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1. Introduction

Studies have suggested that enhanced ozone (O_3) deposition over sea-water and halogen chemistry are important atmospheric processes (Chang et al., 2004) and Read et al., 2008). However, these processes are currently not accounted for in most air quality models. Here, we examine the impacts of these processes on O_3 using the hemispheric the Community Multiscale Air Quality (CMAQ) model.

2.1 Model configuration

Feature	Description	
Domain	Northern hemisphere	
Horizontal grid size	108-km x 108-km	
Vertical layers	44 layers (1000-50 mb)	
Simulation period	May-August, 2006	
Meteorological model	WRFv3.3	
Anthropogenic emissions	Emissions Database for Global Atmospheric Research	
Biogenic emissions	Global Emissions InitiAtive	
Boundary conditions	Static profile	
Initial conditions	Previous model results	
Model simulations		
Case A	CB05TUCI	
Case B	Case A + halogen chemistry	
Case C	Case B + enhanced O_3 deposition	

2.2 Halogen chemistry and emissions

Feature	Description	R
Bromine chemistry	37 chemical reactions	
Iodine chemistry	50 chemical reactions	Sa
Halogen emissions		
Halocarbons	Based on chlorophyll	Ya Or
Br ₂	Based on sea-salt emissions	Ya
HOI and I_2	Based on SST, wind speed, O_3	M

2.3 Enhanced O_3 deposition over sea-water

Feature	Description	F
Deposition velocity	$v_d = \frac{1}{R_a + R_b + R_s}$	
Current R _s	$R_{s} = \frac{\left(\begin{array}{c} S_{c} \\ P_{r} \end{array} \right)^{2/3}}{H_{eff} d_{3} u_{*}}$	
Revised R _s	$\frac{1}{1.75 \frac{(d_3 u_*)}{(S_c / P_r)^{2/3}} + heff (2.0e5 * C_I * Diff_{H_2 O})^{0.5}}$	С

Sea-water iodide (C₁)

Based on sea-surface temperature Carpenter et al., 2014

U.S. Environmental Protection Agency Office of Research and Development

Impact of enhanced ozone deposition and halogen chemistry on model performance

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ang et al., 2005 Saiz-Lopez et al., 2014

arwood, et al., 2012 rdonez et al., 2012 ang et al., 2006

1cDonald et al., 2014

Reference

Chang et al., 2004

3.1 BRO and IO

- predicted IO values tend to be closer to the observed data.



Figure 1: Predicted monthly-mean (August) daytime surface (a) BrO (b) IO

3.2 Impact on surface-level O_3

- Halogen chemistry reduces monthly-mean surface O_3 by 4.0-6.0 ppbv over large areas of marine environments while enhanced deposition reduces O_3 by 3.0 ppbv averaged over the entire domain while the enhanced deposition reduces O_3 by 0.6 ppbv.
- The odd oxygen destruction rate (Saiz-Lopez, et al., 2014) (averaged



Figure 2: (a) Predicted changes (Case A – Case B) in monthly-mean (August) surface O_3 concentrations due to halogen chemistry (b) predicted changes (Case B – Case C) in monthly-mean O_3 concentrations due to enhanced deposition

3.3 Comparison with observed O_3

- Model predictions with these processes improve the comparison with observed O_3 data at remote locations (Table 1 and Figure 3).

 Predicted monthly-mean daytime surface BrO values over marine environments are generally <1.0 pptv while predicted IO values are >1.0 pptv (Figure 1).

 Ship-based measured BrO values are <3.0-3.6 pptv and IO ~3.5 pptv while satellite based measured IO values are ~2.4-3.3 pptv (Saiz-Lopez et al., 2012). Predicted daytime BrO values are generally lower than observations while



reduces O₃ by 1.0-2.0 ppbv (Figure 2). Overall, the halogen chemistry

over all layers in marine environment) due to the halogen chemistry is 0.4 ppbv/day. The majority of the loss was triggered by the iodine chemistry.



• Mean bias for Case A = 8.6 ppbv, Case B = 4.8 ppbv, and Case C = 3.9 ppbv.





4. Summary

- more O_3 than the enhanced deposition.

5. References

Carpenter, R., et al., 2014. The distribution of iodide at the sea surface, Environ. Sci.: Processes Impacts, 16, 1841-1859. Chang, W., et al., 2014. Ozone deposition to the sea surface: chemical enhancement and wind speed dependence, Atmospheric Environment, 38, 1053–1059.

Steyn & Silvia Gastelli (ed.), ITM Air Pollution Modeling and its Application XXI. Springer Netherlands, Netherlands, C Series: 175-179. variables and parameterization for global modeling, ACP, 14, 5841-5852.

Mathur, R., et al., 2011: Extending the applicability of the CMAQ model to hemispheric scales: motivation, challenges, and progress, Douw McDonald S. M., et al., 2014. A laboratory characterization of inorganic iodine emissions from the sea surface: dependence on oceanic Ordonez, C., et al., 2012. Bromine and iodine chemistry in a global chemistry model: description and evaluation of very short-lived oceanic

sources, ACP, 12, 1423-1447. Read, K.A., et al., 2008. Extensive halogen-mediated ozone destruction over the Atlantic Ocean, Nature, 453, 1232-1235. Saiz-Lopez, A., et al., 2012. Estimating the climate significance of halogen-driven ozone loss in tropical marine troposphere, ACP, 12,

3939-3949.

Skamarock, W. C., et al., 2008. A description of the advanced research WRF version 3. NCAR Tech Note NCAR/TN 475 STR, 125 pp. [Available from UCAR Communications, P.O. Box 3000, Boulder, CO 80307.]

Whitten, G. Z., et al., 2010. A new condensed toluene mechanism for CB: CB05-TU. Atmospheric Env., 44, 5346-5355.

Yang, X., et al., 2005. Tropospheric bromine chemistry and its impacts on ozone: a model study, JGR, 110, D23311.

Yarwood, G., et al., 2012. Improving CAMx performance in simulating ozone transport from the Gulf of Mexico, Final Report for the Texas Commission on Environmental Quality project 582-11-10365-FY12-05.



Table 1: A comparison of model and observed O_3 at remote surface locations (ppbv)

Observed	Predictions	Predictions	Predictions
mean	Case A	Case B	Case C
25.9	32.8	29.6	28.8
18.2	25.1	20.9	20.2
27.2	37.3	32.2	31.1
14.2	29.0	24.7	23.9
23.1	31.2	28.2	27.7
38.4	46.6	42.5	41.0
41.4	46.3	43.5	43.2
26.9	35.5	31.7	30.8

Figure 3: A comparison of predicted O_3 concentrations with ozonesonde data

• Both the enhanced deposition and halogen chemistry reduce O_3 concentrations over marine environments. The halogen chemistry reduces

• Model without these processes over-predicts O₃ while the inclusion of these processes improve model performance over marine environments.

Enhanced deposition and halogen chemistry are important processes and need to be incorporated into air quality models.

Saiz-Lopez, A., et al., 2014. Iodine chemistry in the troposphere and it effect on ozone, ACPD, 14, 19985-20044.