSU/NCAR mesoscale model (MM5) simulation conducted by the Texas Commission of Environmental Quality. The EPA's 2001 Clean Air Interstate Rules (CAIR) emission inventory was used as a base to generate emission inputs for the TexAQS 2000 episode. The emission inventory was enhanced by using the episode-specific wildfire emissions from the Center for Energy and Environmental Resources at the University of Texas at Austin (Junquera, Russell et al. 2005). Emissions of olefins from petrochemical industrial point sources were increased by a factor of 5 from the reported values based on recent emission measurements. A revised Sparse Matrix Operator Kernel Emissions (SMOKE) emission processing model (version 2.4) from US EPA was used to process the updated CAIR emission inventory to generate gridded, speciated and temporally allocated VOC emission rates from each emission source for the CMAQ model. Biogenic emissions were generated using the Biogenic Emissions Inventory System, Version 3 (BEIS3) included in the SMOKE distribution. The 1-km resolution BELD3 land cover data with 230 different cover types (Vukovich and Pierce 2002) were used to estimate emission from vegetation.

The emissions were grouped into eight major source categories: biogenic sources, diesel vehicles, highway gasoline vehicles, off-highway gasoline vehicles, petroleum industries, other industries, solvent utilization and wildfires. Other VOC sources do not belong to these explicit source categories were lumped into the "other" source category. To generate emission for a specific source category, a Source Classification Code (SCC) filter was added to the SMOKE code so that only emissions from a set of predefined SCC codes would be processed.

4. RESULTS AND DISCUSSION

4.1 Predicted and Observed O₃ Concentrations

The predicted concentrations were compared with all available surface observation data. In general, results show that the predicted concentrations of O₃ and NO_x agree well with observations. Peak O₃ concentrations in some days are under-predicted at several sites. This under-prediction of peak O₃ in the Houston area has also been previously reported in other modeling studies (Byun, Kim et al. 2006). Fig. 1 shows the comparison of the observed and predicted maximum 8-hour O₃ concentrations at all surface stations within the 4 km model domain. Except for several extreme events, the predictions are generally within 80-120% of the observed values.

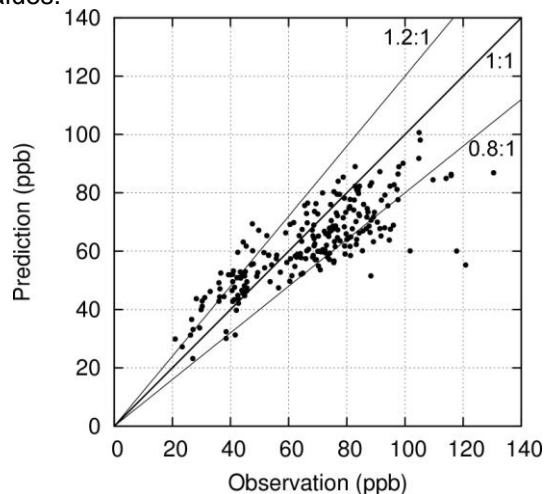


Fig. 1 Predicted and observed daily maximum 8-hour O₃ concentrations at individual observation station during the entire modeling episode.

4.2 Relationship between Net O₃ Formation and NO to NO₂ Conversion Rates

Fig. 2 shows the predicted NO to NO₂ conversion rates due to VOCs (the sum of the contributions from RO₂ and HO₂ radicals) for all grid cells and the corresponding net photochemical O₃ formation rates in the 4 km model domain. The net O₃ formation rate is the difference between the O₃ production rate and the O₃ removal rate in a gas-phase chemistry step. The results are for 1300-1400 Central Standard Time (CST), August 31, 2000 in the surface model layer (0-42 m from the surface). Results in upper layers are similar and thus are not shown here.

Only positive net O₃ formation rates are included in the plot. Grid cells where NO_x concentrations are high actually show negative O₃ formation rates and these negative rates are not included in the plot. The data points are grouped into five different categories based on the VOC (ppbC) to NO_x (ppb) ratio (VOC/NO_x ratio) at each grid cell. In general, the net O₃ formation rate appears to be linearly correlated with the NO to NO₂ conversion rates. Data points deviated from this linear relationship usually have low VOC/NO_x ratio of 0-5. The relatively higher NO_x concentrations in these grid cells enhance the NO titration reaction with O₃ and lead to lower net O₃ formation rates (Kleinman, Daum et al. 2005). To achieve the highest O₃ formation rate, the VOC/NO_x ratio has to be between 5 and 15. The excellent linear relationship between the NO to NO₂ conversion rate and the net O₃ formation rate for most of the data points within the domain at different VOC/NO_x ratios suggests that it is reasonable to attribute the contributions of different VOC sources to net O₃ formation based on their contributions to the NO to NO₂ conversion processes.

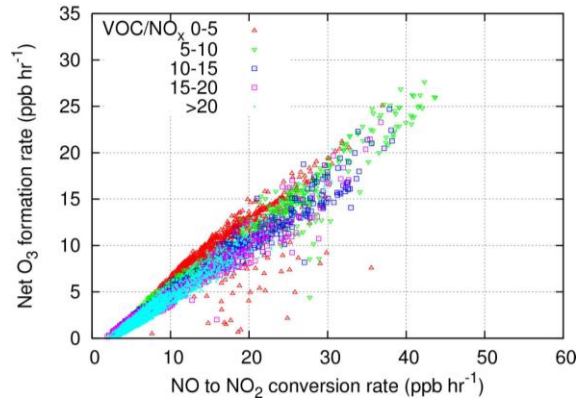


Fig. 2 Net ozone formation rate as a function of NO to NO₂ conversion rate due to peroxy radicals at 1300-1400 CST, August 31, 2000 for the surface model layer.

4.3 Source Contributions of VOC to Net O₃ Formation

The contributions of the *i*th VOC source to the net O₃ formation rate at a grid cell in the *z*th model layer is calculated using Equation (1) and (2):

$$\Delta O_{3,z}^i = f_z^i \cdot \Delta O_{3,z} \delta(\Delta O_{3,z}) \quad (1)$$

$$f_z^i = \frac{R_{NO_2,z}^i}{\sum_{k=1}^K R_{NO_2,z}^k}, i = 1, 2, \dots, K \quad (2)$$

where superscript *i* is the source type index; *z* is the model layer index; $\Delta O_{3,z}$ is the predicted

overall net O₃ formation rate (ppb hr⁻¹); $\Delta O_{3,z}^i$ is the apportioned contribution of VOC source *i* to the overall net O₃ formation rate (ppb hr⁻¹) and R_{NO_2} is the NO to NO₂ conversion rate (ppb hr⁻¹); f_z^i is the relative contribution of source *i* to the NO to NO₂ conversion. The delta function takes a value of 1 when the net O₃ formation rate is greater than zero otherwise it takes a value of 0. *K* is the total number of VOC sources.

O₃ formation in the layers above surface layer contributes to the surface ozone concentration through rapid vertical turbulent diffusion within the mixing layer (Byun, Kim et al. 2007; Song, Vizuete et al. 2008). Thus, it is useful to define a column-averaged source contribution expression as illustrated in Equation (3):

$$\Delta O_{3,avg}^i = \sum_{z=1}^{Z_{pbl}} w(z) \Delta O_{3,z}^i \quad (3)$$

where $\Delta O_{3,avg}^i$ is the column-averaged contribution of source *i* to net ozone formation, *Z*_{pbl} is the layer where mixing layer height is located and the weighting function *w*(*z*) is the ratio of the thickness of the *z*th layer to the mixing layer height.

Fig. 3 shows the regional distribution of column-averaged net O₃ formation rate between 1300-1400 CST from August 16 to September 6, 2000. Highest O₃ formation rate of 16 ppb hr⁻¹ is located in the downwind area of the urban and industrial regions in the model domain. In general, areas north of the Houston Ship Channel area show most significant O₃ formation rate at this hour. Several grid cells near the Ship Channel region shows low net O₃ formation rate. Further analysis of the results indicates that VOC/NO_x ratios in these source areas are close to zero due to high NO_x concentrations during daytime hours.

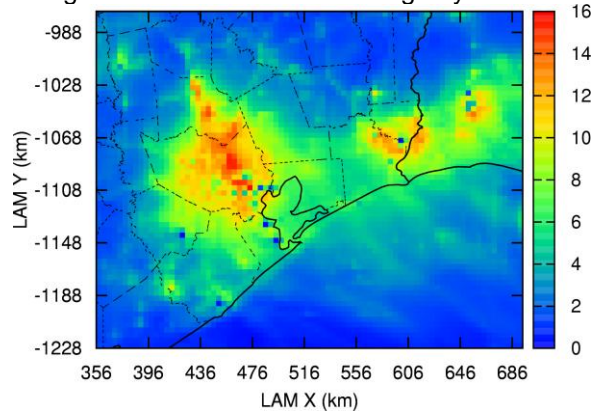


Fig. 3 Net ozone formation rate below the mixing height at 1300-1400 CST averaged from August 16 – September 6, 2000. Units are ppb hr⁻¹.

Fig. 4 shows the regional distribution of VOC source contributions to the column-averaged net O₃ formation rate at 1300-1400 CST averaged from 16 August to September 6, 2000 (see Fig. 3) based on equations (1-3). Fig. 4(a) shows that contribution from biogenic sources is highest in areas north of the Houston metropolitan area with a maximum net O₃ formation rate of approximately 12.2 ppb hr⁻¹. In the urban areas, the net O₃ formation rate due to biogenic sources is approximately 6 ppb hr⁻¹. The biogenic VOC concentrations are highest in areas near the northern boundary of the 4km domain but the O₃ formation rates are low due to low NO_x concentrations (VOC/NO_x ratio typically exceeds 500 in these areas). Fig. 4(b) shows that VOCs emitted from diesel vehicles do not contribute significantly to the net O₃ formation process. The highest contribution is approximately 0.1 ppb hr⁻¹ in the urban Houston area. Fig. 4(c) shows that the spatial distribution of contributions from freeway gasoline vehicles is similar to that of diesel vehicle with highest contribution in the urban Houston region. The maximum net O₃ formation rate is approximately 2.5 ppb hr⁻¹. Fig. 4(d) shows that off-highway gasoline vehicles contribute to the net O₃ formation process with a maximum rate of approximately 1.4 ppb hr⁻¹ in the Galveston and Port Arthur area. An inspection of the emission inventory indicates that emissions from pleasure marine crafts (boats powered by 2- and 4- stroke gasoline engines) are significant in these regions. Fig. 4(e) shows that petroleum industries contribute to a maximum of approximately 1.8 ppb hr⁻¹ to the net photochemical O₃ formation process. The influence of the petroleum industries is obvious in the HGB and BPA areas as well as over the waters of the Galveston Bay and the Gulf of Mexico. Fig. 4(f) shows that other industrial sources contribute significantly to the O₃ formation processes with a peak value of 6.6 ppb⁻¹ in the Ship Channel region. Fig. 4(g) shows that the contribution from solvent utilization is mainly located west part of the urban Houston region, with a maximum conversion rate of 1.4 ppb hr⁻¹. Fig. 4(h) shows that the influence of wildfires averaged over the entire model episode is only limited in a few areas with a highest contribution of approximately 1.2 ppb hr⁻¹. The actual contribution of a wildfire near the source is much higher than the episode average contribution shown here. Fig. 4(i) shows that sources unaccounted for by the above eight source categories contribute to the O₃ formation process mainly over the off-shore drilling areas in the Gulf of Mexico where the peak

contribution can be as high as 3.1 ppb hr⁻¹. An inspection of the emission inventory indicates that emissions from reciprocating engines powered by natural gas are the most significant contributors to the “other” source categories in this area.

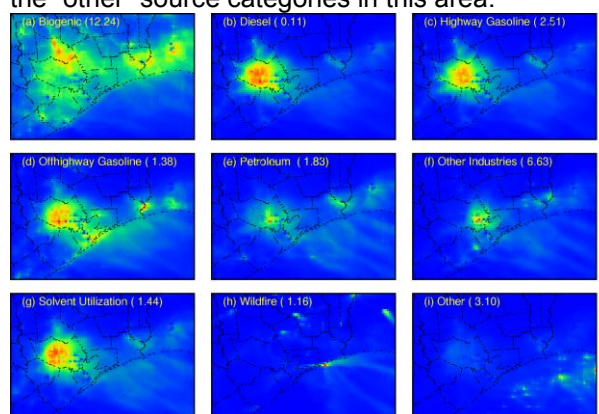


Fig. 4 Regional distributions of net photochemical ozone formation rates below the mixing height for 1300-1400 CST averaged from August 16 – September 6, 2000. The number on each panel is the maximum conversion rate for each source category in units of ppb hr⁻¹. The color bar on the right bottom can be used to estimate the rates on each sub-panel. The color scale is linear, starting from zero to the maximum value noted on the figure.

Fig. 5 shows the averaged hourly O₃ formation rate from August 16 to September 6, 2000 attributed to each VOC emission source. Top 10% of the air columns with highest column-averaged O₃ formation rate were included in the calculation. The highest column O₃ formation rate occurs at early afternoon hours. Contributions from anthropogenic sources are highest in the afternoon hours.

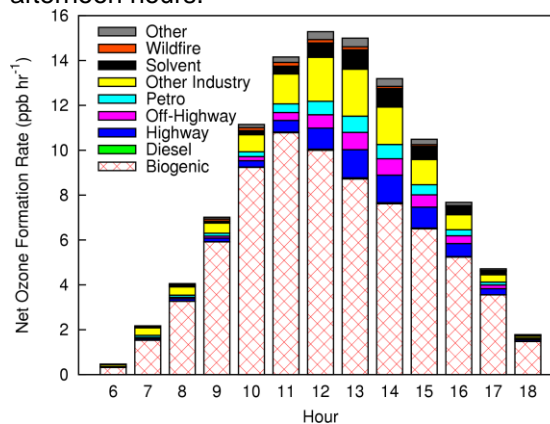


Fig. 5 Hourly VOC source contributions to net column ozone formation rate averaged from August 16 to September 7, 2000.

4.4 Back-trajectory Analysis of High O₃ Events

Although regional analysis discussed in the previous sections could determine VOC source contributions to O₃ formation rate at any given location at any given time, it does not directly determine which sources contribute to the observed O₃ concentration because O₃ builds up over time and the air parcel arriving at a specific receptor location passes through different regions with different emission signatures. Back-trajectories of air parcels arriving at locations when high O₃ concentrations occur can be used to better understand the sources that contribute to the elevated O₃ concentrations at specific locations. By investigating the O₃ concentrations, net O₃ formation rates and the source contributions of NO to NO₂ conversion along an air parcel trajectory, the contributions from major emission sources to a high O₃ event can be determined.

The relative contribution of VOCs from source *i* to surface O₃ concentration at the end of a trajectory can be calculated using Equation (4):

$$F^i = \frac{\sum_{t=1}^T d(t) \Delta O_{3,avg}^i(t)}{\sum_{k=1}^K \sum_{t=1}^T d(t) \Delta O_{3,avg}^k(t)}, i=1,2,\dots, K \quad (4)$$

where F^i is the relative contribution to O₃ formation due to source *i*, $\Delta O_{3,avg}^i(t)$ is the contribution of source *i* to the column-averaged net O₃ formation rate at time *t*, as defined in Equation (3), *T* is the number of time steps used in the trajectory calculation, *d*(*t*) is a decay function that scales the contribution of O₃ formation at time *t* to the final O₃ concentration and *K* is the total number of VOC sources. In this study, *d*(*t*) is taken as 1 to simplify analysis. Analysis shows that this is good first approximation.

Equation (4) was applied to quantify the source contributions to the maximum 8-hour O₃ concentrations in the HGB and BPA areas. All the grid cells on land within the Texas state boundary in the 4 km domain were included in this analysis. For each grid cell, if the daily maximum 8-hour average O₃ concentration exceeds 90 ppb, 8 back-trajectories arriving at this grid cell during the 8-hour span were calculated and the contributions of VOCs to O₃ formation along each trajectory were determined. Although the 8-hour O₃ standard has been revised to 75 ppb, a cut concentration of 90 ppb is selected to represent extreme high O₃ exceedance cases. Most of the O₃ exceedance cases occur on August 25-31 and September 3-5,

2000. 300 grid cells were predicted to have daily maximum 8-hour O₃ concentration higher than the 90 ppb limit and thus a total number of 2400 trajectories were included in the analysis.

Fig. 6 shows the cumulative frequency distribution of the relative contributions of biogenic sources, internal combustion engines (diesel vehicles, highway gasoline vehicles, and off-highway gasoline vehicles combined), industries (petroleum industries and other industries combined), solvent utilization, wildfires and other sources to the net O₃ formation rate based on the 2400 10-hour trajectories. Biogenic sources are the major contributors to high O₃ concentrations for most of the 8-hour exceedance events predicted by the model. In approximately 50% of the cases biogenic sources contribute to over 55% of the net O₃ formation. Industrial sources and internal combustion engines are both significant anthropogenic source of VOC to O₃ formation. In approximately 50% of the cases, their respective contributions exceed 10% and in approximately 20% of the cases their contributions exceed 30%. Solvent utilization is also a large anthropogenic source, contributing over 10% for 20% of the cases considered. Maximum relative vehicle contribution to O₃ formation is approximately 20%. Contributions from wildfires vary significantly. In some cases, its contribution can be as high as 34%, exceeding the maximum contribution from solvent utilization.

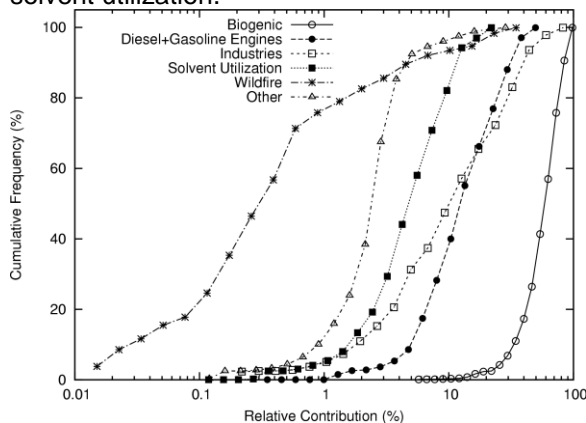


Fig. 6 Cumulative frequency distribution of the relative VOC source contributions to net ozone formation for surface grid cells where predicted daily maximum 8-hour average ozone concentration is greater than 90 ppb.

5. SUMMARY

In this paper, we illustrated that during the TexAQS 2000 O₃ episode, the net photochemical O₃ formation rate was linearly related with the NO

to NO₂ conversion rate due to peroxy and hydroperoxy radicals generated from the photochemical oxidation of VOCs. A source-oriented SAPRC-99 photochemical mechanism was developed and incorporated into the CMAQ model to directly determine the contributions of eight explicit VOC sources to the predicted net O₃ formation in the entire model domain based on their contributions to the NO to NO₂ conversion rate. Of the eight sources studied, biogenic sources were found to be the largest contributor to regional O₃ formation. Industrial sources are the largest anthropogenic source to O₃ formation, followed by gasoline vehicles and solvent utilization. Wildfires could contribute to fast ozone formation near the fire locations but their influence is generally localized.

Vukovich JM and Pierce T (2002) The Implementation of BEIS3 within the SMOKE modeling framework. 2002, / MCNC-Environmental Modeling Center, Research Triangle Park and National Oceanic and Atmospheric Administration, City.

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